
Total Synthesis of TMC-95A and -B via a New Reaction Leading to Z-Enamides.

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J. Am. Chem. Soc. **2004**, 126, 6347-6355

Related References:

Early works:

Angew. Chem. Int. Ed. **2001** 40, 1967 (core synthesis)

Angew. Chem. Int. Ed. **2002** 41, 512 (total synthesis)

J. Am. Chem. Soc. **2003** 125, 5111 (computation)

Other finished synthesis:

Hirama, Masahiro et al. *Angew. Chem. Int. Ed.* **2003**, 42, 2654

Williams, Robert et al. *Org. Lett.* **2003**, 5, 197

Yu Zhang

August 26, 2004

Introduction to TMC-95 Family

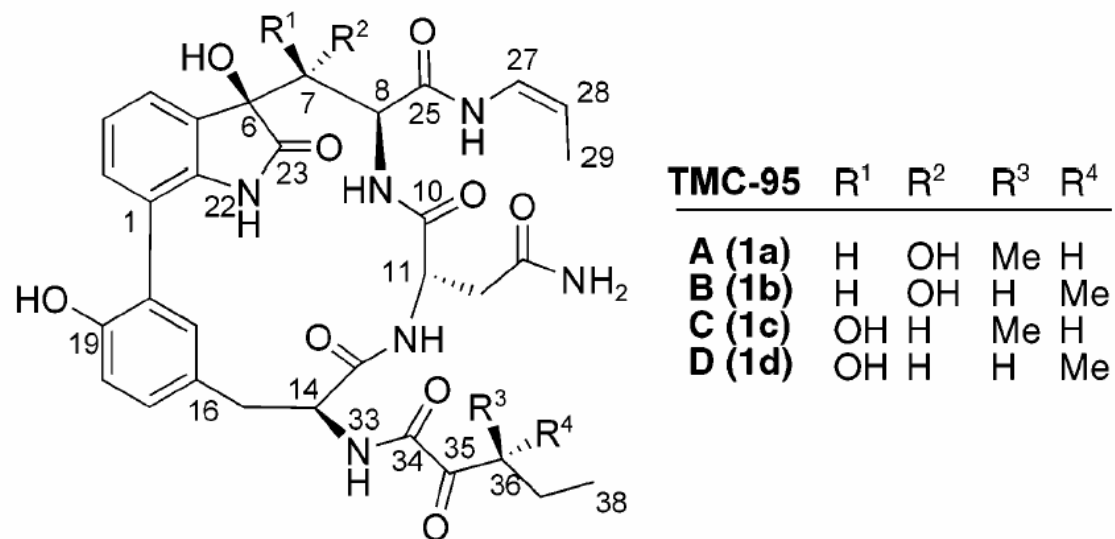


Figure 1. Structures of TMC-95A-D.

- ◆ Isolation: Isolated as fermentation products of *Apoispora Montagnei* in 2000.
- ◆ Biological Activity: TMC-95A inhibited the chymotrypsin-like, trypsin-like, and post-glutamyl peptide hydrolytic activities of proteasome with IC₅₀ of 5.4, 200 and 60 nM. Also showed cytotoxic activities against human cancer cells HCT-116 & HL-60 with IC₅₀ of 4.4 and 9.8 μM.

Structural Characteristics of TMC-95A

Highly oxidized
L-tryptophan moiety

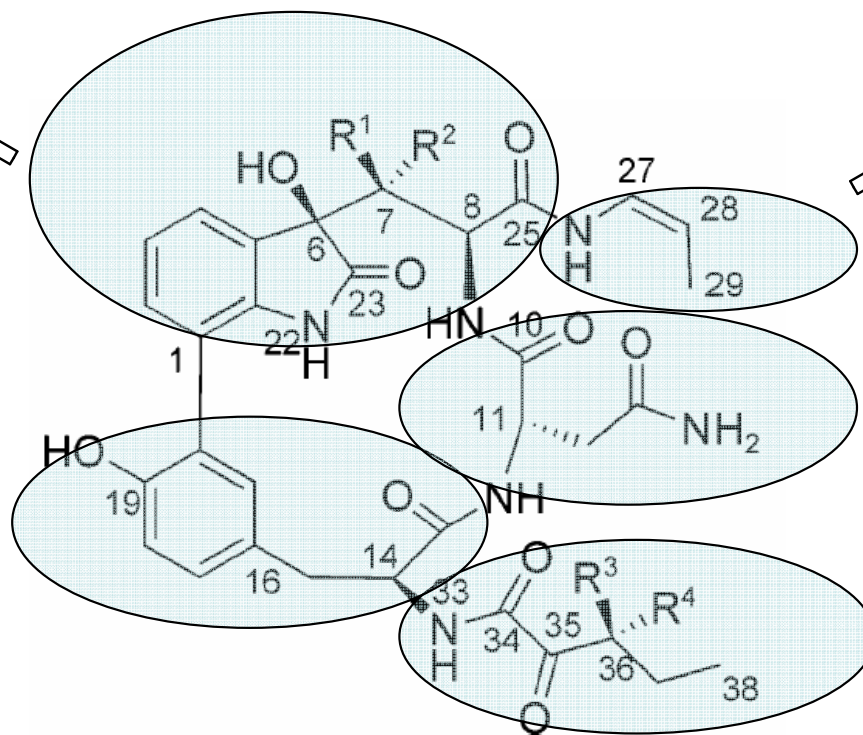
Z-Enamide

L-asparagine

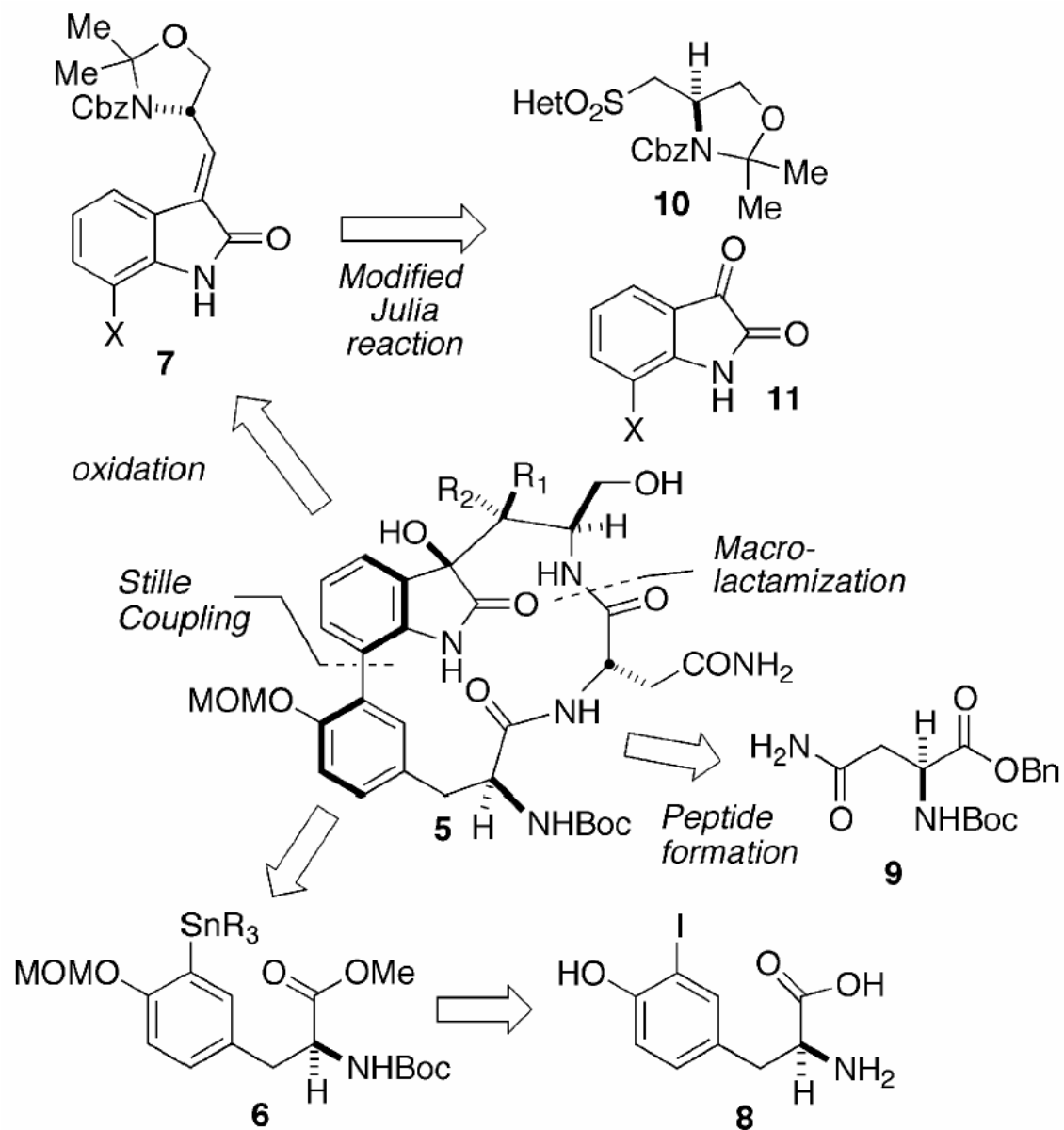
L-Tyrosine

3-methyl-2-oxopentanoic acid

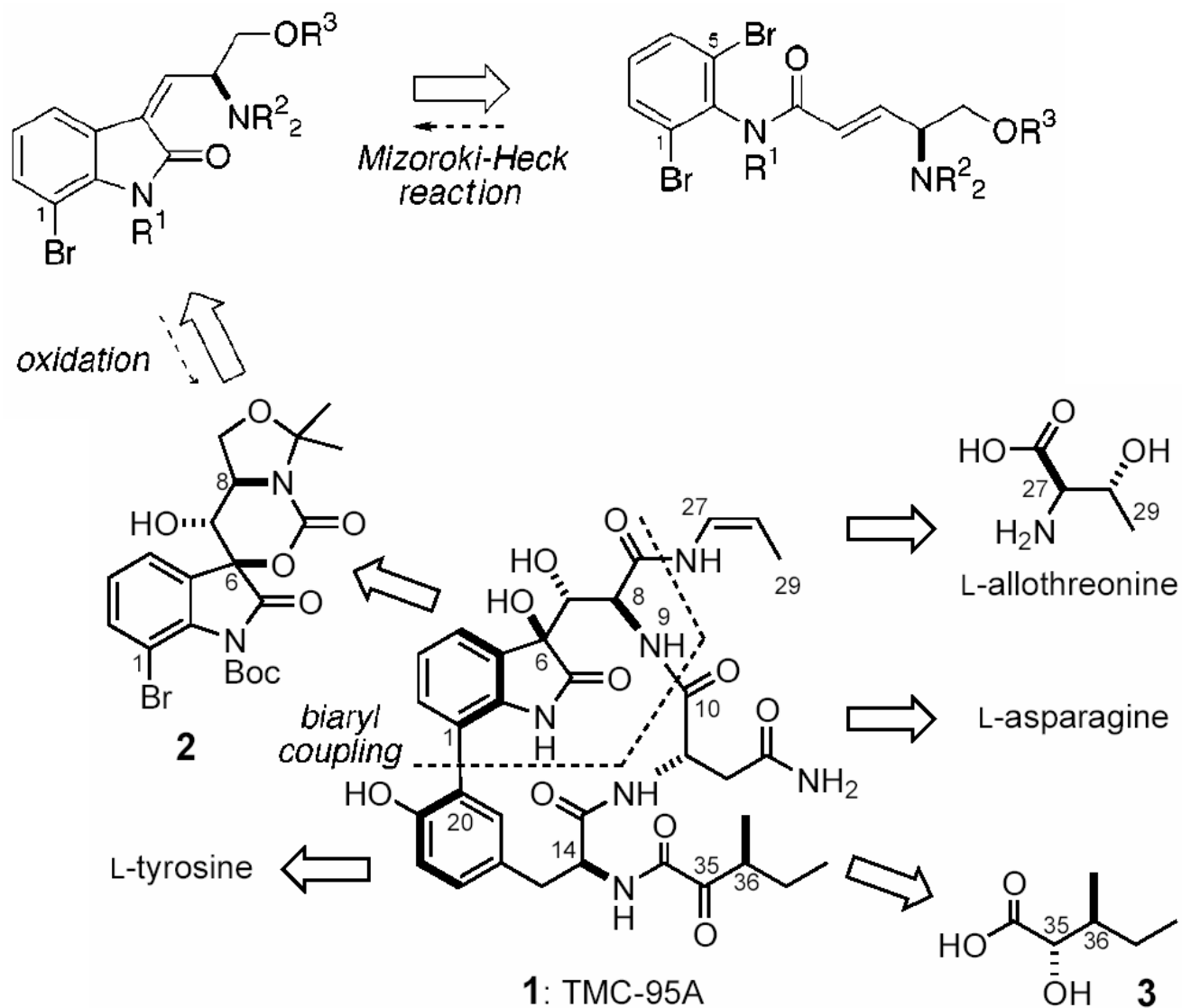
TMC-95	R ¹	R ²	R ³	R ⁴
A (1a)	H	OH	Me	H



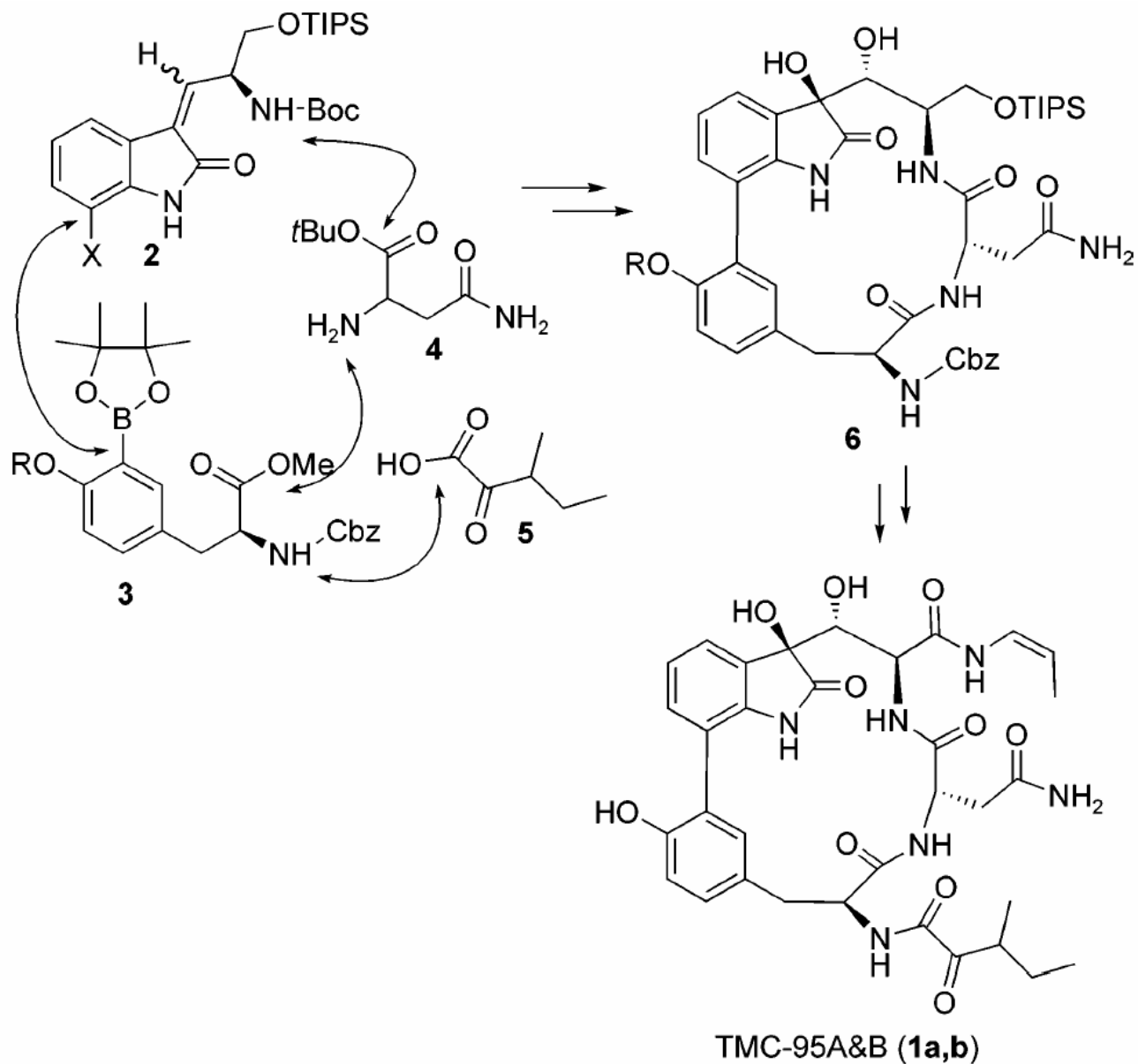
William's Synthetic Strategy



Hirama's Synthetic Strategy

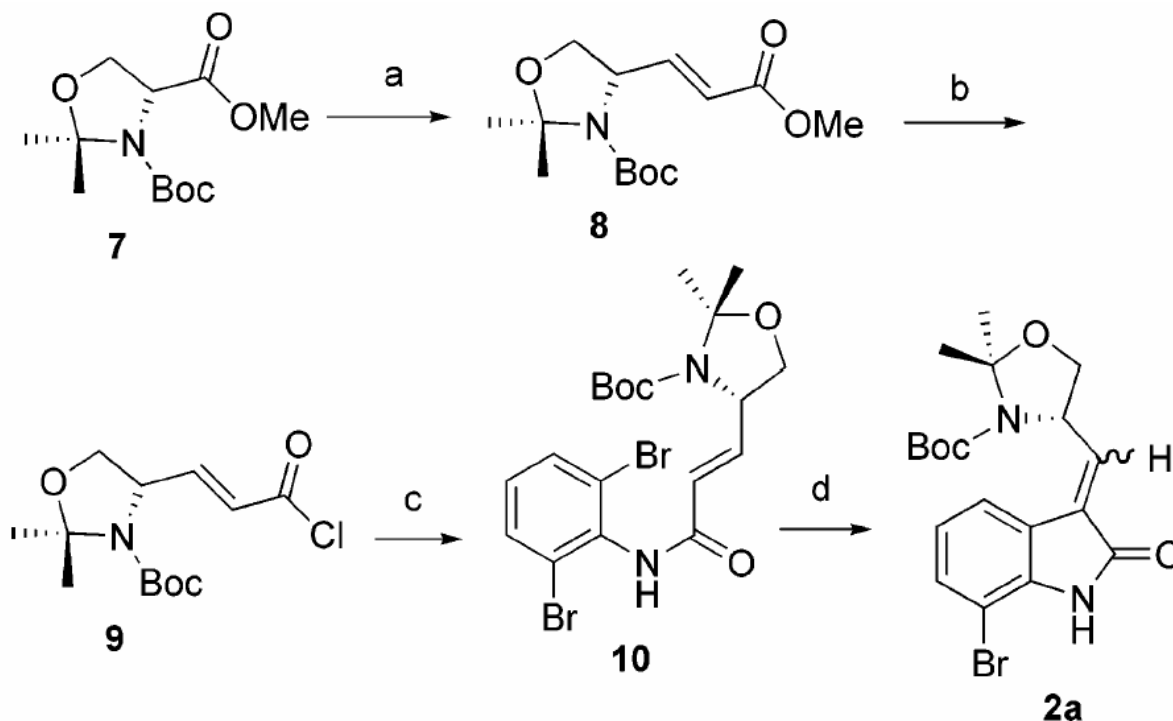


Danishefsky's Synthetic Strategy



Danishefsky, S. et al. *J. Am. Chem. Soc.* **2004**, *126*, 6347.

Attempted Synthesis of 7-Bromooxindole **2a**



Conditions:

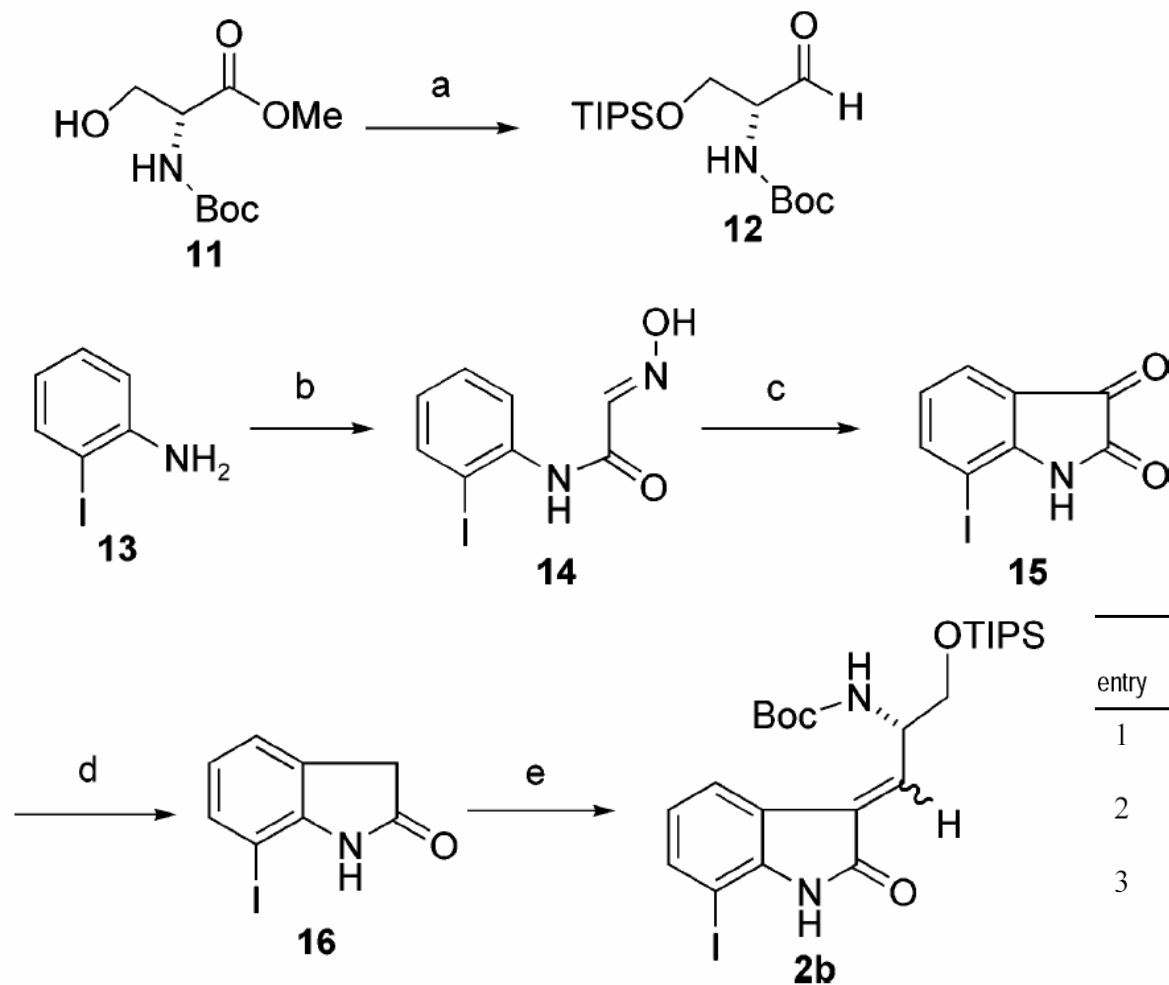
(a) (1) DIBAL/toluene, -78°C , 1 h; (2) methyl (triphenylphosphoranylidine)acetate, CH_2Cl_2 , rt, 88%;

(b) (1) LiOH, THF/MeOH/ H_2O , (2) TBSCl, Et_3N /DMAP, (3) $(\text{COCl})_2$, DMF (cat.);

(c) 2,6-dibromoaniline, NaH/DMF/THF, 75°C , 1.5 h, 44%;

(d) $[\text{Pd}(\text{PPh}_3)_4]$ or $\text{Pd}(\text{OAc})_2$, 5-15%.

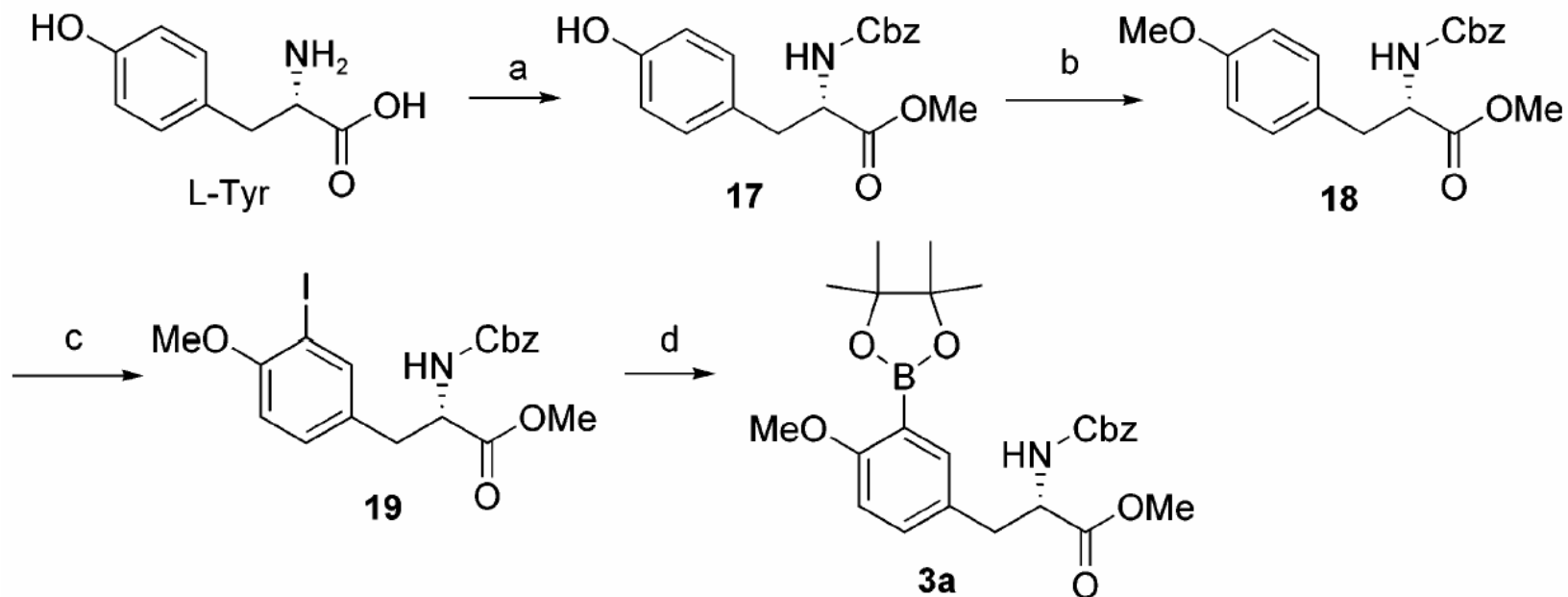
Synthesis of 7-Iodoindole **2b**



- (a) (1) TIPSCl, imidazole, 94%;
DIBAL, -78°C, 96%;
(b) CCl₃CH(OH)₂, NH₂OH HCl, Na₂SO₄,
H₂O, 45°C, 12 h, 66%;
(c) H₂SO₄, 70°C, 15 min, 88-98%;
(d) (1) NH₂NH₂·H₂O, 125 °C, 1 h,
(2) HCl (6 N), 60°C, 2 h, 81% (two steps);
(e) **12**, see Table 1.

entry	conditions	yield ^a (%)	ELZ ^b	ee ^c (%)
1	piperidine (cat.), MeOH or EtOH, 65 °C, 2–3 h	44–50	2.0/1	0
2	piperidine (cat.), THF, rt, 17–44 h	55	1.7/1	~10
3	(i) LDA (2.1 equiv), THF, -78 °C, 16 , then 12 , 1 h; (ii) TEA (2.5 equiv), MsCl (1.2 equiv), CH ₂ Cl ₂ , -60 to -30 °C, 2 h	76	1.3/1	92
4	(i) LDA (2.0 equiv), THF, -78 °C, 16 , then 12 , 1.5 h; (ii) TEA (3 equiv), MsCl (1.5 equiv), CH ₂ Cl ₂ , -70 to -50 °C, 1.5 h	74	1.2/1	96

Synthesis of Aryl Borate 3a



Conditions:

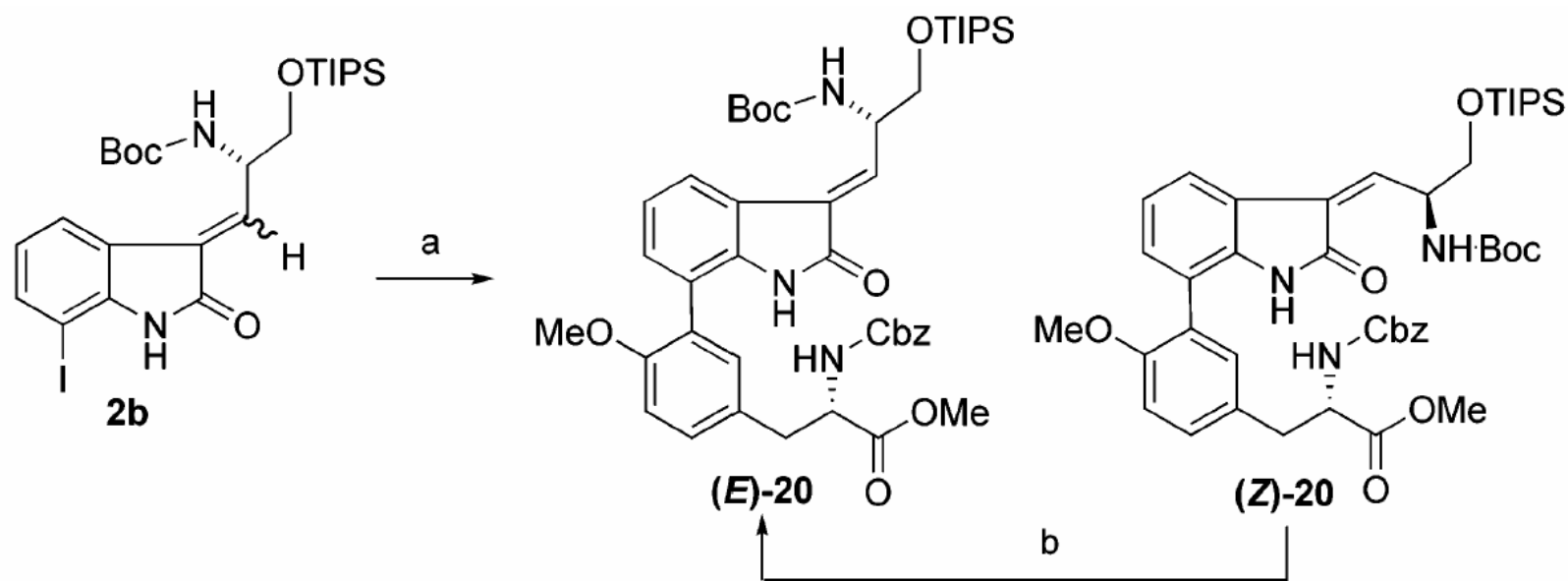
(a) (1) MeOH/SOCl₂, (2) CbzCl, K₂CO₃, H₂O/acetone, 96% (two steps);

(b) LiOH, Me₂SO₄, 86%;

(c) I₂, Ag₂SO₄, MeOH, rt, 1.5 h, 93%;

(d) bis(pinacolato)diboron, [PdCl₂(dppf)]·CH₂Cl₂, KOAc, DMSO, 80°C, 13 h, 95%

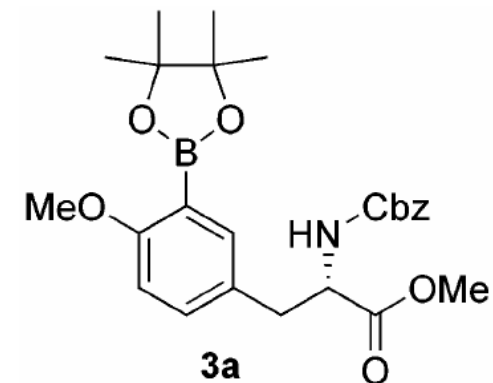
Biaryl Linkage Formation by Suzuki Coupling



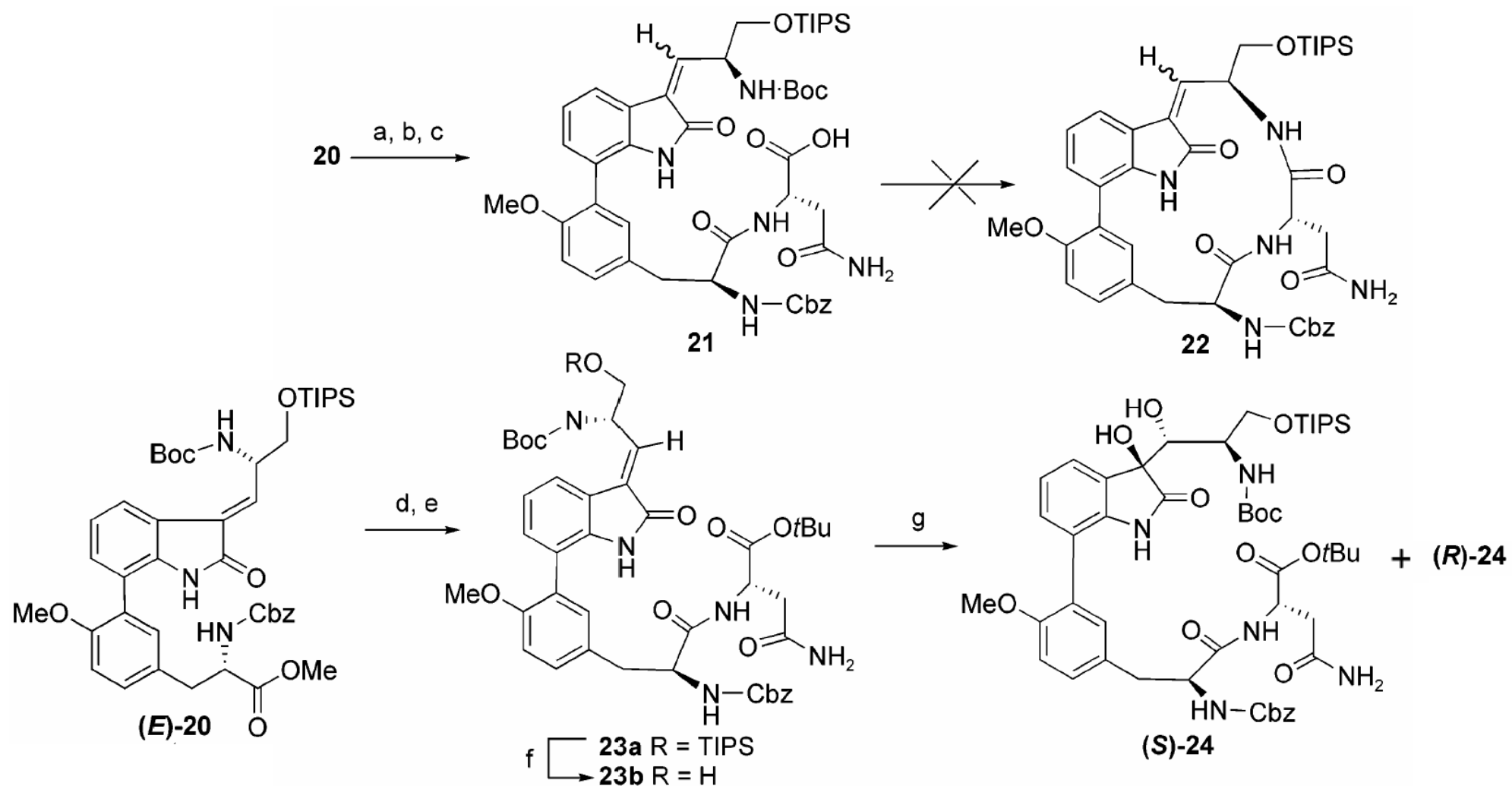
Conditions:

(a) **3a**, [PdCl₂(dppf)]·CH₂Cl₂, KOAc, DME, 80°C, 2 h, 72% (*E/Z* ~ 2/1);

(b) I₂ (cat.), DME, 80°C, 1 d, 87% (63% conv).



Synthesis of Diols 24



(a) LiOH, THF/MeOH/H₂O; (b) hydroxysuccinimide, DCC, THF, 55% (two steps);

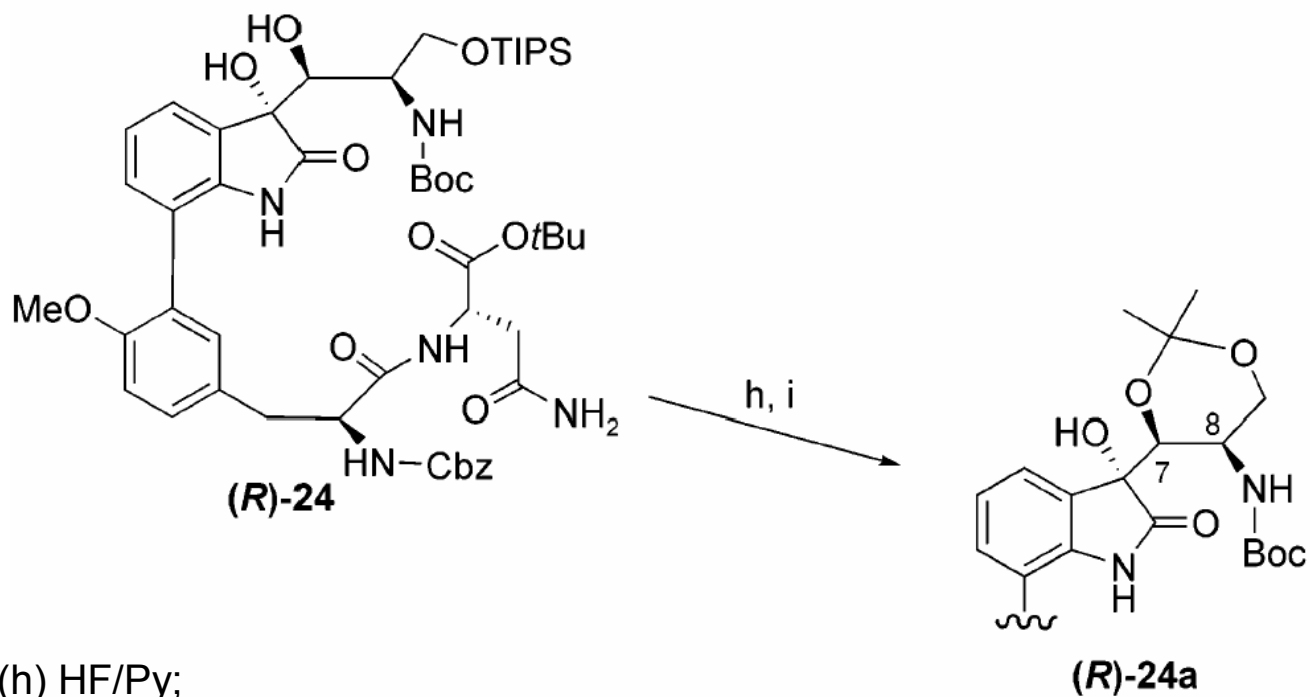
(c) L-Asn·H₂O, Et₃N, THF/H₂O, rt, 4 h, 70%;

(d) LiOH, THF/H₂O, 0°C, 1.5 h; (e) H-Asn-O-*t*-Bu, EDC/HOAt, THF, rt, 2 h, 70% (two steps); (f) HF/Py, 84%;

(g) for **23a**: OsO₄/NMO, (DHQD)₂PHAL, *t*-BuOH/H₂O, rt, 1 h, 84% (*S/R* ~ 1/1.8);

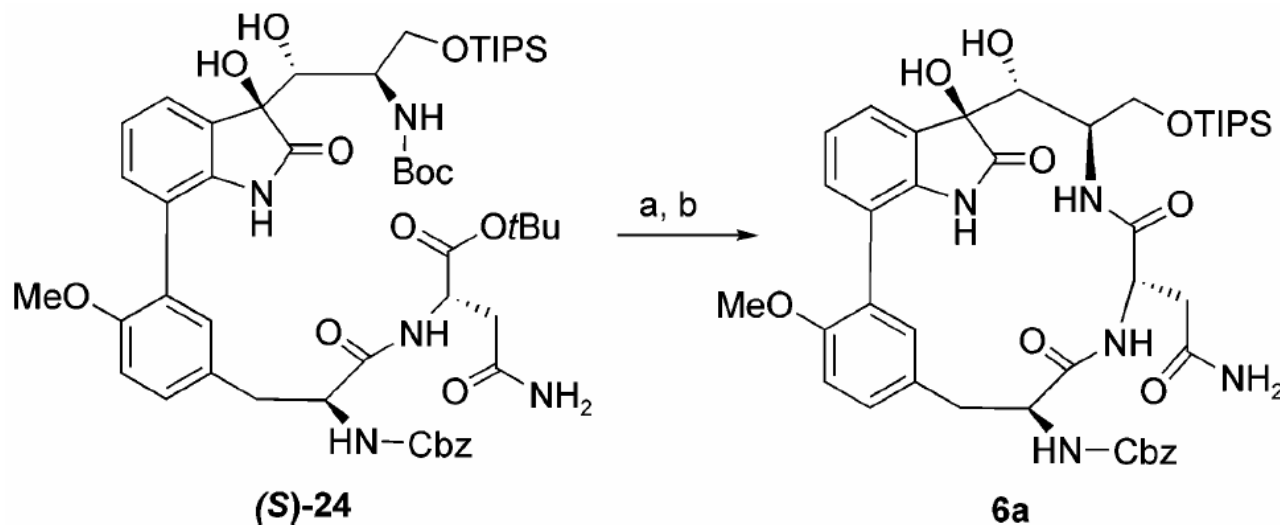
for **23b**: (1) OsO₄/NMO, (DHQ)₂PHAL, *t*-BuOH/H₂O, rt, 4 h, (2) TIPSCl, imidazole/DMAP, 5 h, 81% (*S/R* ~ 1/1.8);

Identification of Absolute Configuration of **24**



(h) HF/Py;
(i) DMP/PPTS, CH₂Cl₂.

Macrolactamization of (S)-24



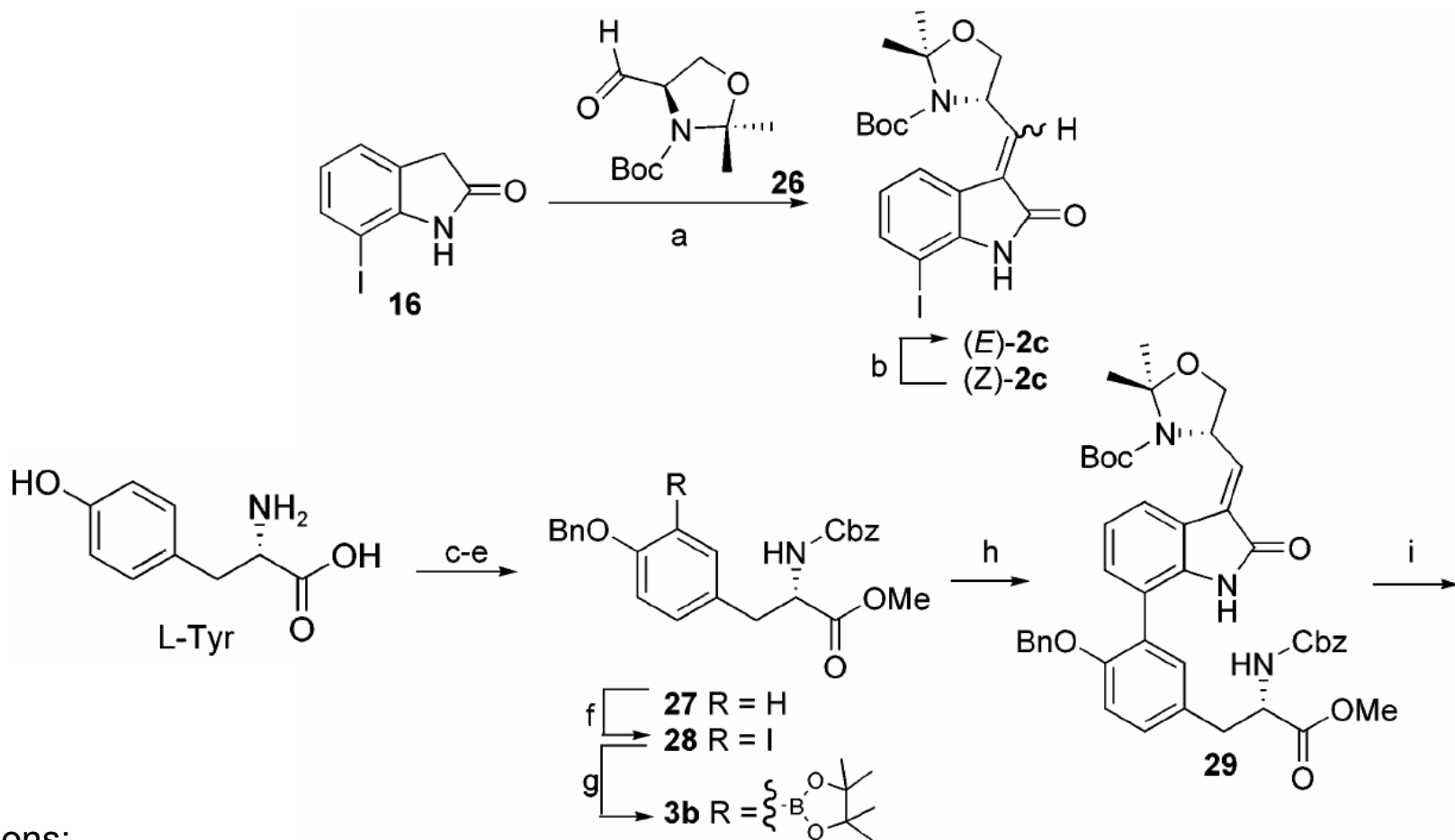
Conditions:

(a) TFA/CH₂Cl₂ (4:1), rt, 2 h;

(b) EDC, HOAT, DIEA, CH₂Cl₂/DMF (4/1, 4 mM), 20 h, 55% (two steps).

◆ Treatment of **24R** under same condition did not afford any cyclization product.

Synthesis of Macrolactam **6b**



Conditions:

(a) LDA (2.0 equiv), THF, -78°C , 1.5 h; then Et_3N , MsCl, CH_2Cl_2 , -70 to -50°C , 1.5 h, 81% (*E/Z* = 1.3/1);

(b) I_2 (cat.), benzene, 80°C , 26 h; DMP/PPTS, toluene, 65°C , 5 h; 85% (60% conv);

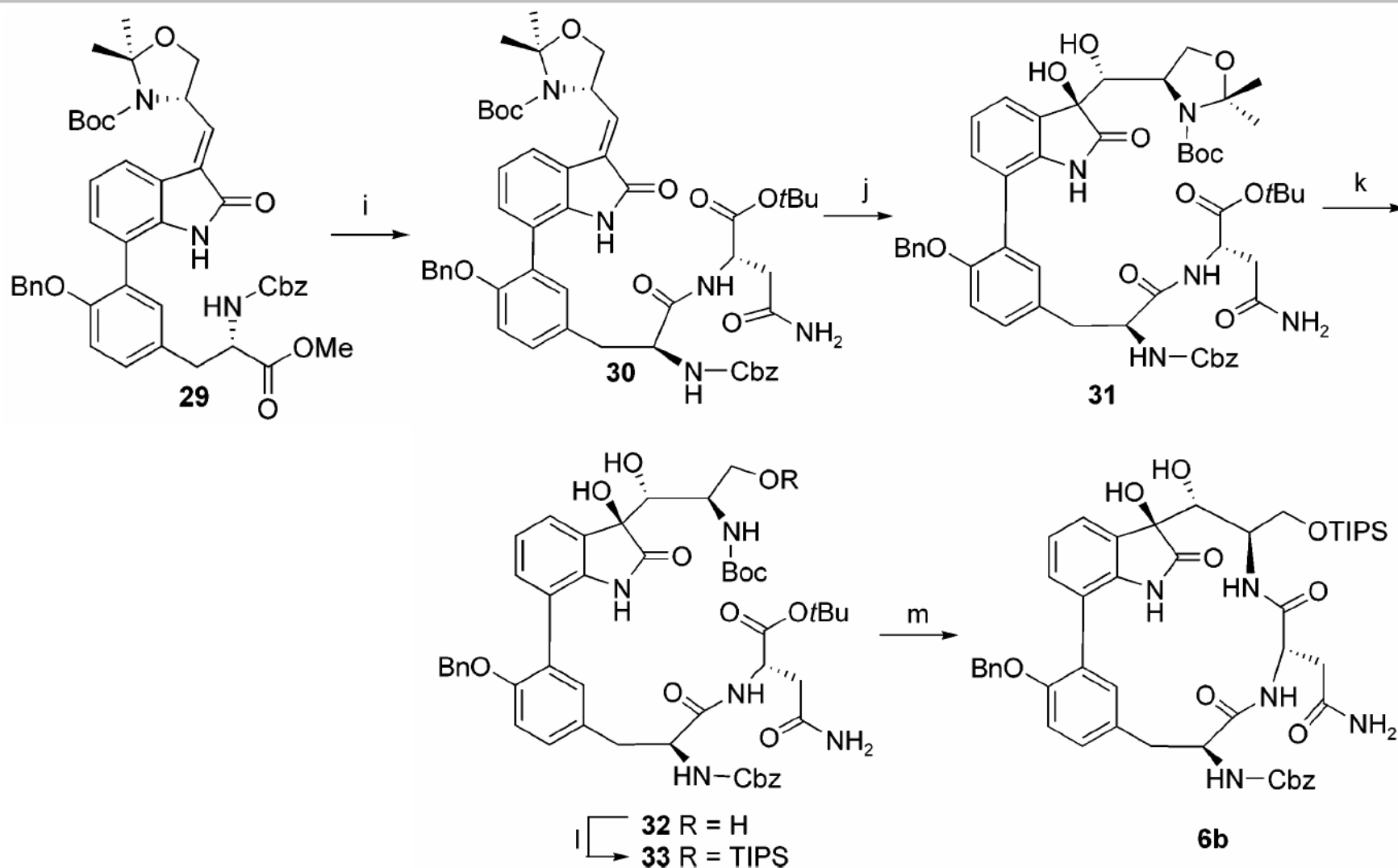
(c) $\text{MeOH}/\text{SO}_2\text{Cl}_2$; (d) CbzCl/ K_2CO_3 ; (e) BnBr, Cs_2CO_3 , acetone, reflux, 88% (three steps);

(f) $\text{Ag}_2\text{SO}_4/\text{I}_2$, MeOH, rt, 1 h, 99%;

(g) pinacolatodiborane, $[\text{PdCl}_2(\text{dppf})]\text{CH}_2\text{Cl}_2$, K_2CO_3 , DME, 80°C , 10 h, 91%;

(h) *(E)*-**2c**, $[\text{PdCl}_2(\text{dppf})]\text{CH}_2\text{Cl}_2$, K_2CO_3 , DME, 80°C , 2 h, 75%;

Synthesis of Macrolactam **6b**



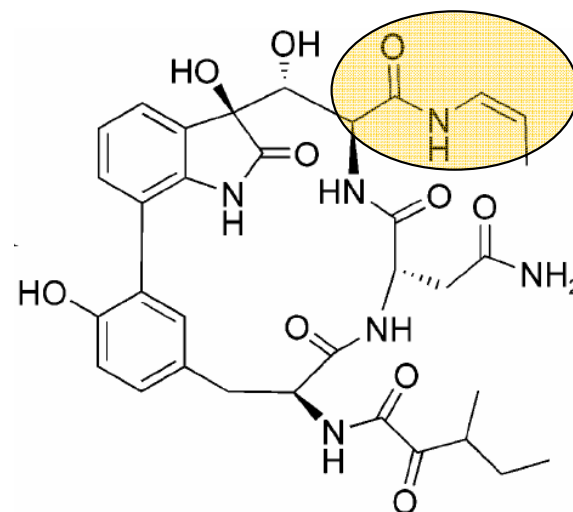
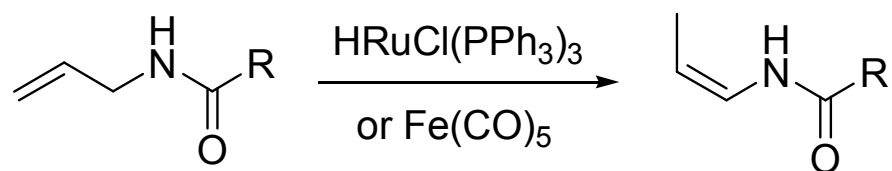
- (i) (1) LiOH, THF/H₂O, 0°C, 1.5 h, (2) H-Asn-O-*t*-Bu, EDC/HOAT, THF, rt, 2 h, 85% (two steps);
 (j) OsO₄/NMO, (DHQD)₂PHAL, *t*-BuOH/H₂O, rt, 12 h, 88% (dr = 5:1);
 (k) (1) PPTS/MeOH, reflux, 2 h; (l) TIPSCl, imidazole/DMAP, CH₂Cl₂, rt, 5 h, 88% (two steps);
 (m) (1) TFA/CH₂Cl₂ (4:1), rt, 2 h, (2) EDC/HOAT/DIEA, CH₂Cl₂/DMF (2 mM), rt, 24 h, 36%.

Danishefsky, S. et al. *J. Am. Chem. Soc.* **2004**, *126*, 6347.

Reported *cis*-Enamide Formations

◆ Isomerization of N-allylamides.

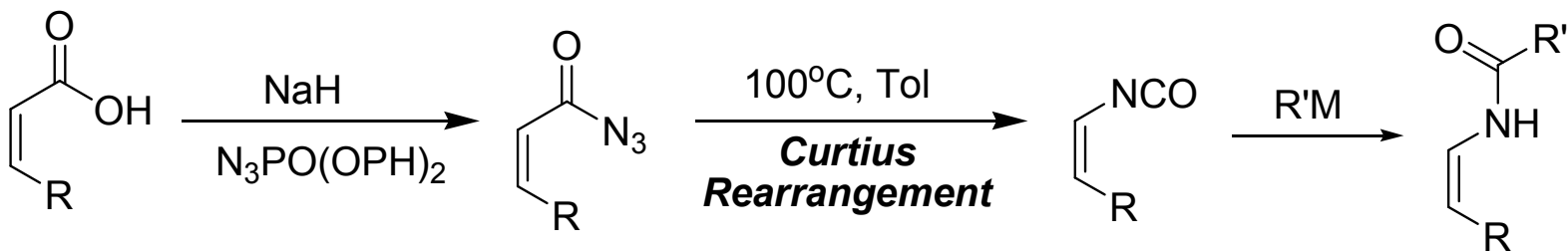
Stille, J. K.; Becker, Y. *J. Org. Chem.* **1980**, *45*, 2139.



TMC-95A&B (1a,b)

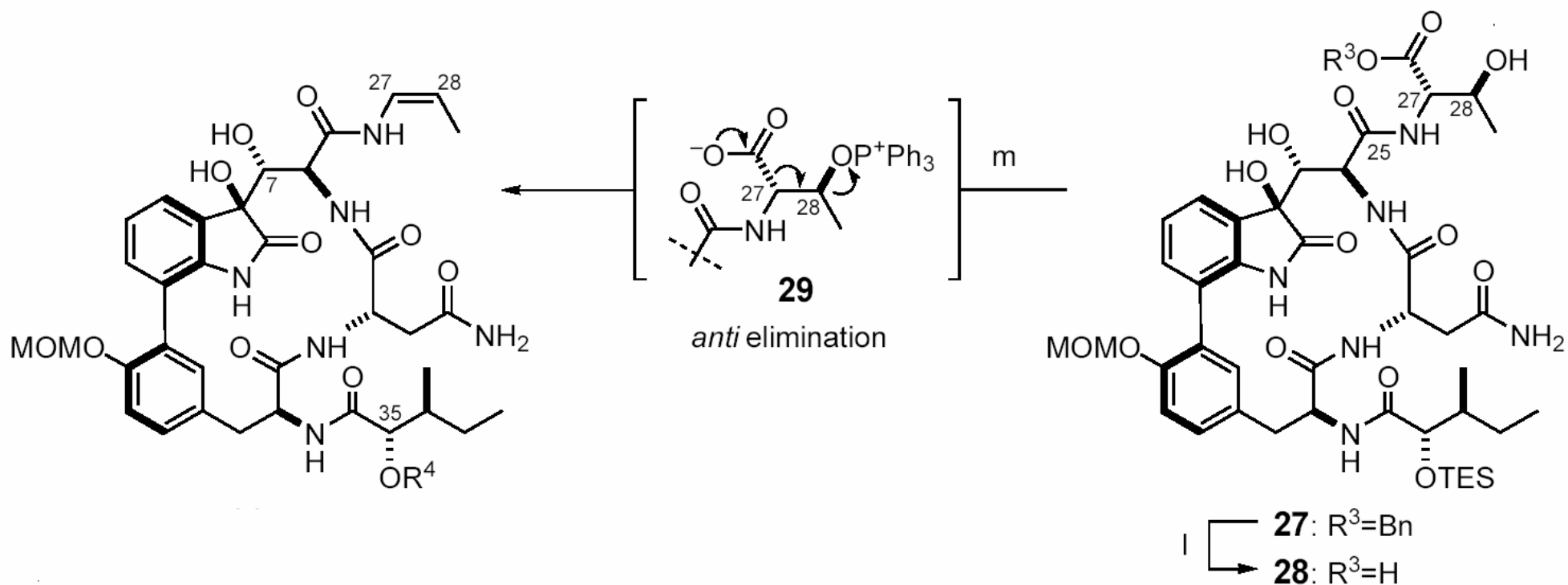
◆ Curtius rearrangement.

Kitahara, T. et al. *Synlett.* **2000**, *3*, 397.



Reported *cis*-Enamide Formations

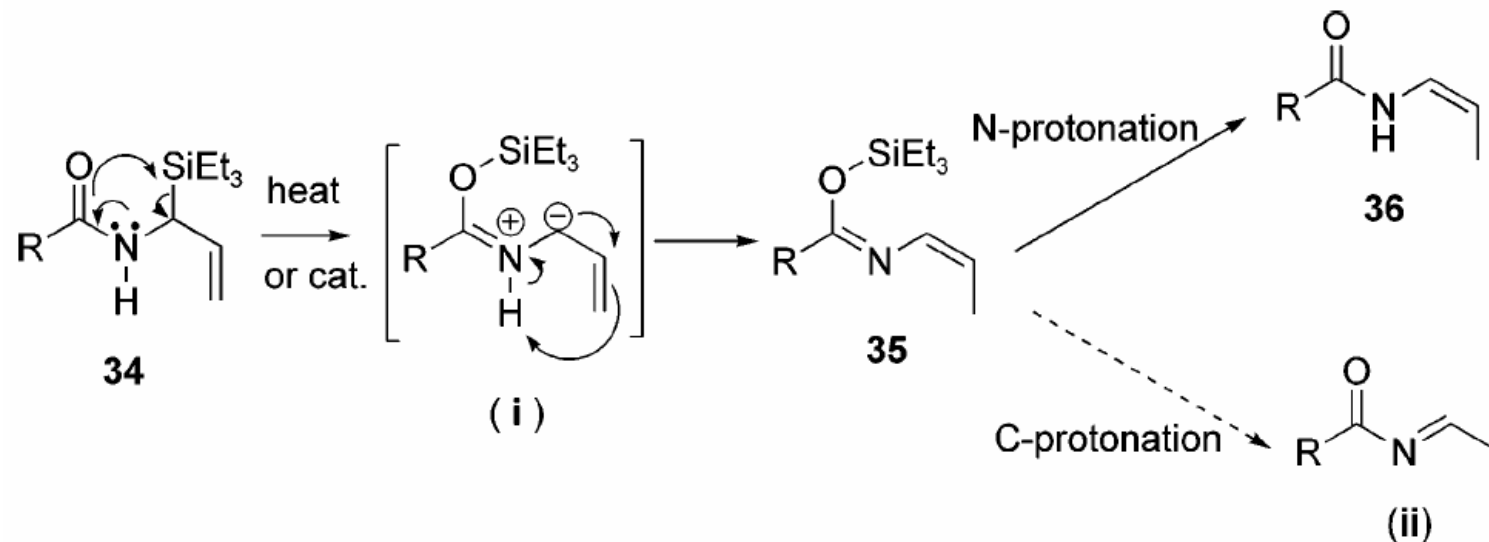
- ◆ *Anti*-elimination of good leaving group.



l) H₂, Pd(OH)₂/C, THF/H₂O (1:2);

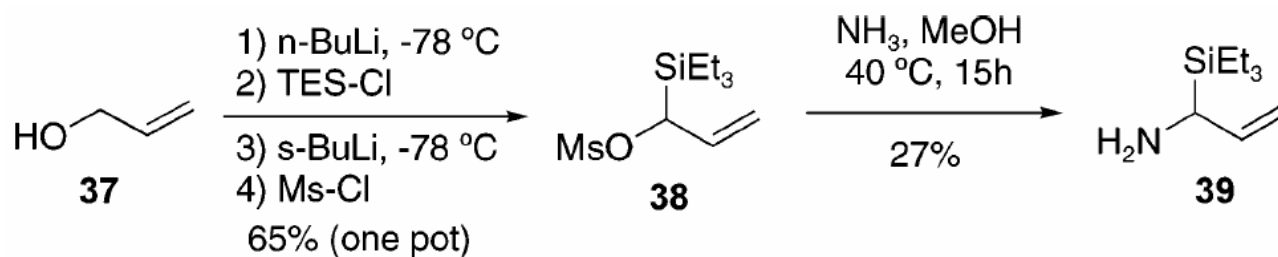
m) DEAD, PPh₃, 4-A MS, 0°C to room temperature, 59% from **27**;

Proposed Strategie for *cis*-Enamide Formations

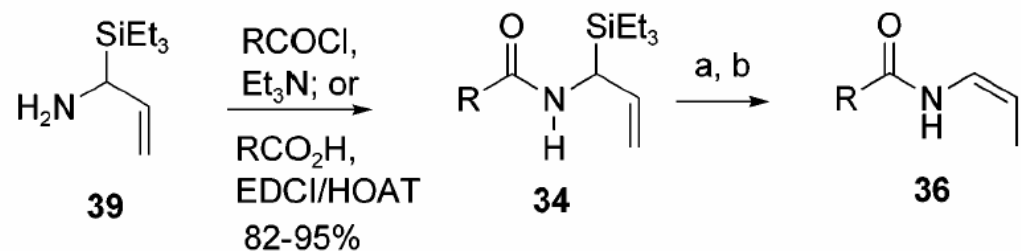


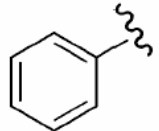
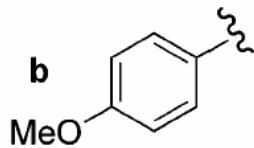
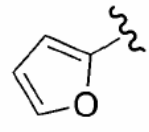
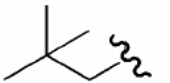
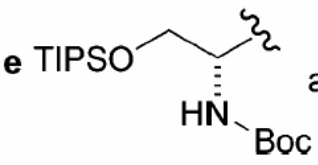
- ◆ It was proposed that **34** might undergo concurrent ene- and silatropic-like bond reorganizations that would lead to **35**.
- ◆ Key step is the transfer of TES from C to O, N-protonation of **35** necessary.

◆ Synthesis of α -silyl Allylamine

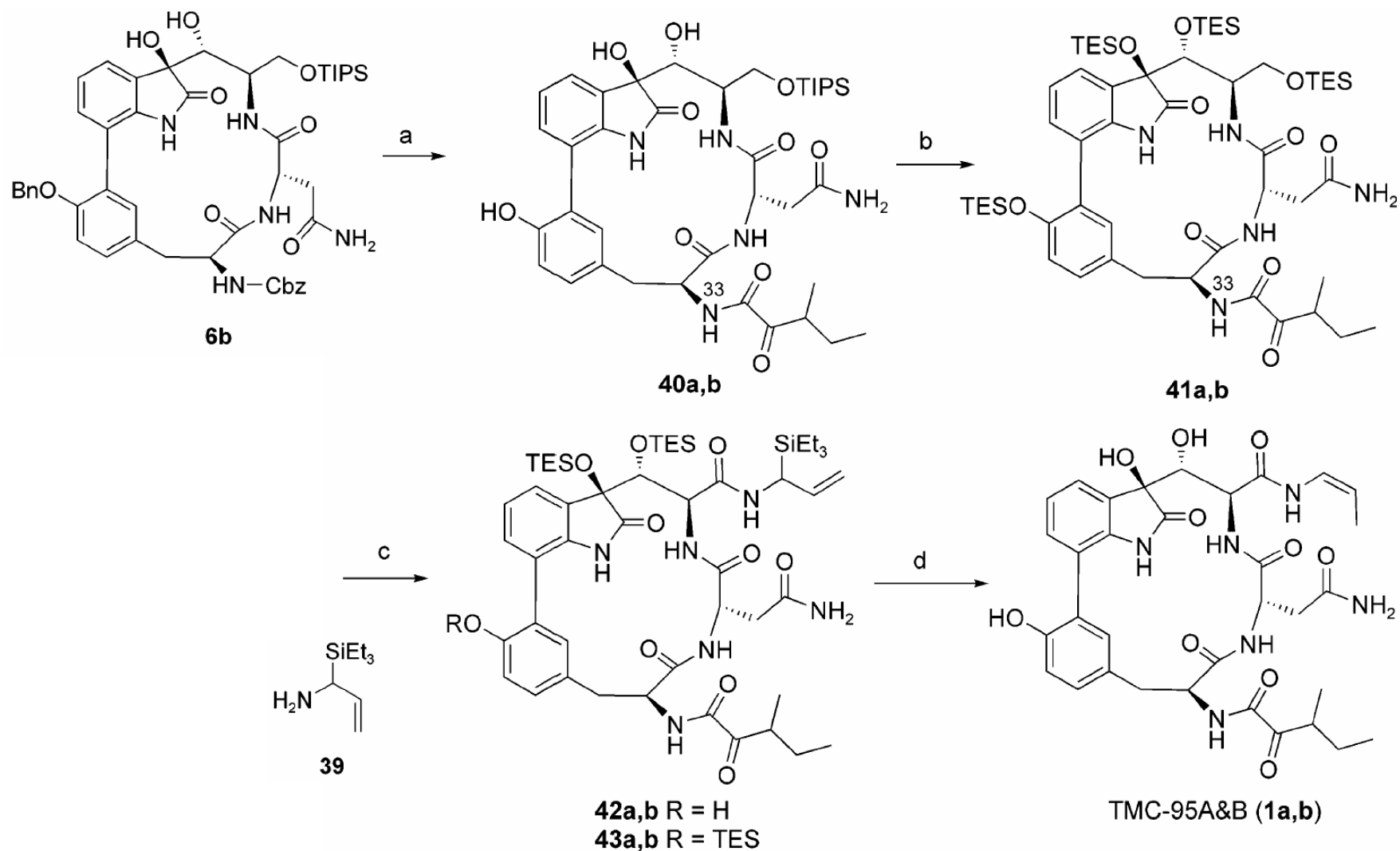


Rearrangement-Hydrolysis of α -Silylallyl Amide **34**



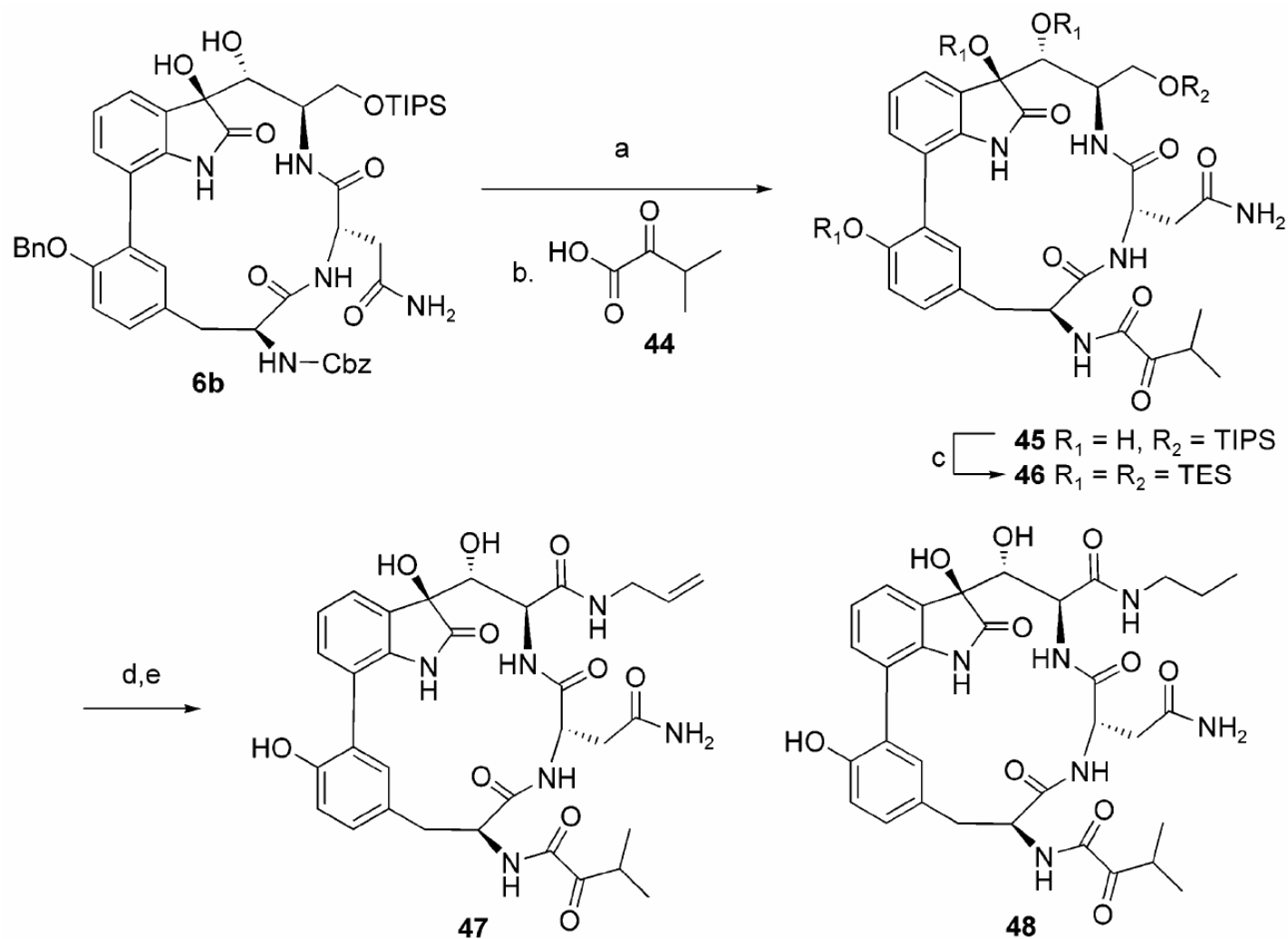
Entry	R =	Conditions	Yield
1		a. toluene, 110 °C, 10 h; b. H_2O	81%
2		a. toluene, 110 °C, 20 h; b. H_2O	73%
3		a. toluene, 110 °C, 27 h; b. H_2O	67%
4		a. toluene, 110 °C, 3 d; b. H_2O	72%
5		a. o-xylene, 110 °C, 4 d; b. H_2O	52%

Completion of the Synthesis of TMC-95A/B



- (a) (1) Pd/C, H₂, EtOH rt, 19 h, (2) (±)-3-methyl-2-oxopentanoic acid (**5**), EDC/HOAT, CH₂Cl₂/DMF, rt, 2 h, 85% (two steps);
- (b) (1) HF/Py, (2) TESOTf, 2,6-lutidine, CH₂Cl₂, 0°C to rt, 15 h, (3) NaHCO₃, (4) citric acid, EtOAc/H₂O, 73%
- (c) (1) Jones reagent, acetone, 0°C, 2 h, (2) **39**, EDC/HOAT, CH₂Cl₂/DMF, rt, 13 h, 45% (two steps);
- (d) (1) *o*-xylene, 140°C, 3 d, (2) HF/py, THF/py, then Me₃SiOMe, 49% (two steps).

Synthesis of Analogues



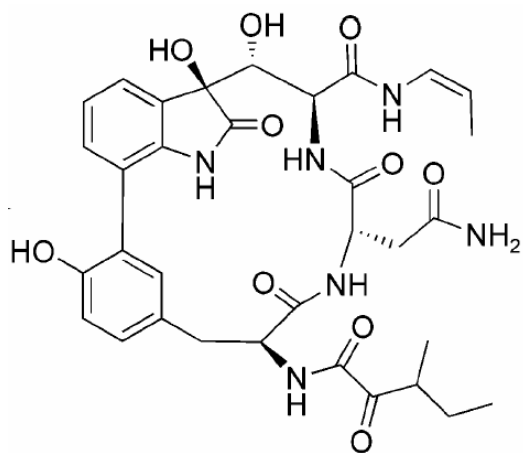
- (a) Pd/C, H₂, EtOH rt, 19 h; (b) **44**, EDC/HOAT, CH₂Cl₂/DMF, rt, 2 h, 57% (two steps);
(c) (1) HF/Py, (2) TESOTf, 2,6-lutidine, CH₂Cl₂, 0°C to rt, 15 h, (3) NaHCO₃, (4) citric acid, EtOAc/H₂O, 56%
(d) (1) Jones reagent, acetone, 0°C, 2 h, (2) allylamine or *n*-propylamine, EDC/HOAT, CH₂Cl₂/DMF, rt, 13 h;
(e) HF/py, THF/py; then Me₃SiOMe, 39% for **47**, 32% for **48** (three steps).

Proteasome Inhibition Studies

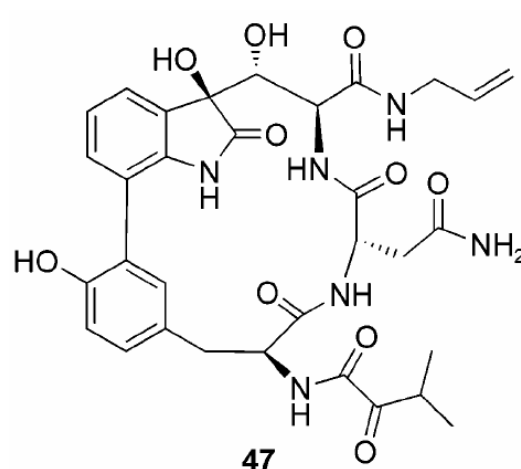
Table 5. Inhibition Constants (K_{iapp}) of Catalytic Activities of the Proteasome by Synthetic Inhibitors^a

	$K_{iapp} = [I]/(v_0/v_s - 1)^b$		
	CT-L activity (nM)	PGPH activity (nM)	TL activity (μ M)
TMC-95A (1a)	1.1	29	0.8
TMC-95B (1b)	1.7	23	1.1
47	1.9	23	1.2
48	24	110	13

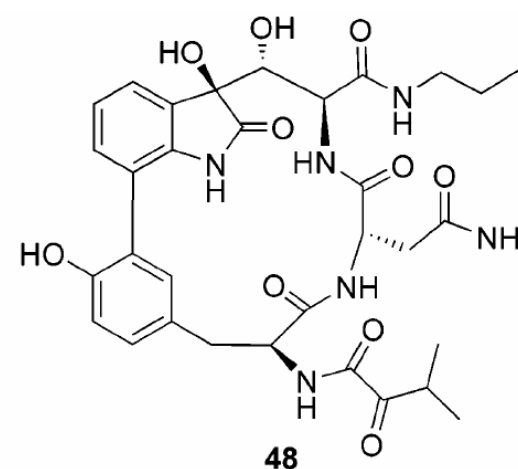
^a The concentrations required for inhibition of the three proteasome catalytic activities were determined for TMC-95A and -B and their synthetic analogues. ^b The value v_0 is the rate of enzyme activity in the absence of inhibitors, and v_s is the steady rate of inhibited enzyme activities.



TMC-95A&B (**1a,b**)

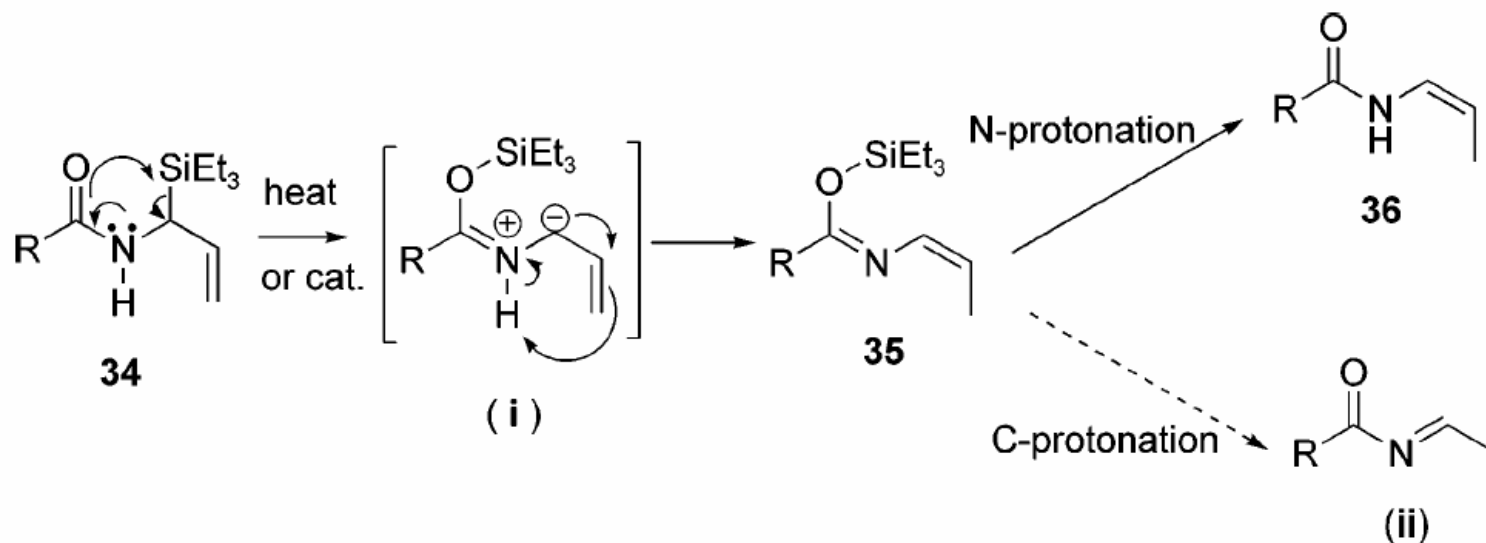


47



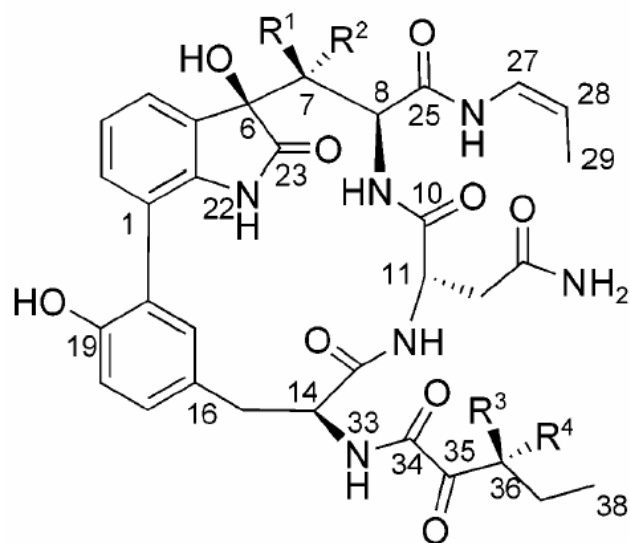
48

On the Mechanism of *cis*-Enamide Formation



- ◆ Silyl imidates **35** was observed by $^1\text{H-NMR}$ analysis.
- ◆ Density functional calculations of the intermediates and TS suggested a stepwise dyotropic rearrangement mechanism involving sequential 1,4-silyl (fast, reversible) and 1,4-hydrogen shifts (rate determining)
- ◆ This mechanism explained well why reactions are much slower when R = alkyl.

Summary



TMC-95	R ¹	R ²	R ³	R ⁴
A (1a)	H	OH	Me	H
B (1b)	H	OH	H	Me

A Total Synthesis

A New Methodology

A Simplified Analogue

More Compounds

More Simplified Structure