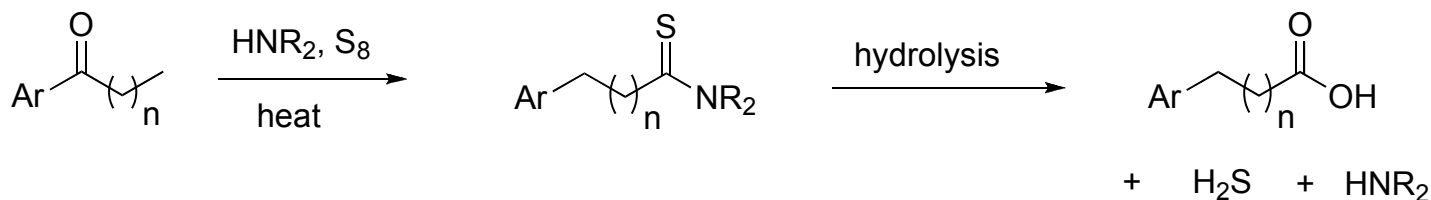


# Willgerodt-Kindler Reaction

Yong Guan

Jan. 30, 2009

# Introduction

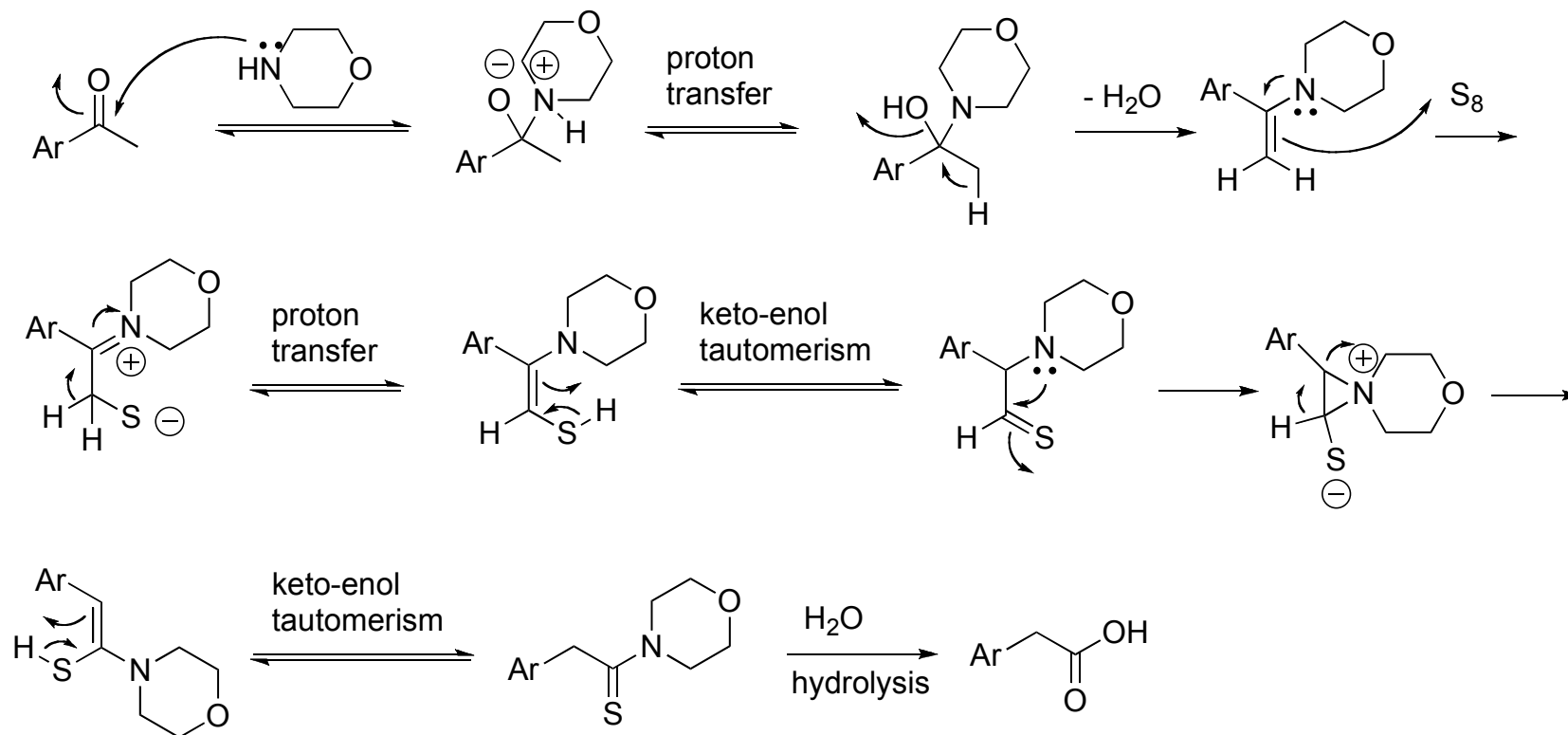


- The Willgerodt rearrangement or Willgerodt reaction is an organic reaction converting an aryl alkyl ketone to the corresponding amide by reaction with ammonium polysulfide, named after Conrad Willgerodt.
- The related Willgerodt-Kindler reaction takes place with elemental sulfur and an amine like morpholine. The reaction is named after Karl Kindler.

(a) Willgerodt, C. *Ber. Dtsch. Chem. Ges.* **1888**, 21, 534–536.

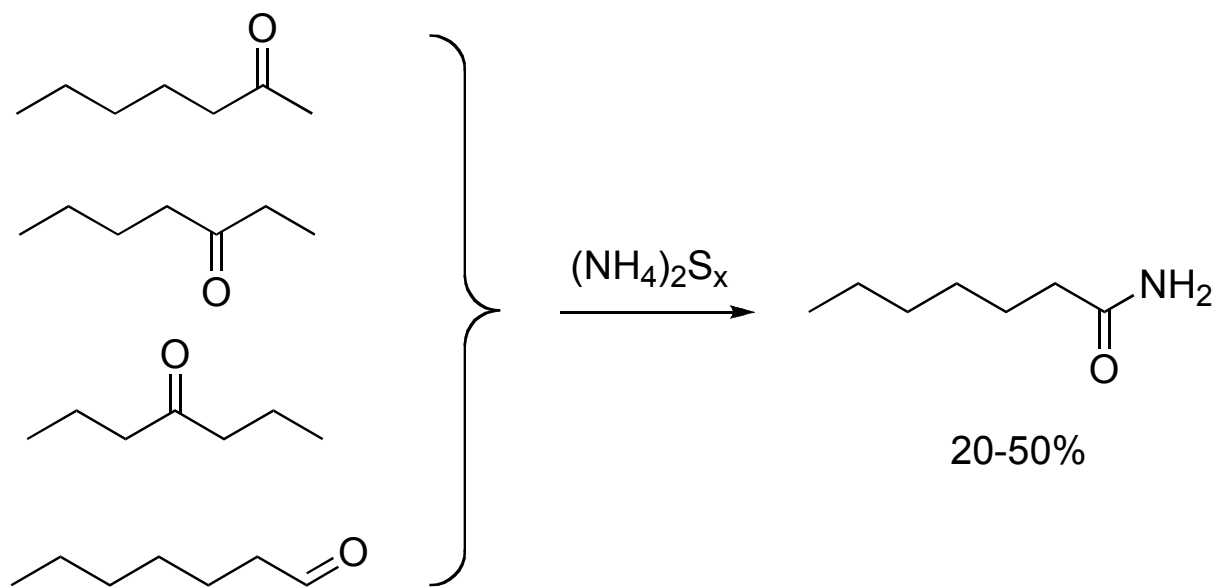
(b) Kindler, K. *Liebigs Ann. Chem* **1923**, 431, 187–207.

# Mechanism



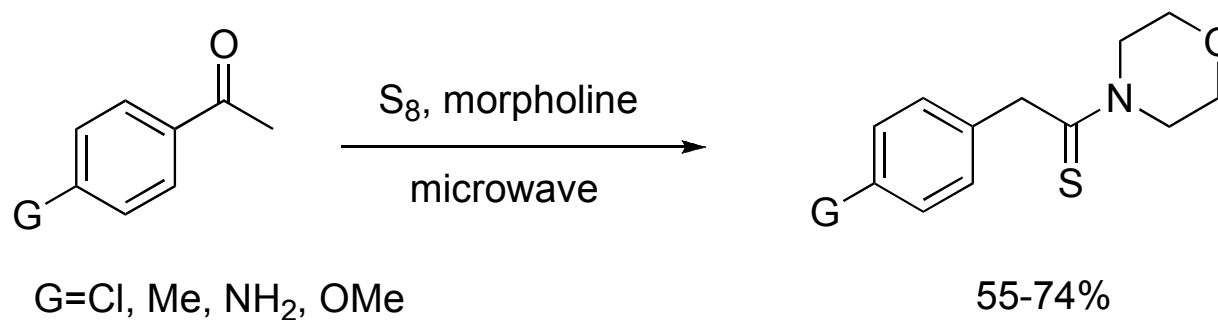
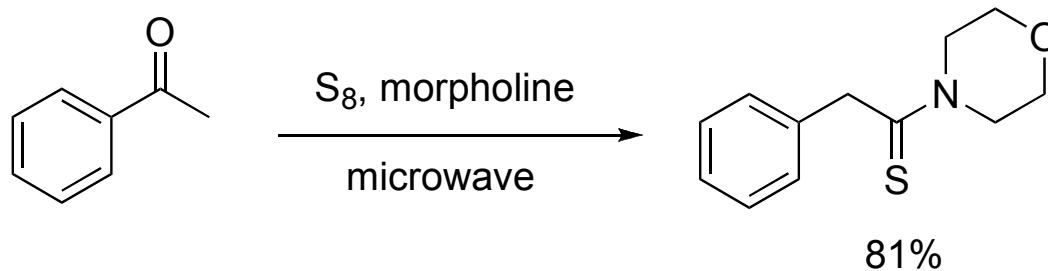
*Name Reactions and Reagents in Organic Synthesis, 2nd ed;*  
Bradford P. Mundy, Michael G. Ellerd, Frank G., Jr. Favaloro; Wiley-Interscience, pp 690.

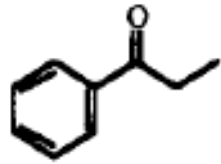
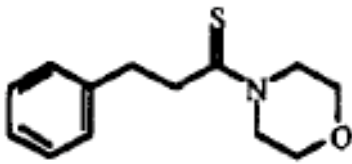
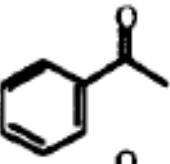
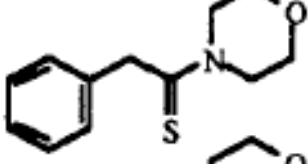
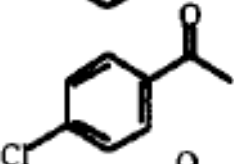
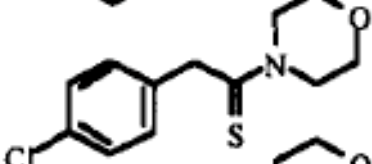
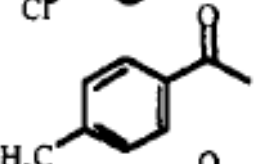
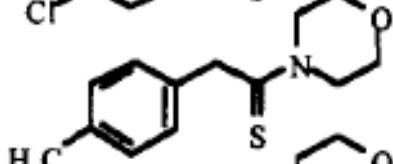
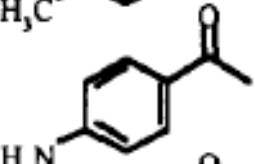
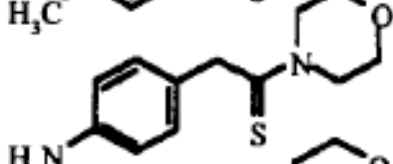
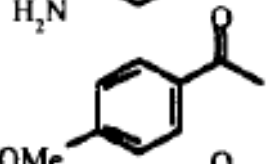
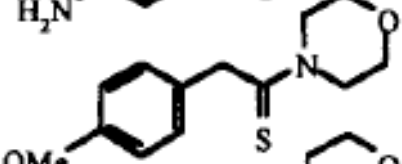
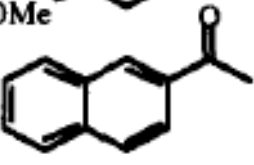
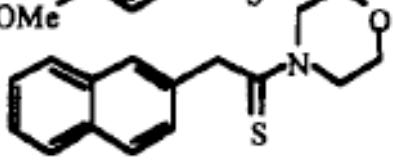
# Aliphatic Substrates



Cavalieri, L; Pattison, D. B.; Carmack, M. *J. Am. Chem. Soc.*, **1945**, 67,1783.

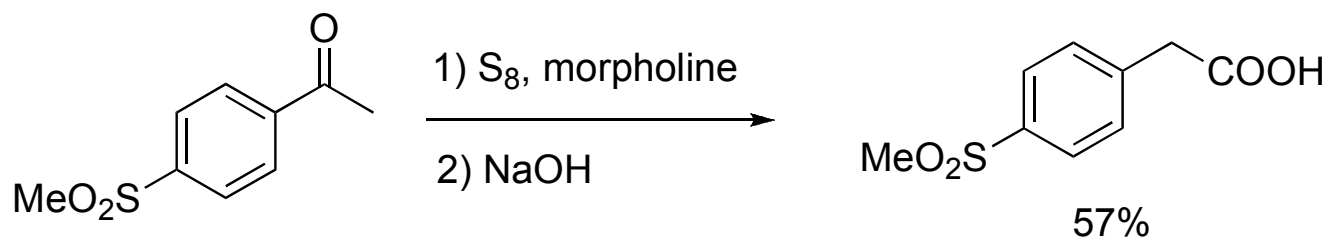
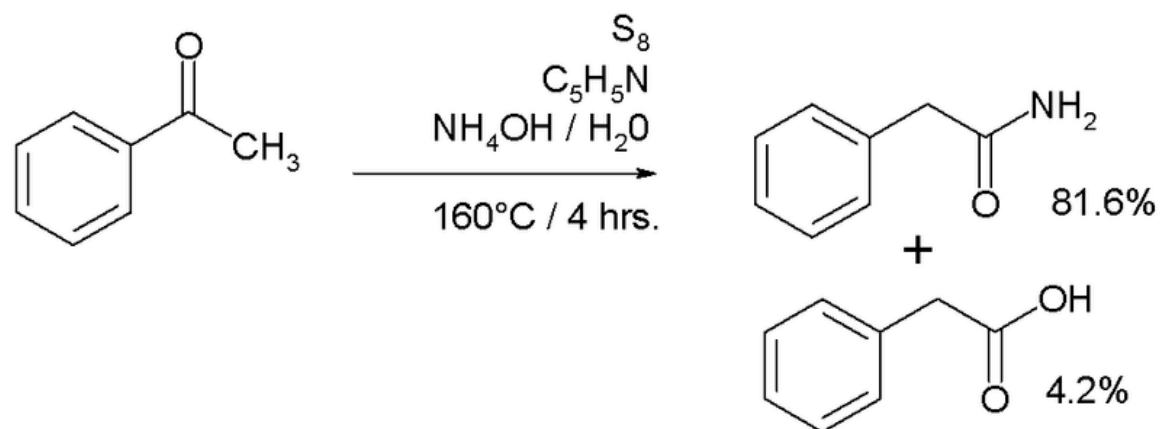
# Aromatic Substrates



Entry	Substrates <b>1</b>	Products <b>2<sup>a</sup></b>	Time (min)	Isolated Yield (%)		
				Series A <sup>b</sup>	Series B <sup>c</sup>	Series C <sup>d</sup>
a			4	10	16	40
b			4	50	56	81
c			6	40	45	55
d			4	44	53	74
e			3.5	42	47	65
f			5	49	56	72
g			4.5	56	61	75

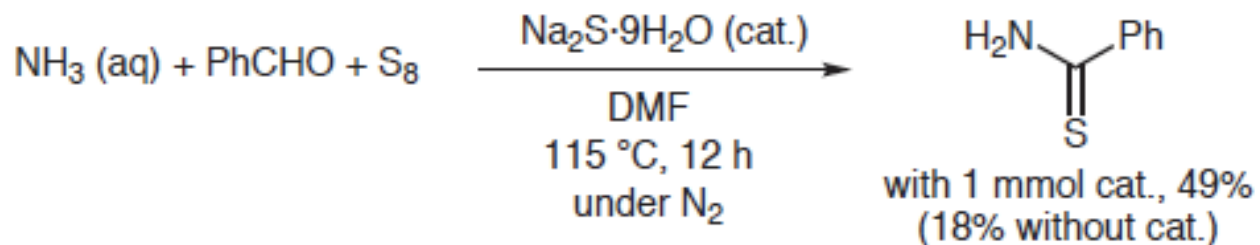
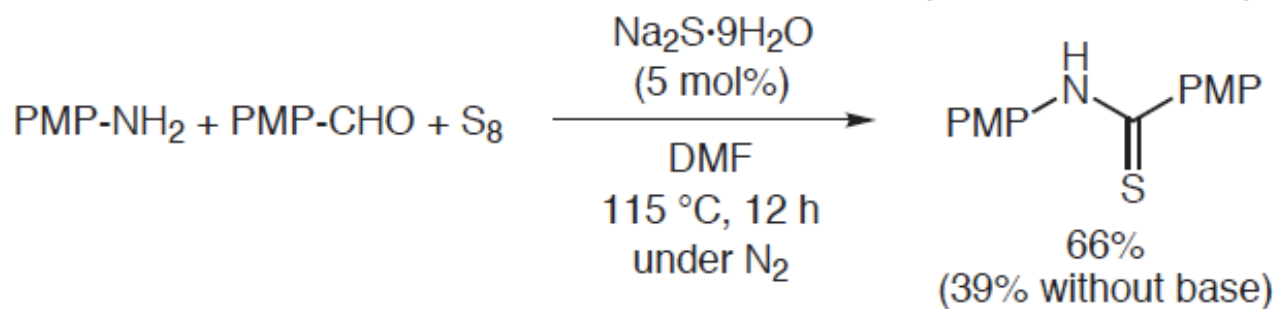
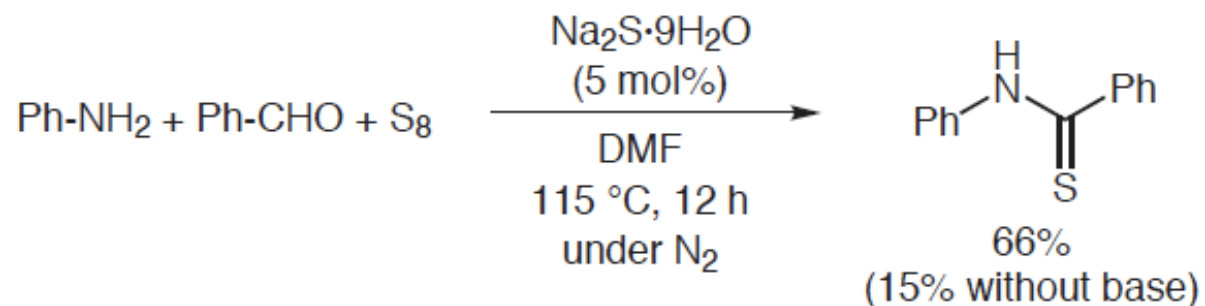
Darabi, H., *et al. Tetrahedron Lett.*, **1999**, *40*, 7549.

# Preparation of Amides and Acids

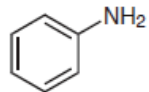
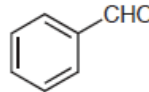
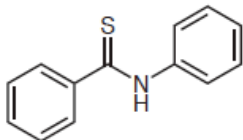
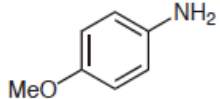
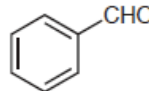
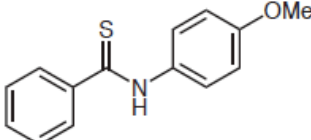
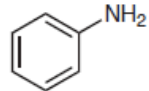
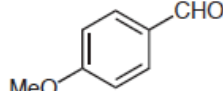
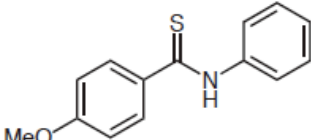
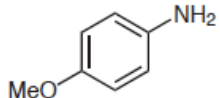
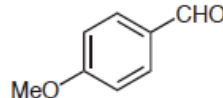
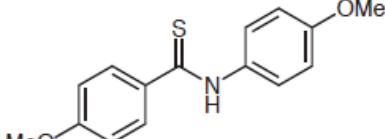
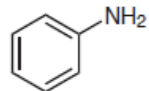
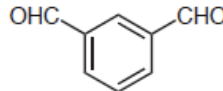
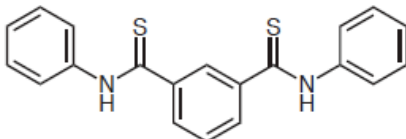
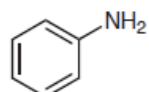
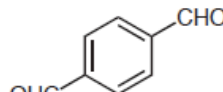
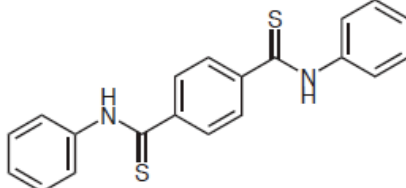
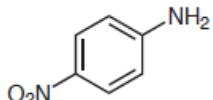
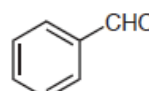
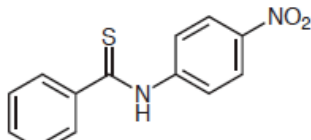


DeTar, D. F.; Carmack, M. *J. Am. Chem. Soc.* **1946**, *68*, 2025.  
Davies, I. W., *et al.* *J. Org. Chem.*, **2000**, *65*, 8415.

# Three Components Reactions

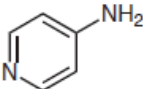
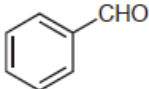
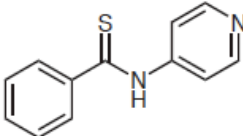
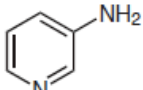
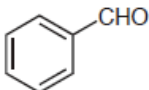
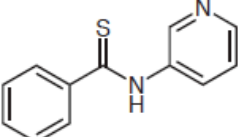
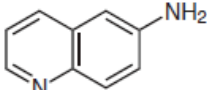
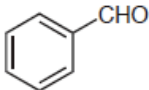
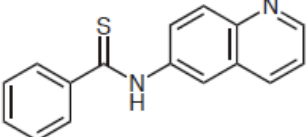
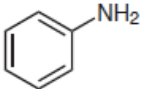
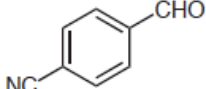
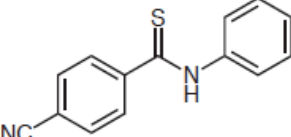
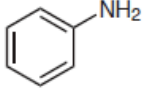
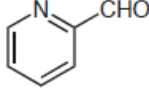
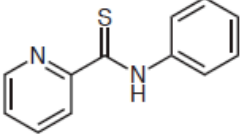




Entry	Amine	Aldehyde	Product	Yield (%) <sup>b</sup>
1				91 (15)
2				92 (40 <sup>c</sup> )
3				78 (7)
4				88 (39)
5 <sup>d</sup>				48 (0 <sup>e</sup> )
6 <sup>d</sup>				40 (0 <sup>e</sup> )
7 <sup>f</sup>				40 (0 <sup>e</sup> )

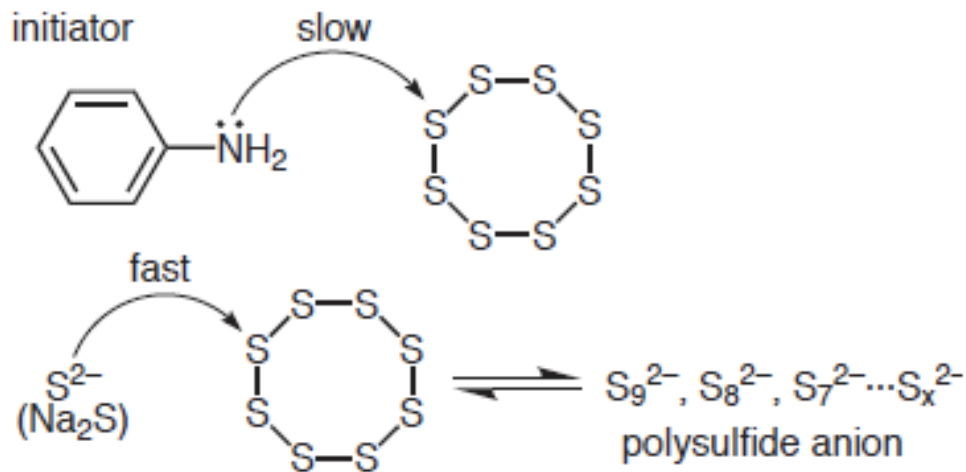
Kanbara, T., *et al. Synlett*, **2007**, 17, 2687.

# Three Components Reactions

Entry	Amine	Aldehyde	Product	Yield (%) <sup>b</sup>
8				70 (51)
9				65 (15)
10				45 (6)
11 <sup>f</sup>				51 (0 <sup>e</sup> )
12				88 (71)

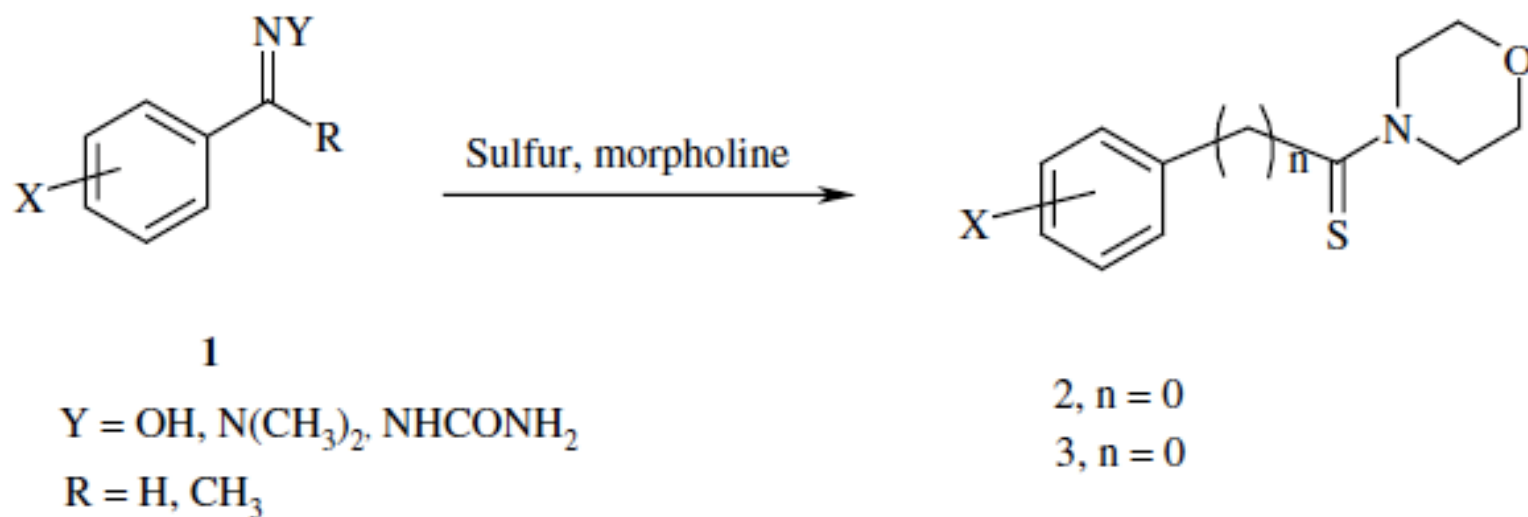
Kanbara, T., *et al. Synlett*, **2007**, 17, 2687.

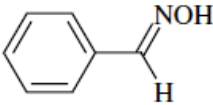
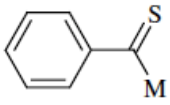
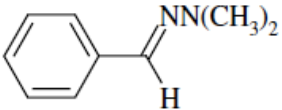
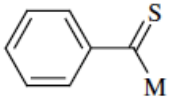
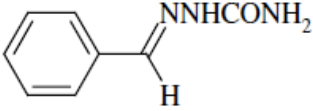
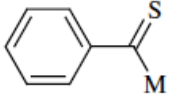
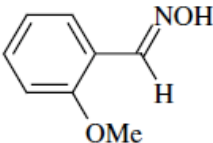
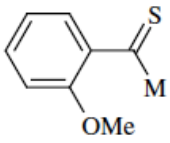
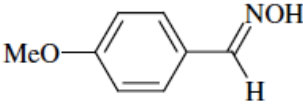
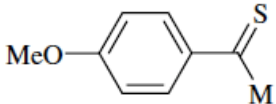
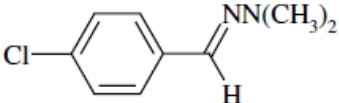
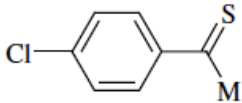
# Three Components Reactions



The addition of a small amount of  $\text{Na}_2\text{S} \cdot 9\text{H}_2\text{O}$  into the reaction mixture of sulfur with aniline caused a color change of the reaction system, from pale yellow to deep blue

# Protected Carbonyl Compounds



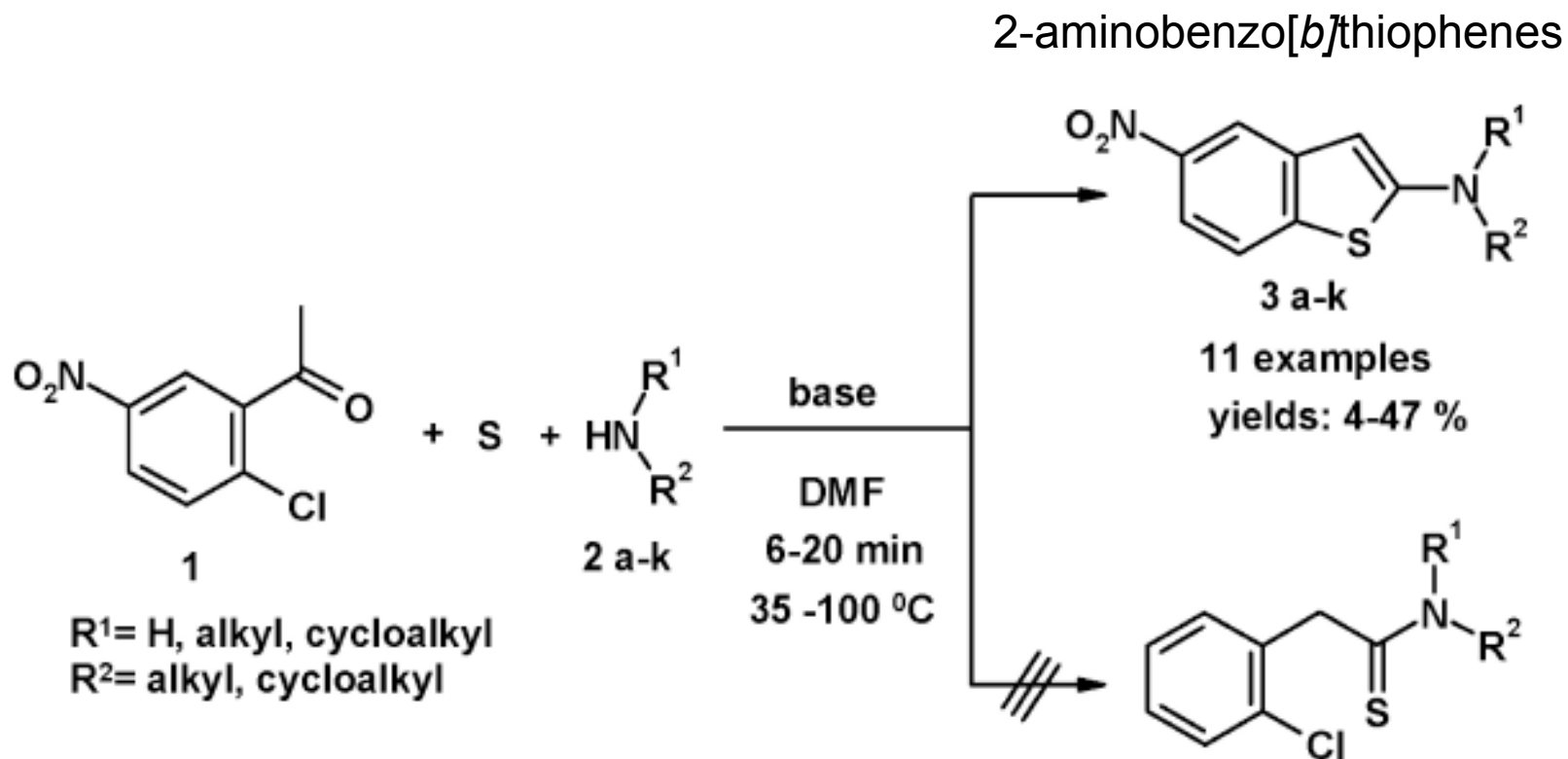
Entry	Substrates 1	Products <sup>a</sup>	Yield (%) <sup>b</sup>	
			Conventional heating <sup>c</sup>	MW <sup>d</sup> (min)
a			65	70 (7)
b			72	79 (10)
c			83	72 (5)
d			43	48 (7)
e			75	80 (7)
f			60	70 (10)

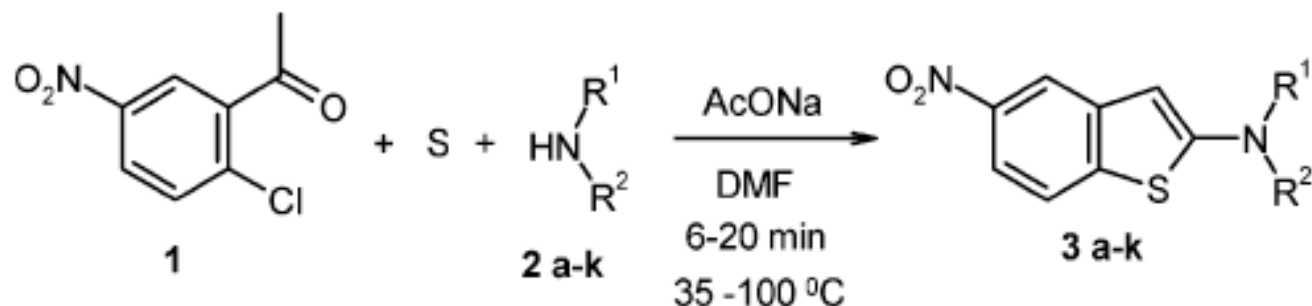
Darabi, H., *et al. Tetrahedron Lett.*, **2004**, *45*, 4167.

Entry	Substrates 1	Products <sup>a</sup>	Yield (%) <sup>b</sup>	
			Conventional heating <sup>c</sup>	MW <sup>d</sup> (min)
g			55	72 (7)
h			71	60 (7) <sup>13</sup>
i			61	55 (7)
j			41	44 (7)
k			55	57 (7)
l			65	73 (10)

Darabi, H., *et al. Tetrahedron Lett.*, **2004**, *45*, 4167.

# Domino Reactions





**TABLE 1.** Reaction Conditions and Yields of 2-Aminobenzo[*b*]thiophenes

entry	R <sup>1</sup>	R <sup>2</sup>	ratio of amine/S/base	temp (°C)	time (min)	yield of <b>3</b> (%)
<b>3a</b>	Me <sup>a</sup>	H	3/5/3	60	10	46
<b>3b</b>	allyl	H	2.17/5/0	45	10	47
<b>3c</b>	<i>n</i> -butyl	H	3.2/3/2	60	8	36
<b>3d</b>	benzyl	H	3.8/5/0	60	10	30
<b>3e</b>	<i>iso</i> -propyl	H	2.3/5/0	60	10	14
<b>3f</b>	cyclopentyl	H	2.15/5/0	60	10	40
<b>3g</b>	cyclohexyl	H	1.95/5/0	35	6	19
<b>3h</b>	Me <sup>a</sup>	Me	1.5/1.5/1.5	100	180	4
<b>3i</b>	-(CH <sub>2</sub> ) <sub>5</sub> -		3.7/3/2	60	15	31
<b>3j</b>	-(CH <sub>2</sub> ) <sub>6</sub> -		1.2/1.5/1.5	100	20	10
<b>3k</b>	-(CH <sub>2</sub> CH <sub>2</sub> OCH <sub>2</sub> CH <sub>2</sub> )-		1.4/5/1.5	100	12	12



**Thank You**