
Reductive Enolate Generation From Enones - Application in C-C bond Formation

1. Cauble, D, F.; Gipson, J, D.; Krische, M, J. *J. Am. Chem. Soc.* **2003**, 125, 1110.
2. Huddleston, R, R.; Krische, M, J. *Org. Lett.* **2003**, 5, 1143.
3. Jang, H, Y.; Huddleston, R, R.; Krische, M, J. *J. Am. Chem. Soc.* **2002**, 124, 15156.

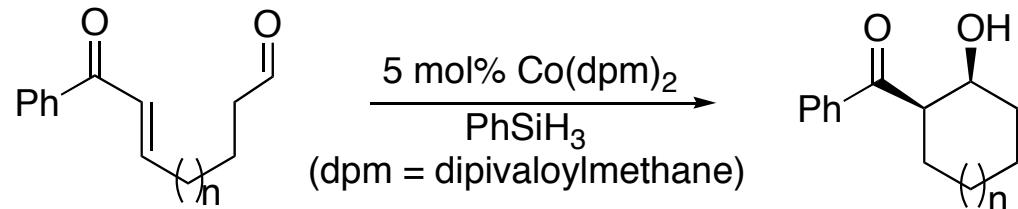
Group Presentation

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Nov 20, 2003

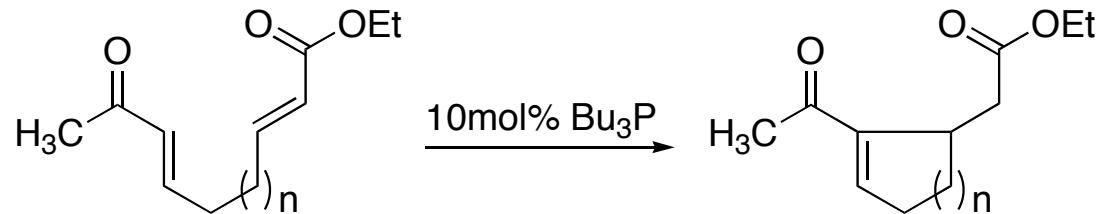
Catalytic Nucleophilic Activation of Enones

- Hydrometallative nucleophilic activation via conjugate reduction



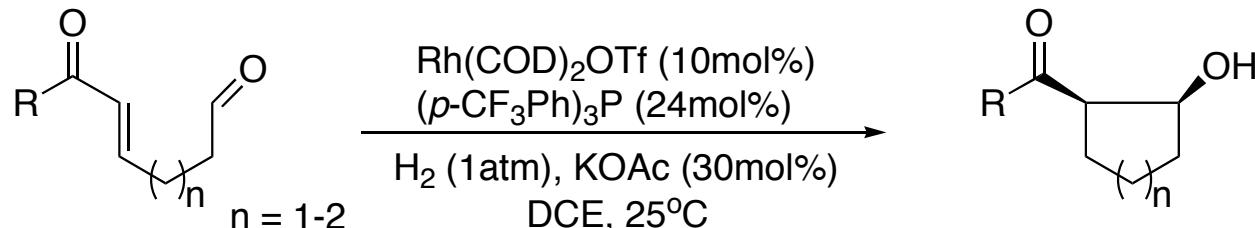
* Co, Ni, Rh catalyst systems, silane, borane, alane, and stannane, H_2 as reductants

- Nucleophilic organocatalysis via reversible conjugate addition

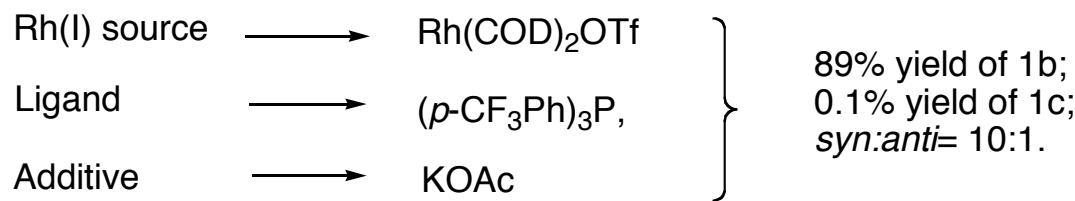
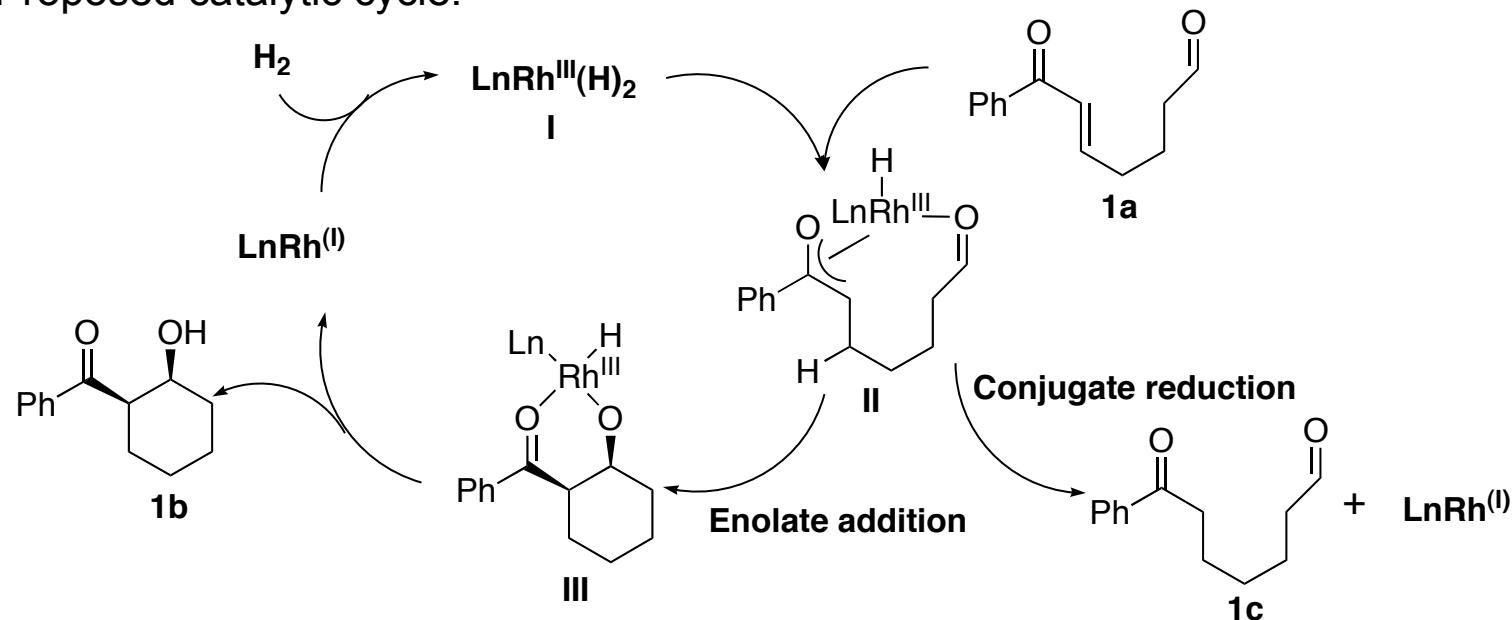


* Intramolecular Baylis-Hillman reaction, Intramolecular Rauhut-Currier reaction

Catalytic C-C bond Formation of Monoenone and Monoaldehyde



➤ Proposed catalytic cycle:



Catalytic C-C bond Formation of Monoenone and Monoaldehyde

➤ Hydrogenative aldol cycloreduction of aromatic, heteroaromatic, and aliphatic enone substrates to form 5- and 6-membered rings.

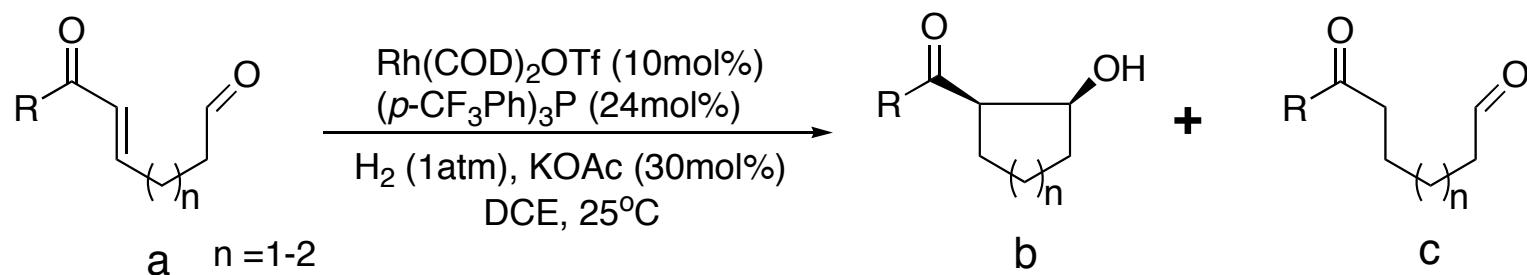


Table 1. Rh-Catalyzed Hydrogenative Aldol Cycloreduction of Monoenone Monoaldehydes **1a-7a**

substrate	product (syn:anti)	1,4-reduction
1a , $n = 2$, R = Ph	1b , 89% (10:1)	1c , 0.1%
2a , $n = 2$, R = <i>p</i> -MeOPh	2b , 74% (5:1)	2c , 3%
3a , $n = 2$, R = 2-naphthyl	3b , 90% (10:1)	3c , 1%
4a , $n = 2$, R = 2-thiophenyl	4b , 76% (19:1)	4c , 2%
5a , $n = 2$, R = 2-furyl	5b , 70% (6:1)	5c , 10%
6a , $n = 1$, R = Ph	6b , 71% (24:1)	6c , 1%
7a , $n = 2$, R = CH_3	7b , 65% (1:5)	

Intermolecular Hydrogenative Aldol Condensation

➤ Phenyl vinyl ketone as prenucleophile.

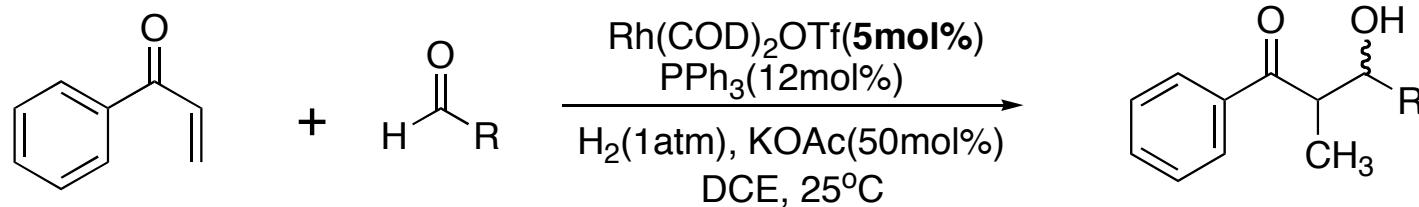
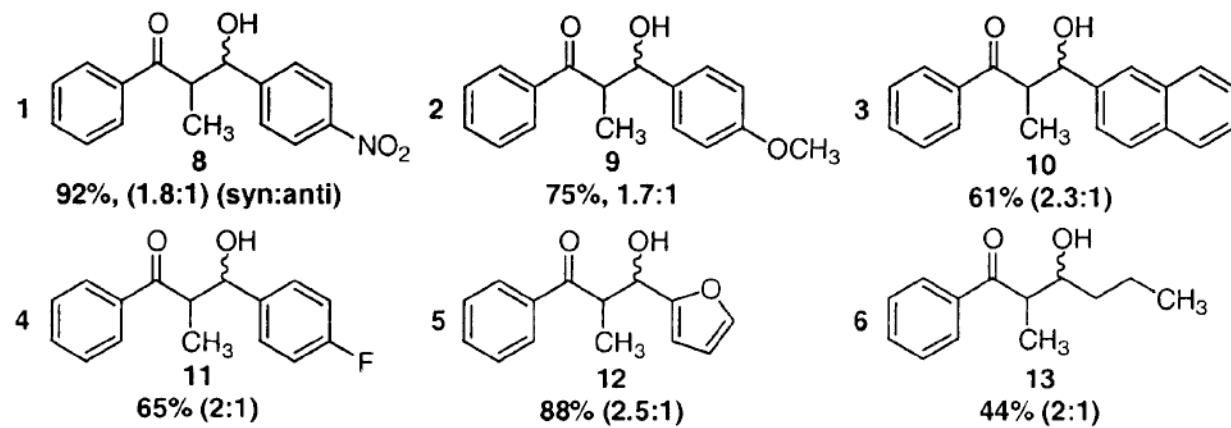


Table 2. Intermolecular Rh-catalyzed hydrogenative aldol condensation of phenyl vinyl ketone and various aldehydes



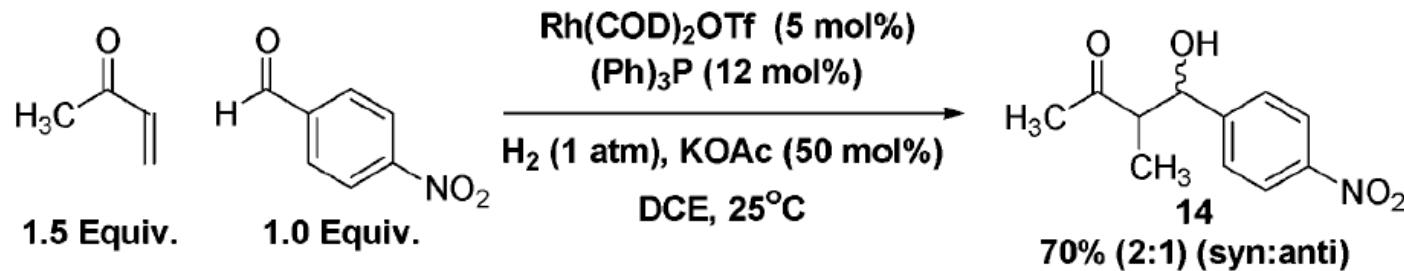
* Changes of conditions:

Intramolecular: enone: 100mol%, catalyst: 10mol%, conc: 0.5mol/L.

Intermolecular: enone: 150mol%, catalyst: 5mol%, conc: 0.1mol/L.

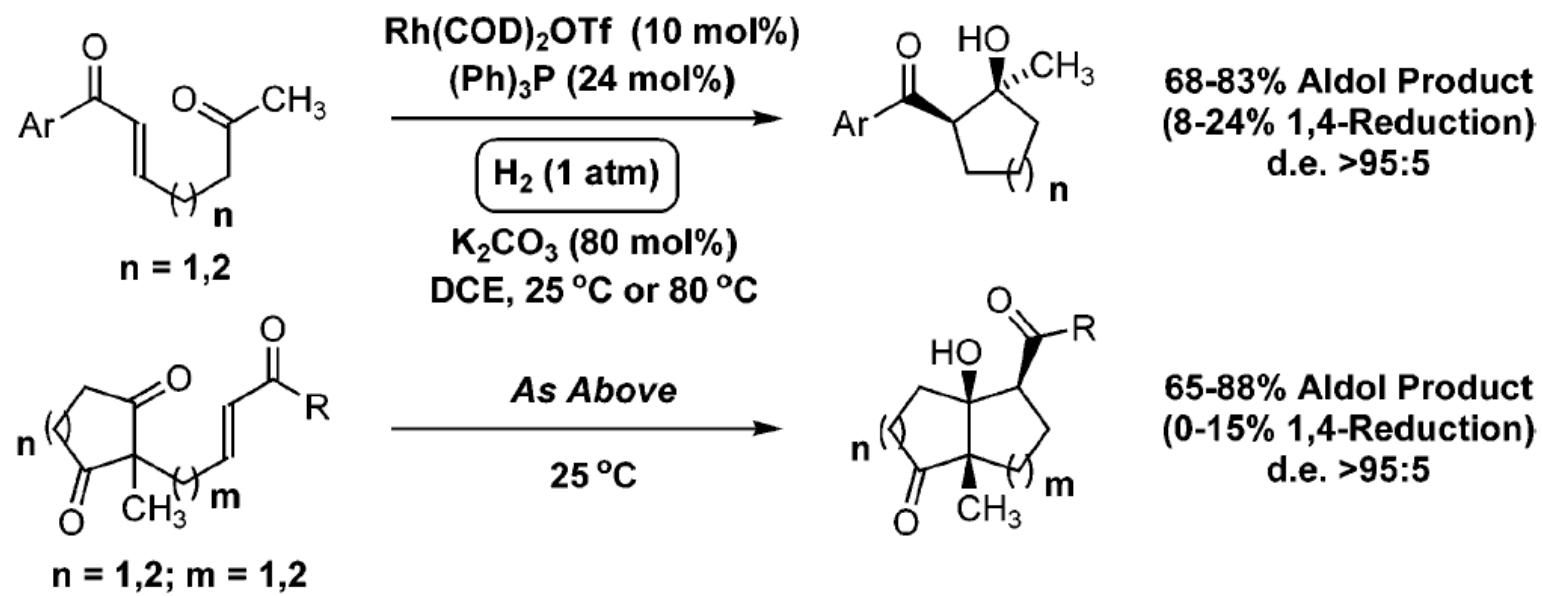
Variation of the Nucleophilic Partners

- Propiophenone doesn't result in aldolization;
- Ethyl acrylate exclusively give 1,4-reduction product;
- Methyl vinyl ketone works.



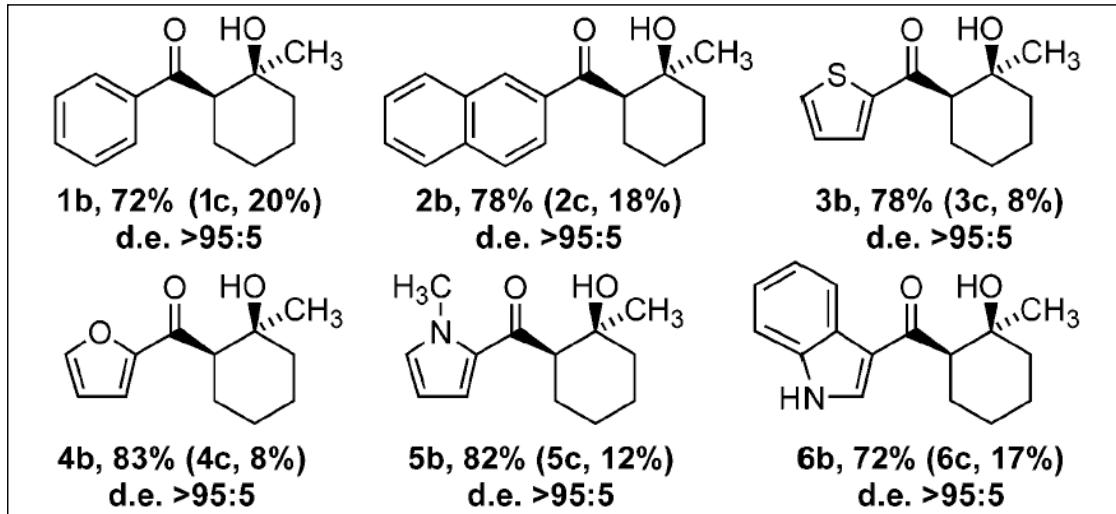
Catalytic Hydrogenative Cyclo reduction of Keto-enones

➤ Ketone and dione partners as electrophiles

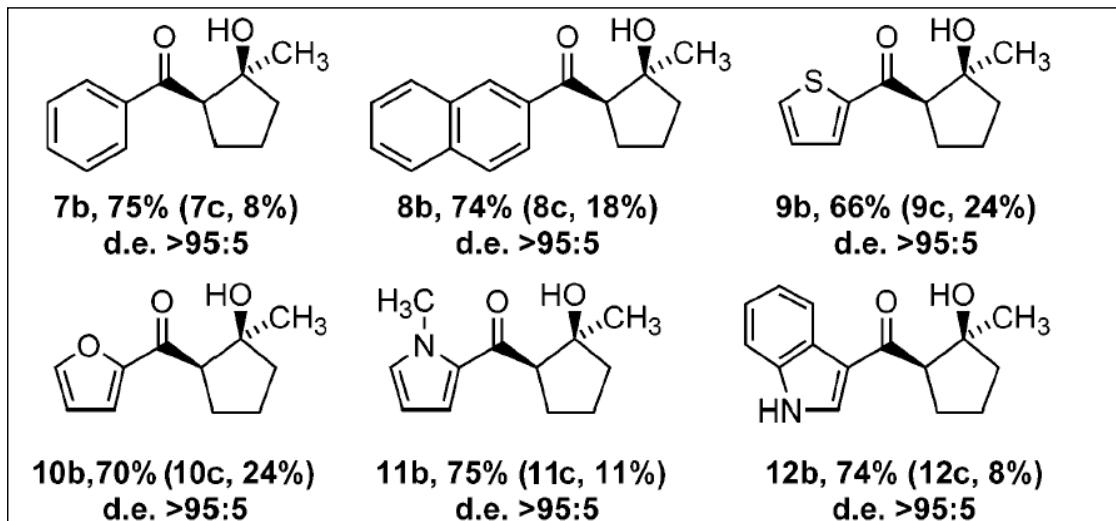


Catalytic Hydrogenative Cyclo reduction of Keto-enones

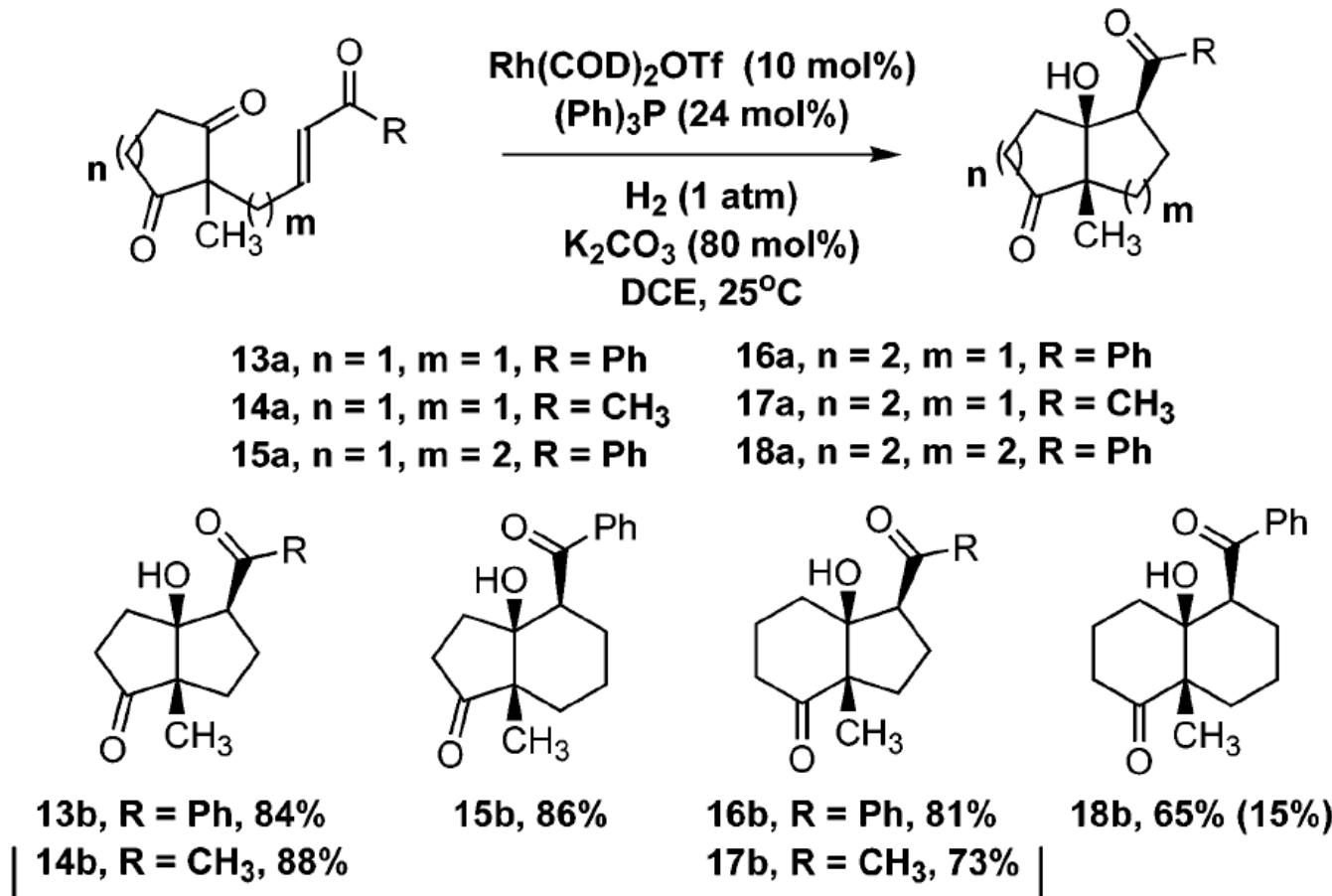
$n = 2$



$n = 1$



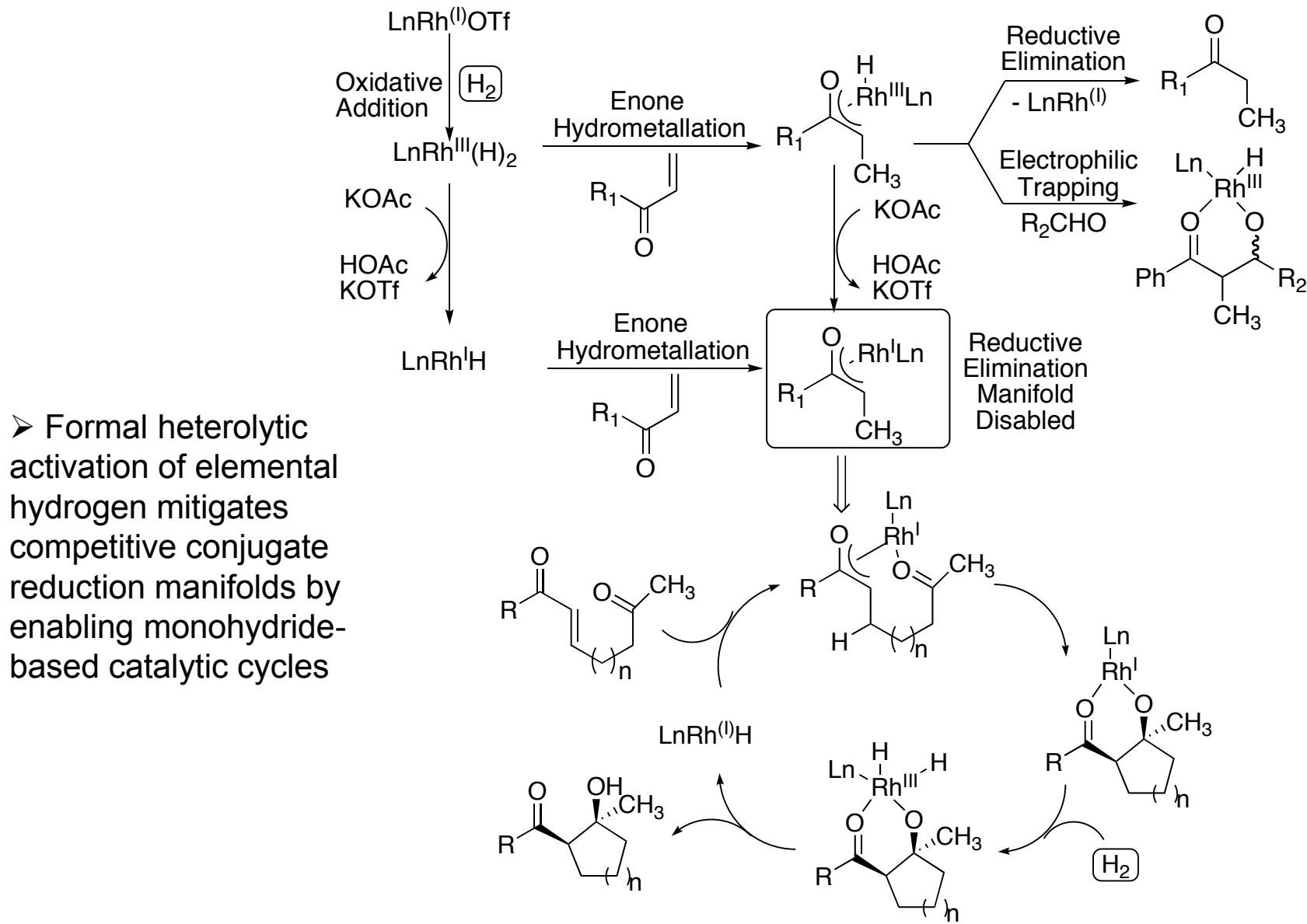
Cycloreduction of Dione-enones



No Conjugate Reduction
d.e. >95:5

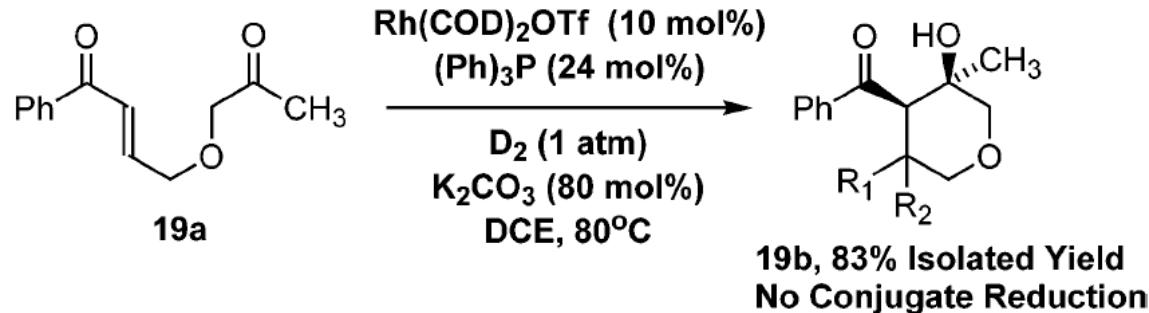
➤ Diastereoselective formation of 3-contiguous stereogenic centers, including 2-contiguous quaternary centers.

Monohydride-based Catalytic Cycle



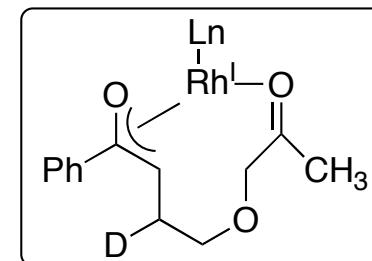
Mechanism Study on the Conjugate Reduction

➤ Catalytic cycloreduction employing elemental deuterium

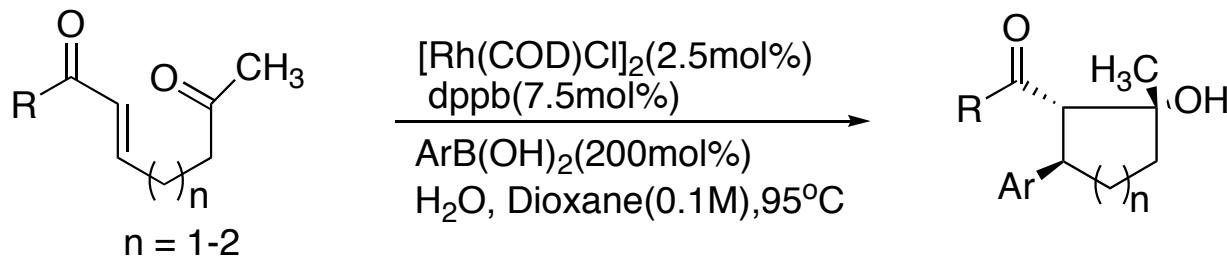


$R_1 = R_2 = H, 11\% \pm 5\%$
 $R_1 = D, R_2 = H, 81\% \pm 5\%$
 $R_1 = R_2 = D, 8\% \pm 5\%$

The enone hydrometalation is reversible.
Such as β -hydride elimination of Rh-enolate.



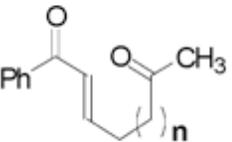
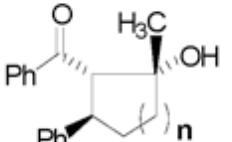
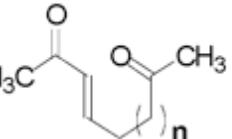
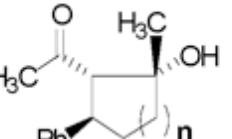
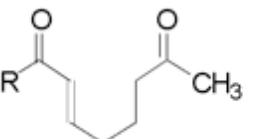
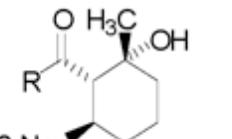
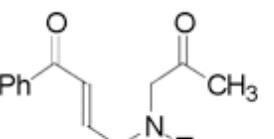
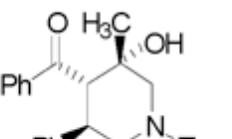
Carbometallative Aldol Cyclo reduction: Tandem Conjugate Addition Aldol Cyclization



- * Rh-catalyzed conjugate addition is performed:
 1. Aqueous organic media \Rightarrow adding water (5 equiv with respect to substrate)
 2. Avoid the addition of arylboronic acids to aldehyde \Rightarrow methyl ketones
- * Three contiguous stereogenic centers are created in one manipulation with high relative and absolute stereochemical control. A single diastereomer was obtained, no epimeric material detected.
- * The stereochemical assignment was corroborated by X-ray analysis.

Carbometallative Aldol Cycloreduction

Table . Catalytic diastereoselective carbometallative aldol cycloreduction

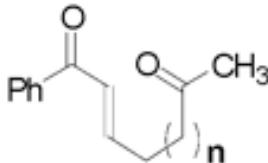
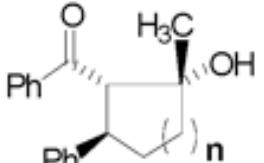
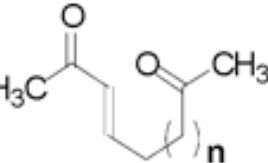
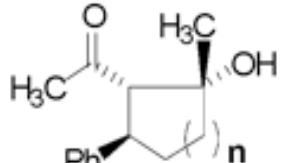
Entry	Substrate	Product	Isolated Yield
1	 $1a, n = 1$ $2a, n = 2$	 $1b$ $2b$	73% 87%
2	 $3a, n = 1$ $4a, n = 2$	 $3b$ $4b$	75% 45%
3	 $4a, R = \text{CH}_3$ $2a, R = \text{Ph}$	 $4c$ $2c$	40% 70%
4	 $5a$	 $5b$	84%

Catalytic Enantioselective Carbometallative Aldol Cyclo reduction

➤ Optimization of the enantioselective carbometallative cyclo reduction of **2a**

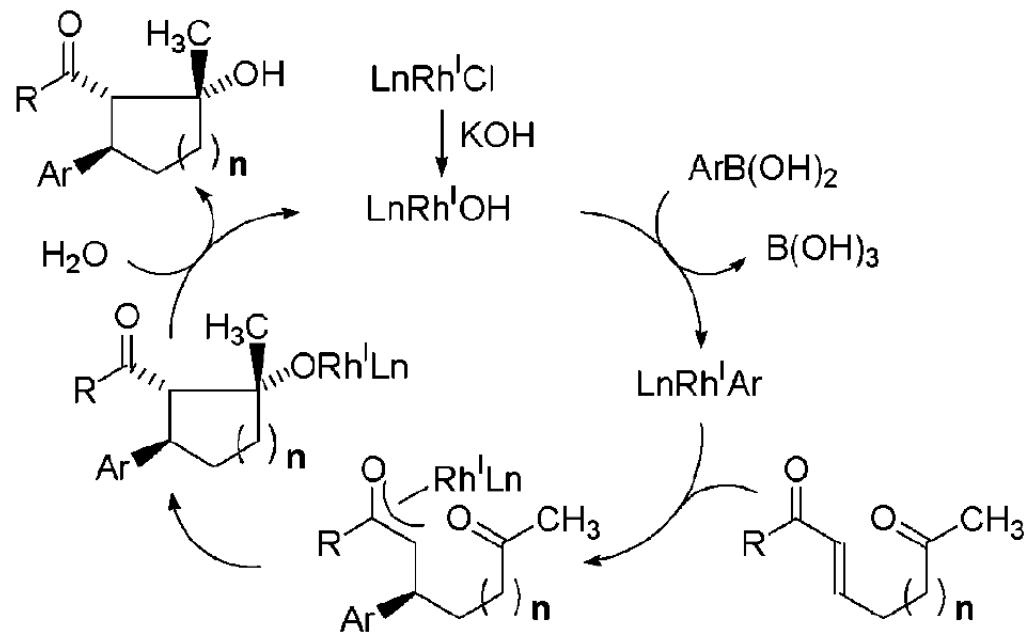
Ligand	Yield (ee %)
(R,S)-Josiphos	57% (0)
(R,R)-MeDuphos	67% (0)
(R)-Phanephos	79% (5)
(R)-Tol-BINAP	94% (62)
(R)-BINAP	90% (77)
(R)-BINAP ^c	80% (87)
(R)-BINAP ^{d,e}	88% (88)

Table . Catalytic enantioselective carbometallative aldol cyclo reduction

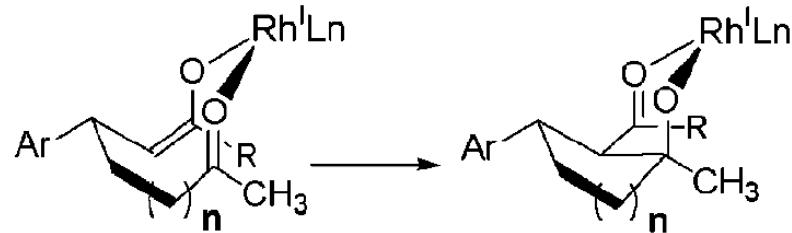
Entry	Substrate	Product	Isolated Yield (ee%)
1	 1a, n = 1 2a, n = 2	 1b 2b	78% (77) 88% (88)
2	 3a, n = 1 4a, n = 2	 3b 4b	88% (94) 69% (95)

Proposed Mechanism

➤ Proposed catalytic cycle



➤ Model of Z-enolate and Zimmerman-Traxler type Transition state



Summary

- A catalytic C-C bond formation under hydrogenative conditions via selectively generate and transform transition metal enolates.
 - A mild and economical hydrometallative method which circumvents 1,4-reduction byproduct.
 - Aldehyde, ketone, and dione used as electrophiles in aldol cycloreduction to form 5- or 6- membered rings.
 - The ability of creating three contiguous stereogenic centers in a single manipulation with high relative and absolute stereochemical control.
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