

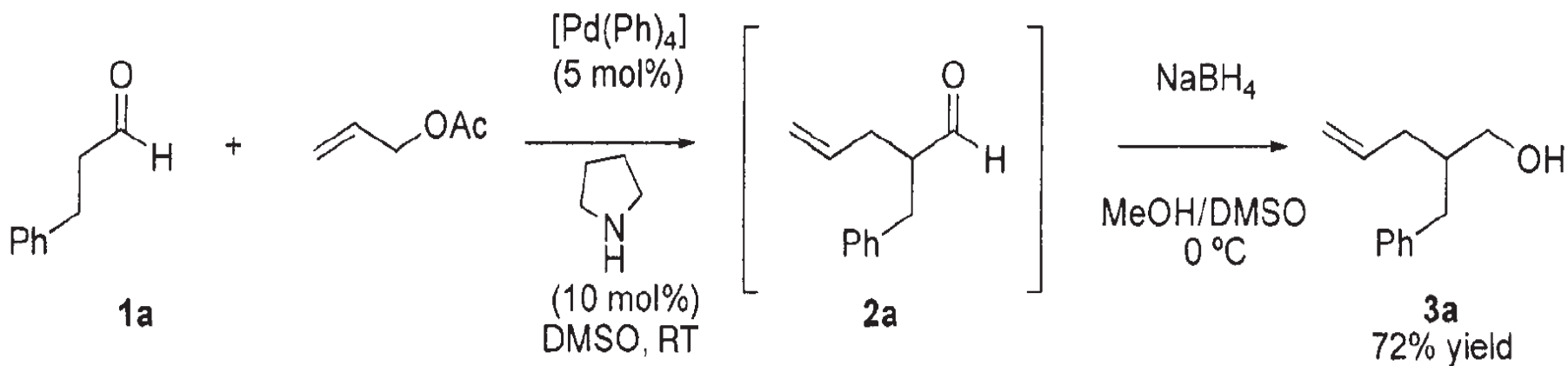
# Enantioselective $\alpha$ -Allylation of Aldehydes and Ketones

Li HUANG  
Dec 7, 2007

# Outline

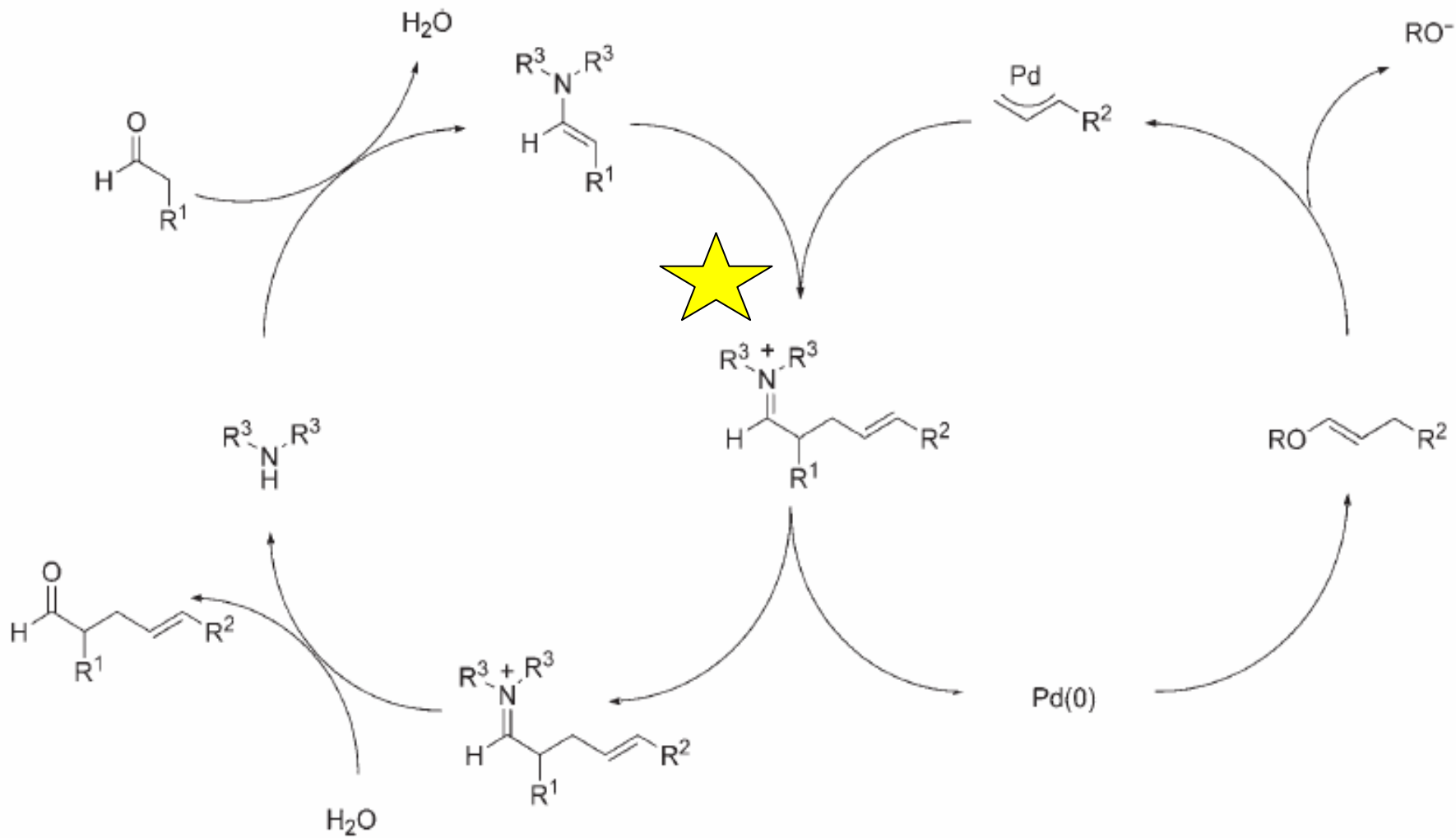
- Direct  $\alpha$ -Allylation of Aldehydes
  - Using ACDC
  - Using SOMO Activation
  
- Indirect  $\alpha$ -Allylation of Ketones
  - Enolates
  - Phase Transfer Catalysts

# Basis for Direct $\alpha$ -Allylation of Aldehydes

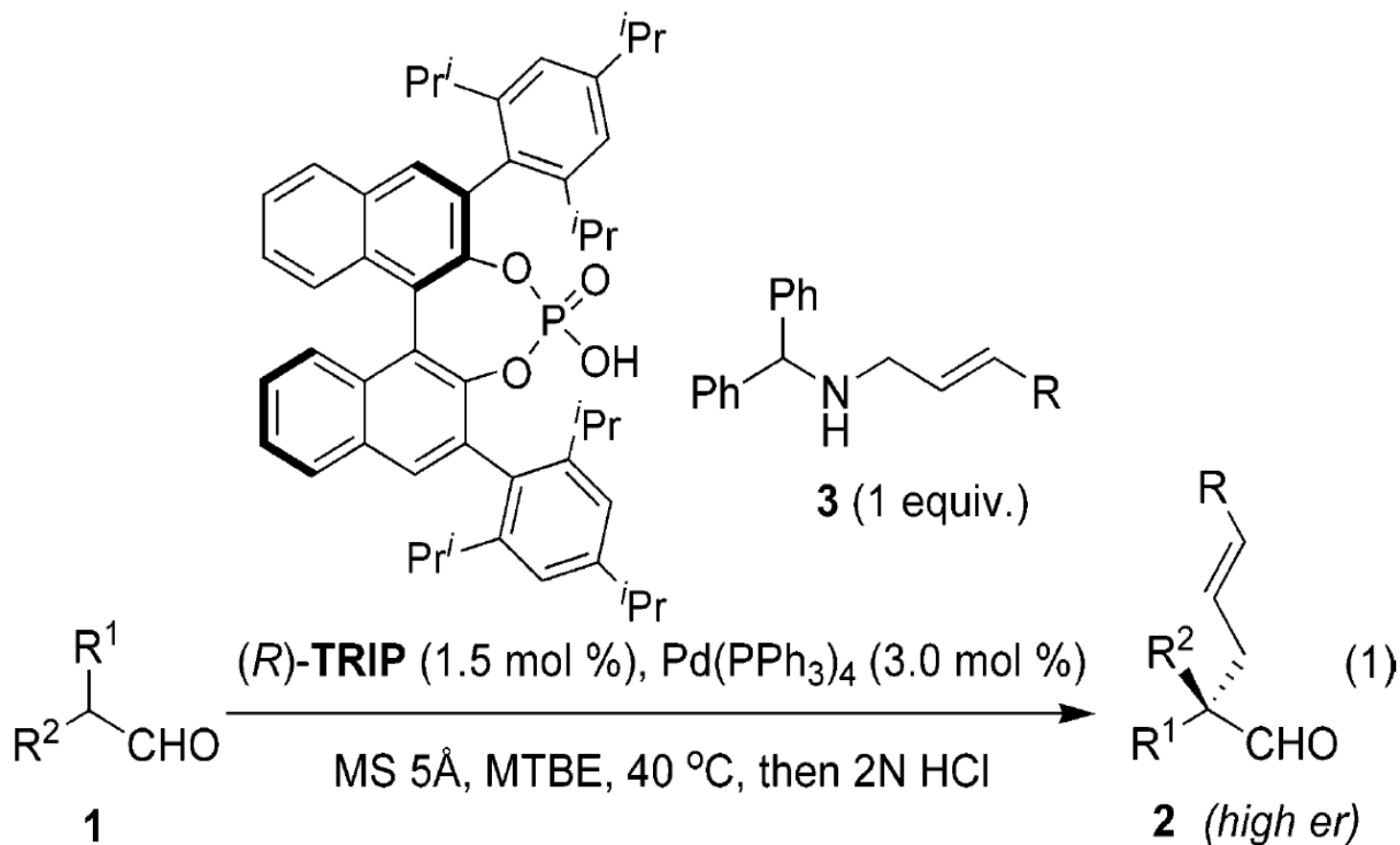


Direct catalytic intermolecular  $\alpha$ -alkylation of aldehydes by combination of transition metal and organocatalysis

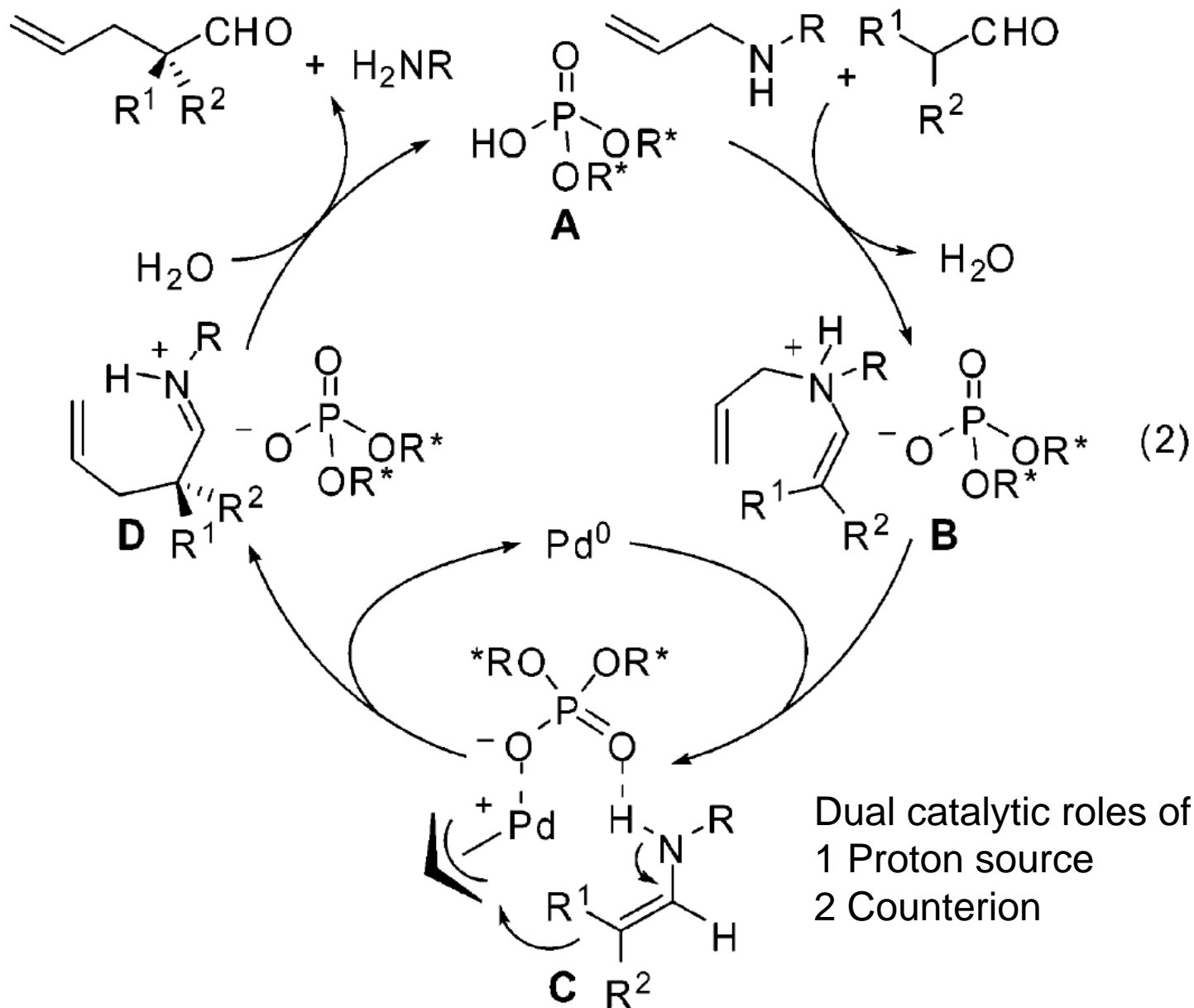
# Basis for Direct $\alpha$ -Allylation of Aldehydes



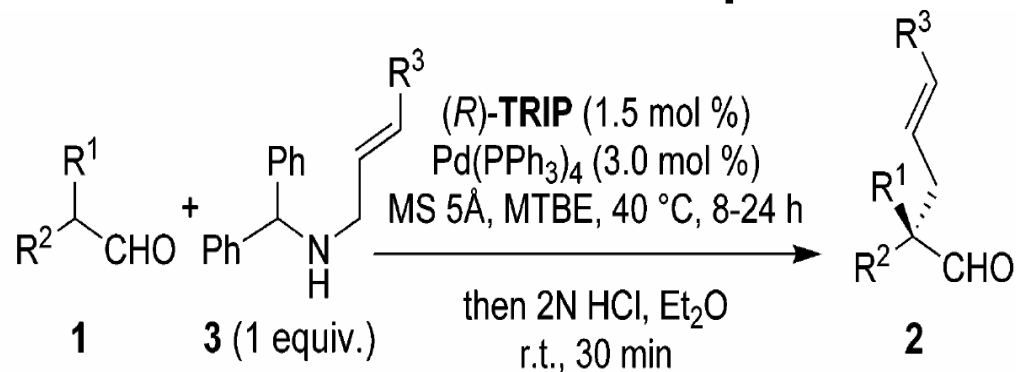
# ACDC in Transition Metal Catalysis

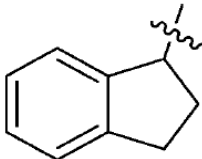


# Mechanism for the Reaction

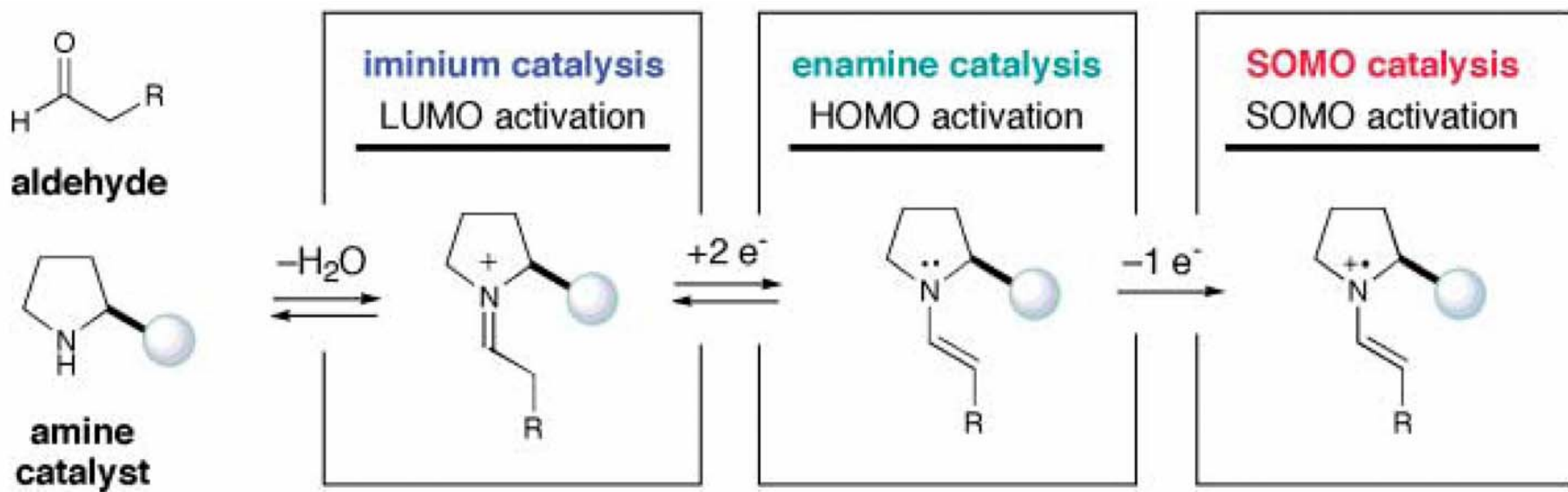


# Substrate Scope



entry	R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>		yield (%)	er <sup>a</sup>
1	Me	Ph	H	<b>2a</b>	85	98.5:1.5
2	Me	4-Me-C <sub>6</sub> H <sub>4</sub>	H	<b>2b</b>	89	97:3
3	Me	3-Me-C <sub>6</sub> H <sub>4</sub>	H	<b>2c</b>	84	98:2
4	Me	3-F-C <sub>6</sub> H <sub>4</sub>	H	<b>2d</b>	85	98:2
5 <sup>b</sup>	Me	2-F-C <sub>6</sub> H <sub>4</sub>	H	<b>2e</b>	74	97:3
6	Me	4- <i>i</i> -Bu-C <sub>6</sub> H <sub>4</sub>	H	<b>2f</b>	76	97.5:2.5
7	Me	2-naph	H	<b>2g</b>	71	97:3
8	Me	2-thiophenyl	H	<b>2h</b>	80	93:7
9			H	<b>2i</b>	45	95:5
10 <sup>c</sup>	Me	<i>c</i> -hex	H	<b>2j</b>	65	85:15
11 <sup>d,e</sup>	Me	Ph	Me	<b>2k</b>	40	96:4
12 <sup>d,e</sup>	Me	Ph	Ph	<b>2l</b>	82	91:9

# Organocatalysis Using SOMO Activation

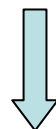
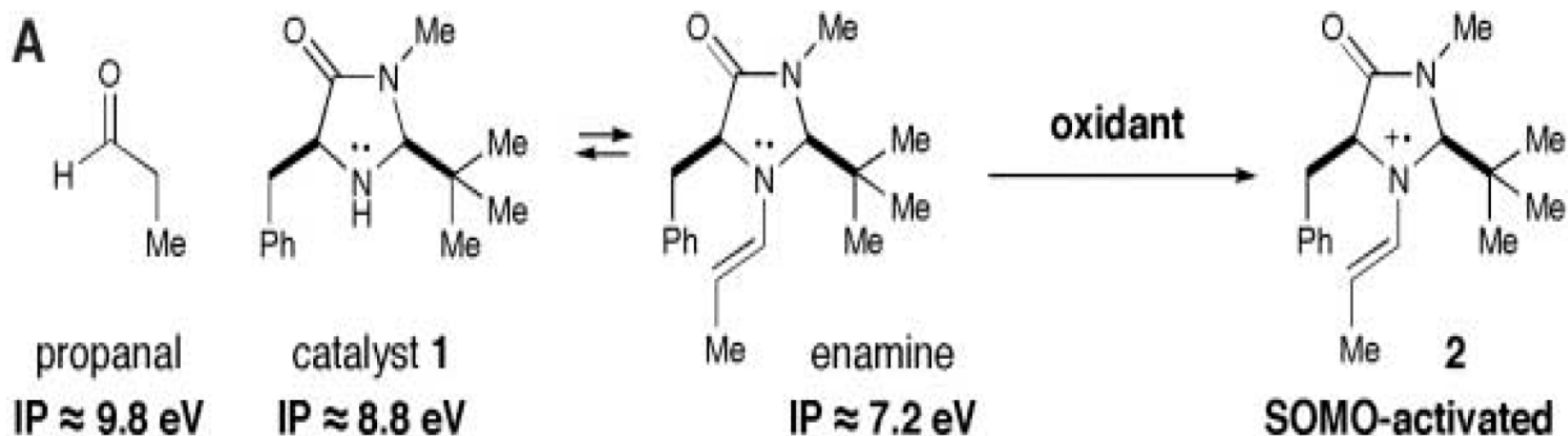


SOMO: a singly occupied molecular orbital



# SOMO Activation

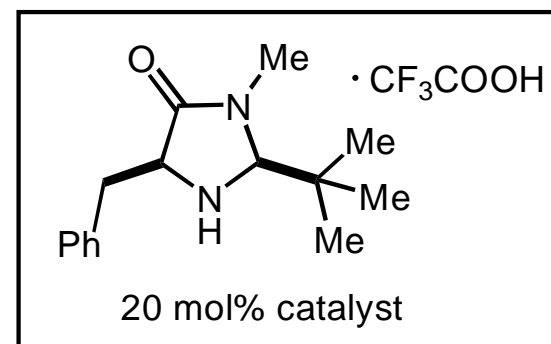
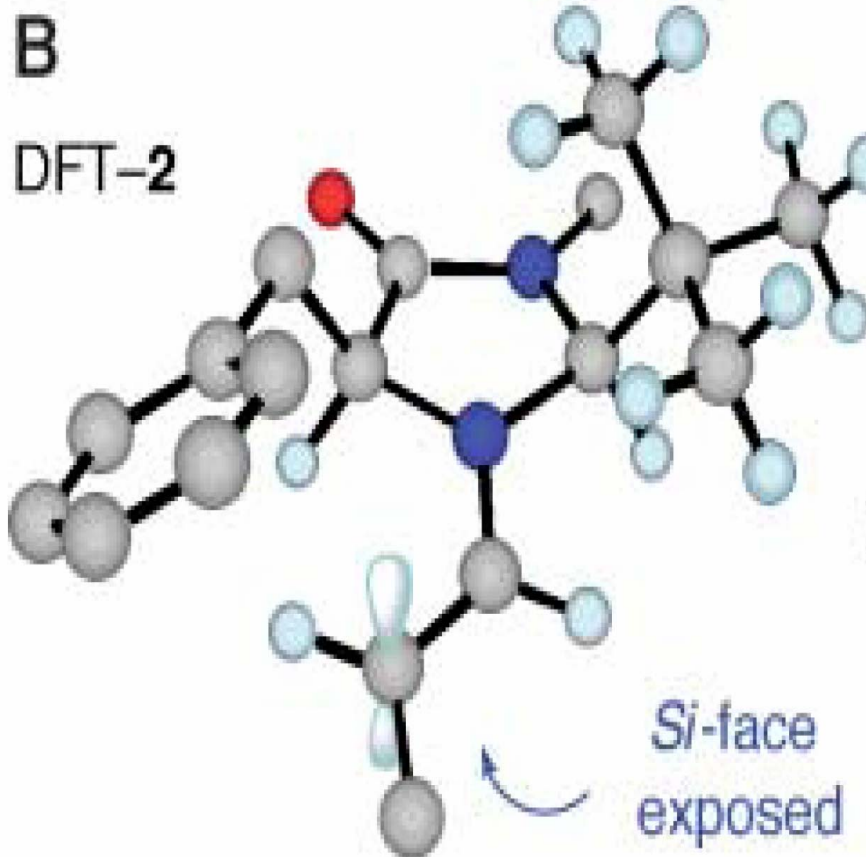
Key elements to support the proposal



The transient enamine component to be sufficiently more susceptible to oxidation than the accompanying reaction partners.

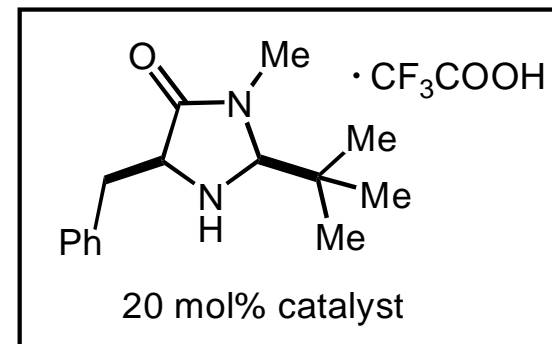
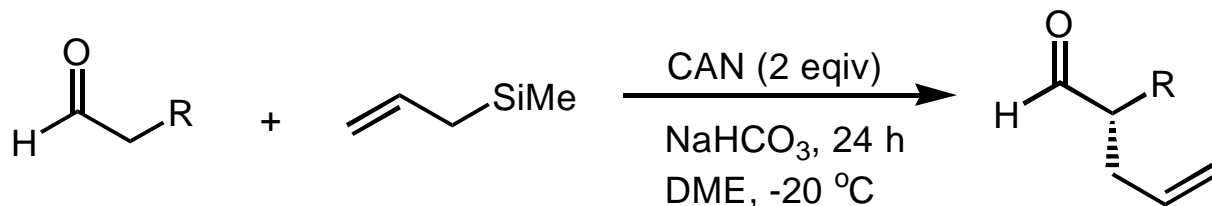
# SOMO Activation

Key elements to support the proposal



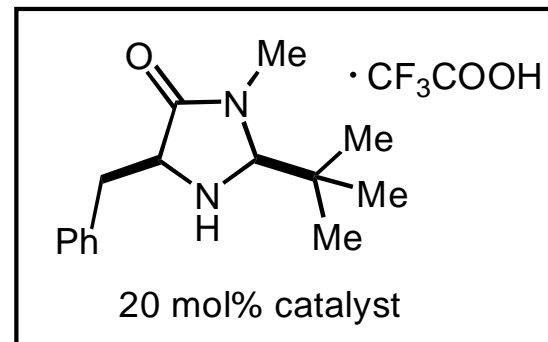
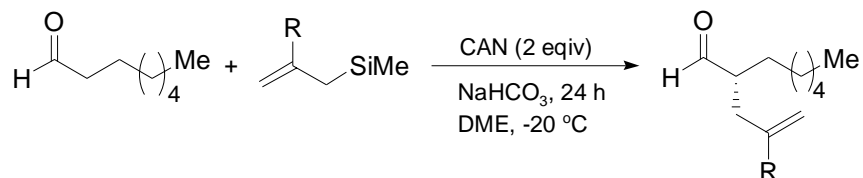
- ✓ Three  $\pi$  electron system away from the bulky t-Bu group
- ✓ Radical-centered carbon selectively populates an E configuration to minimize nonbonding interactions with the ring

# Enantioselective Aldehyde $\alpha$ -Allylation



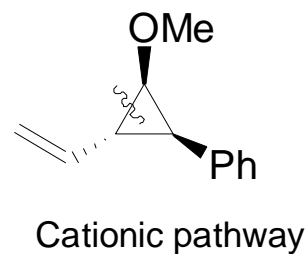
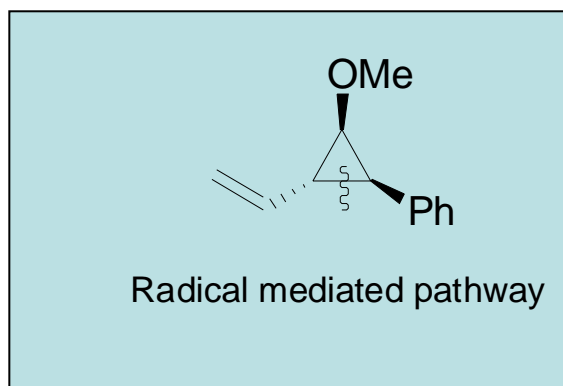
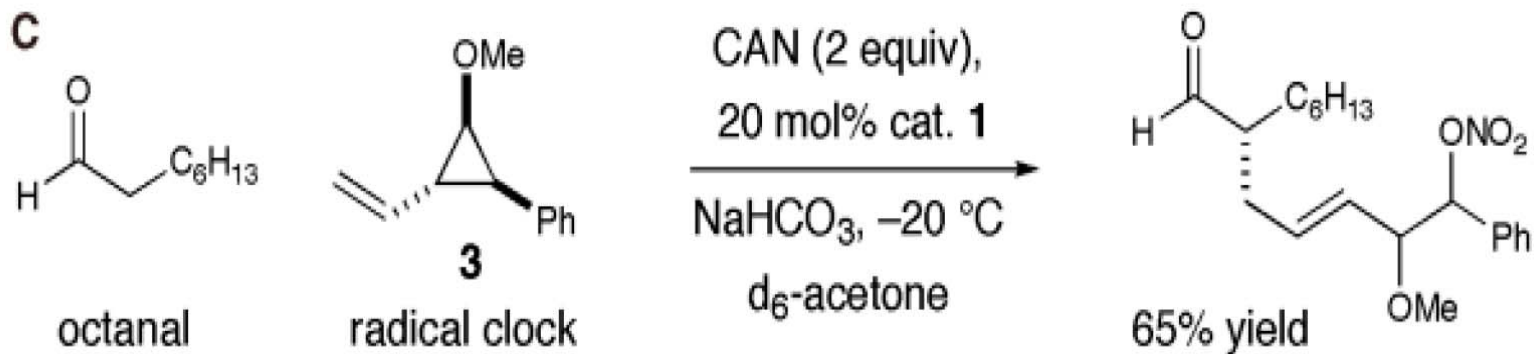
Aldehyde	Product	Aldehyde	Product
<p>Yield: 81% ee: 91%</p>		<p>Yield: 72% ee: 87%</p>	
<p>Yield: 75% ee: 92%</p>		<p>Yield: 75% ee: 94%</p>	
<p>Yield: 72% ee: 95%</p>		<p>Yield: 70% ee: 93%</p>	

# Enantioselective Aldehyde $\alpha$ -Allylation



Allylsilane	Product	Yield	ee
		88%	91%
		87%	90%
		77%	88%
		81%	90%

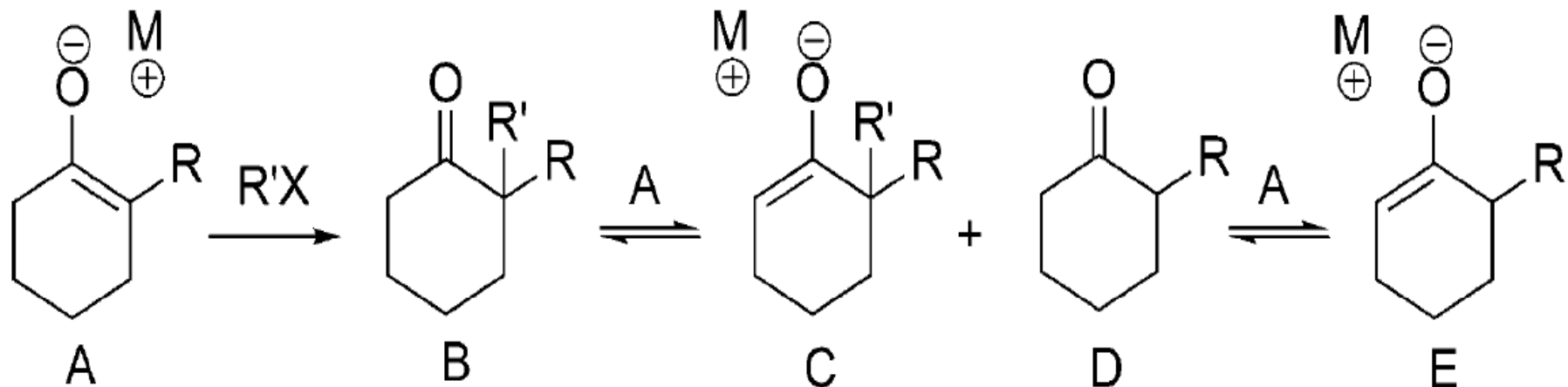
# Evidence for SOMO



# Outline

- Direct  $\alpha$ -Allylation of Aldehydes
  - Using ACDC
  - Using SOMO Activation
  
- Indirect  $\alpha$ -Allylation of Ketones
  - Enolates
  - Phase Transfer Catalysts

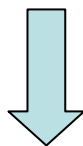
# Enantioselective Allylation of ketones



Problems:

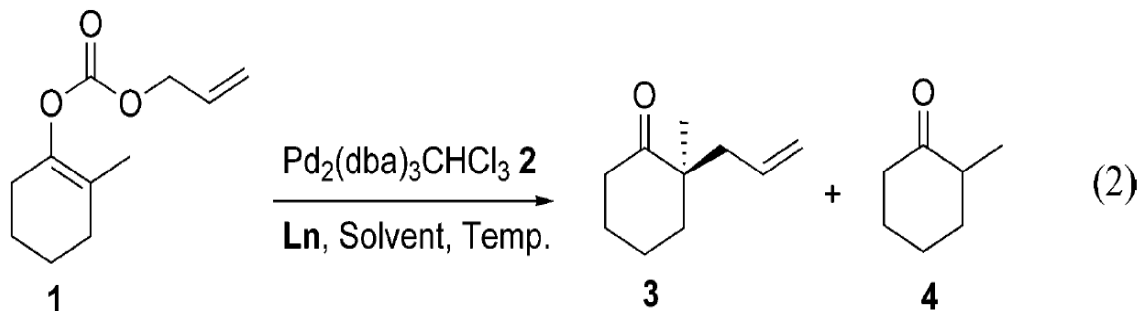
Enolate equilibrium can lead to loss of regioselectivity (A and E)

$R = H$ , the chiral product is subjected to racemization

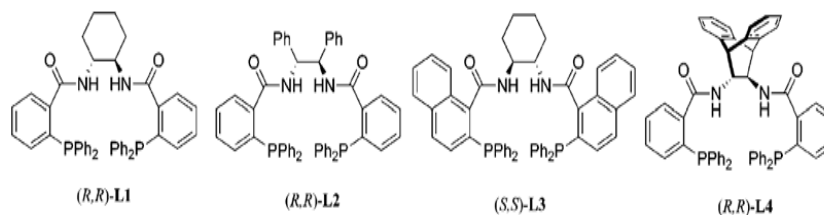


Neutral condition  
Low concentration of enolate

# Enantioselective Pd-Catalyzed Allylation

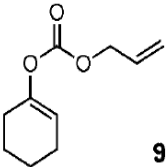
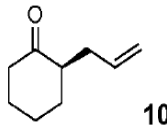
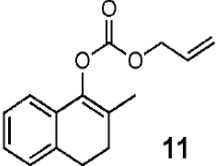
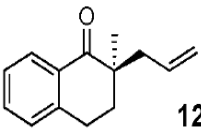
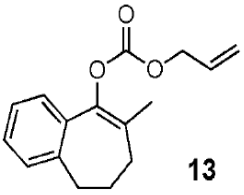
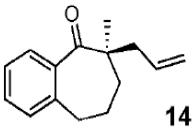
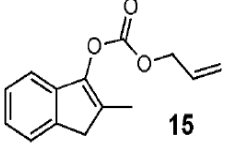
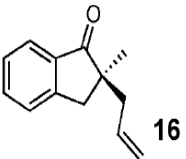
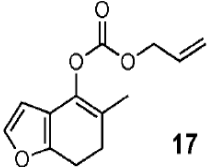
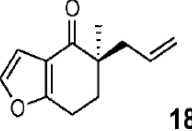
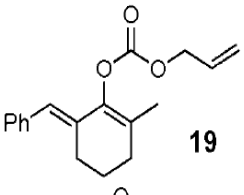
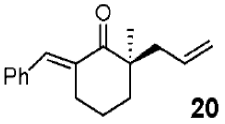


entry	ligand	solvent	ee <sup>b</sup>	yield <sup>c</sup> of 3	yield <sup>c</sup> of 4
1	L1	DME	66	81	8
2	L3	DME	76	87	2
3	L1	toluene	31	73	0
4	L2	toluene	61	73	2
5	L3	toluene	60	85	1
6	L4	toluene	85	88	0
7	L4	CH <sub>2</sub> Cl <sub>2</sub>	84	64	26
8	L4	dioxane	80	99	0
9	L4	DME	84	87	7
10	L4	THF	81	85	1
11	L4	DME (1% H <sub>2</sub> O)	NA	20	3.7
12	L3	DME (1% H <sub>2</sub> O)	NA	1.5	0

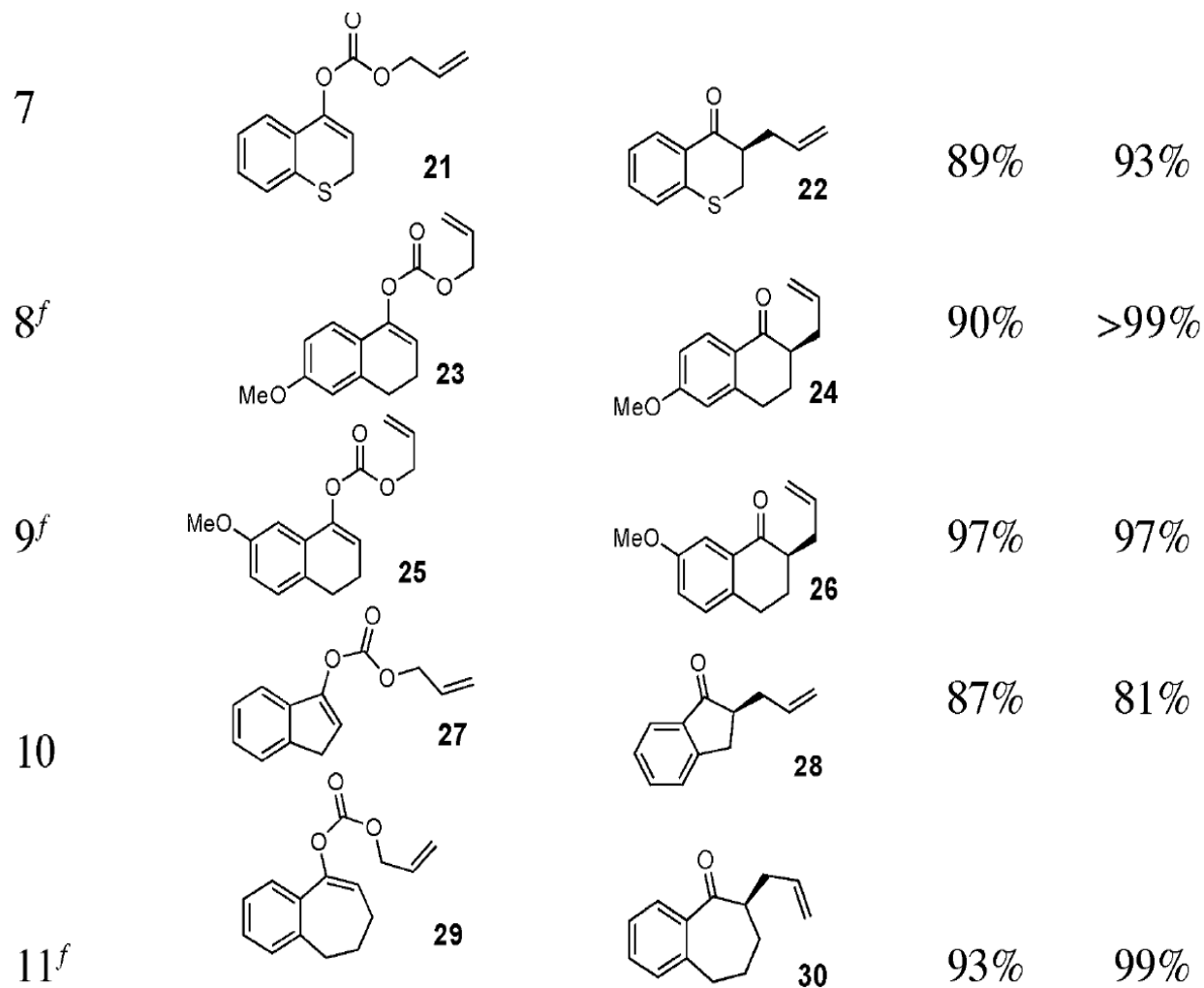




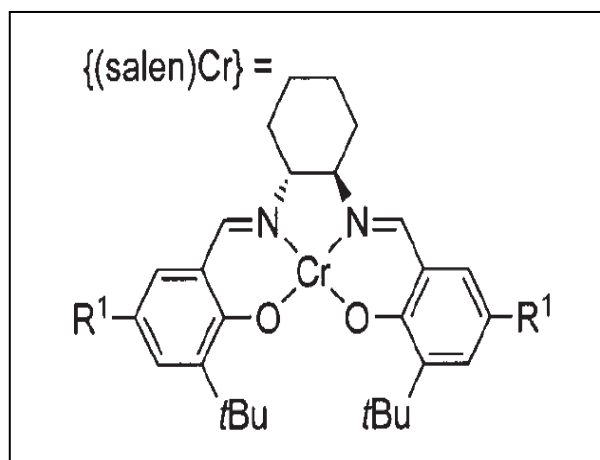
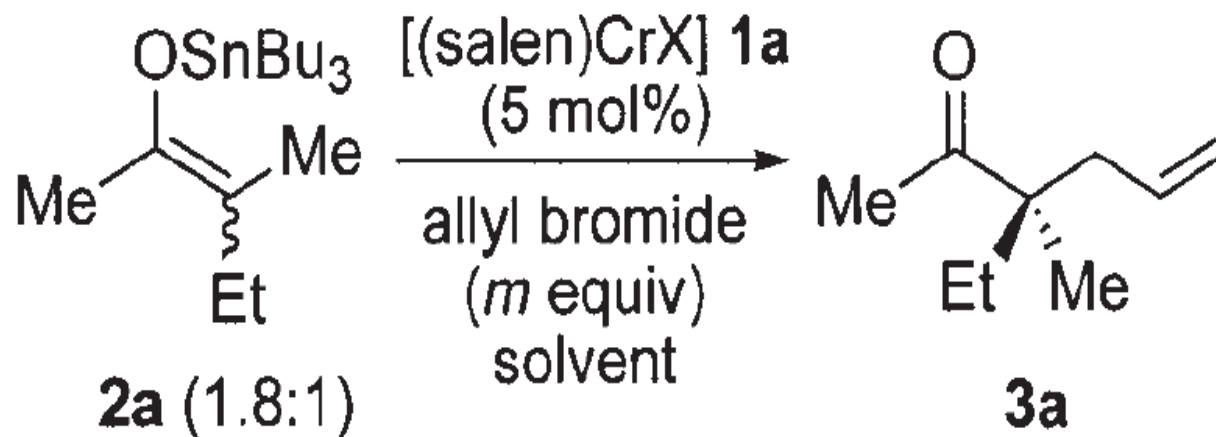
# Substrate Scope

entry	substrate	product	yield <sup>b</sup>	ee <sup>c</sup>
1 <sup>d</sup>	 <b>9</b>	 <b>10</b>	78%	78% <sup>e</sup>
2 <sup>d</sup>	 <b>11</b>	 <b>12</b>	88%	>99%
3	 <b>13</b>	 <b>14</b>	94%	91% <sup>e</sup>
4 <sup>b</sup>	 <b>15</b>	 <b>16</b>	98%	76%
5	 <b>17</b>	 <b>18</b>	64%	82%
6	 <b>19</b>	 <b>20</b>	99%	95%

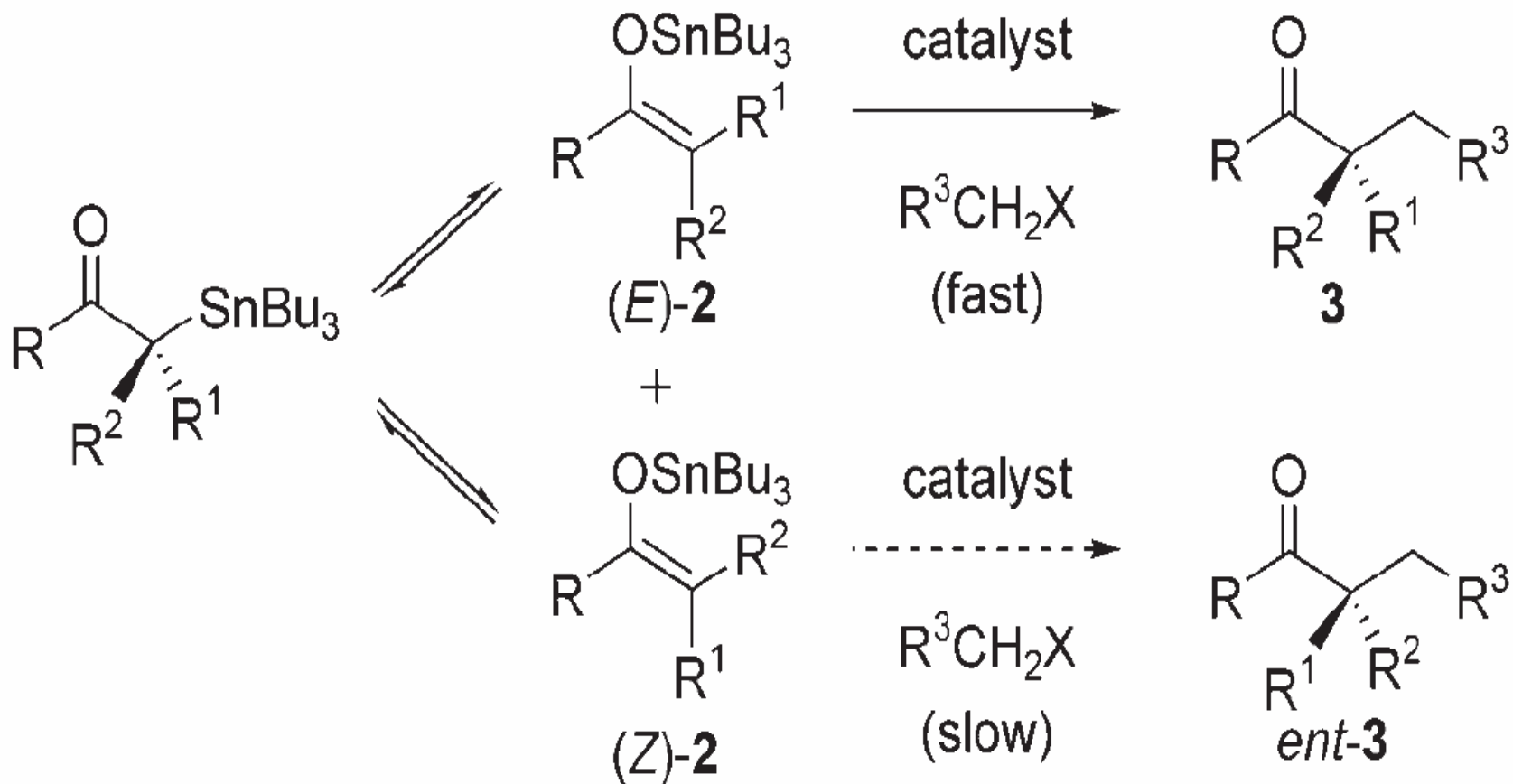
# Substrate Scope



# Enantioselective Allylation of Acyclic $\alpha,\alpha$ -Disubstituted Tributyltin Enolates



# How could they come up with the idea?

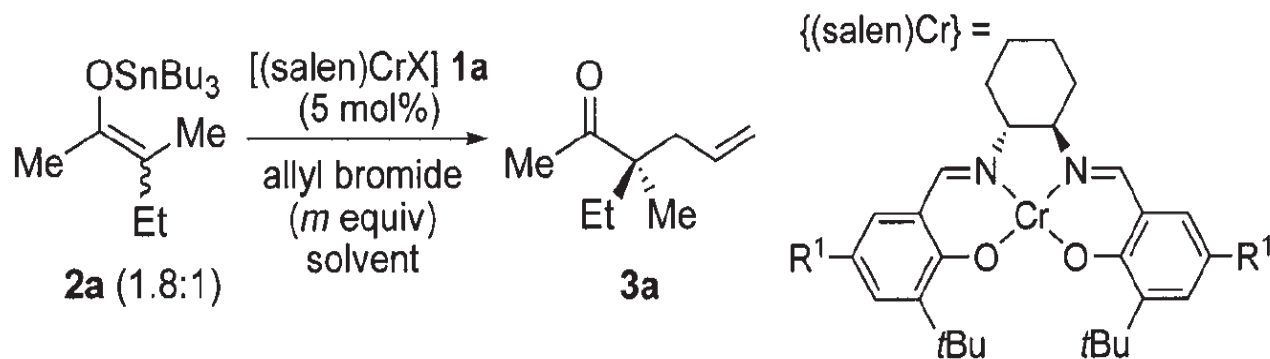


Tin enolates are known to undergo tautomerization between their O-stannyl and C-stannyl forms in solution

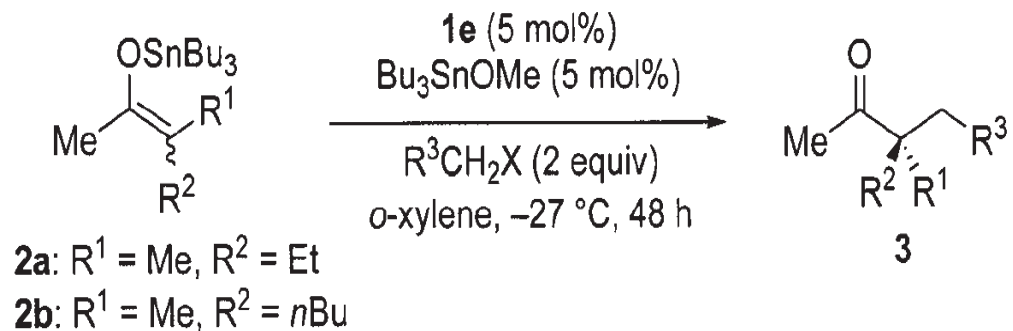
The mixture of acyclic tin enolates might undergo reaction selectively through one geometric isomer

Dynamic control

# Catalyst Screening

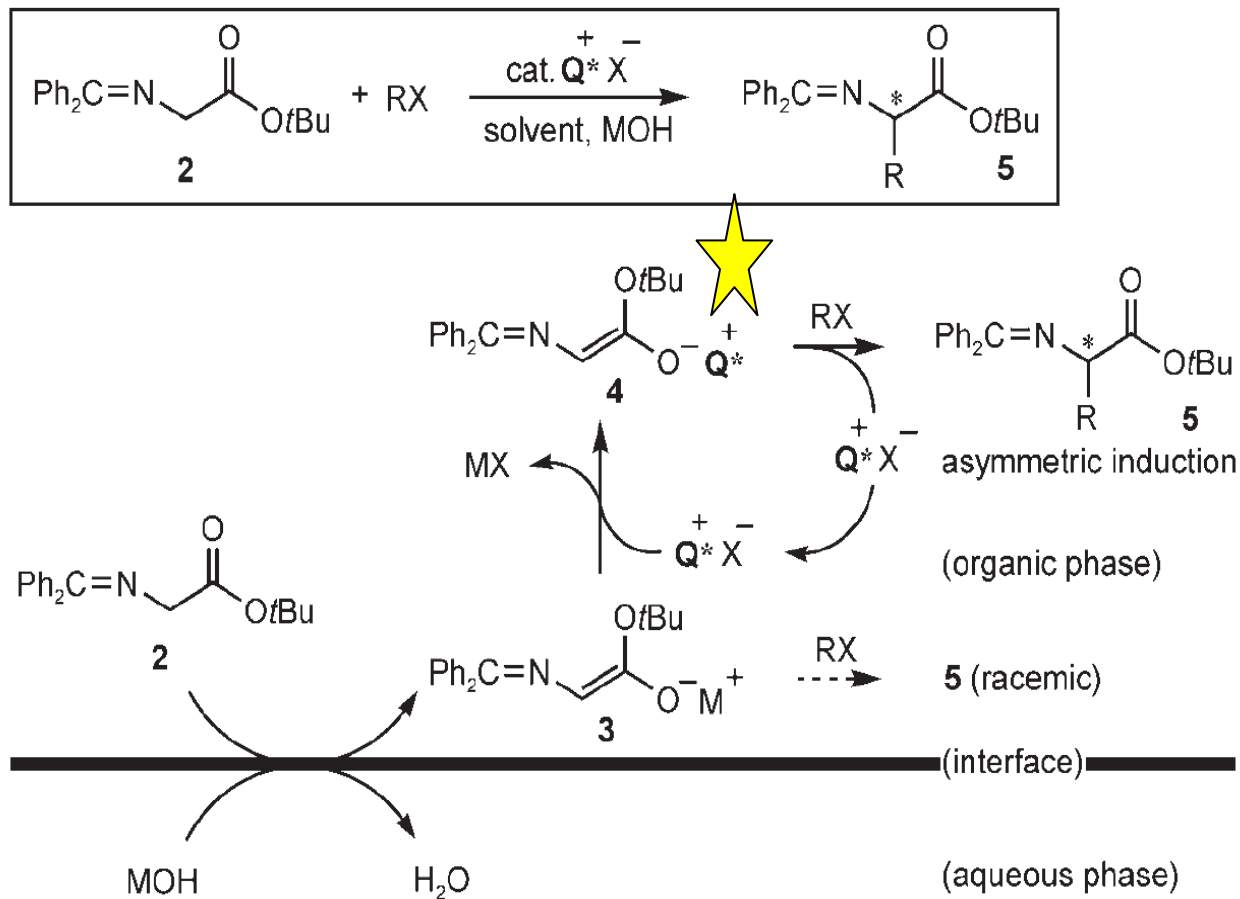


Entry	Catalyst	X	R <sup>1</sup>	Solvent	<i>m</i> [equiv]	T [°C]	Yield [%] <sup>[b]</sup>	ee [%] <sup>[c]</sup>
1	<b>1a</b>	Cl	<i>t</i> Bu	benzene	4	4	80	21
2	<b>1b</b>	Cl	OTIPS	benzene	4	4	84	36
3	<b>1c</b>	Br	OTIPS	benzene	4	4	90	49
4	<b>1d</b>	I	OTIPS	benzene	4	4	93	56
5	<b>1d</b>	I	OTIPS	benzene	2	4	90	60
6 <sup>[d]</sup>	<b>1d</b>	I	OTIPS	<i>o</i> -xylene	2	4	95	65
7 <sup>[d]</sup>	<b>1e</b>	I	OSiThMe <sub>2</sub>	<i>o</i> -xylene	2	4	94	68
8 <sup>[d]</sup>	<b>1e</b>	I	OSiThMe <sub>2</sub>	<i>o</i> -xylene	2	-27	94	79

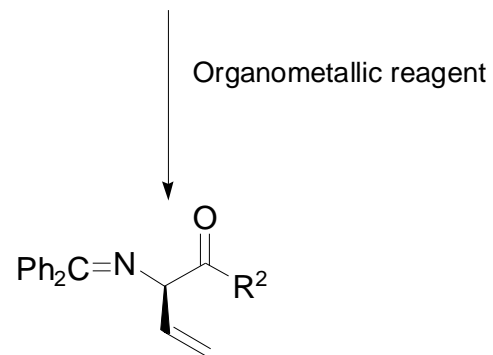
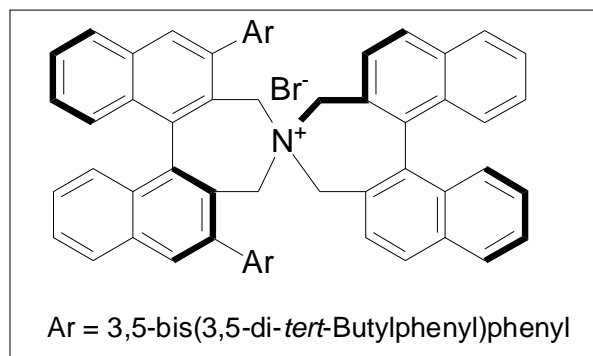
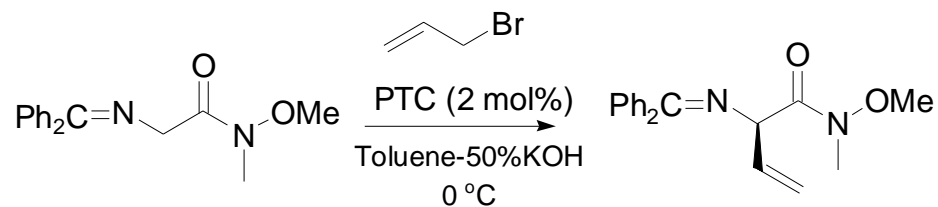


Entry	Tin enolate ( <i>E/Z</i> ratio)	$R^3CH_2X$	Adduct	Yield [%] <sup>[a]</sup>	<i>ee</i> [%] <sup>[b]</sup>
1	<b>2a</b> (1.8:1)		<b>3a</b>	80	79
2			<b>3a</b>	83	82
3			<b>3b</b>	86	81
4			<b>3c</b>	73	76
5	<b>2b</b> (1.5:1)		<b>3d</b>	92	87
6			<b>3e</b>	83	86
7			<b>3f</b>	77	84 <sup>[c]</sup>
8			<b>3g</b>	97	78 <sup>[d]</sup>

# Allylation Using Phase Transfer Catalysts



# One Example





# Conclusions

- There are more than one way to get to the destination.
- How you get there really matters.

One shortcut



*Standing on the shoulder of giants*