
Deracemization/Dynamic Kinetic Resolution of Quaternary Stereocenters

Shair, M. D. et al, *Angew. Chem. Int. Ed.* **2005**, 44, 2259

Stoltz, B. M. et al, *Angew. Chem. Int. Ed.* **2005**, 44, 6924

Nakamura, E. et al, *Angew. Chem. Int. Ed.* **2005**, 44, 7248

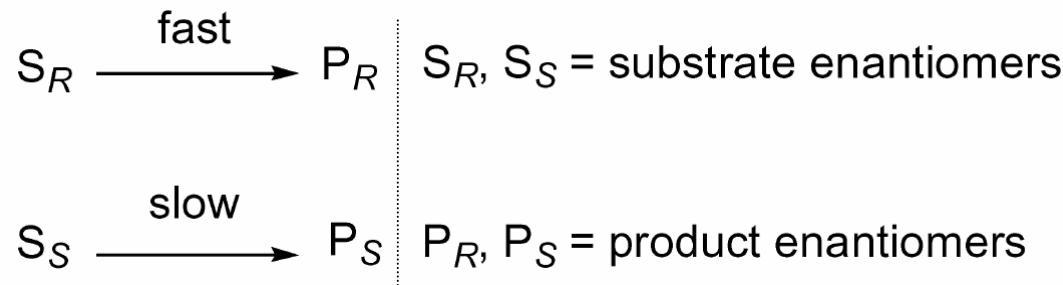
Literature Presentation

Yu Zhang

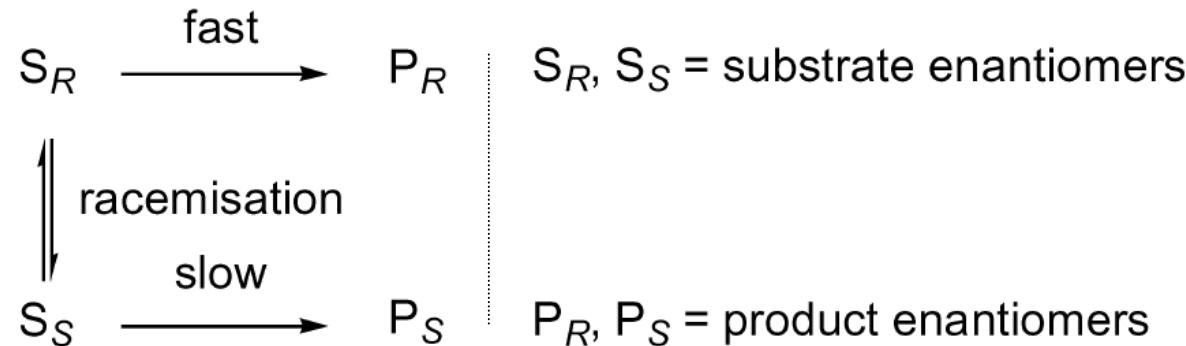
Nov. 10, 2005

Introduction to Dynamic Kinetic Resolution (DKR)

- ◆ Classical kinetic resolution:



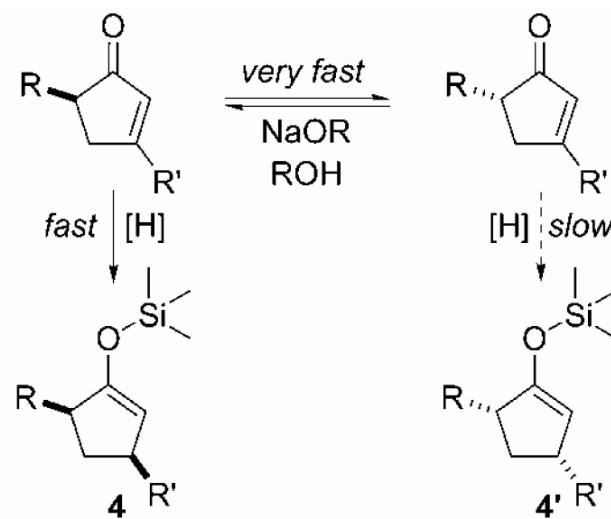
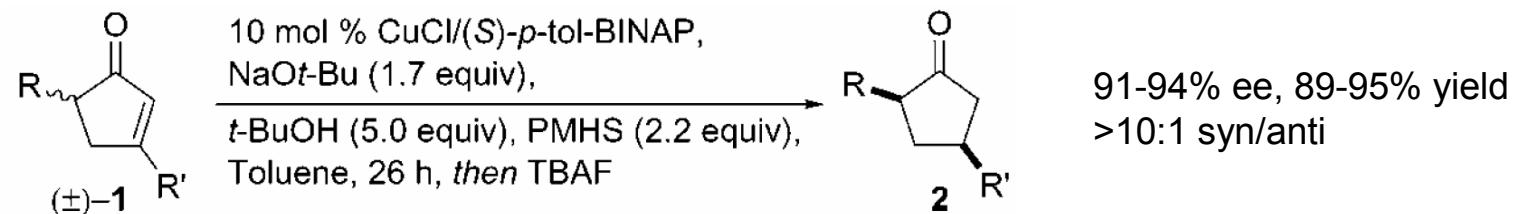
- ◆ Dynamic kinetic resolution (DKR):



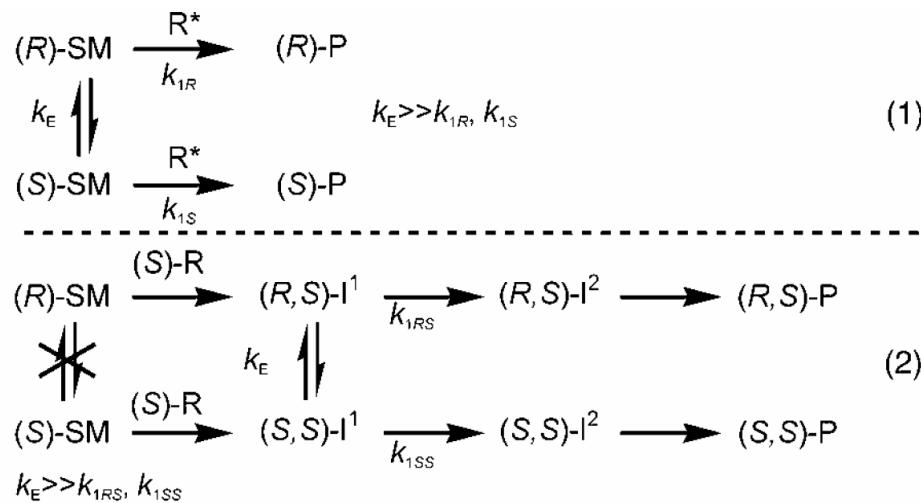
- DKR combines the resolution step of kinetic resolution, with an in situ equilibration or racemization of the chirally labile substrate.
- Racemization of the substrate can be performed either chemically, biocatalytically or even spontaneously; conditions must be chosen to avoid the racemization of the product.

Example of DKR

◆ DKR using proton transfer:



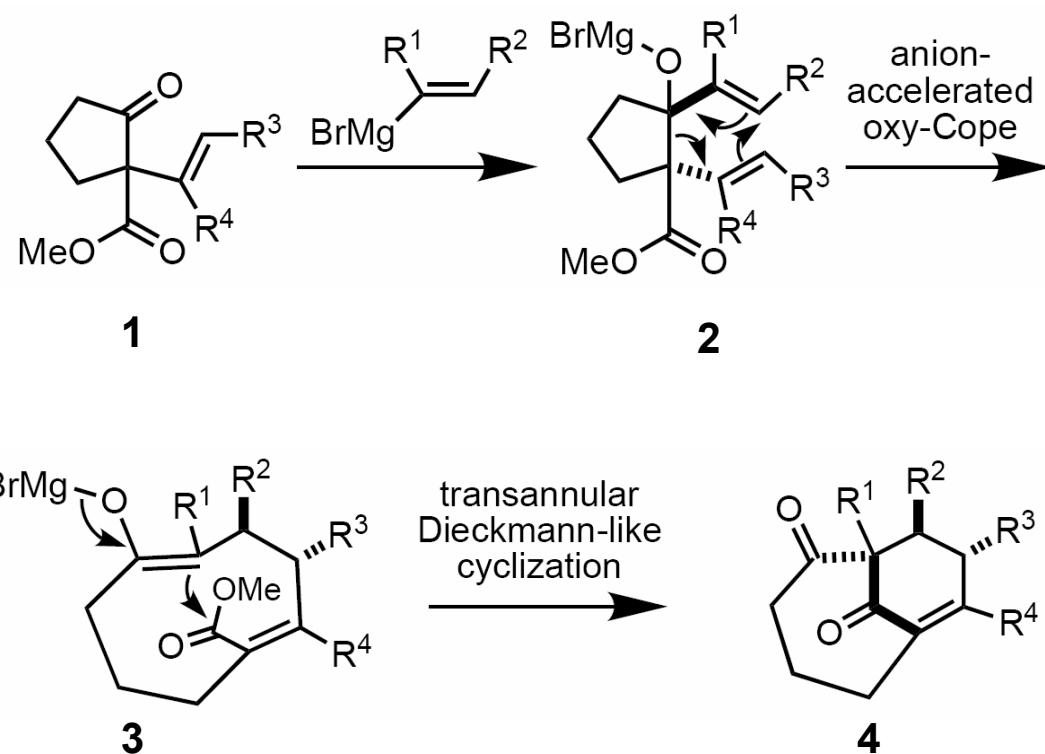
DKR of Chiral All-Carbon Quaternary Centers



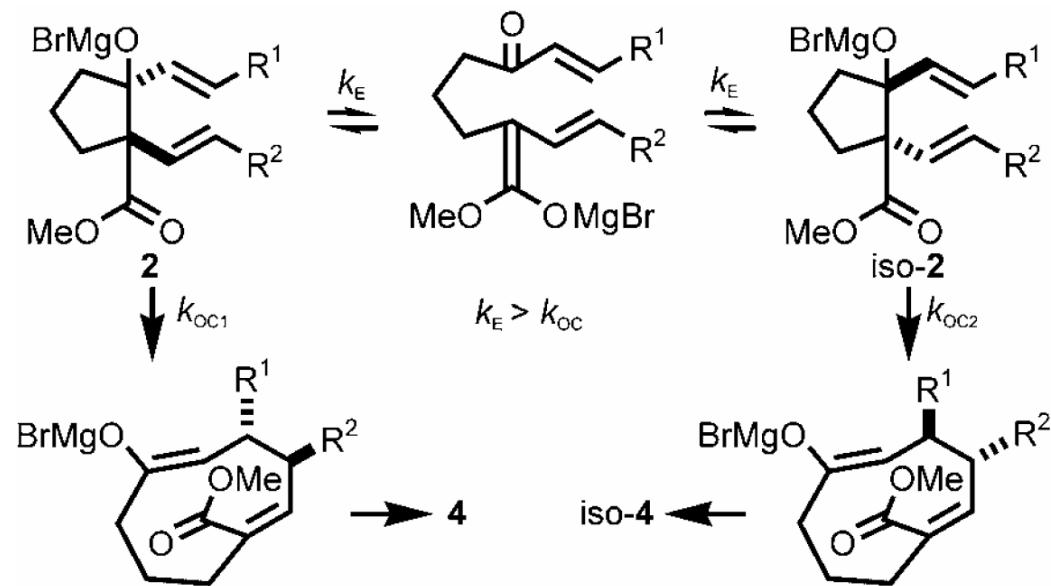
- ◆ Dynamic kinetic resolution with enantiomerization of the SM. Eq. (1).
- ◆ Dynamic kinetic resolution with epimerization in the second step (Eq. (2)).
 I^1 =first intermediate, I^2 =second intermediate, P=product.

DKR of Chiral All-Carbon Quaternary Centers

◆ A cascade reaction for the synthesis of polycyclic bridgehead enone compounds. (Shair, *Ang*, **2000**, 39, 2714)



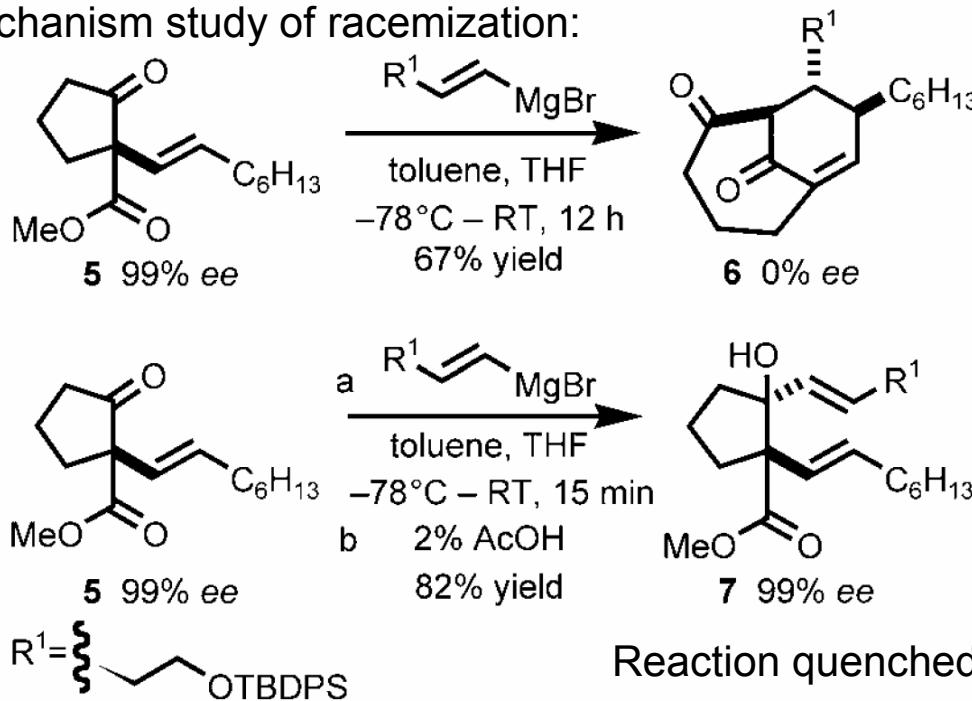
DKR of Chiral All-Carbon Quaternary Centers



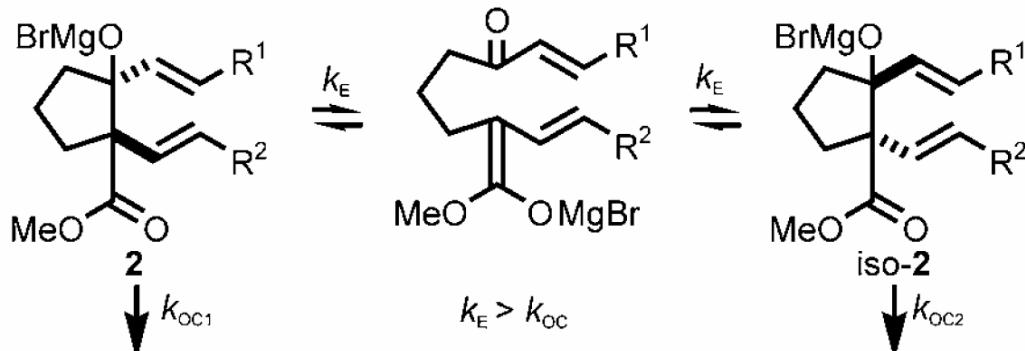
- ◆ Possibility of **2** to undergo retro-aldo/aldol equilibrium?
- ◆ Retro-aldo or Dieckmann reaction – Which one is faster?

DKR of Chiral All-Carbon Quaternary Centers

- ◆ Mechanism study of racemization:

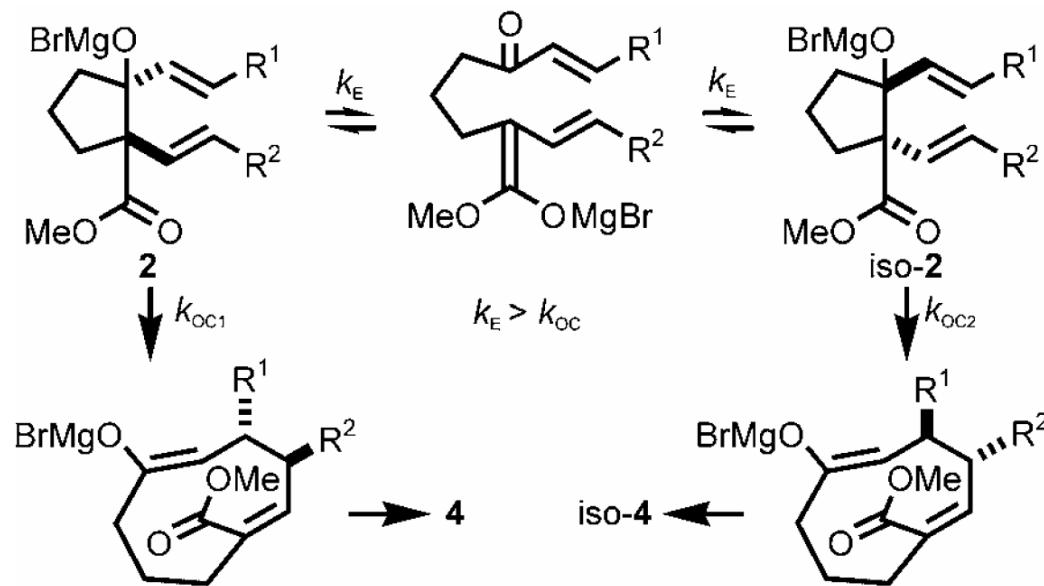


- ◆ Result: $K_E \gg K_{OC}$



DKR of Chiral All-Carbon Quaternary Centers

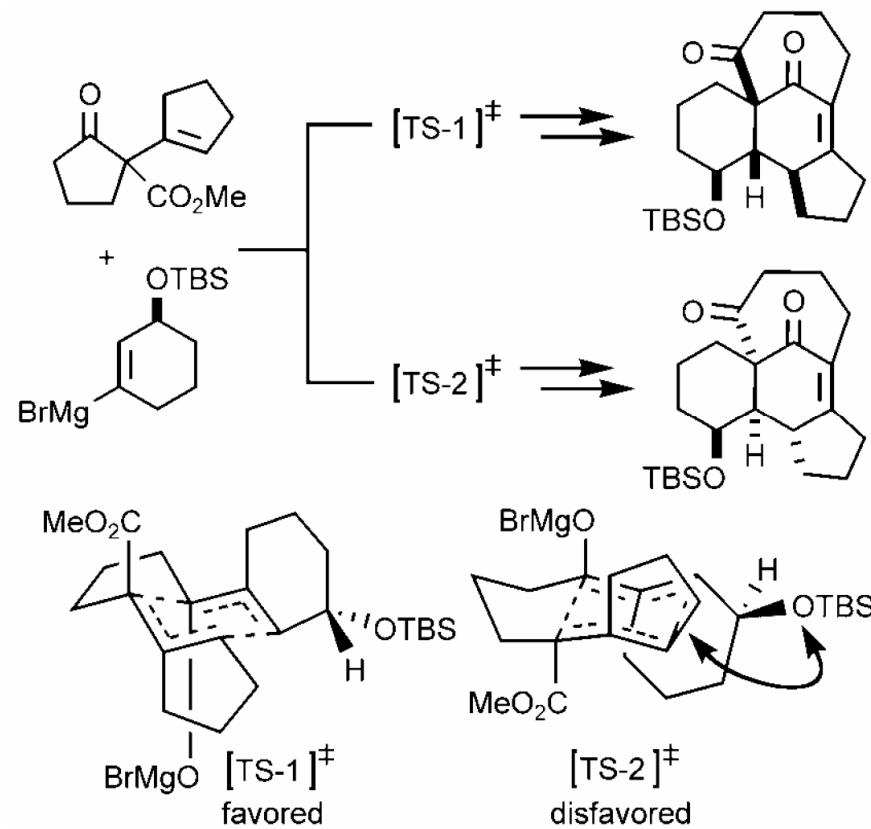
- ◆ Racemization of cascade reaction:



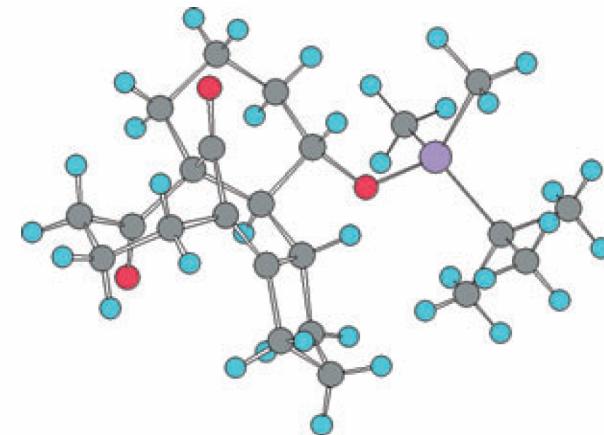
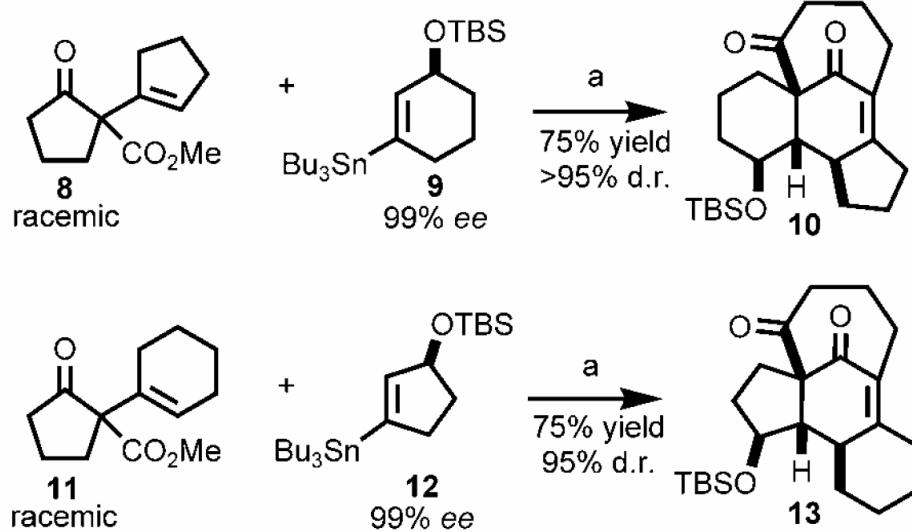
- Retro-aldol/aldol much faster than anionic oxy-Cope(AOC);
- AOC involves highly organized TS, control relative rate of K_{OC1} and K_{OC2} possible?

DKR of Chiral All-Carbon Quaternary Centers

◆ Proposed DKR of the polycyclic products:

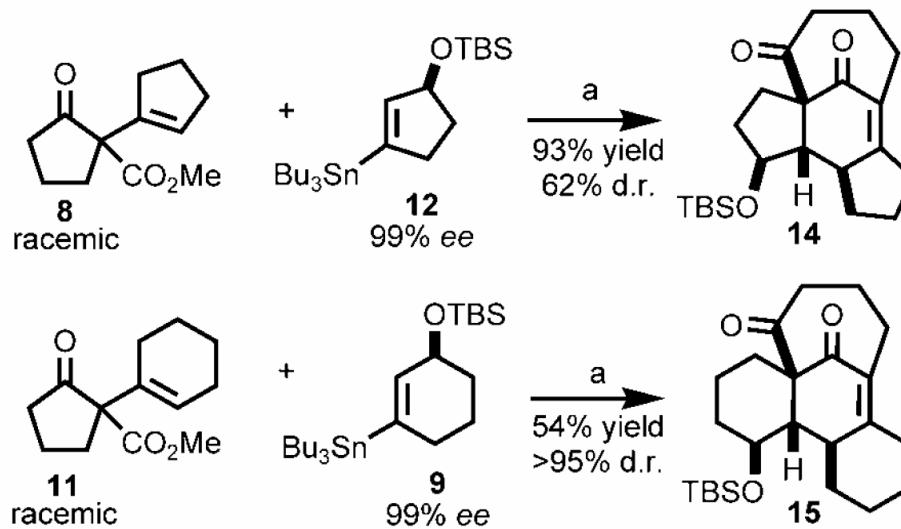


DKR of Chiral All-Carbon Quaternary Centers



a) $n\text{BuLi}$, THF, -78°C ; MgBr_2 , $\text{Et}_2\text{O}/\text{benzene}$, 0°C ; THF/toluene, 23°C , 18 h.

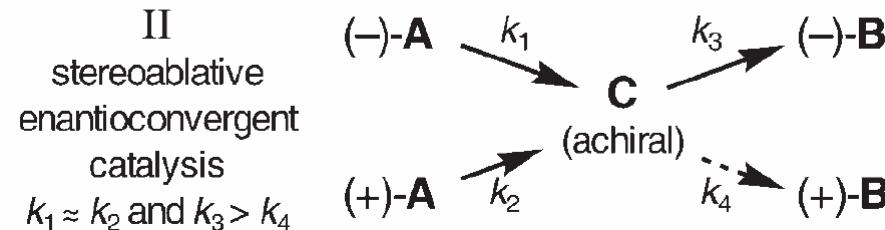
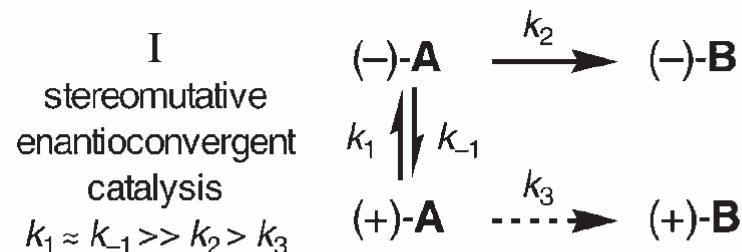
DKR of Chiral All-Carbon Quaternary Centers



a) $n\text{BuLi}$, THF, -78°C ; MgBr_2 , $\text{Et}_2\text{O}/\text{benzene}$, 0°C ; THF/toluene, 23°C , 18 h.

Deracemization of Quaternary Centers - Introduction

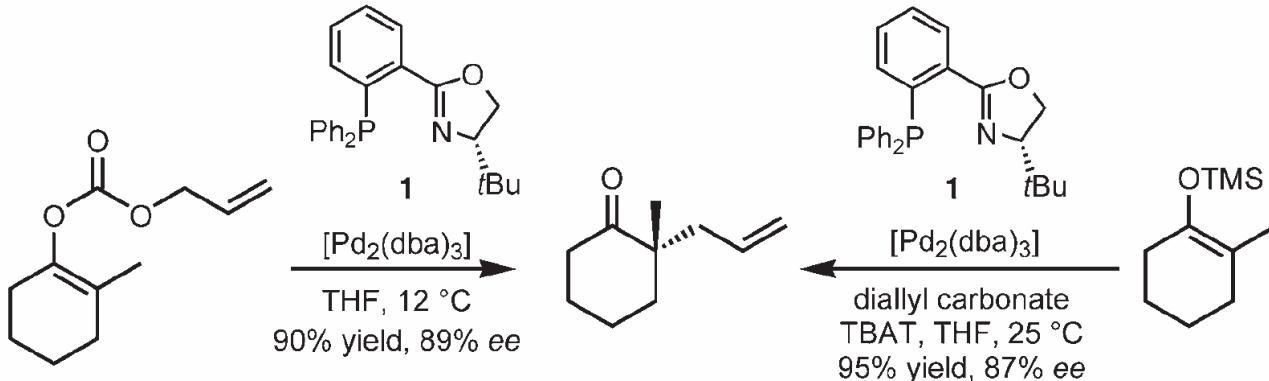
- ◆ Stereomutative versus stereoablative enantioconvergent catalysis.



- The term “stereoablative” describes the conversion of a chiral molecule to an achiral molecule.

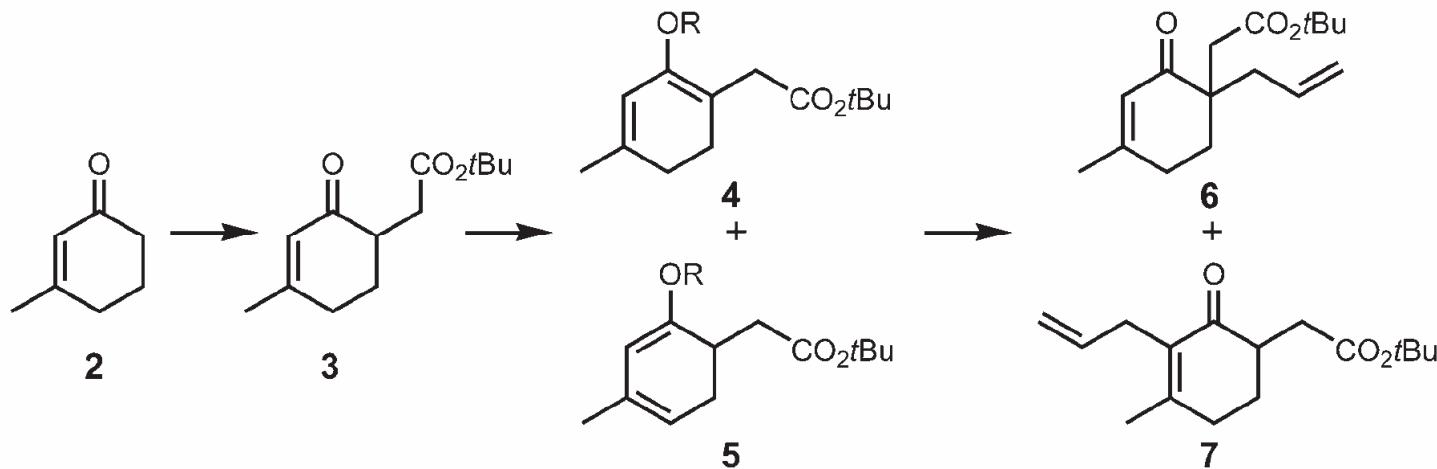
Deracemization of Quaternary Centers - Introduction

- ◆ Enantioselective Tsuji Allylations:



Stoltz, JACS, 2004, 126, 15044

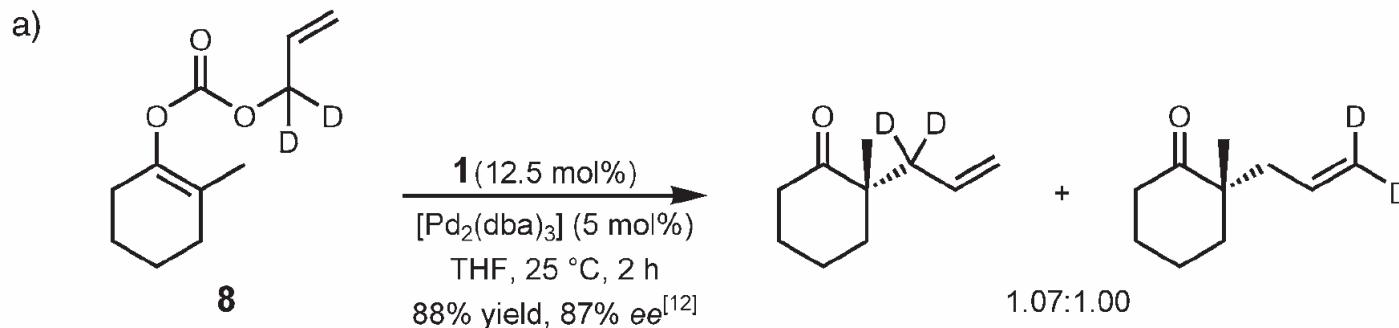
- ◆ However, enolization not selective for the synthesis of 4 ($\text{R} = \text{CO}_2\text{allyl}$ or TMS):



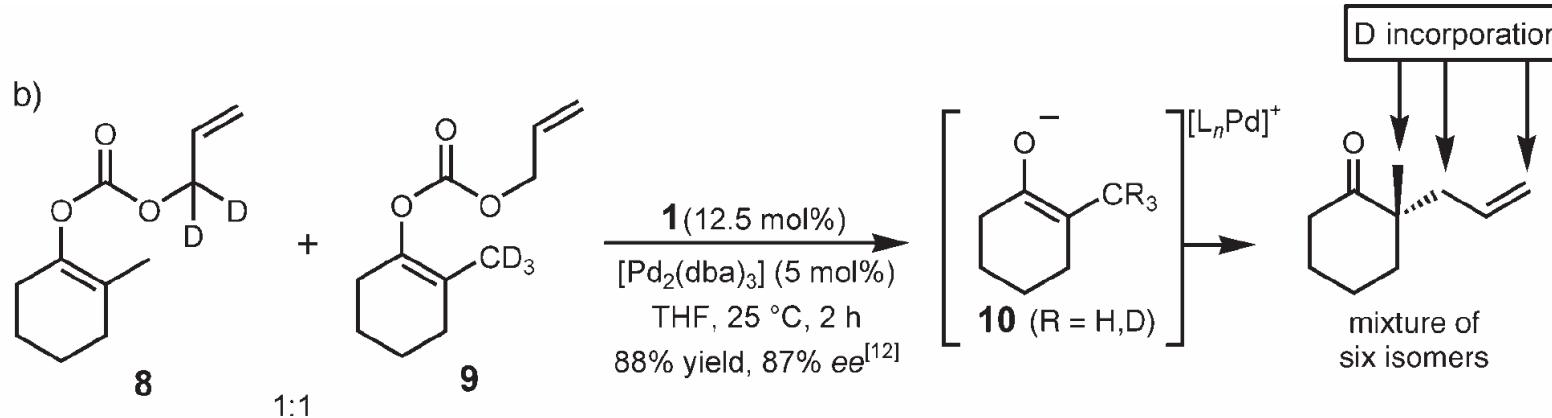
Mohr, J. T.; Behenna, D. C.; Harned, A. M.; Stoltz, B. M, Angew. Chem. Int. Ed. 2005, 44, 6924

Deuterium-labeling experiments

- ◆ Deuterium-labeling experiment with **8**:



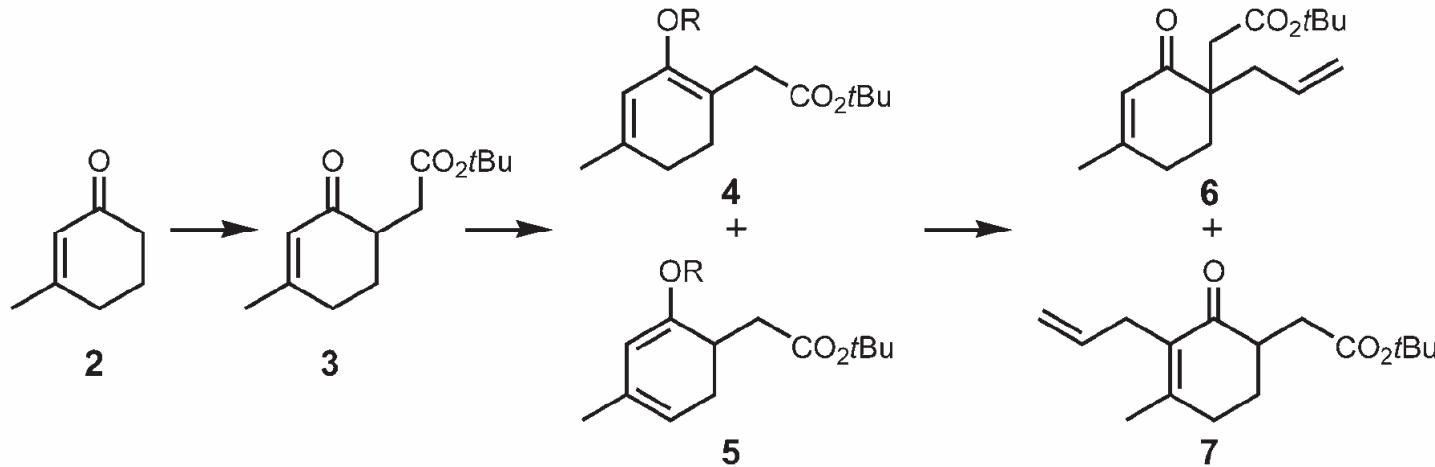
- ◆ Crossover experiment with **8** and **9**:



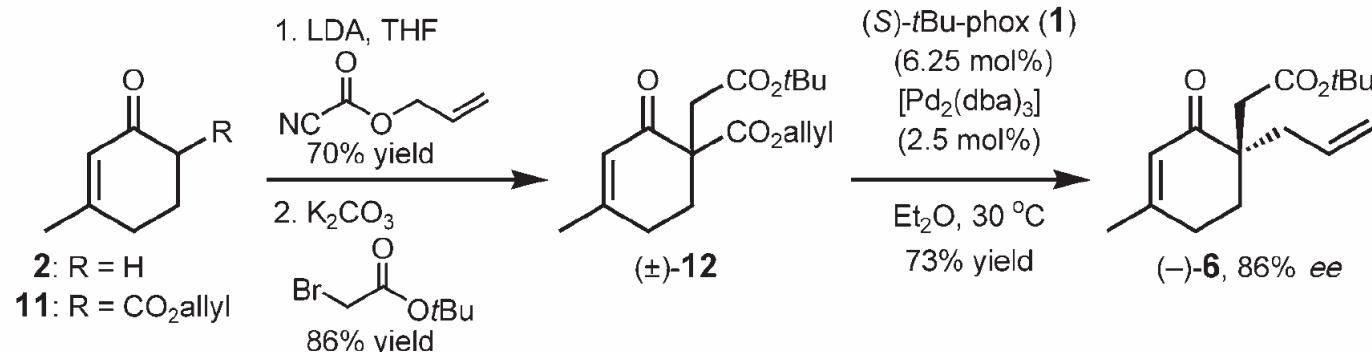
-- A discrete achiral ketone enolate **10** must exist.

β -Ketoester synthesis and decarboxylative allylation

- ◆ Nonselective synthesis of **6**



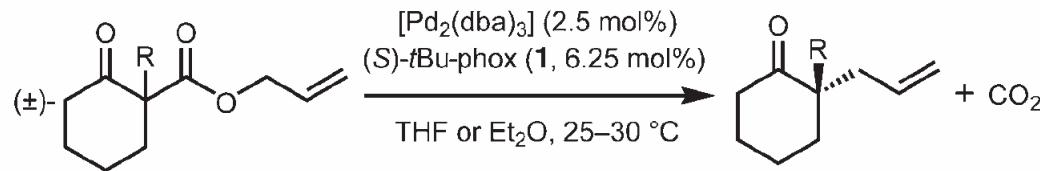
- ◆ Enantioselective synthesis of **6** via decarboxylative allylation.



- Above process solved the nonselective enolization of **3**;
- **12** was racemic throughout the reaction.

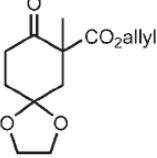
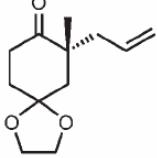
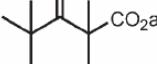
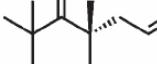
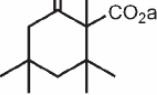
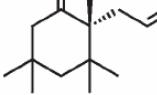
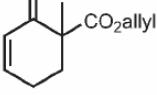
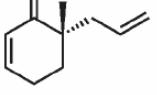
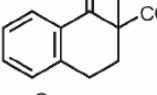
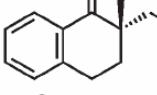
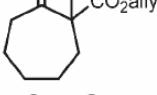
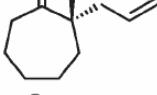
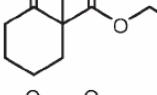
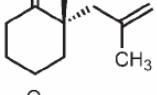
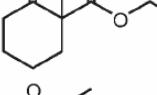
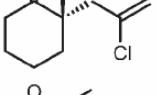
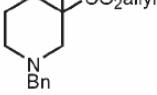
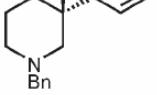
Catalytic enantioconvergent decarboxylation allylation

◆ Enantioconvergent decarboxylation allylation of 2-carboxyallylcyclohexanones.



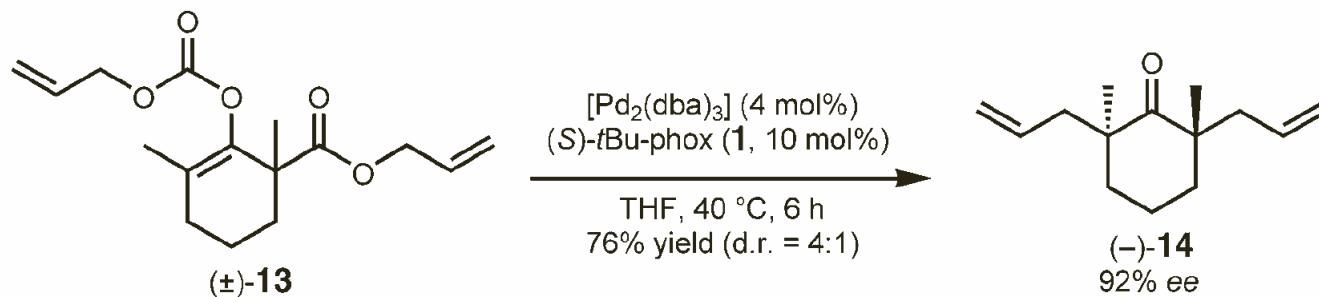
Entry	R	Solvent	T [°C]	t [h]	Yield [%] ^[a]	ee [%] ^[b]
1	CH ₃	THF	25	7.5	85	88
2	CH ₃	Et ₂ O	25	4.75	89	88
3	prenyl	Et ₂ O	30	6	97	91
4	CH ₂ CH ₂ CN	Et ₂ O	25	6.5	97	88
5 ^[c]	CH ₂ CH ₂ CO ₂ Et	Et ₂ O	25	6	96	90
6	CH ₂ C ₆ H ₅	THF	25	0.5	99	85
7	CH ₂ (4-CH ₃ OC ₆ H ₄)	THF	25	10	80	86
8	CH ₂ (4-CF ₃ C ₆ H ₄)	THF	25	0.5	99	82
9 ^[c]	CH ₂ OTBDPS	THF	25	5	86	81
10	F	Et ₂ O	30	3.5	80	91

Catalytic enantioconvergent decarboxylation allylation

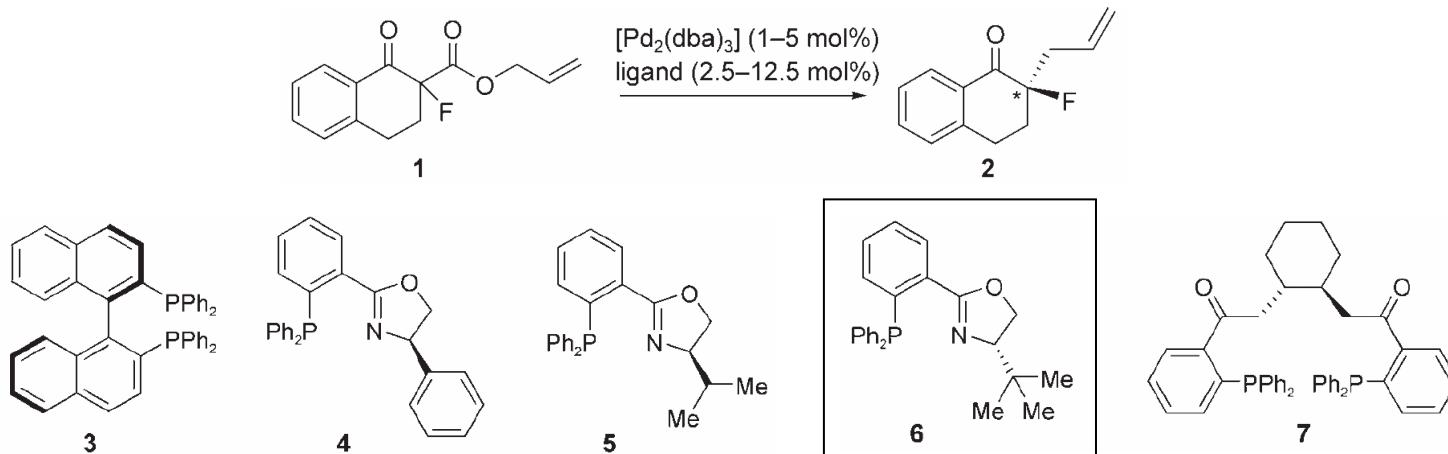
Entry	Substrate	Product	T [°C]	t [h]	Yield [%] ^[a]	ee [%] ^[b]
1 2 ^[c]			25 25	1.5	94	85
				24	94	86
3			30	9	89	90
4			25	5	90	85
5 ^[d,e]			30	4	77	90
6 ^[d]			25	10	97	92
7			25	9.5	83	87
8 ^[d]			35	6.5	87	92
9 ^[d,e]			35	2.5	87	91
10			25	2.5	91	92

Catalytic enantioconvergent decarboxylation allylation

- ◆ The enantioselective allylation cascade generating two quaternary carbon stereocenters:

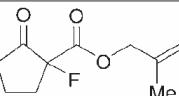
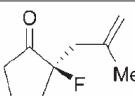
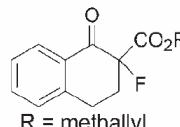
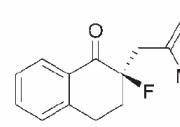
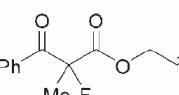
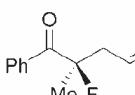
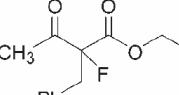
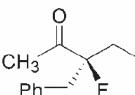
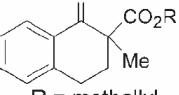
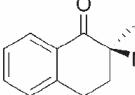
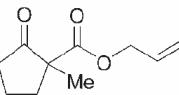
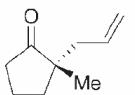
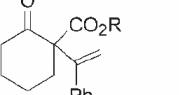
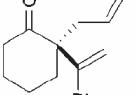


Catalytic enantioselective decarboxylation



Entry	Ligand	<i>t</i> [h]	Solvent	[Pd ₂ (dba) ₃] ^[b] [mol %]	Yield [%] ^[c]	<i>ee</i> [%] ^[d]
1	none	24	THF	5	0	—
2	3	5	THF	5	83	11
3	4	3	THF	5	91	11
4	5	3	THF	5	94	83
5	6	3	THF	5	95	96
6	6	4	THF	2.5	95	96
7	6	10	THF	1	93	94 ^[e]
8	6	4	Et ₂ O	5	93	96
9	6	5	CH ₂ Cl ₂	5	89	25
10	6	4	THF	2.5	94	96 ^[f]
11	7	10	THF	5	9	n.d. ^[g]

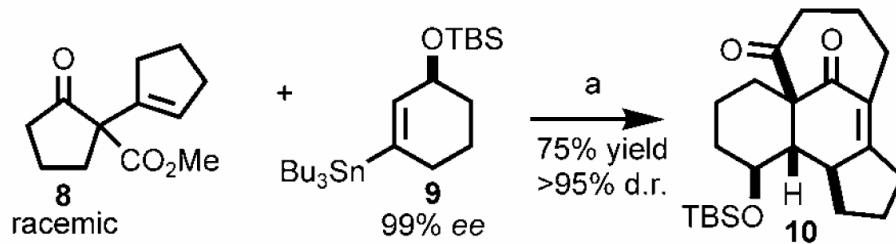
Catalytic enantioselective decarboxylation

Entry	Substrate	Product	Yield [%] ^[b]	ee [%] ^[c]	Entry	Substrate	Product	Yield [%] ^[b]	ee [%] ^[c]
1	1	2	95	96	7			82 ^[e]	85
2 ^[d]			96	99	8			89	51
3	R = methallyl		95	97	9			91	55
4	R = allyl		92	96	10			94	95
5	R = allyl		94	91	11			87 ^[e]	84
6	R = allyl		94	93	12			89	39

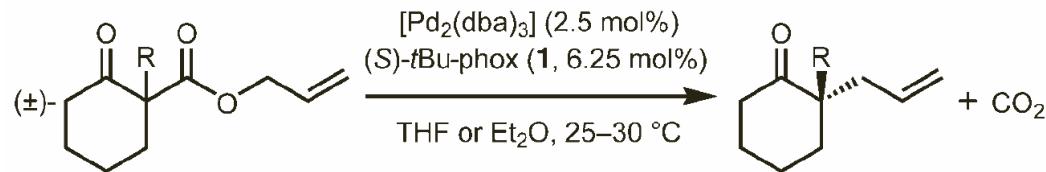
All reactions were performed with a 0.05m substrate concentration in THF for 4–10 h at 22–25°C in the presence of [Pd₂(dba)₃] (2.5 mol%) and ligand **6** (6.25 mol%).

Conclusion

- Shair:



- Stoltz, Nakamura:



◆ Careful consideration – detailed mechanism study – valuable reaction.