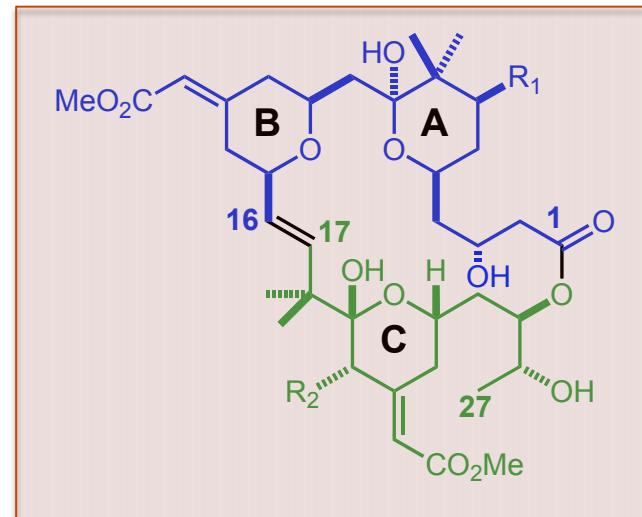
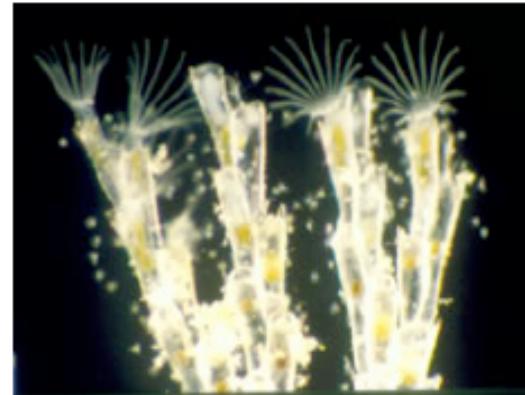


Total Synthesis of Bryostatins

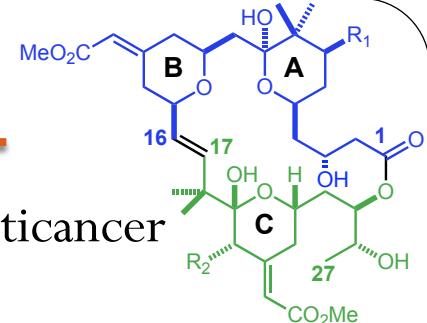
Group Meeting Presentation
Anil Kumar Gupta
The Wulff Group
February 20, 2009

Isolation

- In 1968, new macrolides were isolated from the marine bryozoan invertebrates *Bugula neritina* Linnaeus and *Amathia convulata* nearby Jack Rudloe of the Gulf Specimen Company off the west coast of Florida.
- Bryostatins consist of at least 20 members, which vary at R₁ and R₂.
- Characterised by Pettit *et al* in 1982.



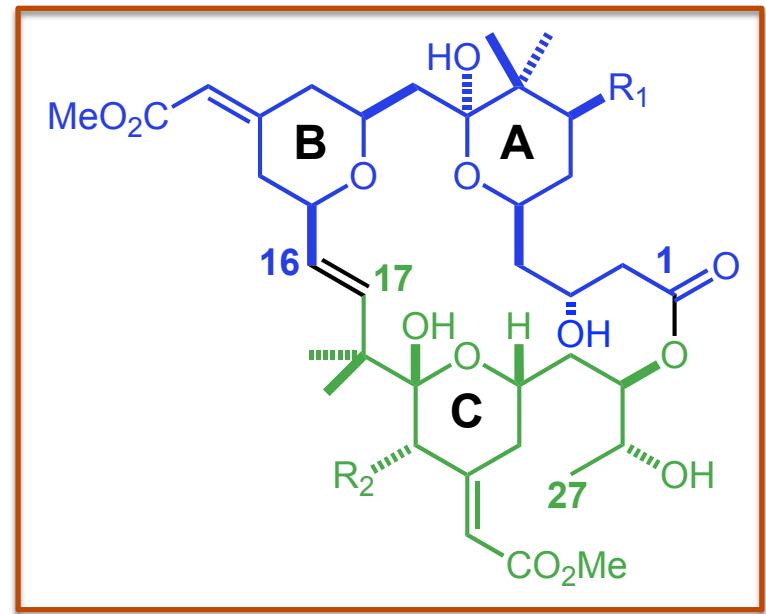
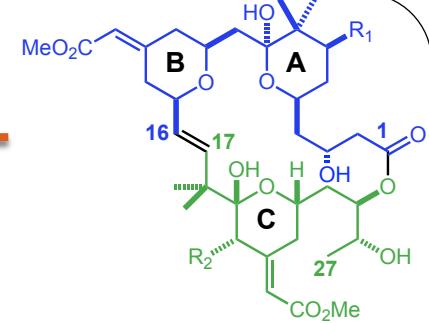
Biological Activity



- They exhibit exceptional biological activity, most notable their anticancer activity *in vivo*.
- Bryostatin significantly affects both cognition and memory enhancement in animals, Hence, treatment of Alzheimer's disease, depression and other cognitive impairments.
- Their clinical advancement is hampered by the limited availability of bryostatins from isolation, due to low yield ($10^{-3}\%$ to $10^{-8}\%$).
- Currently in phase I and phase II clinical trials for melanoma, myeloma, chronic lymphocytic leukemia (CLL), AIDS related lymphoma, non-Hodgkin's lymphoma, colorectal, renal, prostate, head and neck, cervix, ovarian, breast, peritoneal, stomach, esophagus, anus, prostate, and nonsmall cell lung cancer.

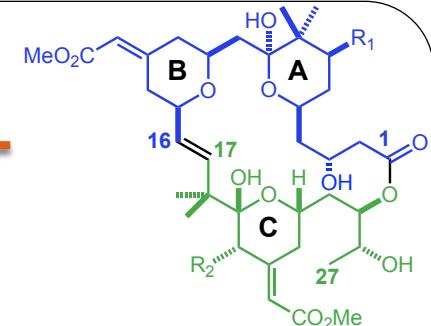
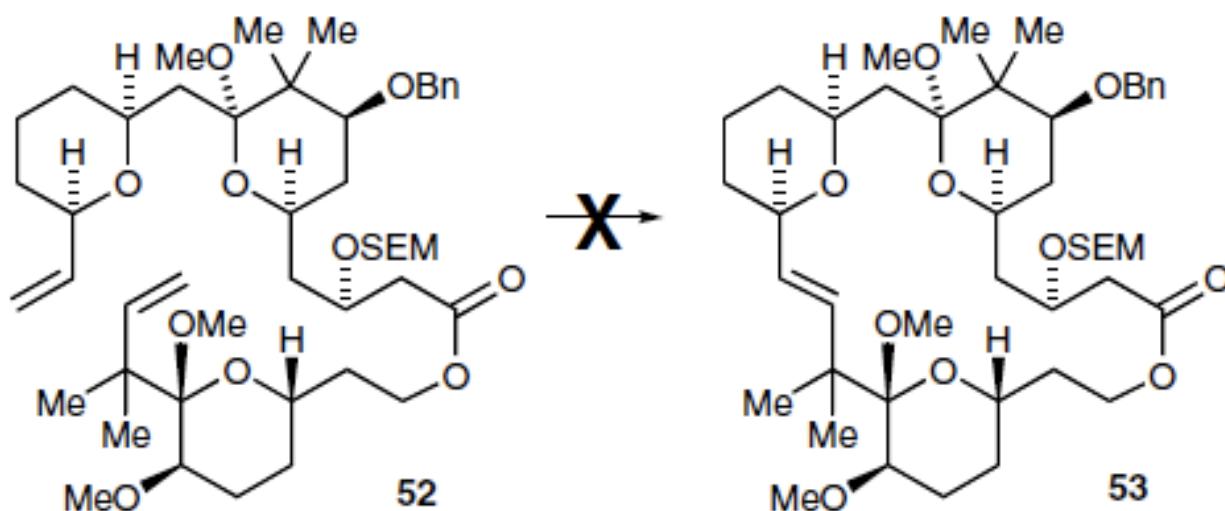
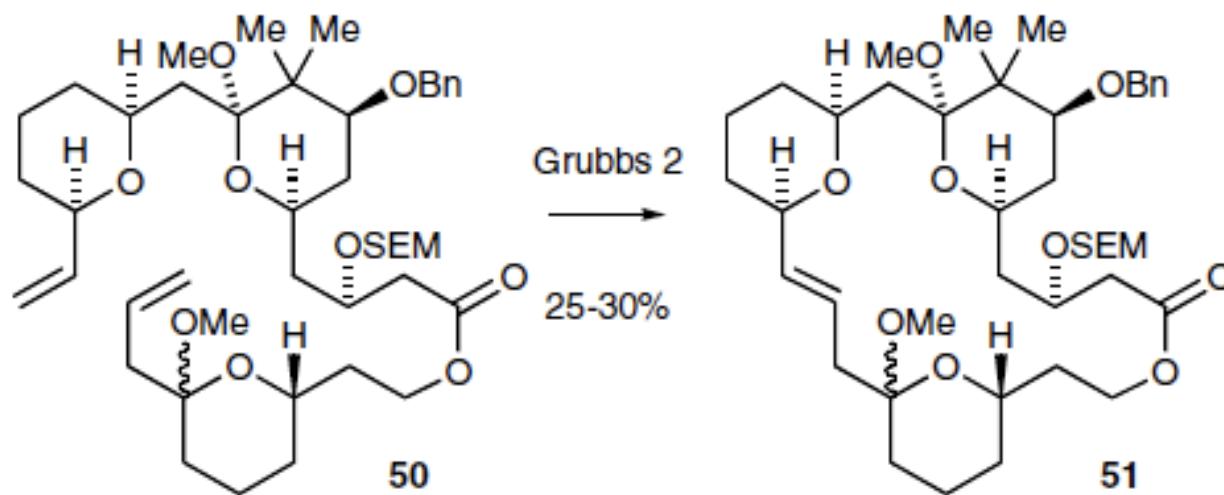
Architectural features

- Two Main Hemispheres:
 - Northern Hemisphere (C1-16)
 - Southern Hemisphere (C17-C22)
- Three heavily substituted tetrahydropyran rings (A, B, C) rings
- Two acid/base-sensitive *exo*-cyclic unsaturated esters
- One congested C16–C17 *trans*-alkene
- Numerous oxygen-containing functionalities on a 26-membered lactone



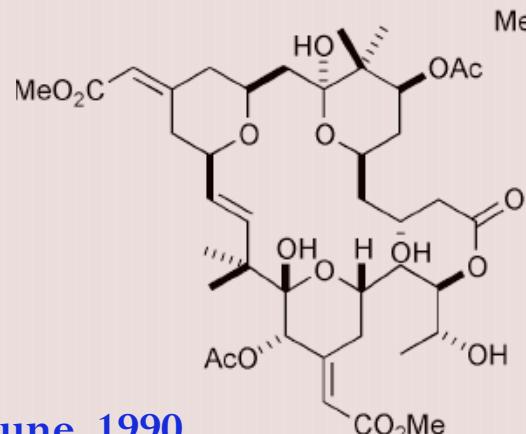
Architectural features

- One congested C16–C17 *trans*-alkene



Approaches towards Bryostatins

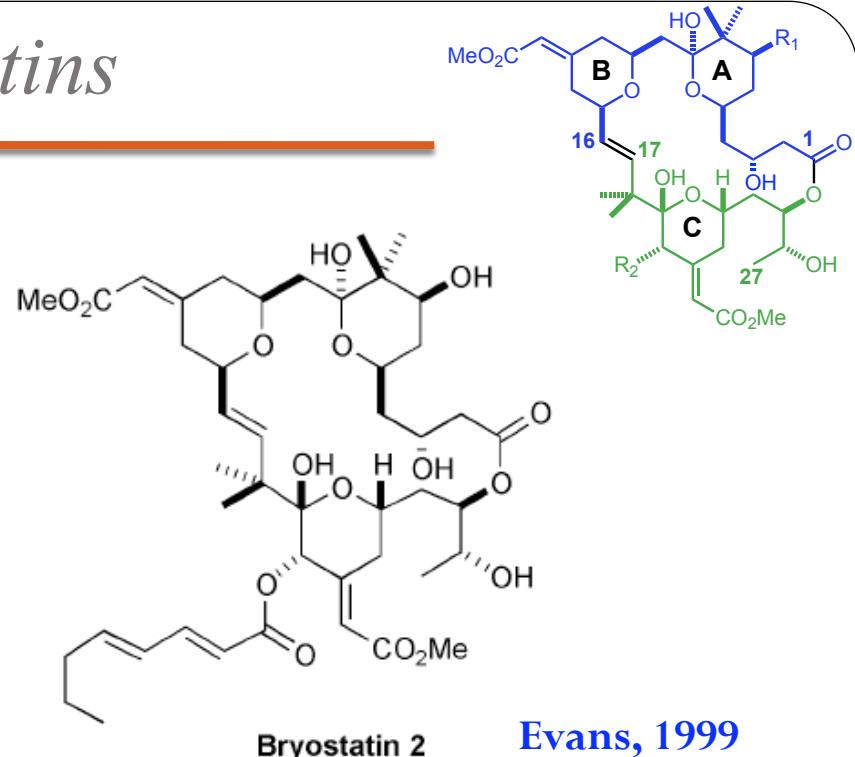
Completed Total Synthesis so far :



Masamune, 1990

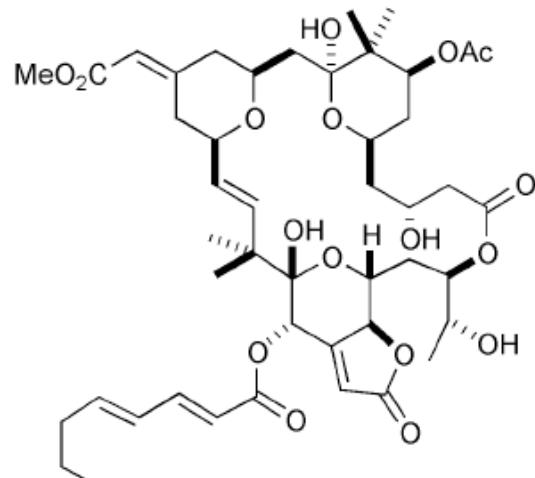
Hale, 2006 (Formal)

Bryostatin 7



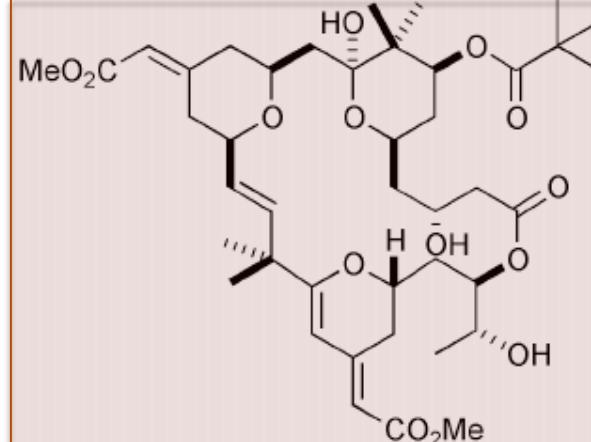
Bryostatin 2

Evans, 1999



Bryostatin 3

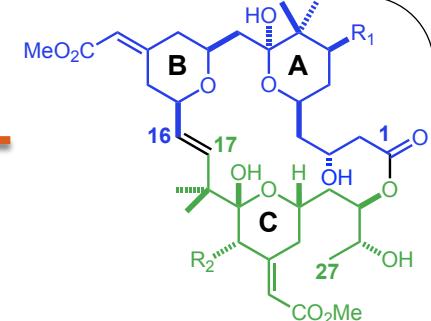
Yamamura, 2000



Bryostatin 16

Trost, 2008

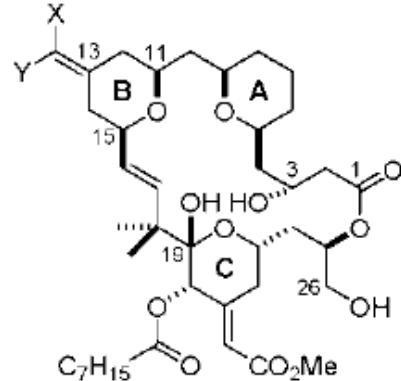
Previous Approaches to Bryostatins



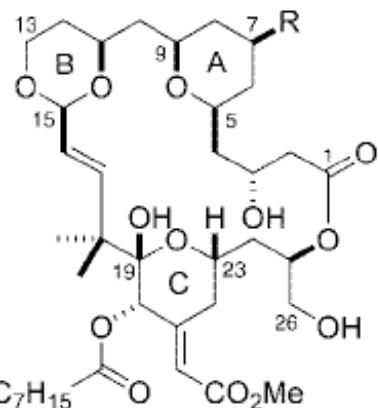
Other Groups Involved:

- E. J. Thomas, Bryostatin 11, C1-C16,C17-C27 (**1989, 1998, 2000, 2004, 2006, 2008**)
- M. Vandewalle, Bryostatin 11,C1-C9, B,C ring (**1991,1994,1997**)
- R. W. Hoffmann, C1-C9 (**1995**)
- M. Kalesse, C1-C9 (**1996**)
- R. Roy, Bryostatin 1, C1-C9, C21-C27 (**1989, 1990**)
- H. M. R. Hoffmann, C1-C16, C1-C9, B ring (**1996,1997,2001**)
- K. D. Janda, Bryostatin 1, C21-C27 (**2000**)
- J. S. Yadav, , Bryostatin 1, C1-C16 (**2001**)
- S. D. Burke, Bryostatin 1, C1-C16, C17-C27 (**2004**)
- G. E. Keck, Bryostatin 1, C1-C16 (**2006**)

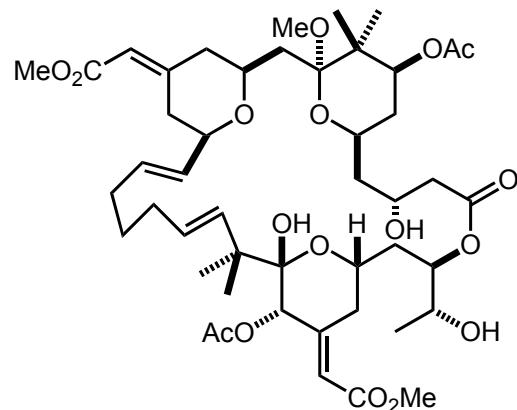
Approaches Towards Bryologs



1 X = Y = H $K_i = 1.6 \text{ nM}$
2 X = CO₂Me, Y = H $K_i = 2.5 \text{ nM}$
3 X = H, Y = CO₂Me $K_i = 0.9 \text{ nM}$

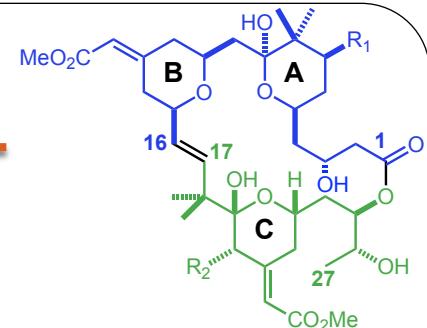


Analog **1**: R = H $K_i = 0.25 \text{ nM}$
 C7 oxygenated analogs: R = OR'

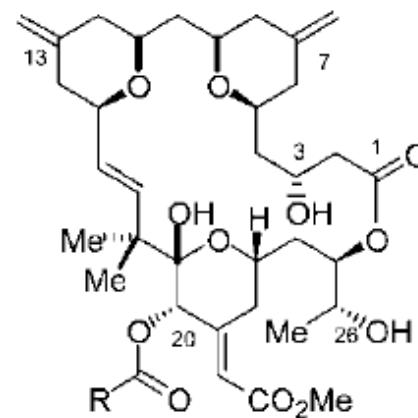


B. M. Trost
 (2007)

Bryostatin 1: R = Ac $K_i = 1.4 \text{ nM}$
Bryostatin 2: R = H $K_i = 5.9 \text{ nM}$



P. A. Wender
 (2008)



R = Ph $K_i = 0.70 \pm 0.01 \text{ nM}$
 R = C₇H₁₅ $K_i = 1.05 \pm 0.04 \text{ nM}$
 R = $K_i = 0.70 \pm 0.06 \text{ nM}$

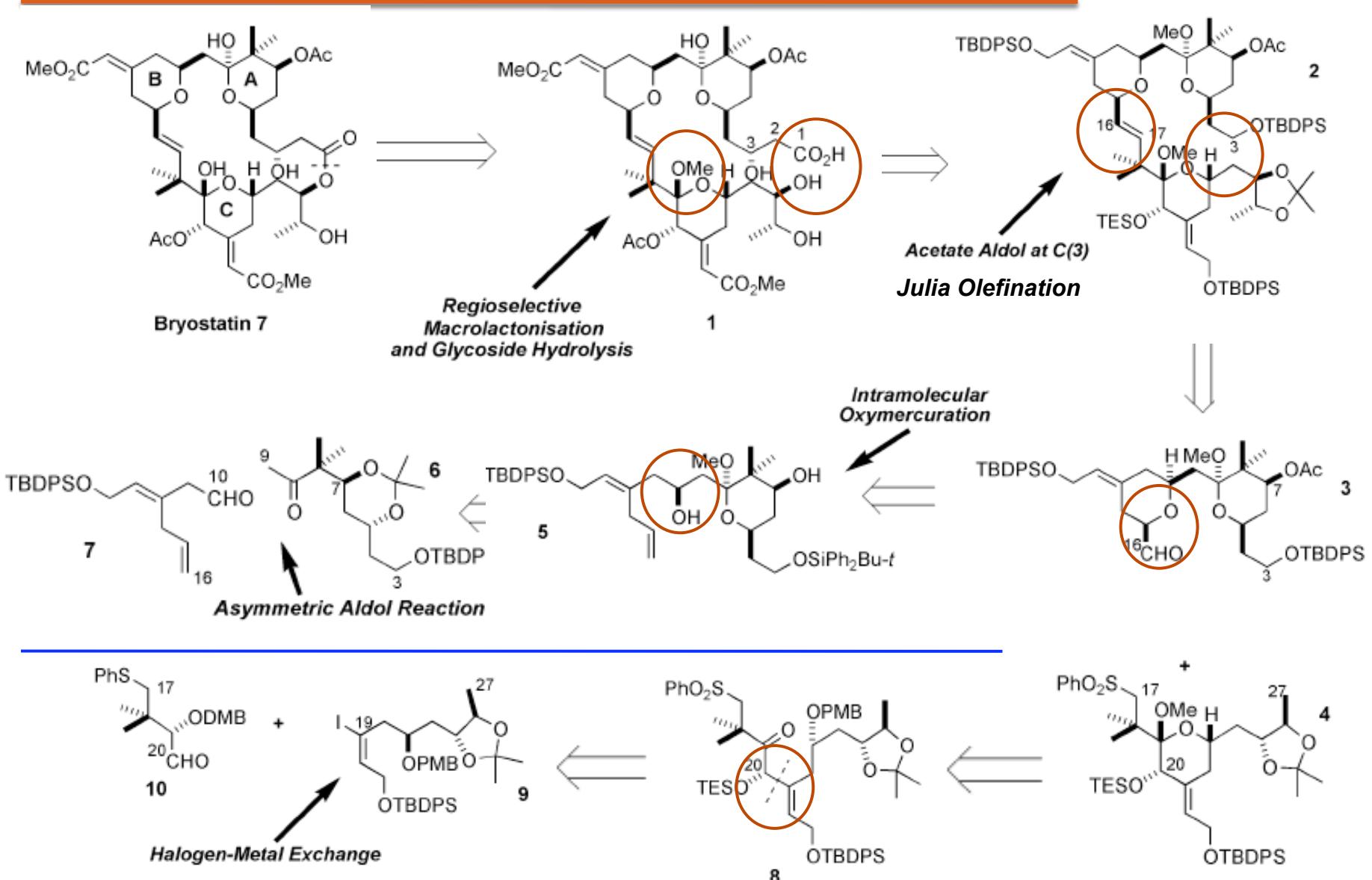
G. E. Keck (2008)

Trost, B. M.; Yang, H.,; Thiel, O. R.; Frontier, A. J.; Brindle, C. S. *J. Am. Chem. Soc.* **2007**, *129*, 2206
 Wender, P. A.; DeChristopher, B. A.; Schrier, A. J. *J. Am. Chem. Soc.* **2008**, *130*, 6658

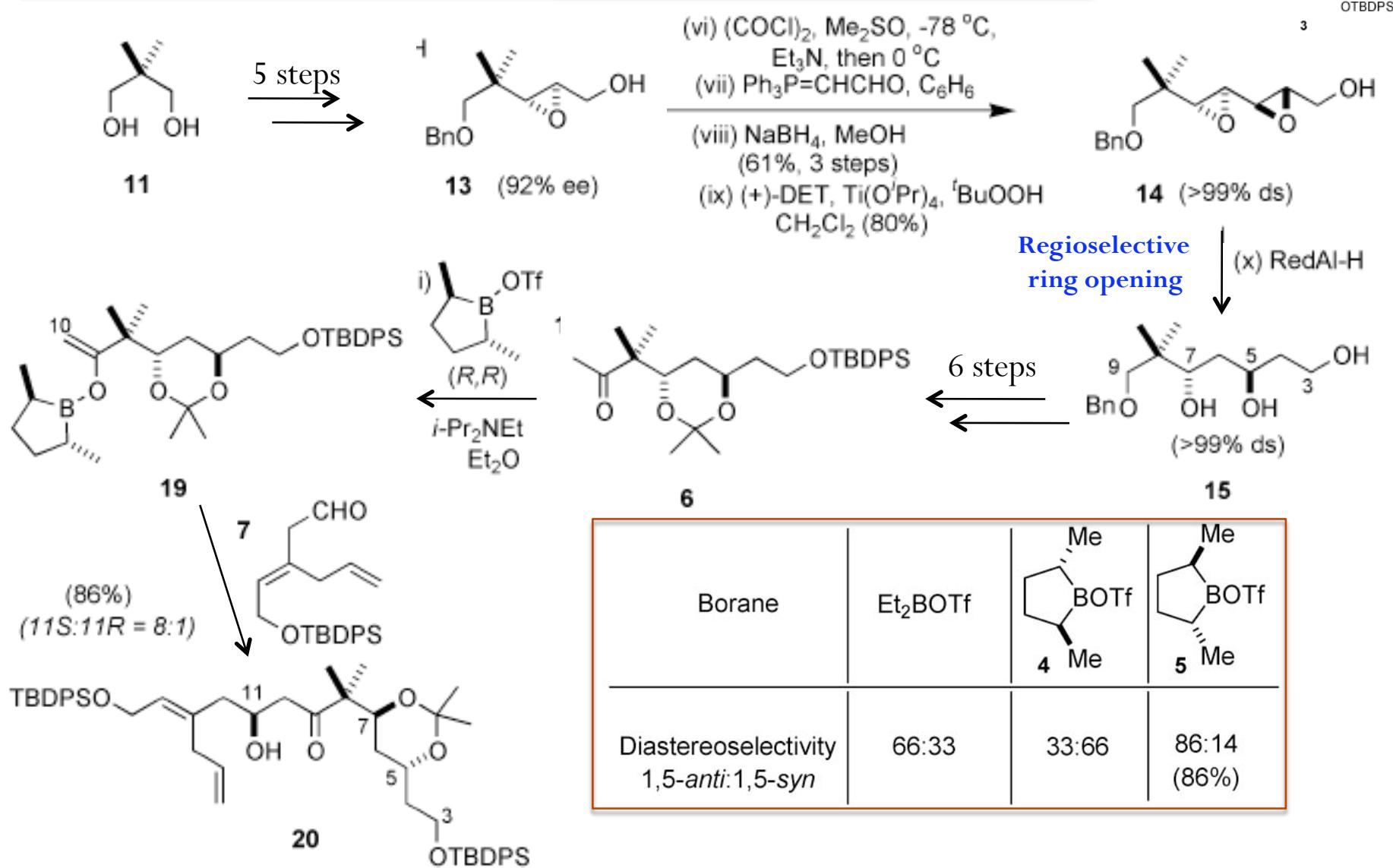
Wender, P. A. & Verma, V. A. *Org. Lett.* **2008**, *10*, 3331

Keck, G. E.; Kraft, M. B.; Truong, A. P.; Li, W.; Sanchez, C. C.; Kedei, N.; Lewin, N. E.; Blumberg, Peter M. *J. Am. Chem. Soc.* **2008**, *130*, 6660

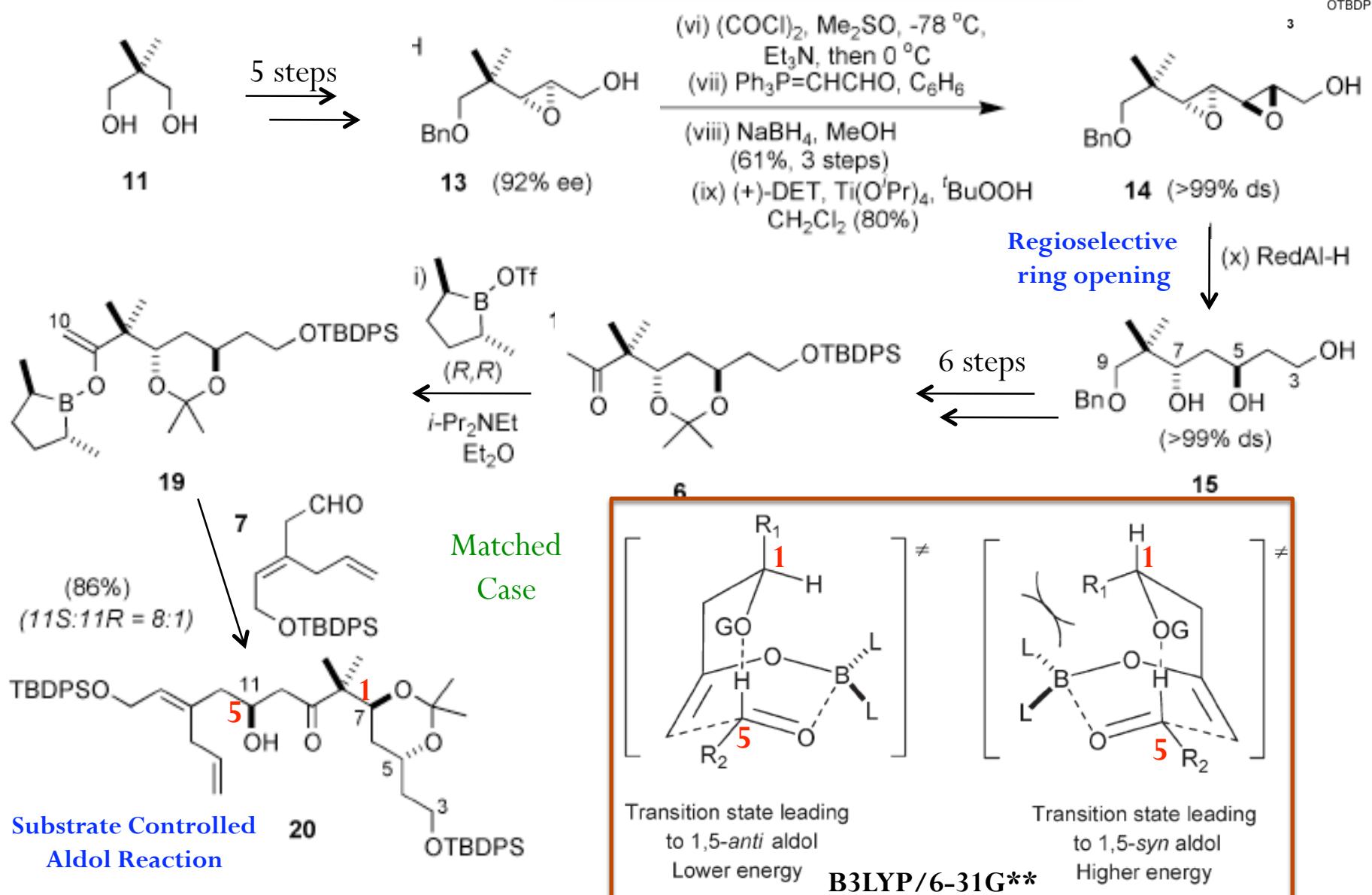
Masamune's Approach to Bryostatin 7



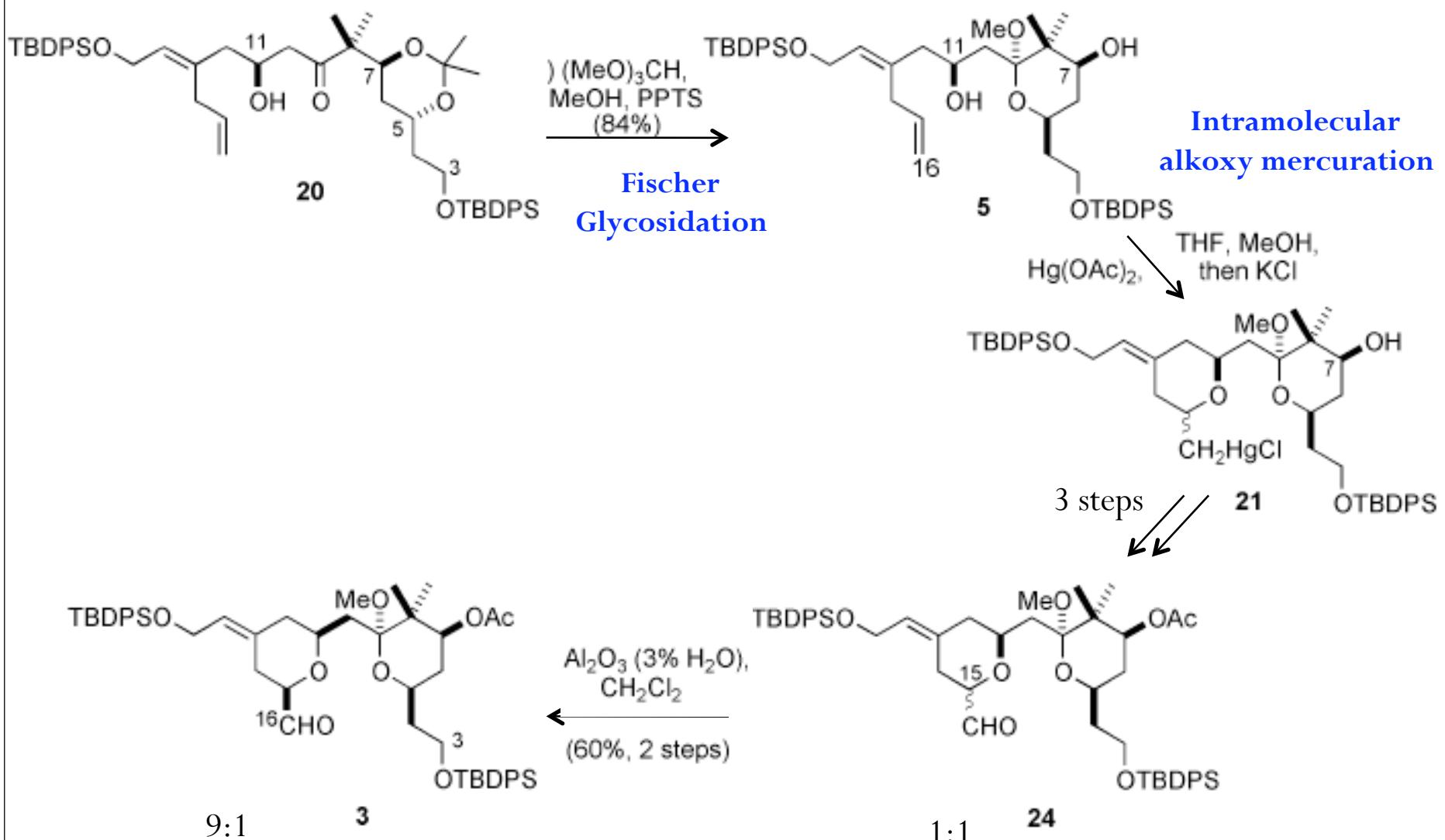
Masamune's Approach to Bryostatin 7



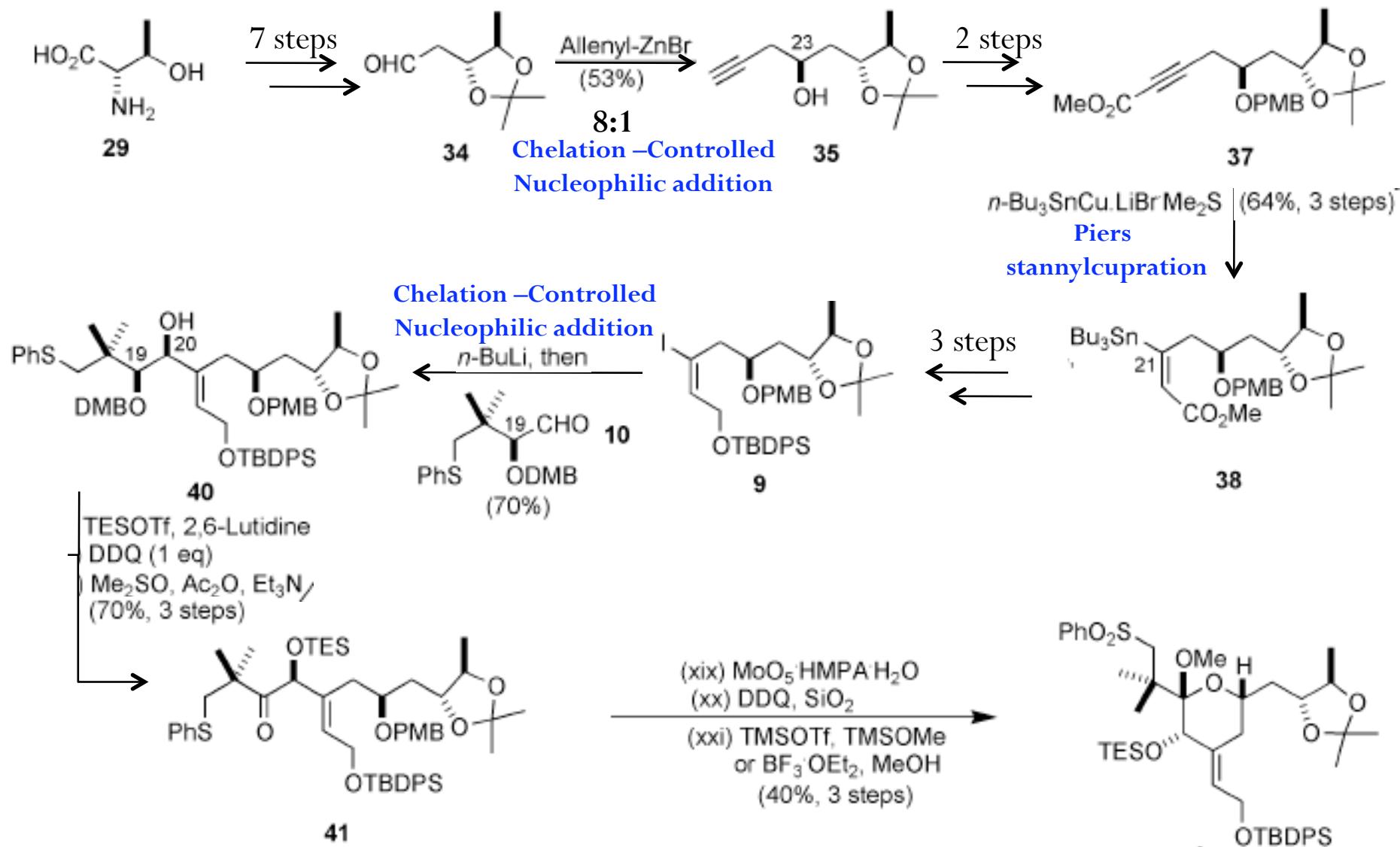
Masamune's Approach to Bryostatin 7



Masamune's Approach to Bryostatin 7

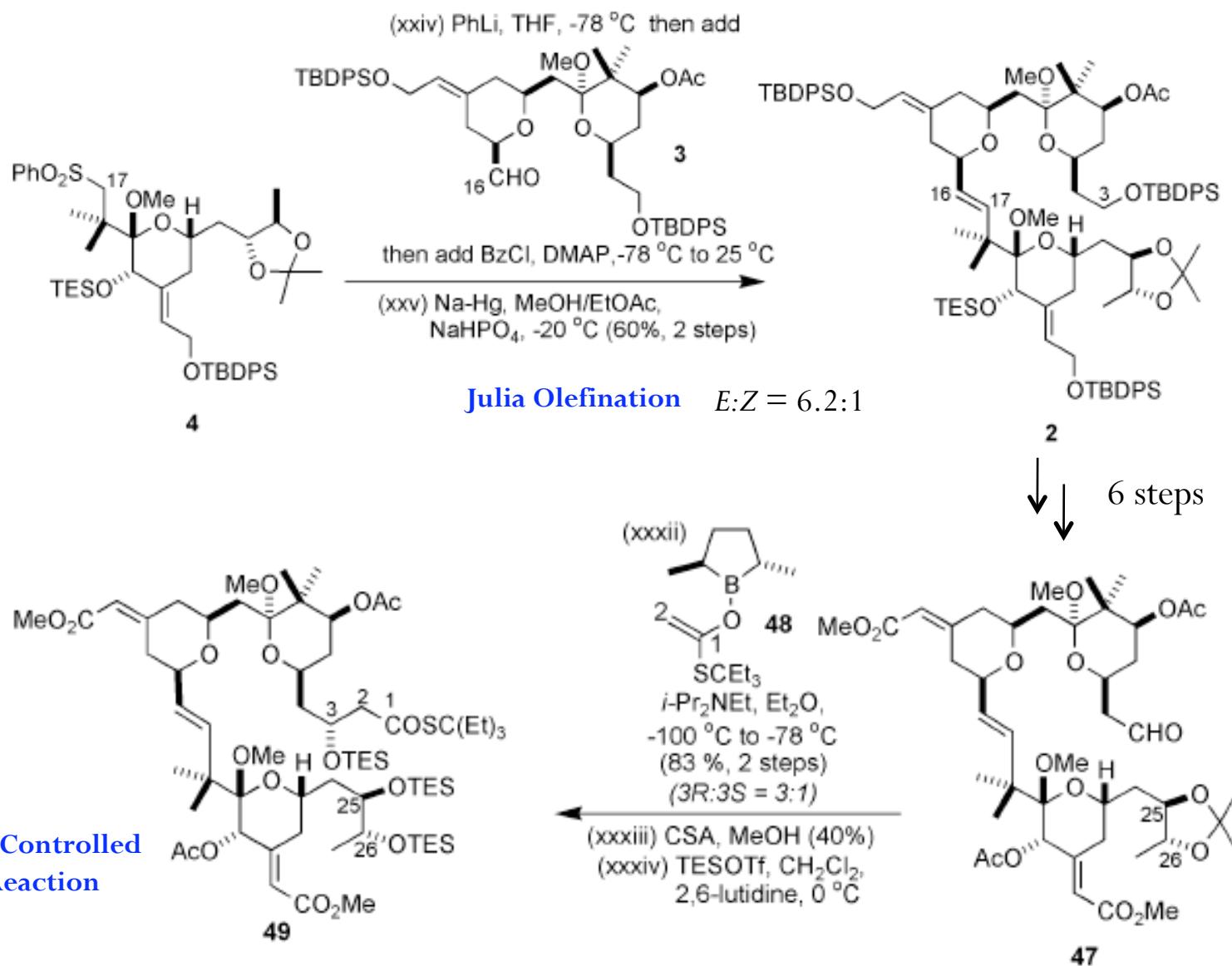
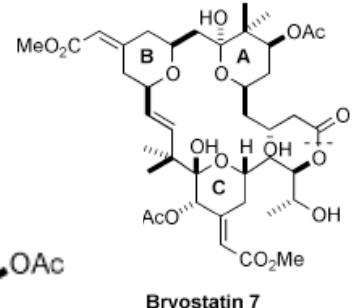


Masamune's Approach to Bryostatin 7



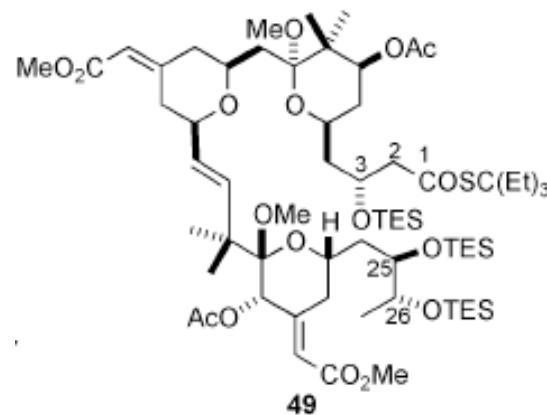
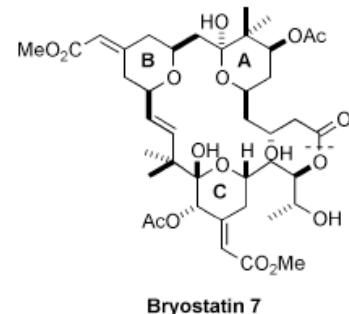
Kageyama, M.; Tamura, T.; Nantz, M. H.; Roberts, J. C.; Somfai, P.; Whritenour, D. C.; Masamune, S. *J. Am. Chem. Soc.* **1990**, *112*, 7407

Masamune's Approach to Bryostatin 7

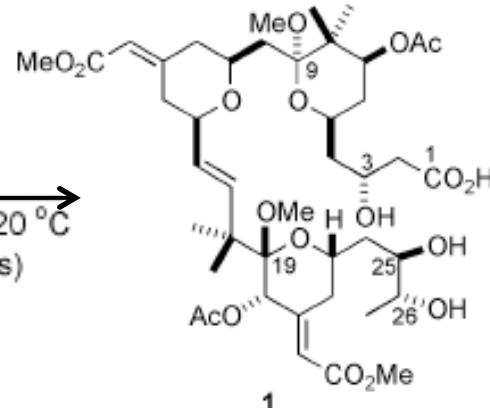


Kageyama, M.; Tamura, T.; Nantz, M. H.; Roberts, J. C.; Somfai, P.; Whritenour, D. C.; Masamune, S. *J. Am. Chem. Soc.* **1990**, *112*, 7407

Masamune's Approach to Bryostatin 7

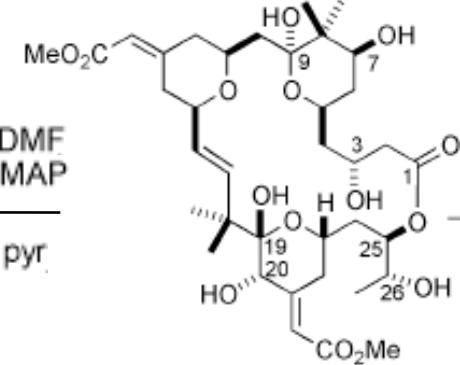
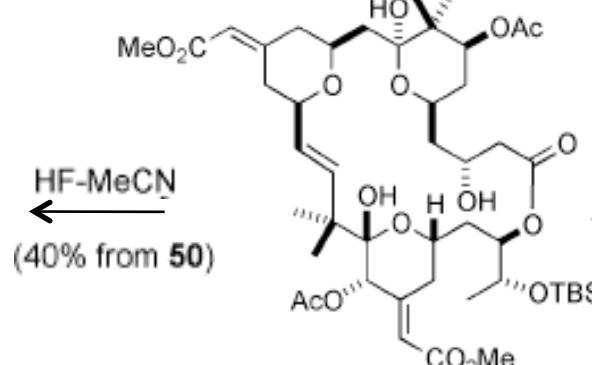
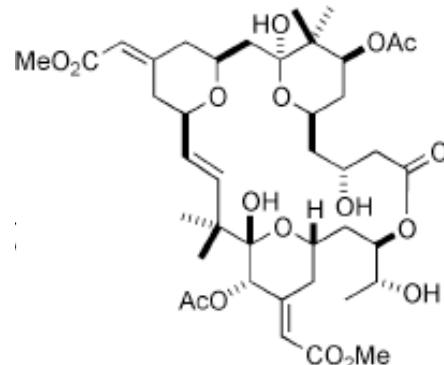


(xxxv) $\text{Hg}(\text{O}_2\text{CCF}_3)_2$
 NaHPO₄, THF
 (xxxvi) HF-py, THF, -20 °C
 (64%, 3 steps)



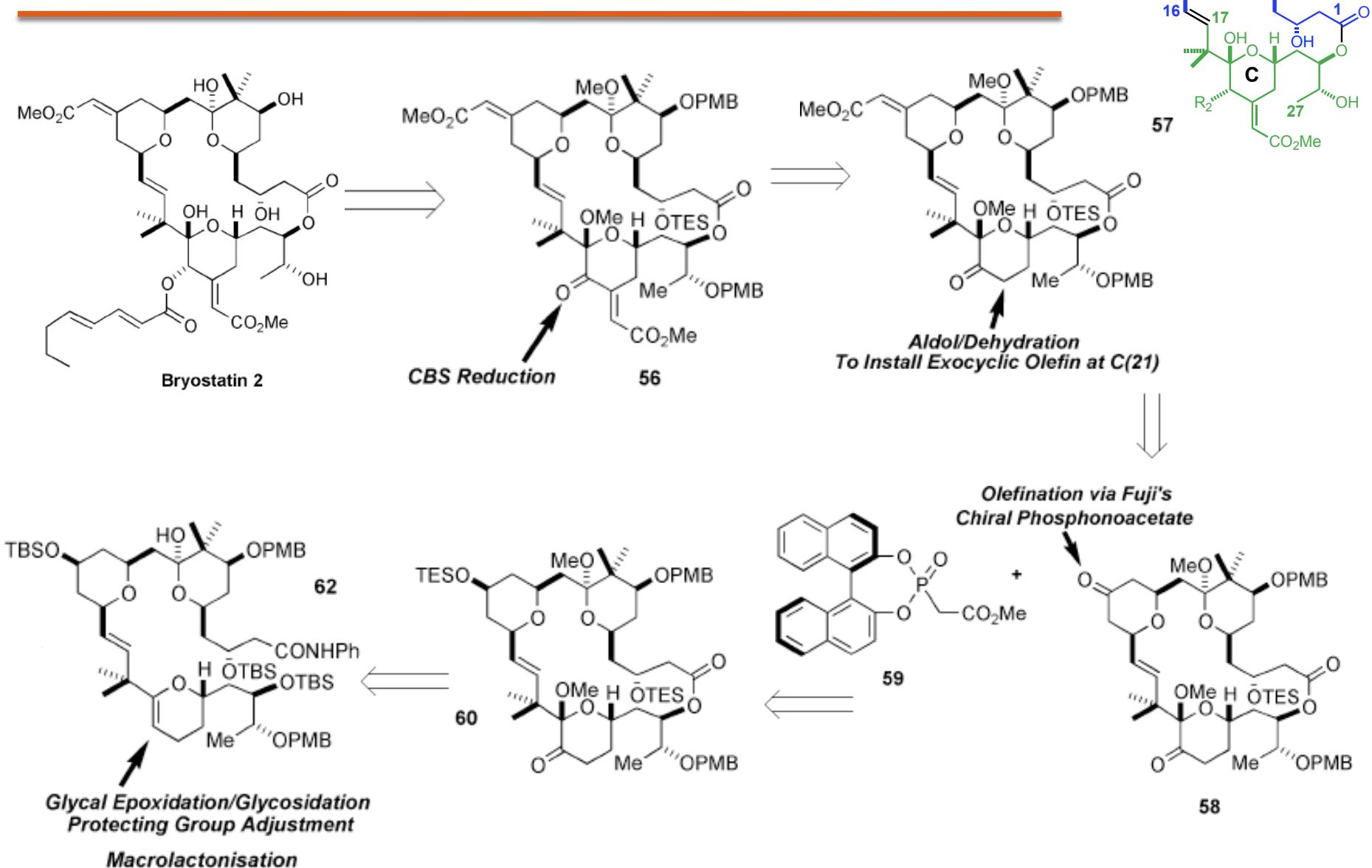
**Regeoselective
 Steglich Macrolactonisation
 & Glycoside hydrolysis
 (C9 vs C19)**

(i) DCC, PPTS, py,
 $\text{CH}_2=\text{CHCH}_2\text{Cl}$, reflux (51%)
 (ii) K_2CO_3 , MeOH, then
 5% aqueous HCl workup (54%)



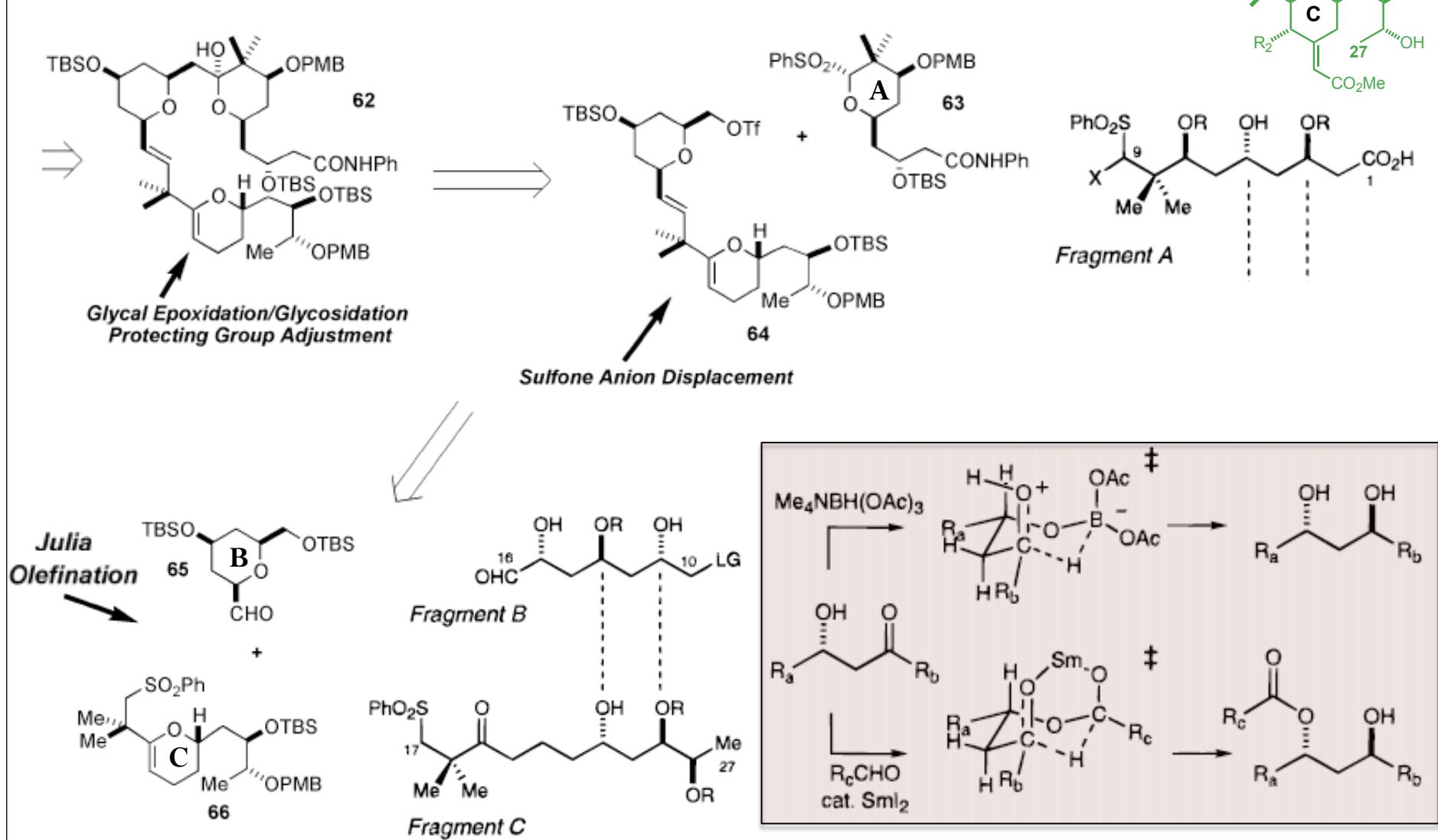
Bryostatin 7

Evans Approach to Bryostatin 2



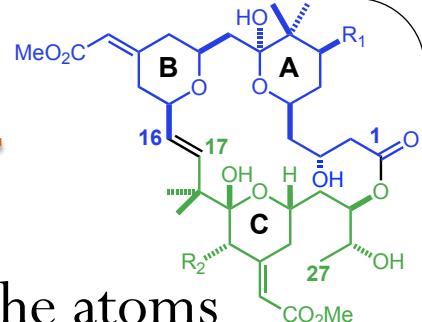
Evans, D. A.; Carter, P. H.; Carreira, E. M.; Charette, A. B.; Prunet, J. A.; Lautens, M. *J. Am. Chem. Soc.* 1999, 121, 7540.

Evans Approach to Bryostatin 2



Evans, D. A.; Carter, P. H.; Carreira, E. M.; Charette, A. B.; Prunet, J. A.; Lautens, M. *J. Am. Chem. Soc.* 1999, 121, 7540.

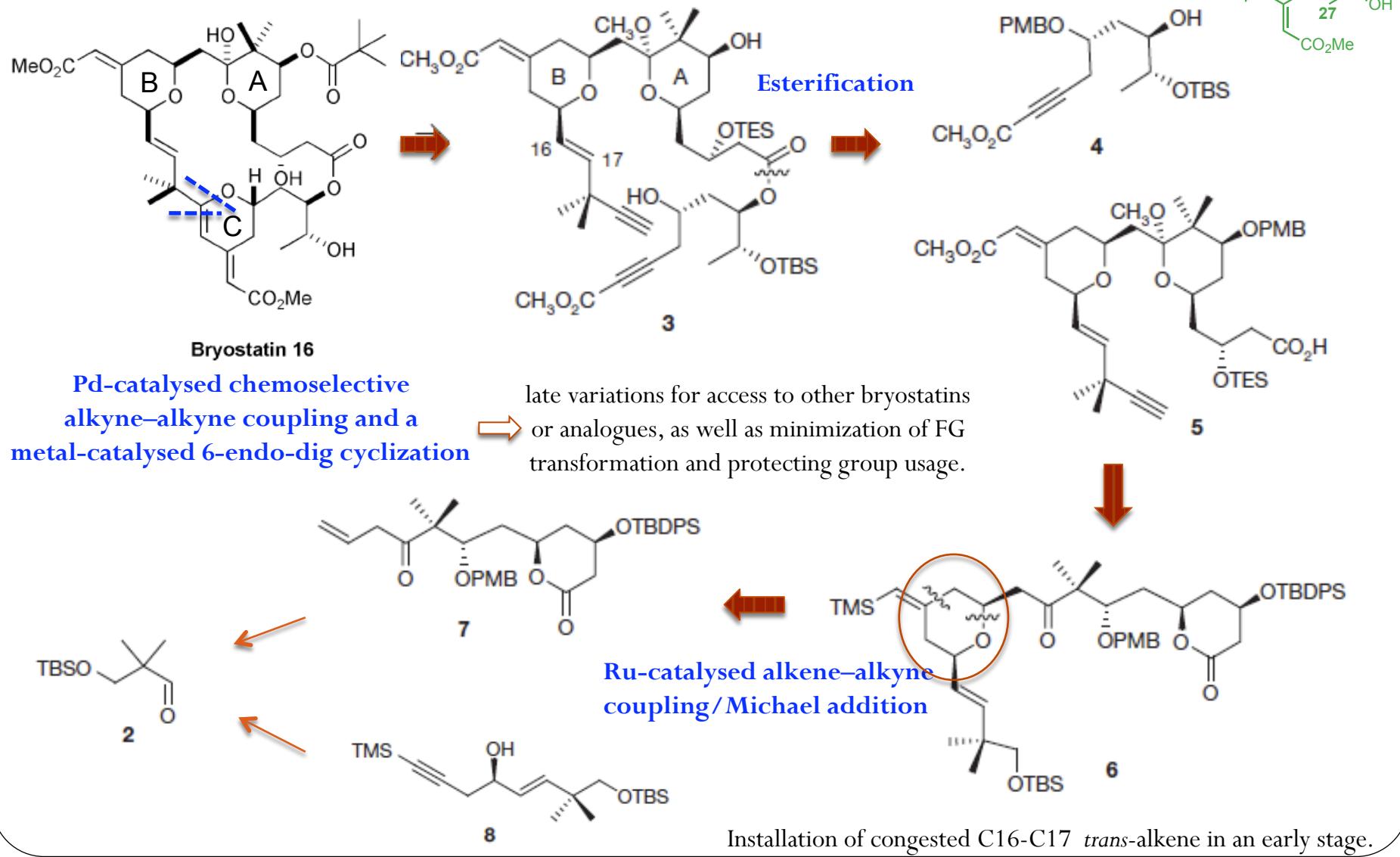
Trost's Approach to Bryostatin 16



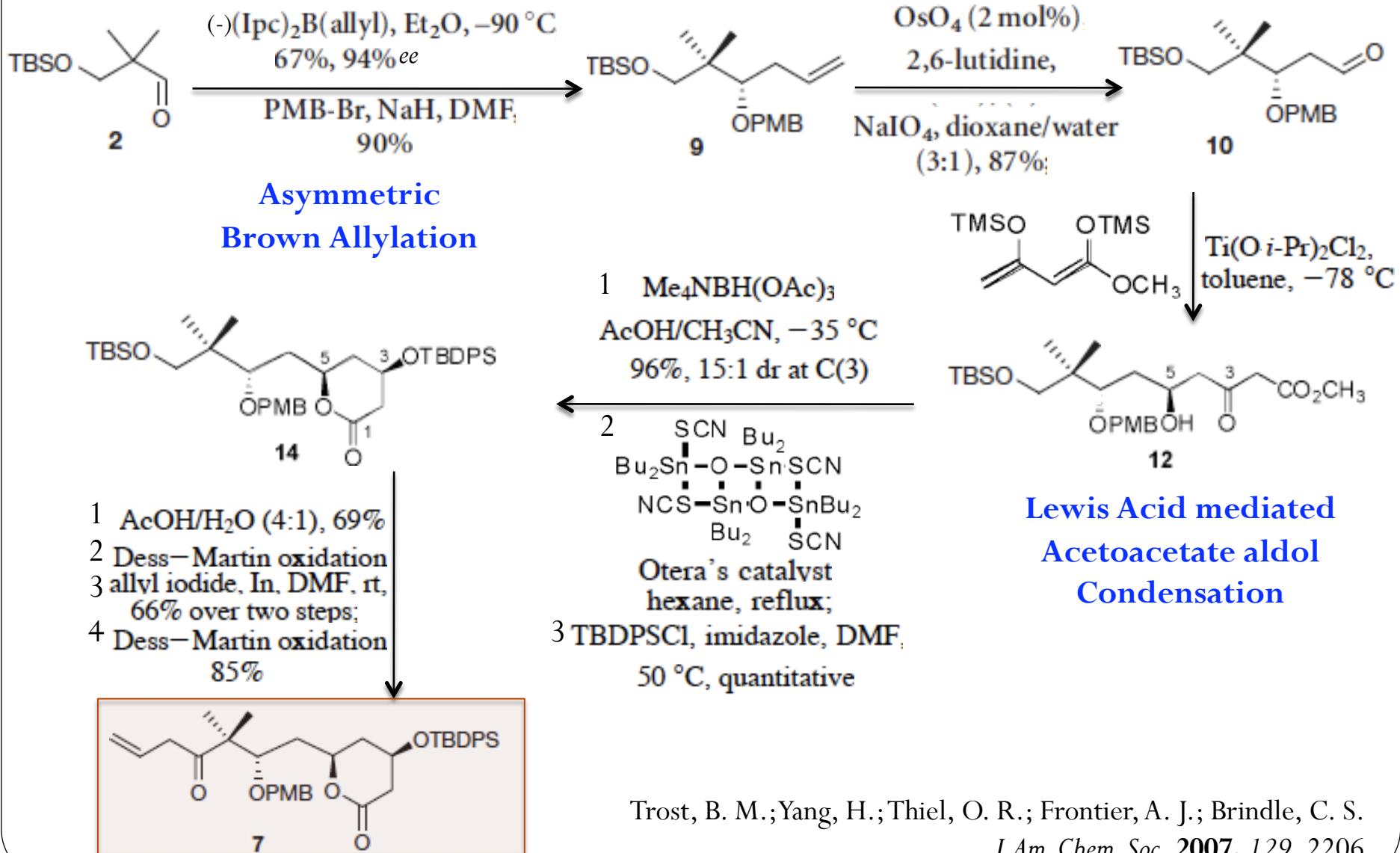
- **Atom economy** (the use of routes in which most of the atoms present in the reactants also end up in the product)
- **Chemoselectivity** (the use of reactions that take place only at desired positions in a molecule).
- A pivotal parent structure allowing access to all other bryostatins
- New analogues, might be readily obtained simply by variations in this natural product's synthesis.
- Palladium catalysed alkyne–alkyne coupling as a macrocyclization method for complex natural product synthesis.
- Gold catalysed conversion of the product of above step into a dihydropyran (the ‘C ring’ of bryostatins)

Trost's Retrosynthesis

Previous Approaches: a difficult Julia olefination followed by a lactonization

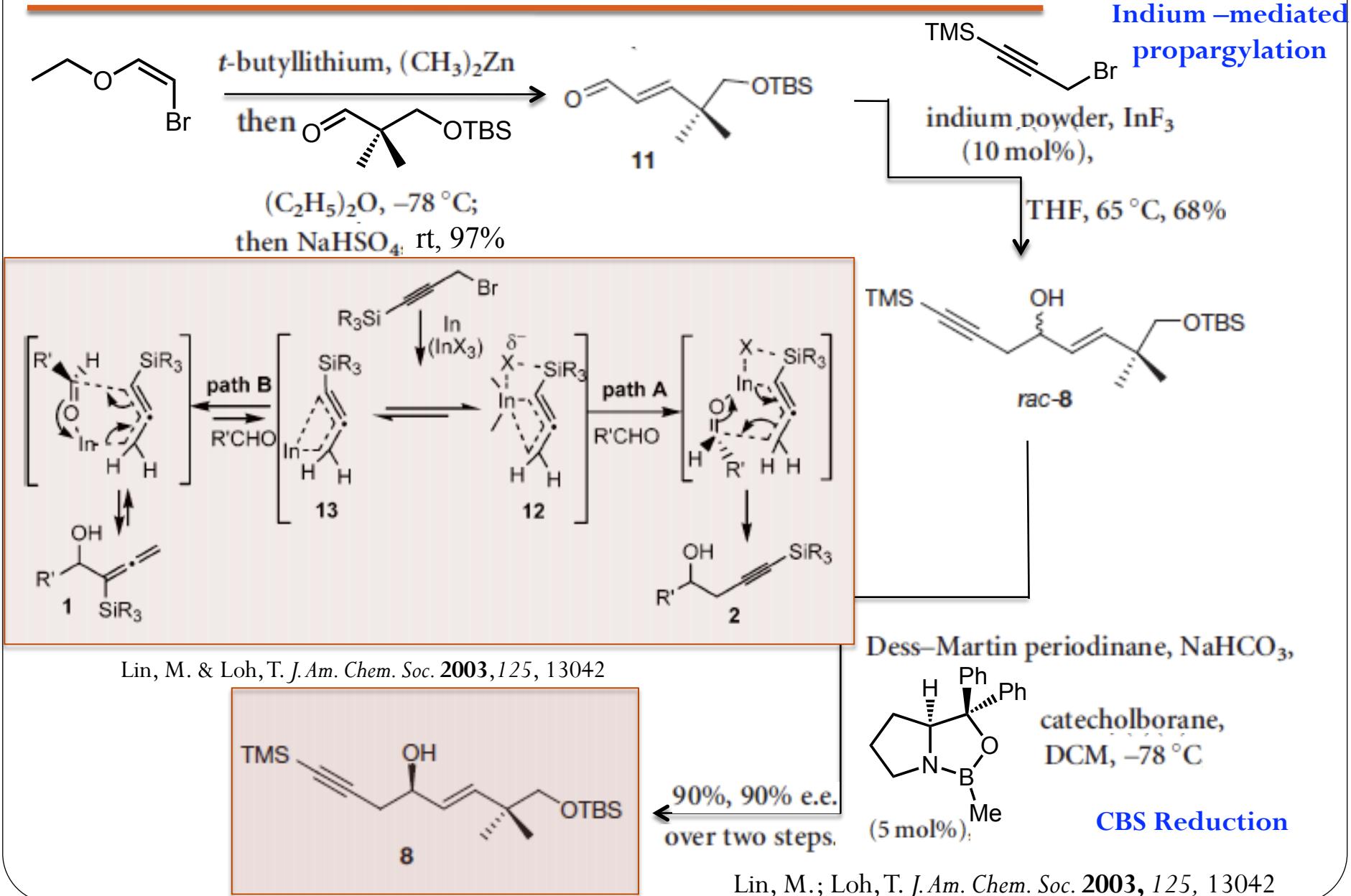


Synthesis of fragment 7

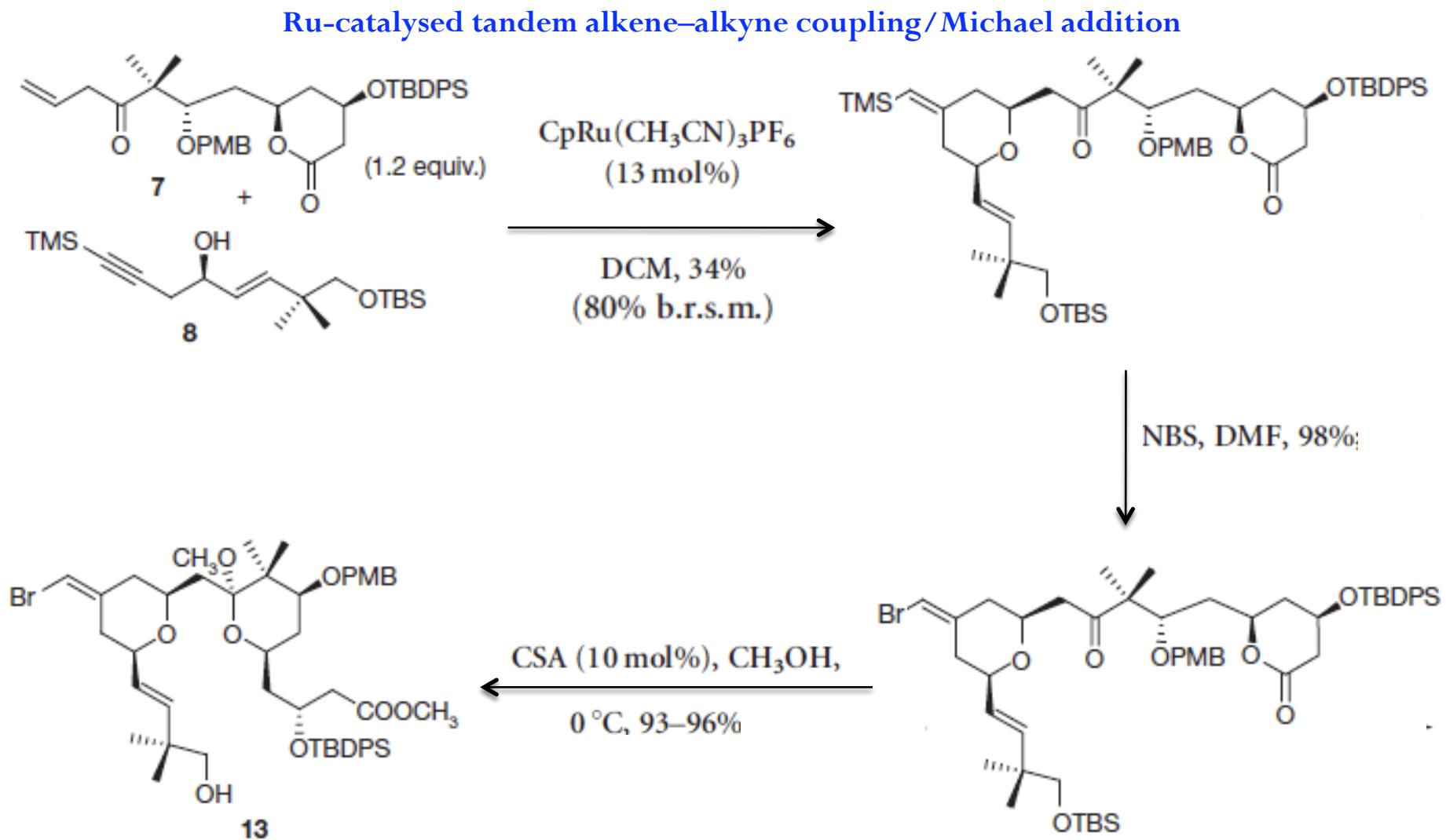


Trost, B. M.; Yang, H.; Thiel, O. R.; Frontier, A. J.; Brindle, C. S.
J. Am. Chem. Soc. **2007**, *129*, 2206

Synthesis of fragment 8

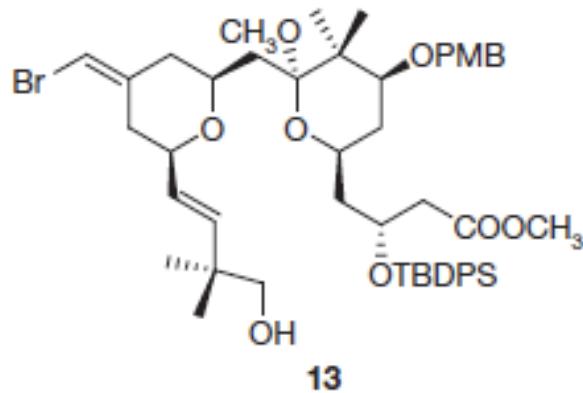


Synthesis of Acid Functionality



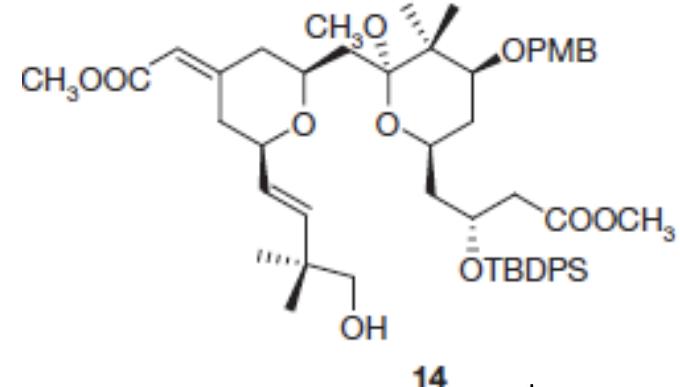
Trost, B. M.; Yang, H.; Wuitschik, G. A. *Org. Lett.* **2005**, *7*, 4761–4764

Synthesis of Acid Functionality



PdCl₂(CH₃CN)₂ (10 mol%)
dppf (30 mol%)CO (1 atm)
CH₃OH, (C₂H₅)₃N, DMF,
80 °C, 83% (90% b.r.s.m.)

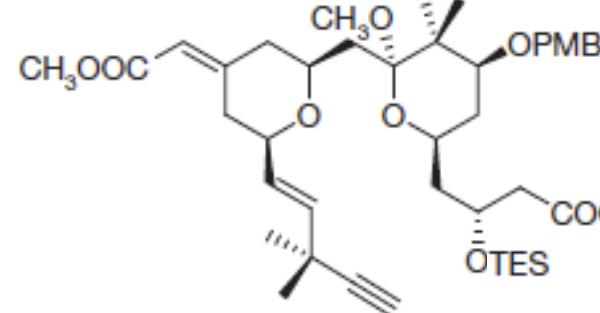
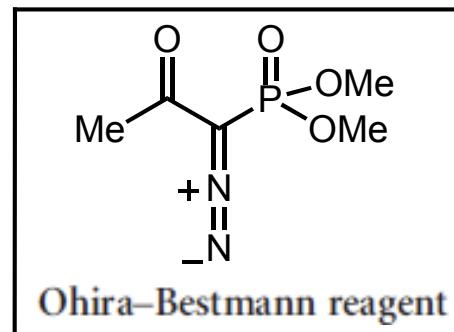
Pd-catalysed carbonylation



Dess–Martin periodinane
NaHCO₃, DCM, 88%

Ohira–Bestmann reagent
K₂CO₃, CH₃OH, 97%

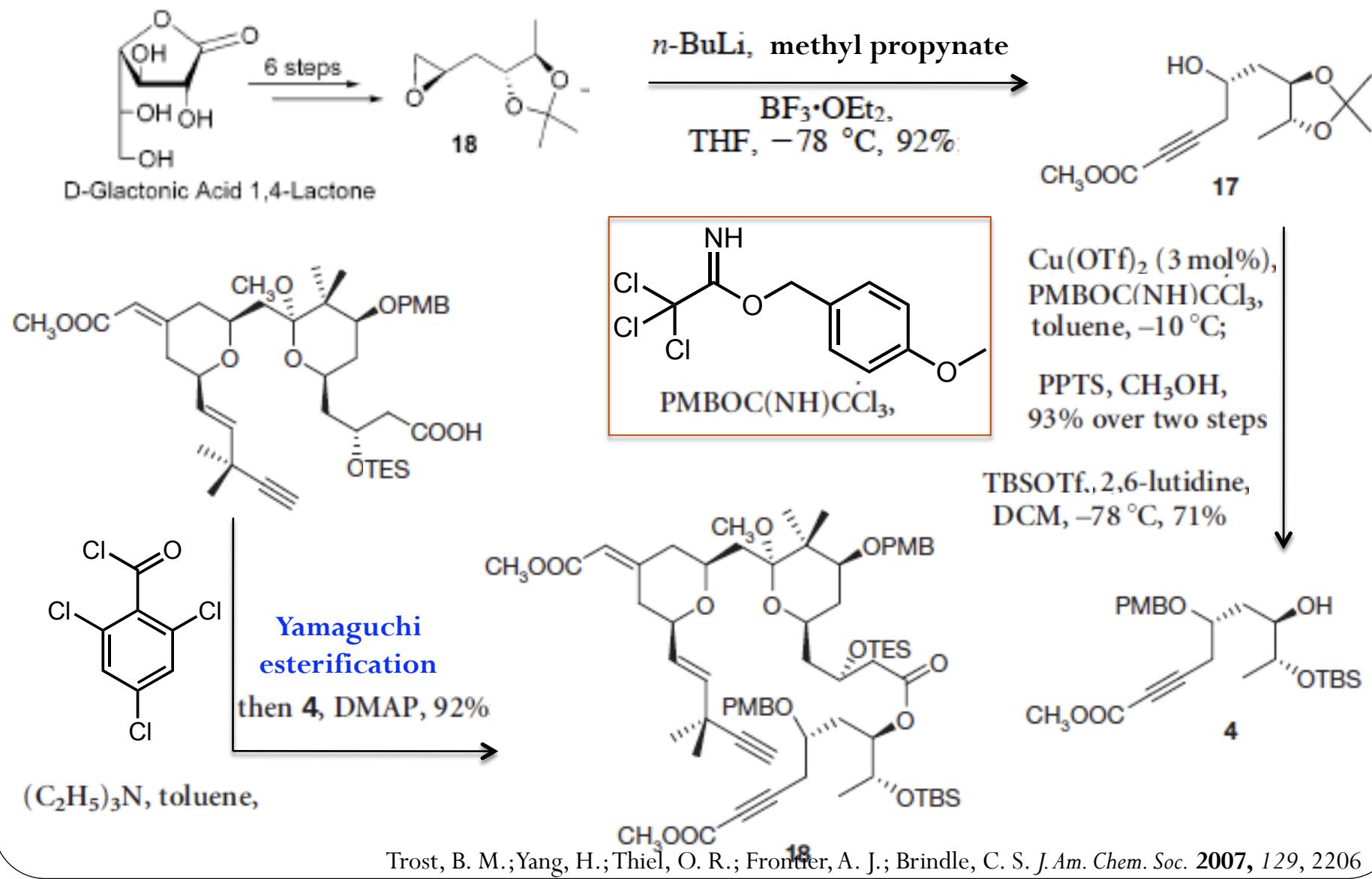
TBAF, HOAc, THF
90% (96% b.r.s.m.)



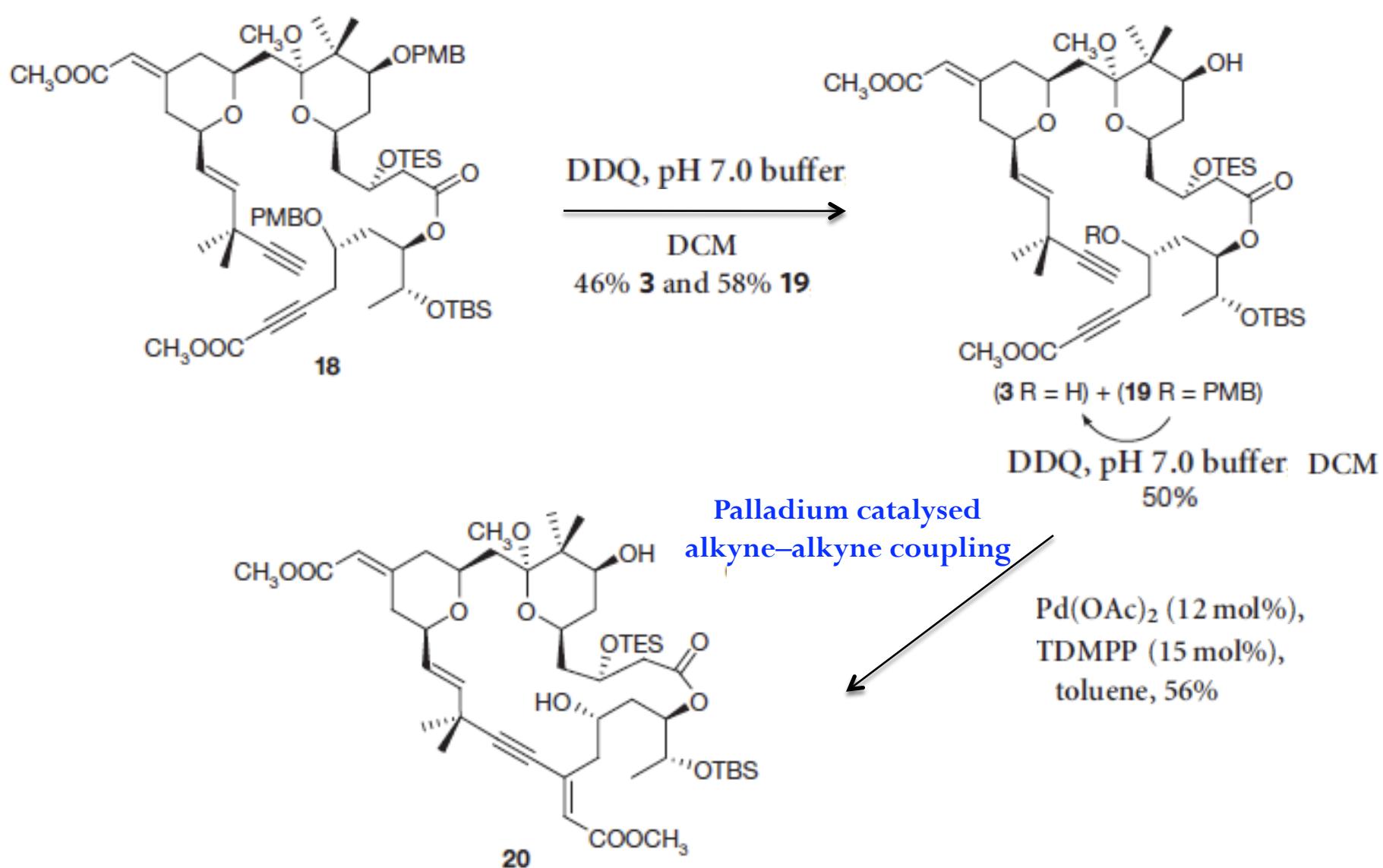
1 (CH₃)₃SnOH, DCE, 80 °C,
84%
2 TESOTf, 2,6-lutidine,
DCM, −10 °C to 0 °C
76–79%

Nicolaou, K. C.; Estrada, A.A.; Zak, M.; Lee, S. H.; Safina, B. S. *Angew. Chem. Int. Ed.* 2005, 44, 1378

Synthesis of Alcohol Functionality

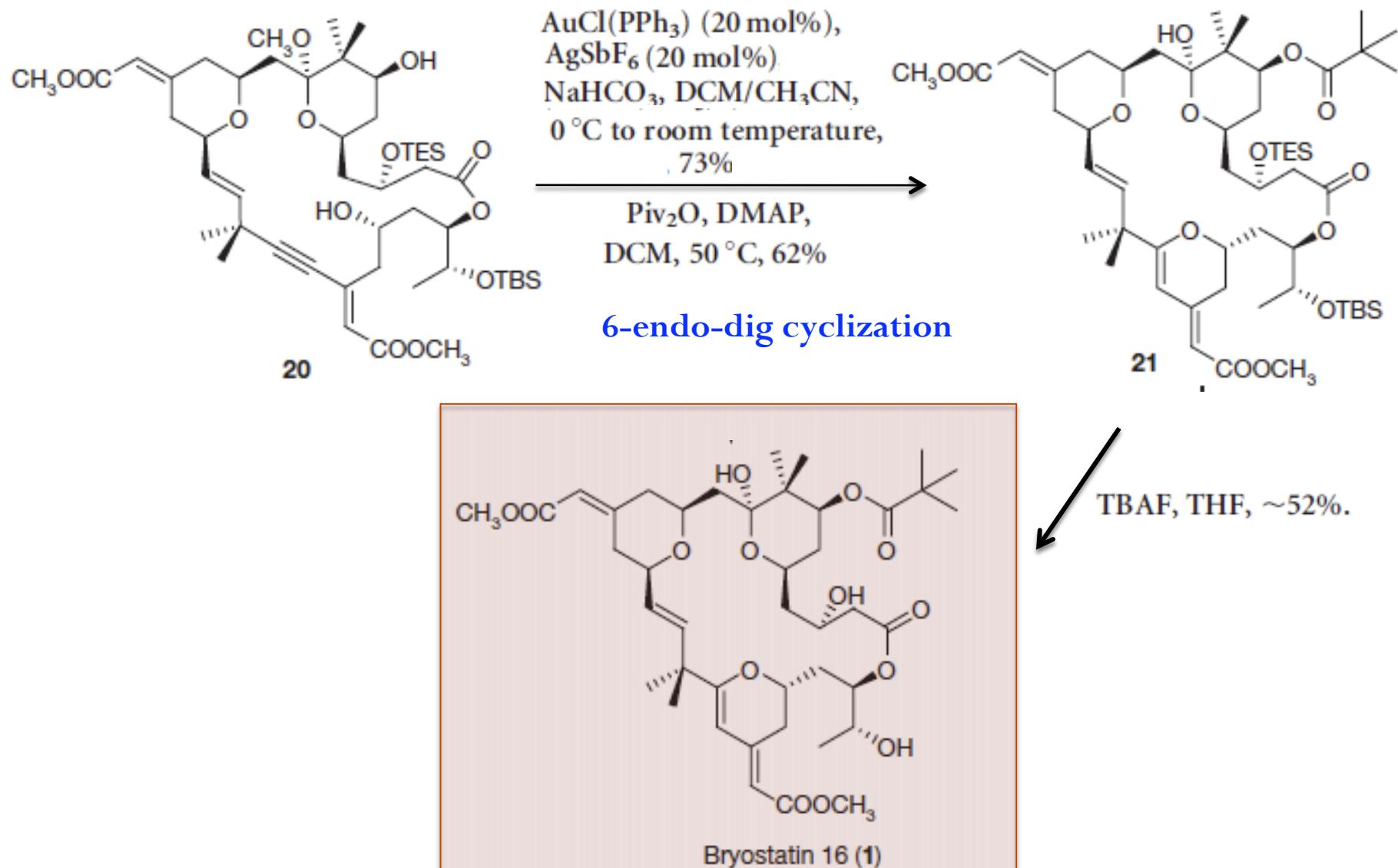


Trost's Approach to Bryostatin 16



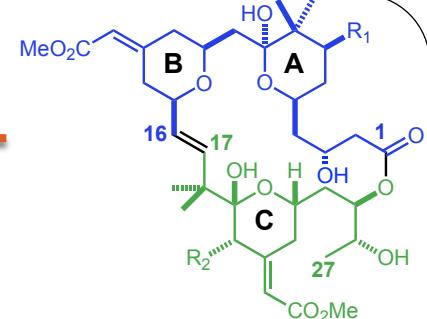
Trost, B. M.; Matsubara, S.; Caringi, J. J. *J. Am. Chem. Soc.* **1989**, *111*, 8745

Trost's End Game to Bryostatin 16

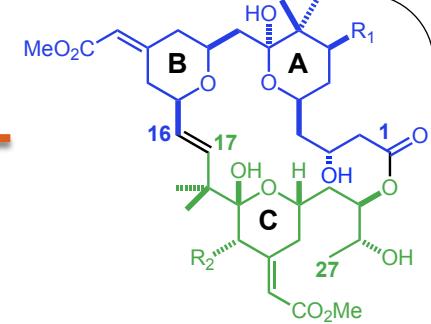


Trost, B. M.; Dong, G. *Nature*, 2008, 456, 485

Conclusion

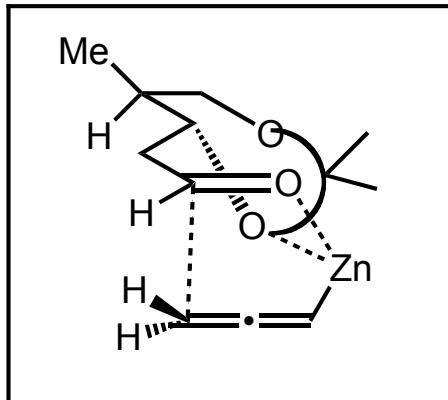
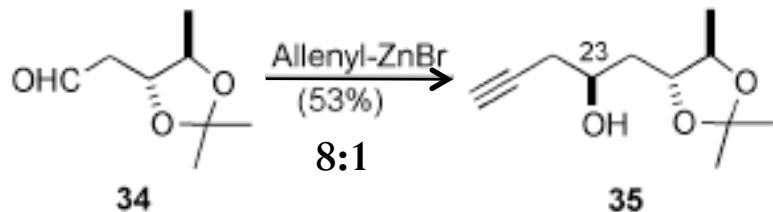
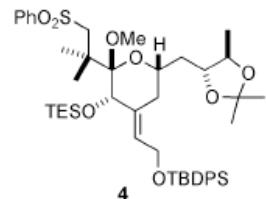


- Previous Syntheses have used more than 60 steps.
- Trost's synthesis is a highly concise strategy (26-step longest linear sequence, 39 total steps from aldehyde 2)
- A pivotal parent structure allowing access to all other bryostatins
- New analogues, might be readily obtained simply by variations in this natural product's synthesis.
- Palladium catalysed alkyne–alkyne coupling as a macrocyclization method for complex natural product synthesis.
- Gold catalysed conversion of the product of above step into a dihydropyran (the ‘C ring’ of bryostatins)



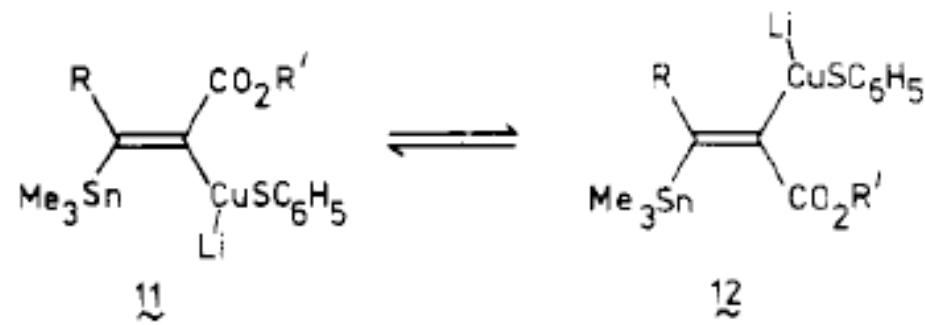
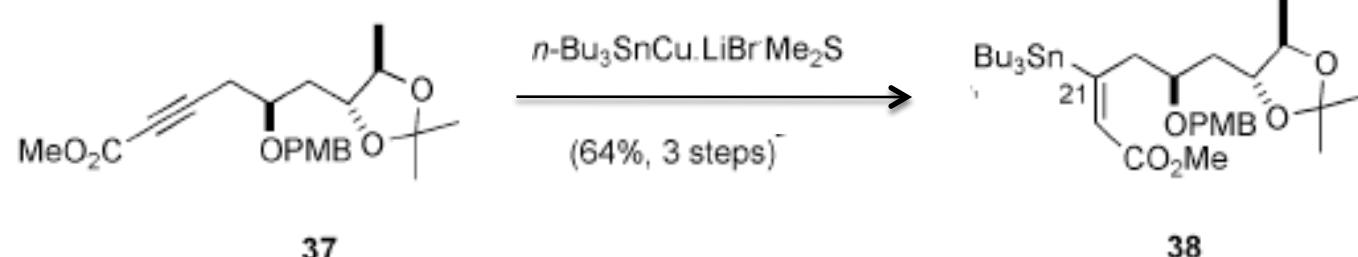
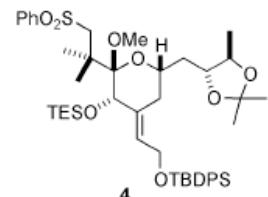
Thanks

Chelation Controlled Nu Attack



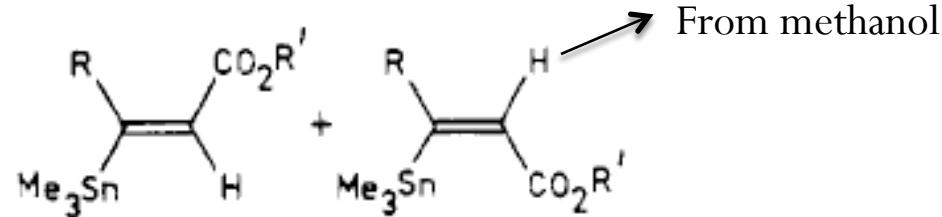
**Chelation –Controlled
Nucleophilic addition**

Piers stannylcupration



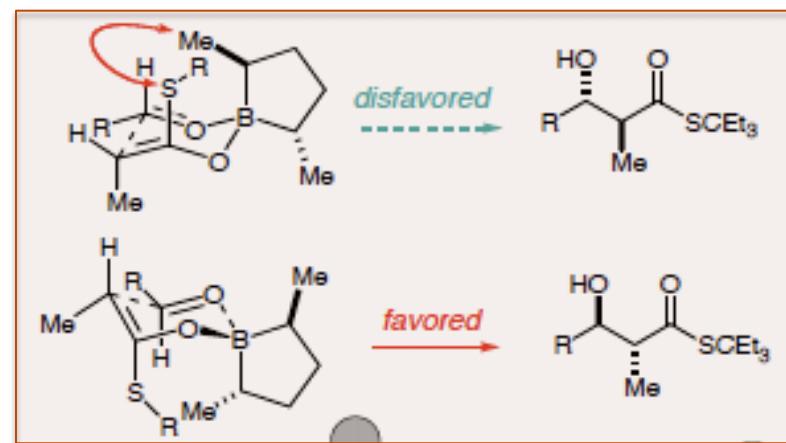
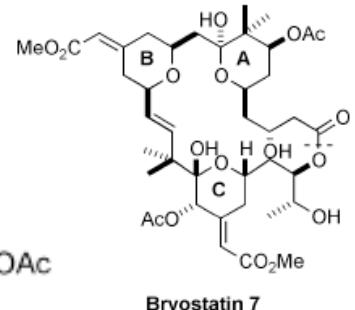
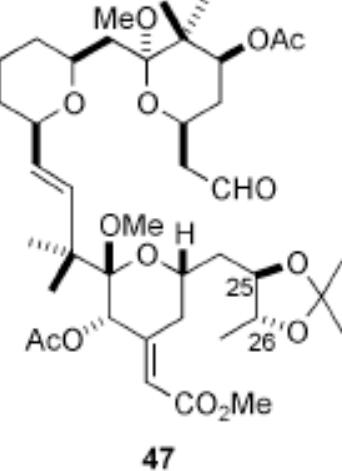
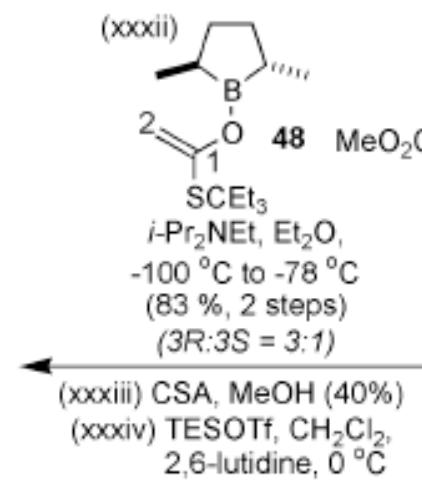
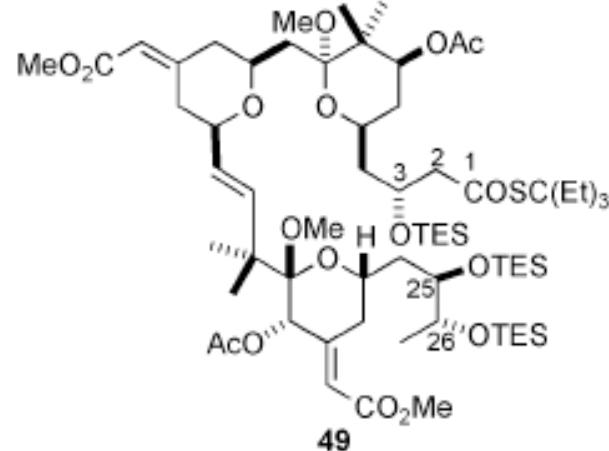
Kinetic Control

Thermodynamic Control



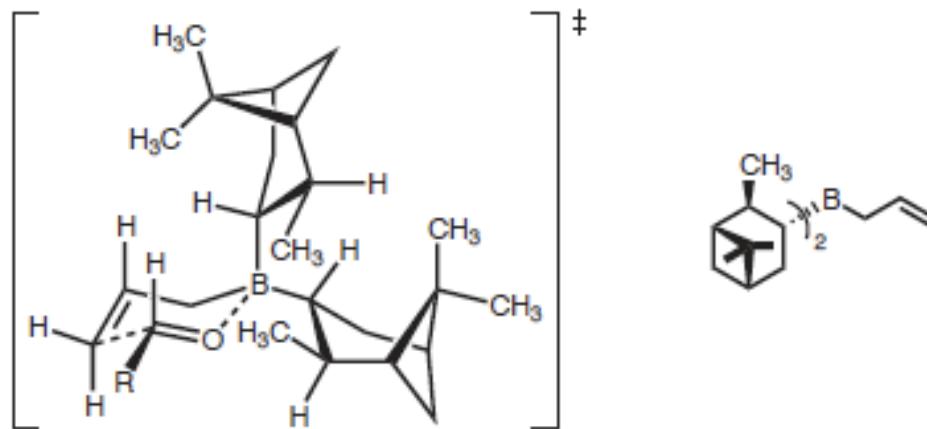
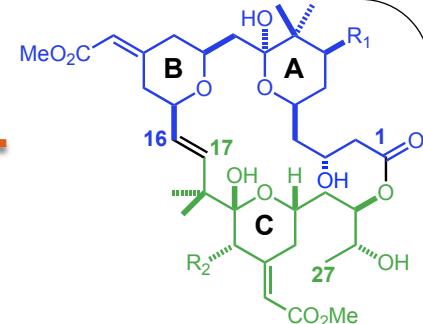
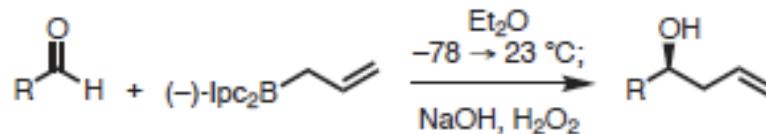
Edward Piers,* Howard E. Morton *J. Org. Chem.* 1980, 45, 4263

Substrate Controlled Aldol Reaction



Brown's Asymmetric Allylation

Enantioselective Allylboration



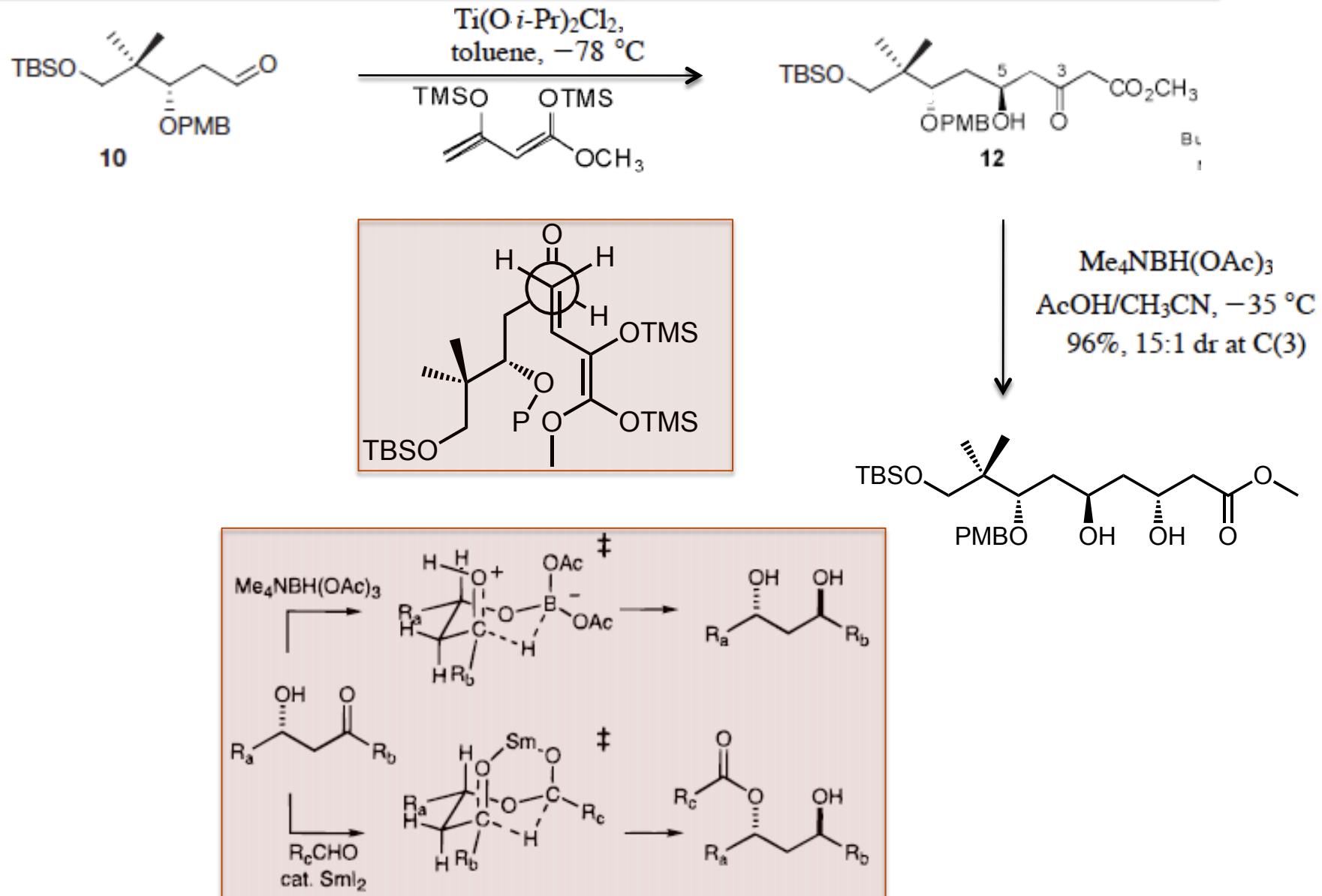
Allylation of aldehydes proceeds through a chair-like TS where R occupies an equatorial position and the aldehyde facial selectivity derives from minimization of steric interactions between the axial Ipc ligand and the allyl group.

Brown, H. C.; Jadhav, P. K. *J. Am. Chem. Soc.* 1983, 105, 2092-2093.

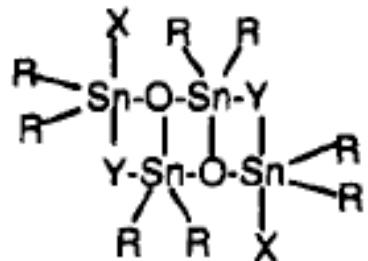
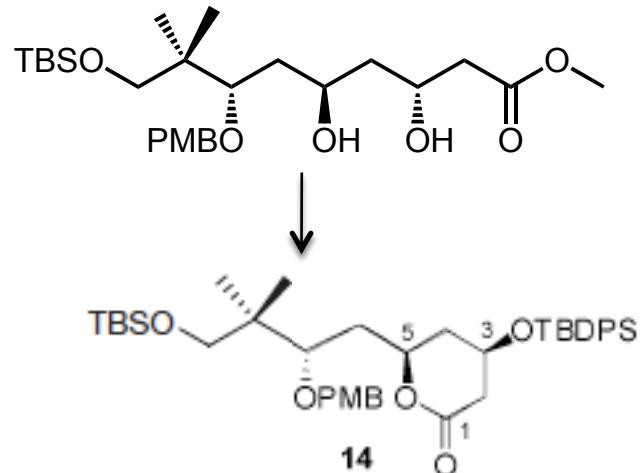
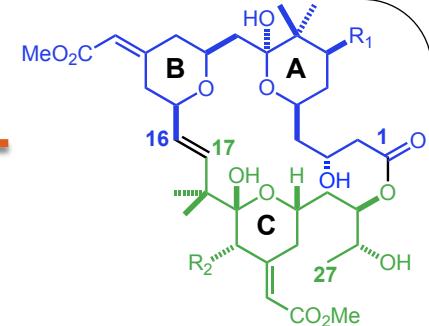
Brown, H. C.; Bhat, K. S. *J. Am. Chem. Soc.* 1986, 108, 5919-5923.

Racherla, U. S.; Brown, H. C. *J. Org. Chem.* 1991, 56, 401-404.

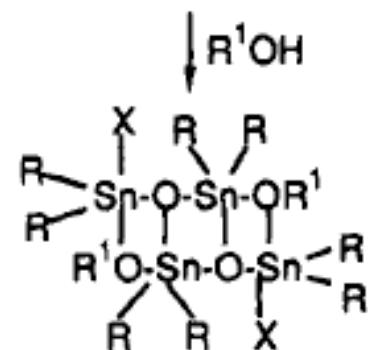
Evans Diastereoselective Vinylogous Aldol Reaction and Hydroxyl Directed Reduction



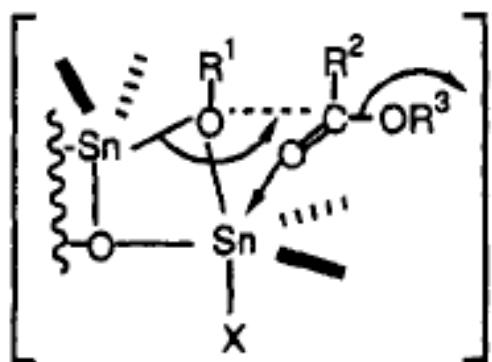
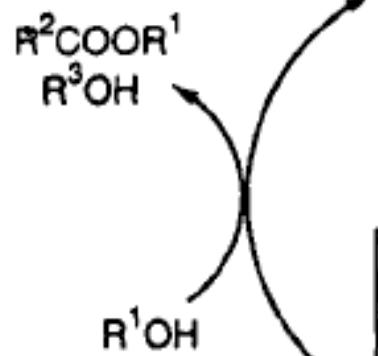
Otera's Catalyst for Transesterification



$R = Bu, X = Y = -NCS$



alkoxydistannoxane



Otera, J et al (1991). "Novel template effects of distannoxane catalysts in highly efficient transesterification and esterification". *J. Org. Chem.* 56 (18): 5307–5311

Otera's Catalyst for Transesterification

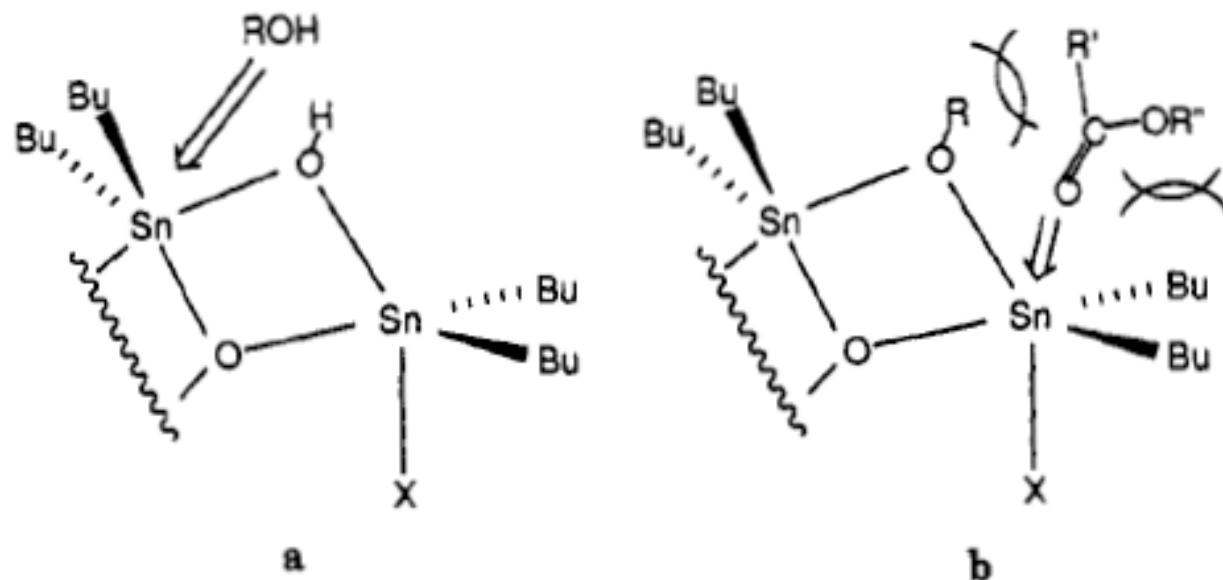
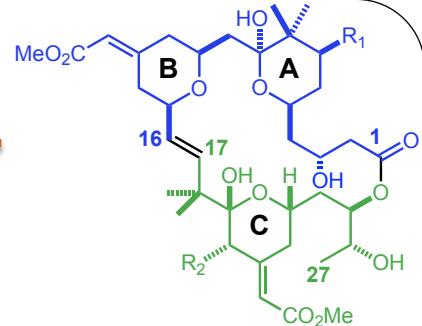
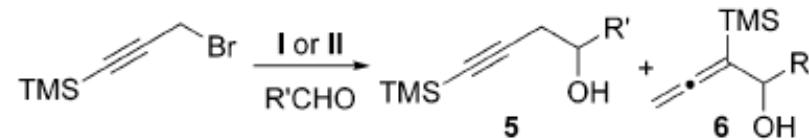


Figure 1. (a) Approach of the alcohol reactant to the distannoxane template. (b) Approach of the ester reactant to the distannoxane template.

Otera, J et al (1991). "Novel template effects of distannoxane catalysts in highly efficient transesterification and esterification". *J. Org. Chem.* 56 (18): 5307–5311

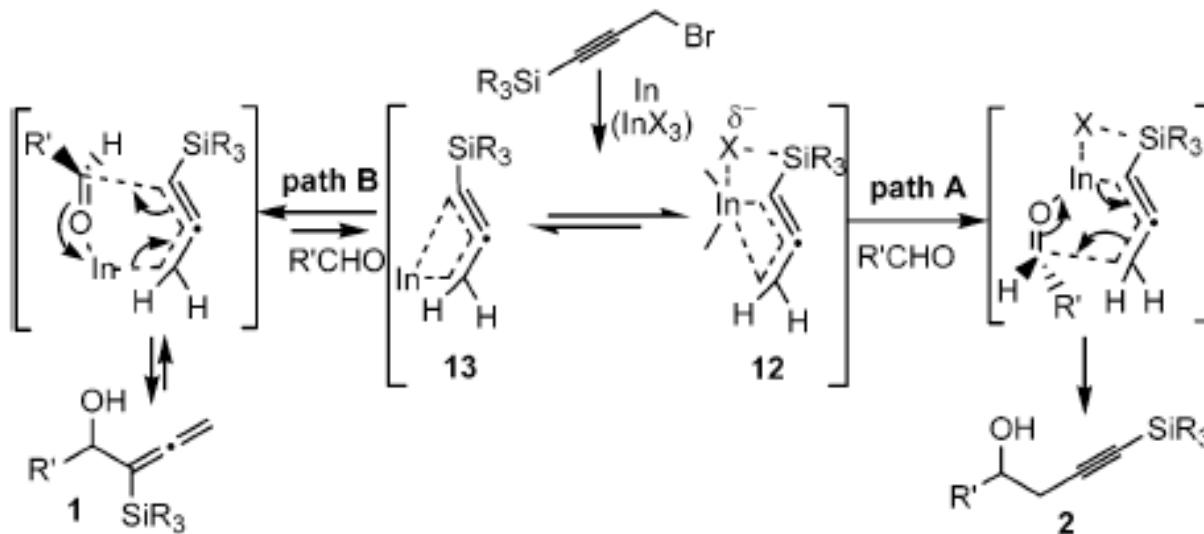
Indium-mediated Propargylation



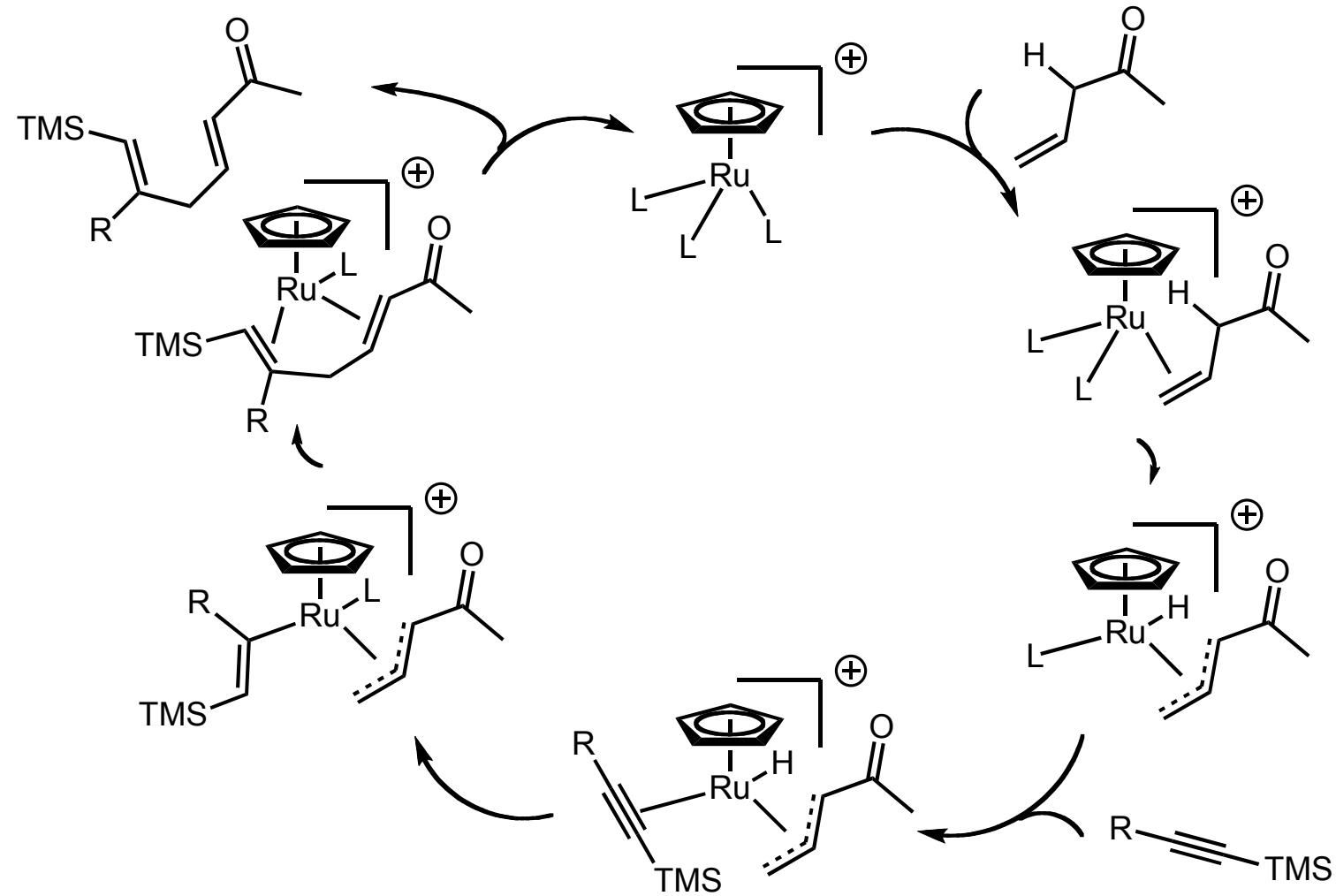
I: In/10mol% InBr_3 , THF, reflux, 20h

II: In/10mol% InF_3 , THF, reflux, 9h

entry	aldehyde	5	I (5:6) yield ^b %	II (5:6) yield ^b %
1	hydrocinnamaldehyde	a	85 (99:1)	89 (99:1)
2	cinnamaldehyde	b	89 (99:1)	92 (99:1)
3	cyclohexanecarbaldehyde	c	90 (99:1)	94 (99:1)
4	benzaldehyde	d	92 (99:1)	92 (99:1)
5	nonyl aldehyde	e	95 (99:1)	93 (99:1)



Ru-catalysed alkene-alkyne coupling



Palladium catalysed alkyne–alkyne coupling

