



Asymmetric [3+2] Cycloaddition of Azomethine Ylides

-- From Chiral Starting Material to Asymmetric Catalysis --

Chen, C.; Li, X.; Schreiber, S. L. *J. Am. Chem. Soc.* **2003**, *125*, 10174

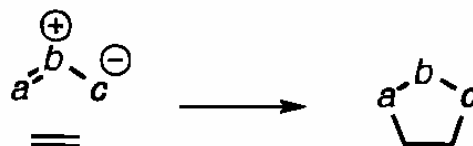
Gothelf, A. S.; Gothelf, K. V.; Hazell, Jørgensen, K. A. *Angew. Chem. Int. Ed.* **2002**, *41*, 4236

Longmire, J. M.; Wang, B.; Zhang, X. *J. Am. Chem. Soc.* **2002**, *124*, 13400

For an early review of asymmetric 1,3-dipolar cycloaddition reactions, see:
Gothelf, K. V.; Jørgensen, K. A. *Chem. Rev.* **1998**, *98*, 863

Introduction to 1,3-Dipolar Cycloaddition

- ◆ 1,3-Dipolar cycloaddition reactions:

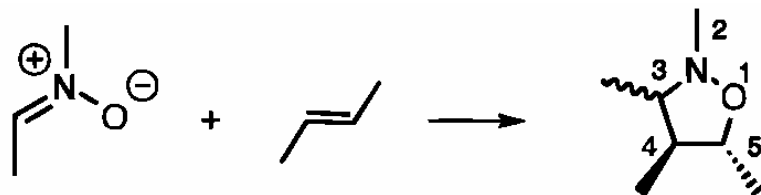


- ◆ History of 1,3-dipoles and 1,3-dipolar cycloaddition reactions:

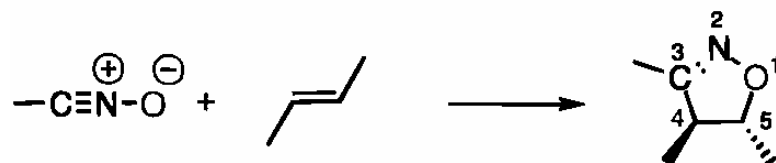
- Curtius discovered diazoacetic ester (1883)
- Buchner studied the reaction of diazoacetic ester with α,β -unsaturated esters and described the first 1,3-DC reaction. (1888)
- Huisgen established general application of 1,3-dipoles in organic synthesis (1960s)
- Woodward and Hoffmann developed conservation of orbital symmetry
- Newest challenge: Control of Stereochemistry

Examples of Typical 1,3-Dipolar Cycloadditions

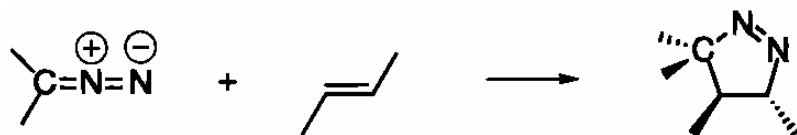
- ◆ 1,3-Dipolar Cycloaddition of Nitrones – Preparation of isoxazolidines



- ◆ 1,3-Dipolar Cycloaddition of Nitrile Oxides – Preparation of 2-isoxazolines

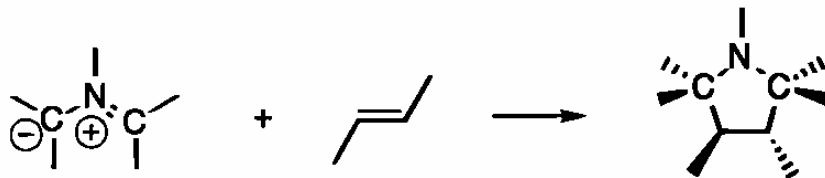


- ◆ 1,3-Dipolar Cycloaddition of Diazoalkanes – Preparation of 1-pyrazoline



1,3-Dipolar Cycloadditions of Azomethine Ylides

◆ 1,3-Dipolar Cycloaddition of Azomethine Ylides



- Reaction forms highly substituted heterocycle.
- Product can be used as new catalyst, or serve as important motifs in biologically active molecules.
- Azomethine ylides are unstable. Generation of azomethine ylides include proton abstraction from imine derivatives, thermolysis or photolysis of aziridines, dehydrohalogenation of imonium salts.

How to perform the reaction
in **asymmetric** fashion ?

↓
Chiral
Azomethine Ylides

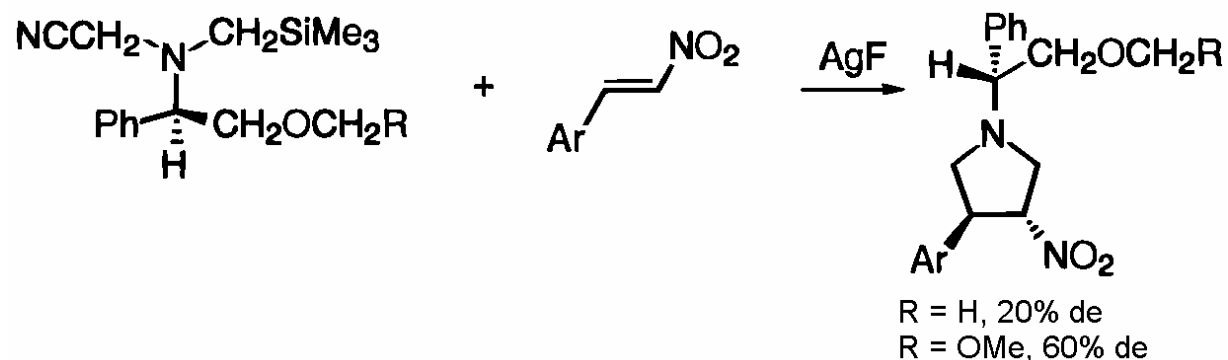
↓
Chiral
Alkenes

↓
Intramolecular
Reactions

↓
Metal-Catalyzed
Reactions

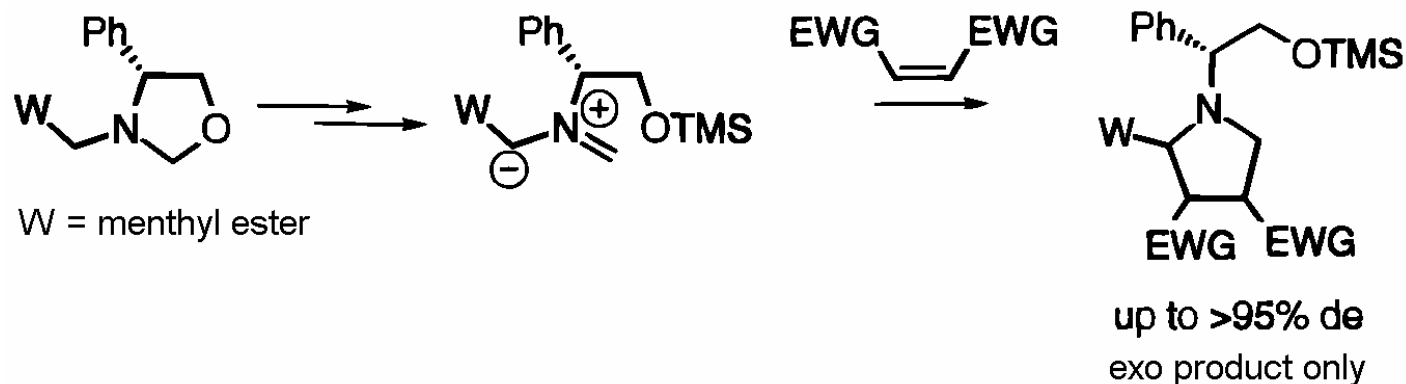
1,3-DC of Azomethine Ylides: Chiral Azomethine Ylides

- ◆ First example of diastereo-facial selective 1,3-DC reaction of chiral azomethine ylides.



Padwa, A.; Chen, Y.-Y.; Chiacchio, U.; Dent, W. *Tetrahedron*. **1985**, *41*, 3529.

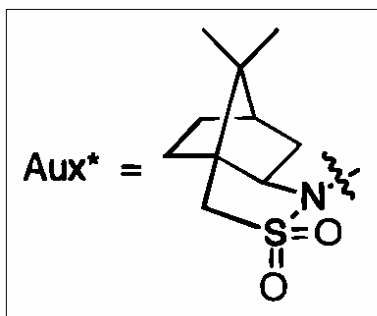
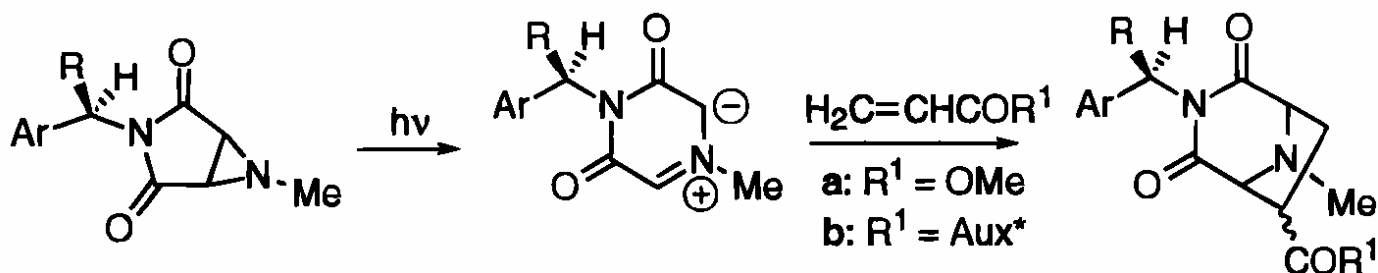
- ◆ Royer's chiral azomethine ylide system:



Deprez, P.; Rouden, J.; Chiaroni, A.; Riche, C.; Royer, J.; Husson, H.-P.
Tetrahedron Lett. **1991**, *32*, 7531.

1,3-DC of Azomethine Ylides: Chiral Azomethine Ylides

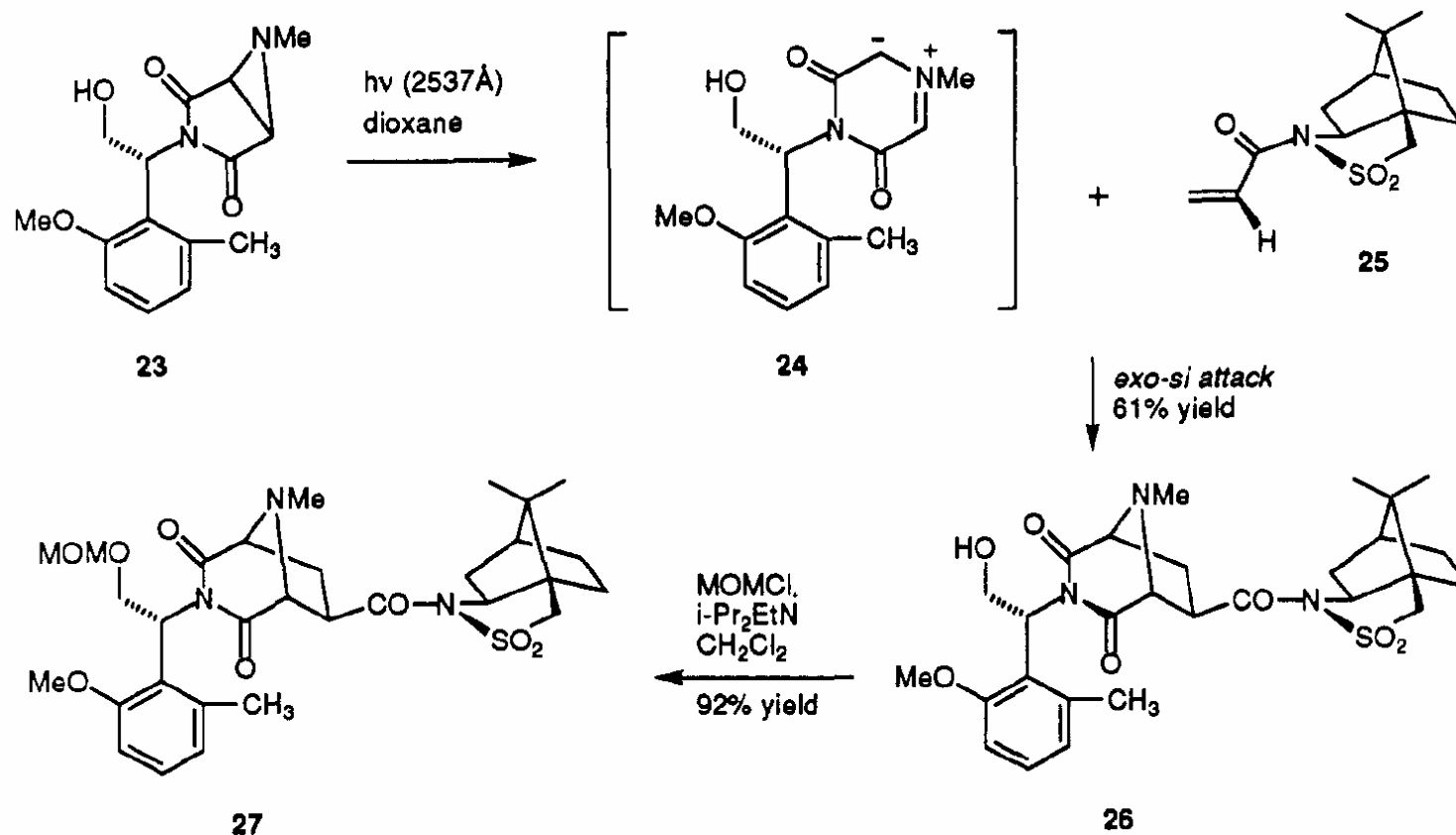
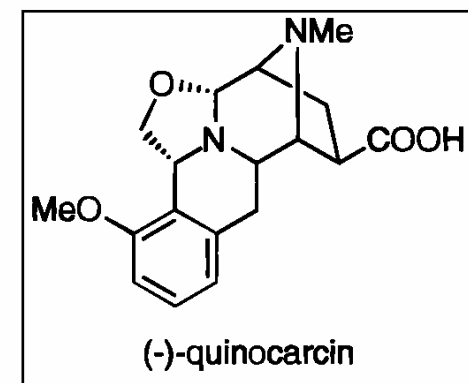
- ◆ Azomethine ylides derived from chiral aziridines.



a: $\text{R}^1 = \text{Me}$, *endo:exo* = 1:5,
low de
b: $\text{R}^1 = \text{Aux}^*$, *endo:exo* = 0:100,
de >90%

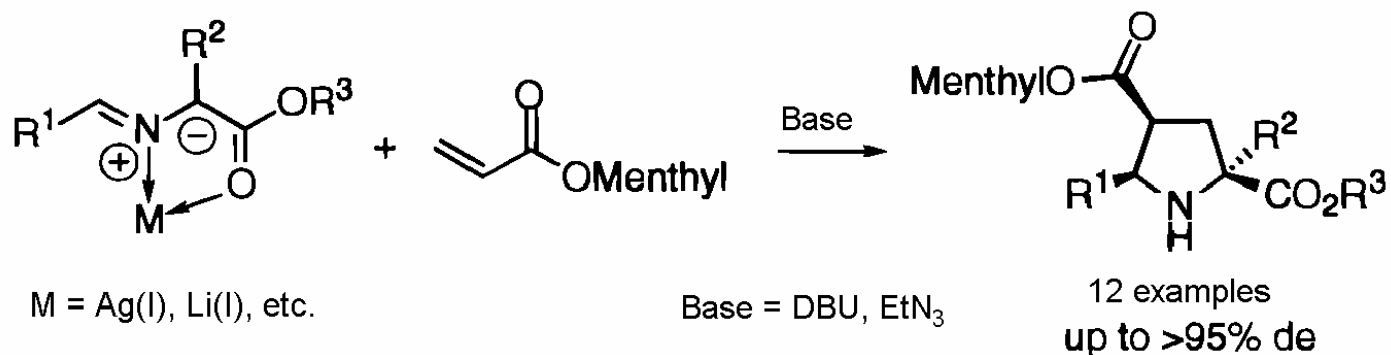
1,3-DC of Azomethine Ylides: Chiral Azomethine Ylides

- ◆ Azomethine ylides derived from chiral aziridines.
 - Application in the total synthesis of (-)-quinocarcin

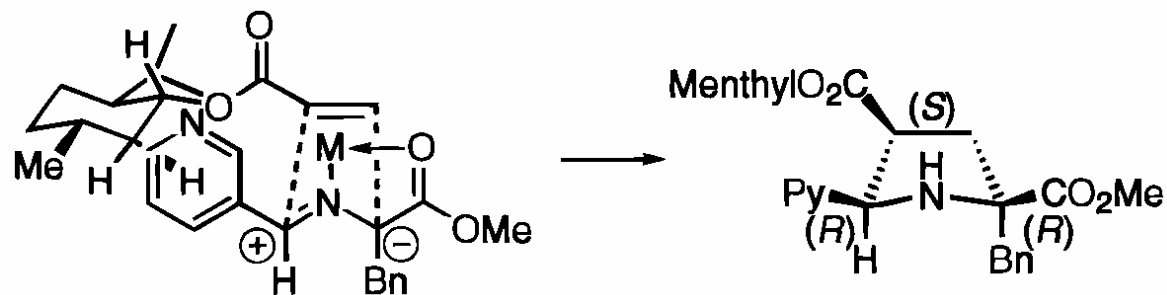


1,3-DC of Azomethine Ylides: Chiral Alkenes

- ◆ Cycloaddition using chiral alkenes: Grigg's system.



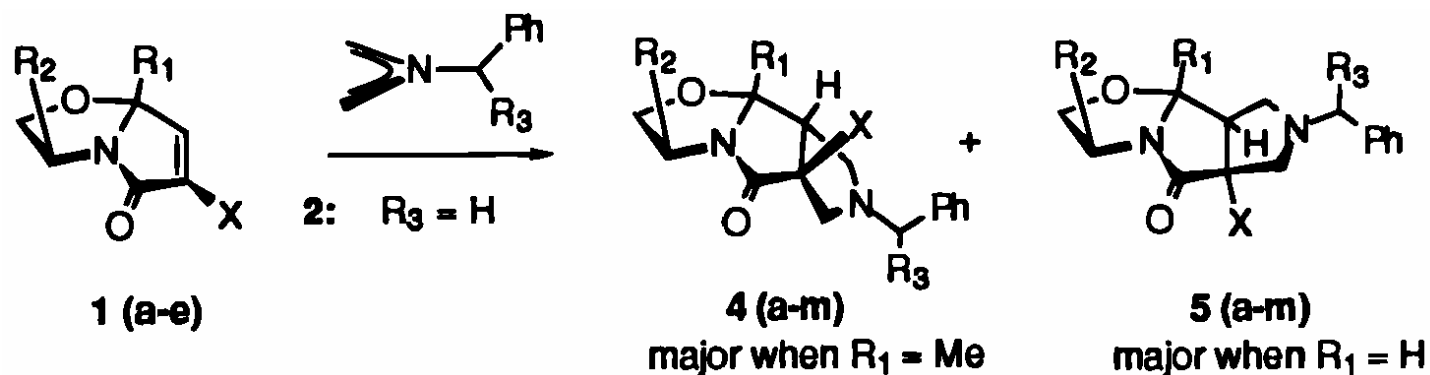
- ◆ Proposed transition state model based on regioselectivities, endo and diastereofacial- selectivities



Barr, D. A.; Dorrity, M. J.; Grigg, R.; Hargreaves, S.; Malone, J. F.; Montgomery, J.; Redpath, J.; Stevenson, P.; Thornton-Pett, M. *Tetrahedron* **1995**, *51*, 273.

1,3-DC of Azomethine Ylides: Chiral Alkenes

◆ Cycloaddition using chiral alkenes: Meyers' system.

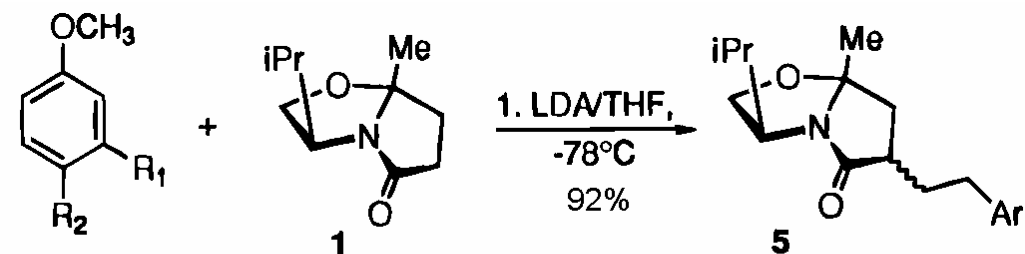


entry	lactam 1			lactam	ylide 2
	R ₂	R ₁	X		(A)- 4:5 ^a
1	<i>i</i> -Pr	Me	H	1a	91:9
2	Ph	Me	H	1b	94:6
3	Ph	H	H	1c	17:83
4	<i>i</i> -Pr	Me	CO ₂ Me	1d	71:29
5	<i>i</i> -Pr	Me	CO ₂ <i>t</i> -Bu	1e	72:28
6	<i>i</i> -Pr	Ph	CO ₂ Me	1m	74:26

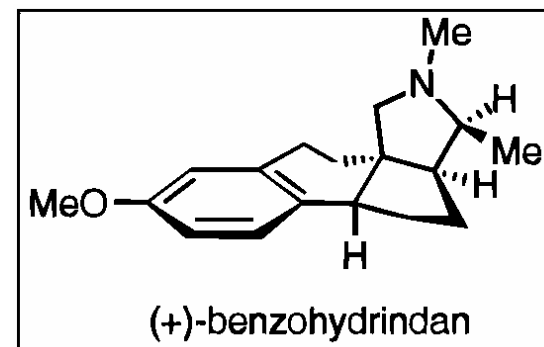
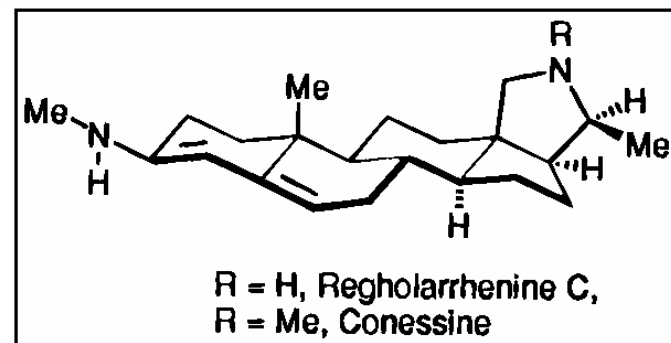
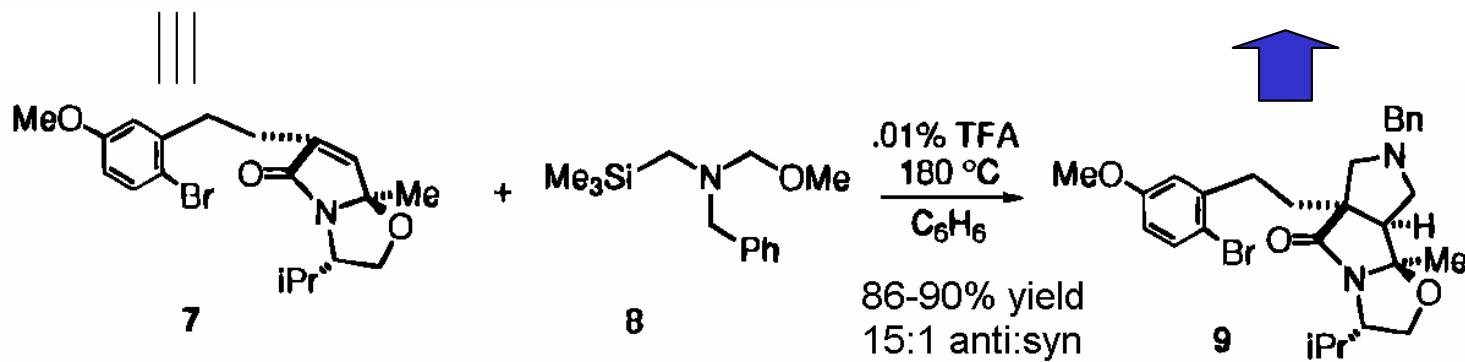
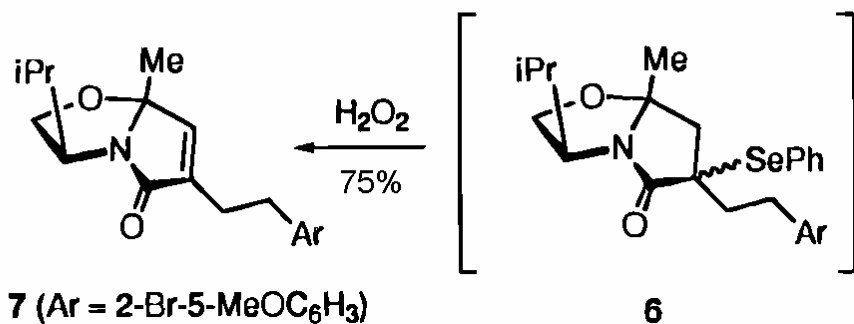
Fray, A. H.; Meyers, A. I. *J. Org. Chem.* **1996**, *61*, 3362.

1,3-DC of Azomethine Ylides: Chiral Alkenes

◆ Application in the total synthesis of benzohydrindan.

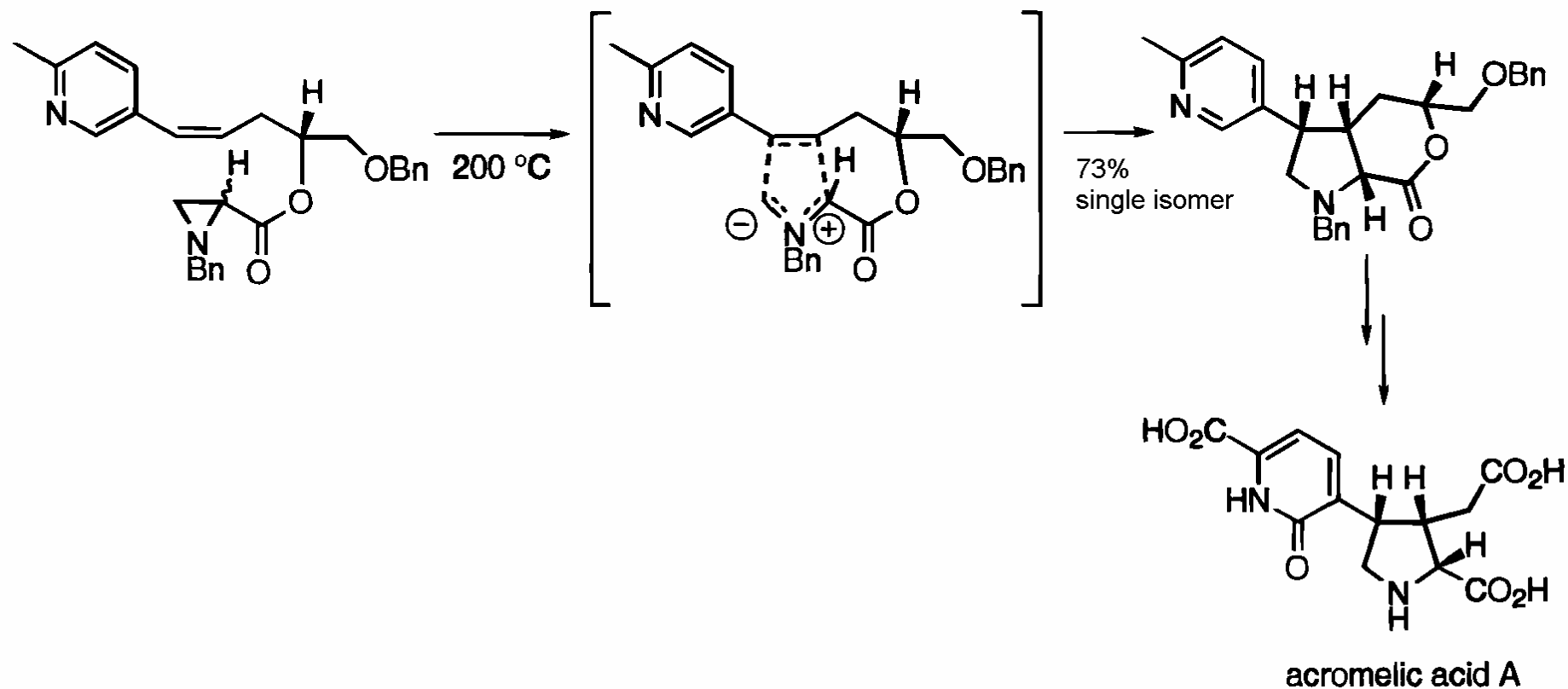


- 2 R₁ = CH₂CH₂OH, R₂ = H
- 3 R₁ = CH₂CH₂I, R₂ = H
- 4 R₁ = CH₂CH₂I, R₂ = Br



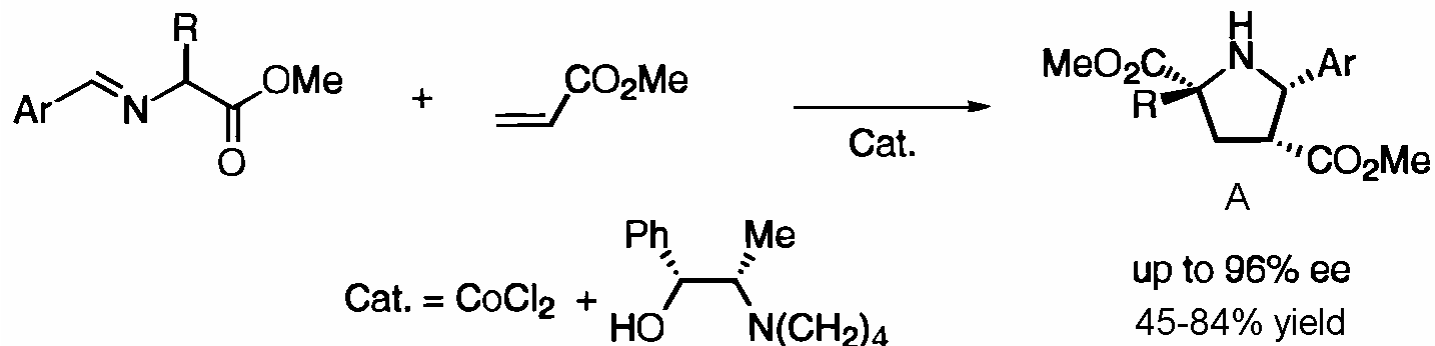
1,3-DC of Azomethine Ylides: Intramolecular Reactions

- ◆ Intramolecular cycloaddition in the application of synthesis of Acromelic acid A.



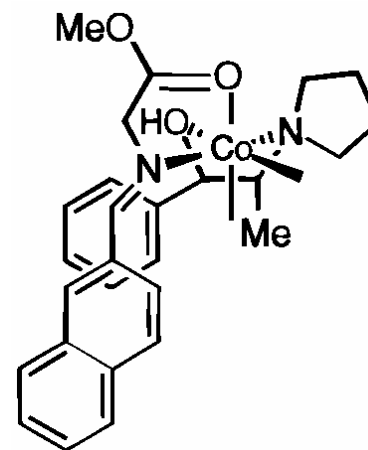
1,3-DC of Azomethine Ylides: Metal-Catalyzed Reactions

- ◆ Early studies (and the only case) on metal-catalyzed reactions.



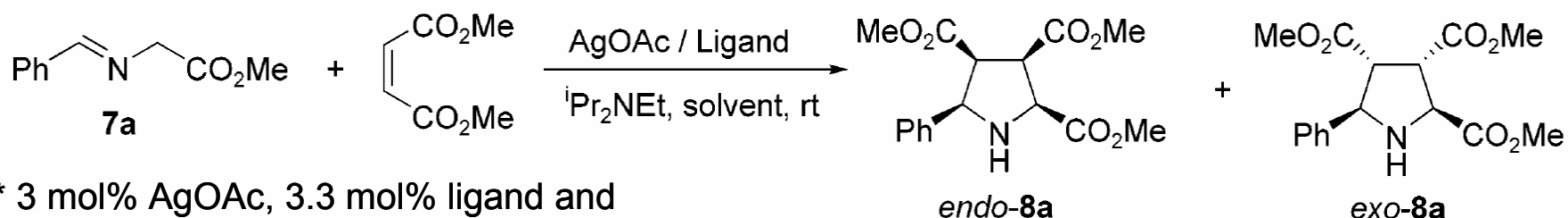
- Higher ee when Ar = 2-naphthyl, 4- BrC_6H_4 , 4- MeOC_6H_4 , and use methyl acrylate as solvent.
- Stoichiometric amount of Co(II) (1 mol) and ligand (2 mol) were used.
- Up to 70% ee was reported using AgOTf and bisphosphine ligand, details not revealed.

Proposed working model:

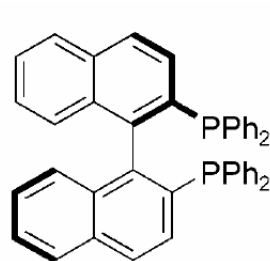


Catalytic Asymmetric 1,3-DC of Azomethine Ylides

◆ First successful substoichiometric catalyst system – Ligand Screening.

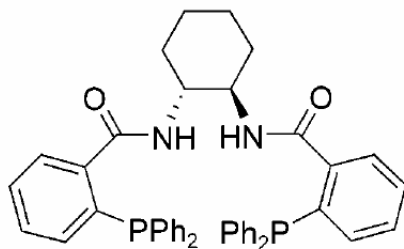


* 3 mol% AgOAc, 3.3 mol% ligand and 10 mol% amine used



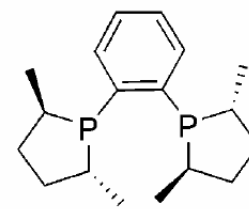
(*R*)-BINAP (1)

13% ee



(*R,R*)-Trosc Ligand (2)

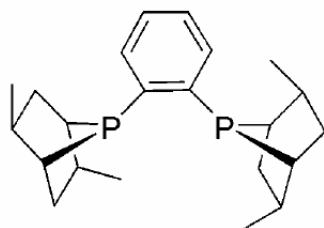
59% ee



(*R,R*)-Me-DuPhos (3)

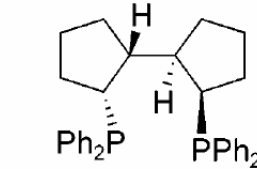
23% ee

* All ligands (except BINAP) give *endo* pdt only.



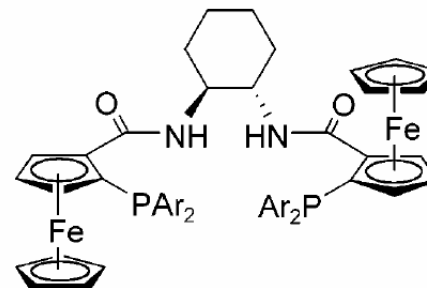
(*R,S,R,S*)-PennPhos (4)

27% ee



(*R,R,R,R*)-BICP (5)

13% ee



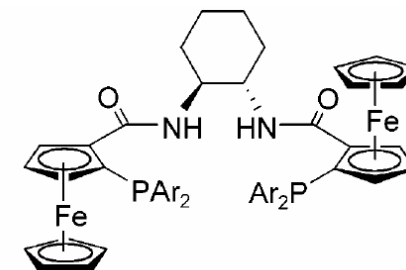
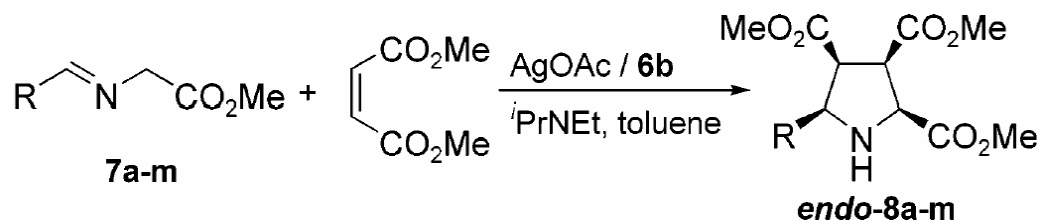
(*S,S,S_p*)-FAP: Ar = phenyl (**6a**)

(*S,S,S_p*)-xylyl-FAP: Ar = 3,5-dimethylphenyl (**6b**)

- 76% ee with **6a**;
- 86% ee with **6b**.

Catalytic Asymmetric 1,3-DC of Azomethine Ylides

◆ Variation of α -iminoester substrates.



(*S, S, S_p*)-FAP: Ar = phenyl (**6a**)
 (*S, S, S_p*)-xylyl-FAP: Ar =
 3,5-dimethylphenyl (**6b**)

entry ^a	7, R=	time (h)	<i>endo</i> - 8	
			% yield ^c	% ee ^d
1	Ph (7a)	7	87	87
2	<i>p</i> -toluyl (7b)	7	93	88
3	<i>p</i> -anisole (7c)	7	98	92
4	4-chlorophenyl (7d)	7	96	92
5	4-fluorophenyl (7e)	7	96	90
6	4-cyanophenyl (7f)	7	90	96
7	2-chlorophenyl (7g)	7	96	86
8	<i>o</i> -toluyl (7h)	7	97	90
9	1-naphthyl (7i)	7	73	85
10	2-naphthyl (7j)	14	98	97
11	3-pyridyl (7k)	7	98	84
12 ^b	<i>i</i> -Pr (7l)	48	82	70
13 ^b	cyclohexyl (7m)	48	82	81

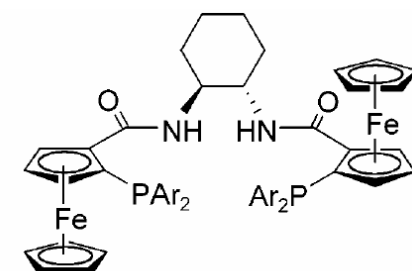
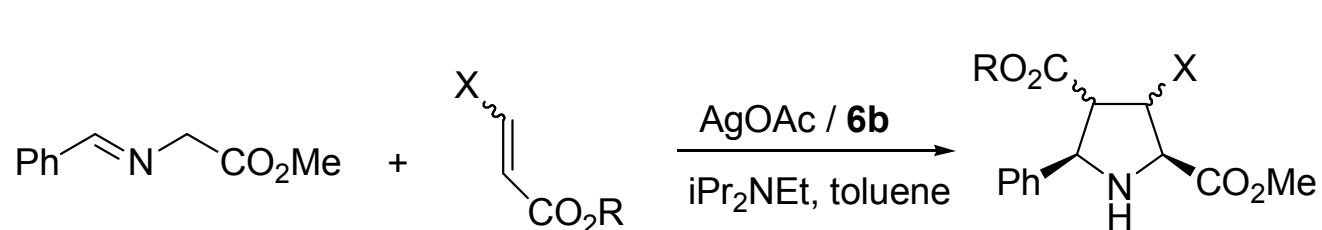
^a Conditions: imine (1.0 equiv), dimethyl maleate (1.2 equiv), AgOAc (3 mol %), ligand **6b** (3.3 mol %), *i*Pr₂NEt (10.0 mol %), toluene (3 mL) at 0°C, unless indicated otherwise.

^b Reactions were run at room temperature.

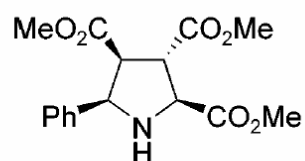
Longmire, J. M.; Wang, B., Zhang, X. *J. Am. Chem. Soc.* **2002**, *124*, 13400

Catalytic Asymmetric 1,3-DC of Azomethine Ylides

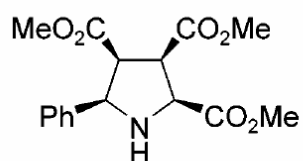
- ◆ Cycloaddition of **7a** with Various Dipolarophile Substrates Catalyzed by Ag(I)-**6b**



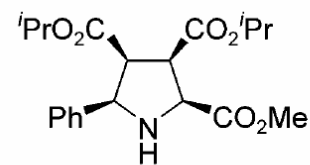
(*S,S,S_p*)-FAP: Ar = phenyl (**6a**)
 (*S,S,S_p*)-xylyl-FAP: Ar =
 3,5-dimethylphenyl (**6b**)



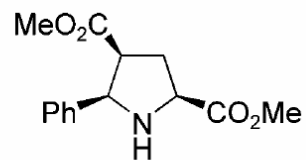
52% ee, 88% yield



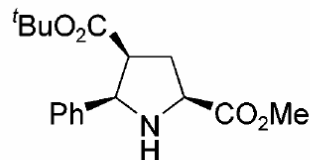
87% ee, 87% yield



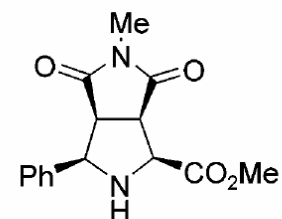
87% ee, 85% yield



60% ee, 90% yield



93% ee, 85% yield

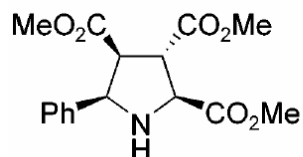


79% ee, 87% yield

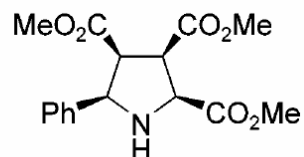
Catalytic Asymmetric 1,3-DC of Azomethine Ylides

◆ Working model of Zhang's system:

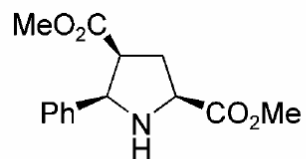
- 1) Coordination of α -iminoester to chiral Ag(I) catalyst;
- 2) Deprotonation with $i\text{Pr}_2\text{NEt}$;
- 3) Ligand blocks one enantiotopic face of azomethine ylide.



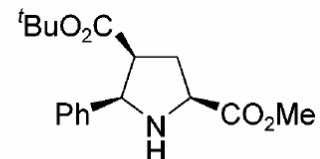
52% ee, 88% yield



87% ee, 87% yield

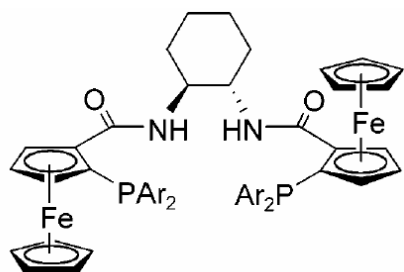


60% ee, 90% yield



93% ee, 85% yield

◆ Difference in ee can be explained by endtransition state model.



(*S,S,S_p*)-FAP: Ar = phenyl (**6a**)

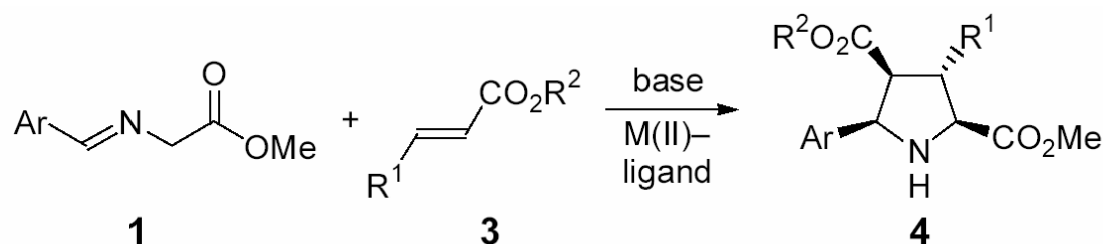
(*S,S,S_p*)-xylyl-FAP: Ar =

3,5-dimethylphenyl (**6b**)

- ## ◆ The 3,5-dimethyl substitution extends the steric environment of ligand **6b**.

Catalytic Asymmetric 1,3-DC of Azomethine Ylides

◆ Jorgensen's Zn-*t*-Bu-BOX catalyst system:



a: Ar = Ph

b: Ar = 2-Np

c: Ar = *p*-BrC₆H₄

a: R¹ = H, R² = Me

b: R¹ = H, R² = Et

c: R¹ = H, R² = *t*Bu

d: R¹ = CO₂Me, R² = Me

a: Ar = Ph, R¹ = H, R² = Me

b: Ar = 2-Np, R¹ = H, R² = Me

c: Ar = 2-Np, R¹ = H, R² = Et

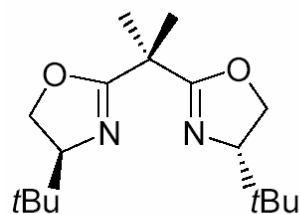
d: Ar = 2-Np, R¹ = H, R² = *t*Bu

e: Ar = *p*-BrC₆H₄, R¹ = H, R² = Me

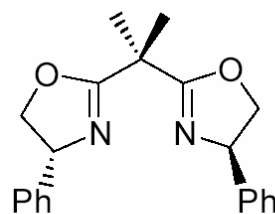
f: Ar = Ph, R¹ = CO₂Me, R² = Me

g: Ar = 2-Np, R¹ = CO₂Me, R² = Me

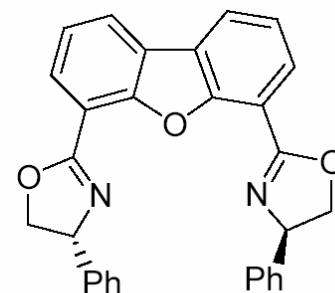
h: Ar = *p*-BrC₆H₄, R¹ = CO₂Me, R² = Me



5a: (*S*)-*t*-Bu-BOX

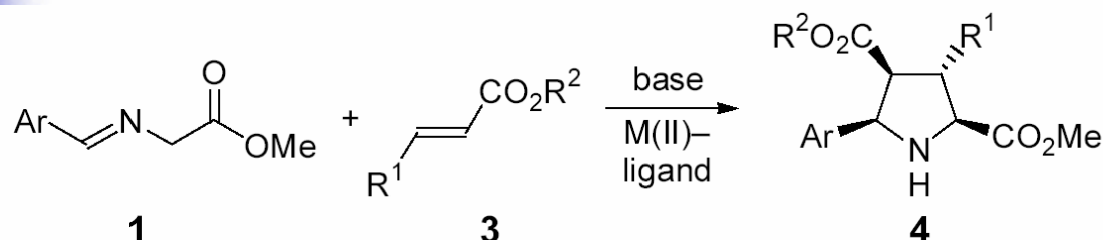


5b: (*R*)-Ph-BOX



5c: (*R,R*)-Ph-DBFOX

Catalytic Asymmetric 1,3-DC of Azomethine Ylides



a: Ar = Ph

b: Ar = 2-Np

c: Ar = *p*-BrC₆H₄

a: R¹ = H, R² = Me

b: R¹ = H, R² = Et

c: R¹ = H, R² = *t*Bu

d: R¹ = CO₂Me, R² = Me

a: Ar = Ph, R¹ = H, R² = Me

b: Ar = 2-Np, R¹ = H, R² = Me

c: Ar = 2-Np, R¹ = H, R² = Et

d: Ar = 2-Np, R¹ = H, R² = *t*Bu

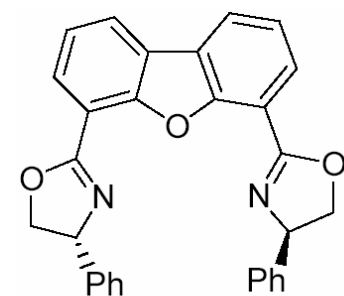
e: Ar = *p*-BrC₆H₄, R¹ = H, R² = Me

f: Ar = Ph, R¹ = CO₂Me, R² = Me

g: Ar = 2-Np, R¹ = CO₂Me, R² = Me

h: Ar = *p*-BrC₆H₄, R¹ = CO₂Me, R² = Me

Entry	Azomethine ylide	Dienophile	Yield ^[b] (%)	<i>ee</i> ^[c] [%]
1	1a	3a	4a (> 95 ^[d])	78
2 ^[e]	1a	3a	4a (80)	88
3	1b	3a	4b (93)	78
4 ^[e]	1b	3a	4b (84)	91
5 ^[e,f]	1b	3a	4b (86)	87
6	1b	3b	4c (76)	68
7	1b	3c	4d (12)	< 5
8	1c	3a	4e (89)	61
9 ^[e]	1c	3a	4e (89)	94
10 ^[e]	1a	3d	4f (78)	76
11	1b	3d	4g (84)	90
12	1c	3d	4h (87)	68



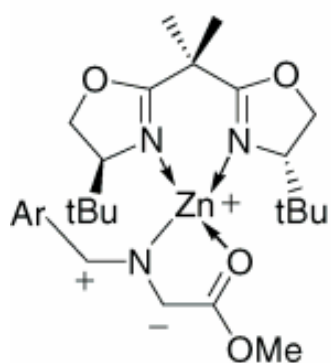
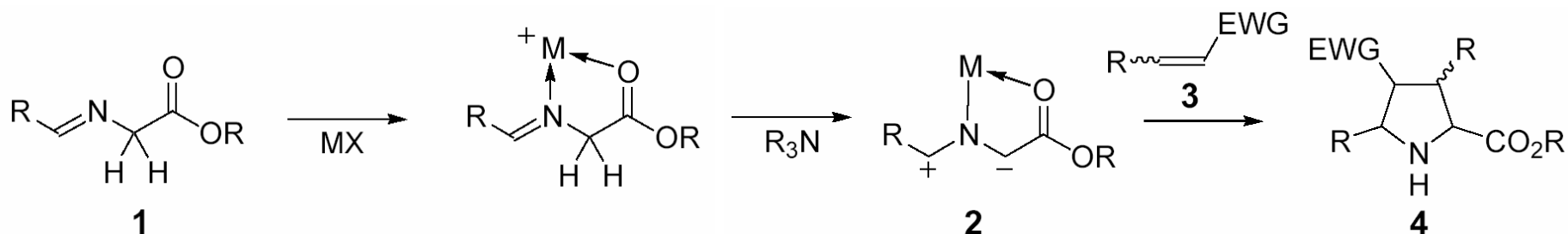
5c: (*R,R*)-Ph-DBFOX

[a] Lewis acid, ligand, and Et₃N (10 mol% each) were used.

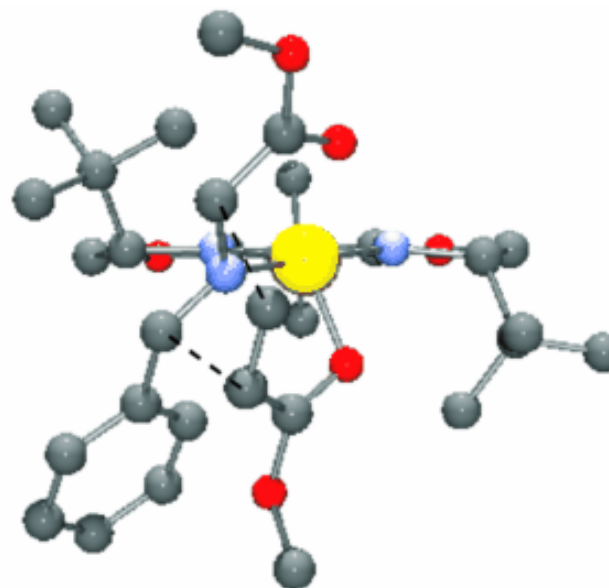
[e] Temperature -20°C. [f] No solvent.

Catalytic Asymmetric 1,3-DC of Azomethine Ylides

◆ Working model for Jorgensen's Zn-*t*-Bu-BOX catalyst system.



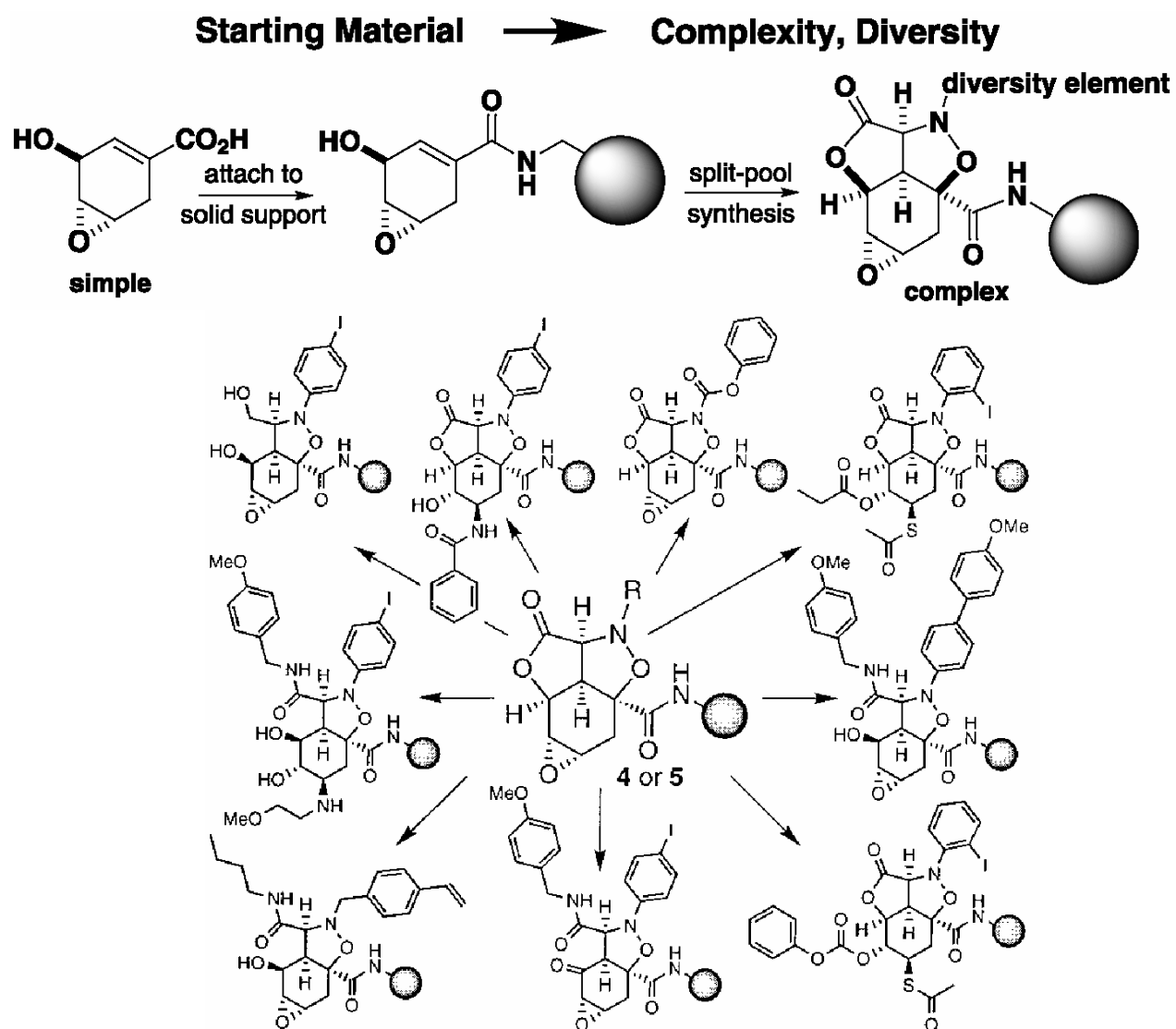
Proposed intermediate:
18-electron complex
-- tetrahedral conformation would give
opposite enantiomer!



Proposed bipyramidal intermediate:
Involves coordination of α,β -unsaturated
ester carbonyl to metal center.
(Acrylnitrile does not react!)

Catalytic Asymmetric 1,3-DC of Azomethine Ylides

◆ Diversity-Oriented Synthesis (DOS)

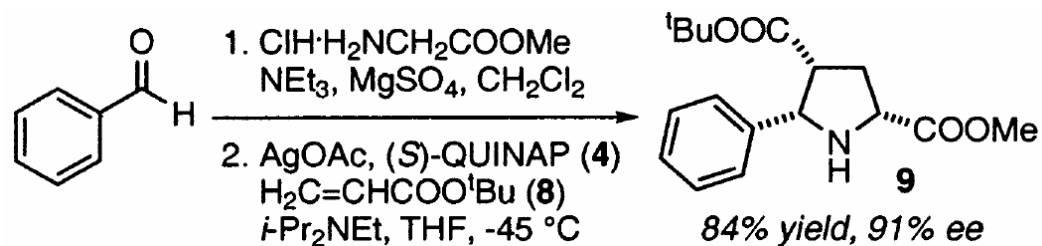
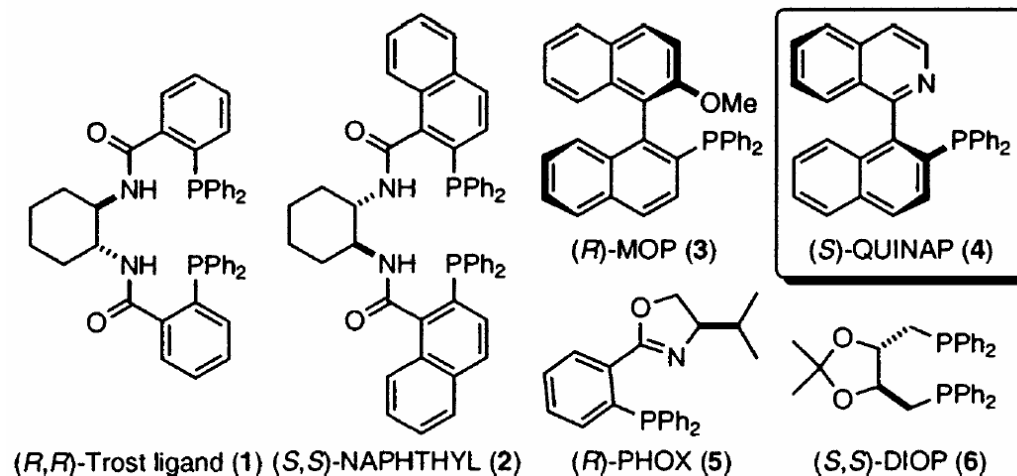


Tan, D. S.; Foley, M. A.; Shair, M. D.; Schreiber, S. L. *J. Am. Chem. Soc.* **1998**, *120*, 8565.
Schreiber, S. L. *Science*, **2000**, *287*, 1964

Catalytic Asymmetric 1,3-DC of Azomethine Ylides

◆ Schreiber's silver(I)-QUINAP catalyst system.

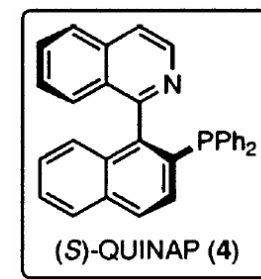
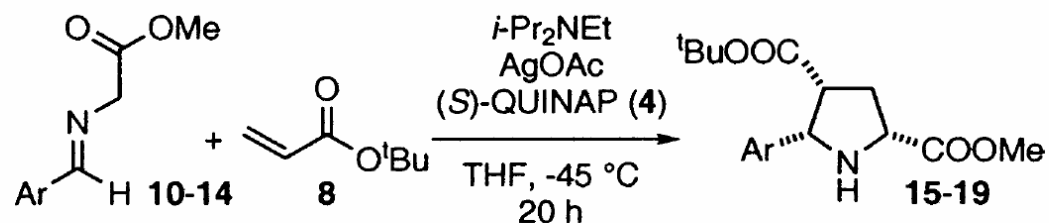
-- Chiral phosphine ligands screened for the azomethine ylide cycloadditions.



* 3 mol% catalyst used.

Catalytic Asymmetric 1,3-DC of Azomethine Ylides

- ◆ Exploration of the reactivity of the aromatic moiety

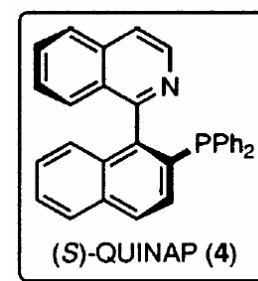
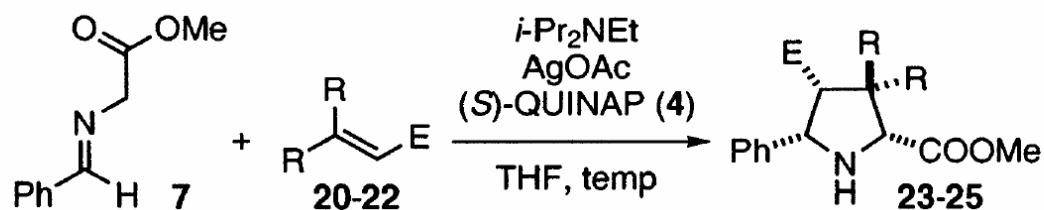


entry	Ar	pyrrolidine	yield ^b	ee ^c
1	4-methoxyphenyl (10)	15	93%	95%
2	4-bromophenyl (11)	16	89%	95%
3	4-cyanophenyl (12)	17	92%	96%
4	2-naphthyl (13)	18	89%	94%
5	2-tolyl (14)	19	95%	89%

^a Catalyst loading: 3 mol %. ^b Isolated yield. ^c Determined by HPLC.

Catalytic Asymmetric 1,3-DC of Azomethine Ylides

- ◆ Exploration of the reactivity of dipolarophiles.

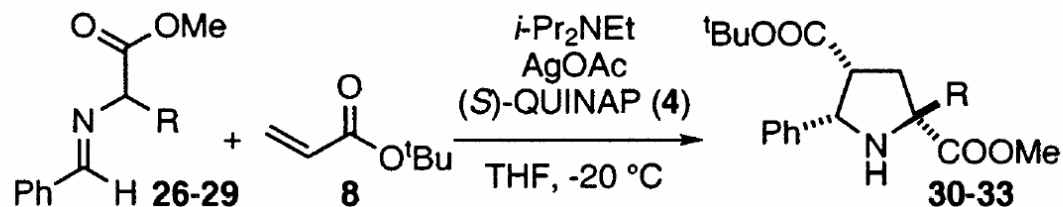


entry	dipolarophile	temp	time	yield ^c	endo:exo ^d	ee ^e
1 ^a	dimethyl maleate (20)	-60 °C	48 h	88%	(23) >20:1	60%
2 ^b	<i>tert</i> -butyl crotonate (21)	-20 °C	85 h	97% ^f	(24) >20:1	84%
3 ^b	<i>tert</i> -butyl cinnamate (22)	-20 °C	85 h	62% ^g	(25a, 25b) 2:1	81%, 50% ^h

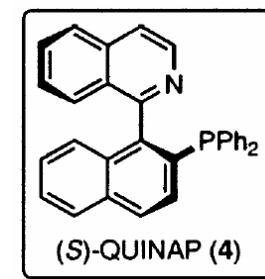
^a Catalyst loading: 3 mol %, solvent: toluene. ^b Catalyst loading: 10 mol %. ^c Isolated yield. ^d Determined by crude ¹H NMR spectra. ^e Determined by HPLC. ^f ca. 95% purity. ^g Combined yield of endo and exo products. ^h Enantioselectivity of the exo product **25b**.

Catalytic Asymmetric 1,3-DC of Azomethine Ylides

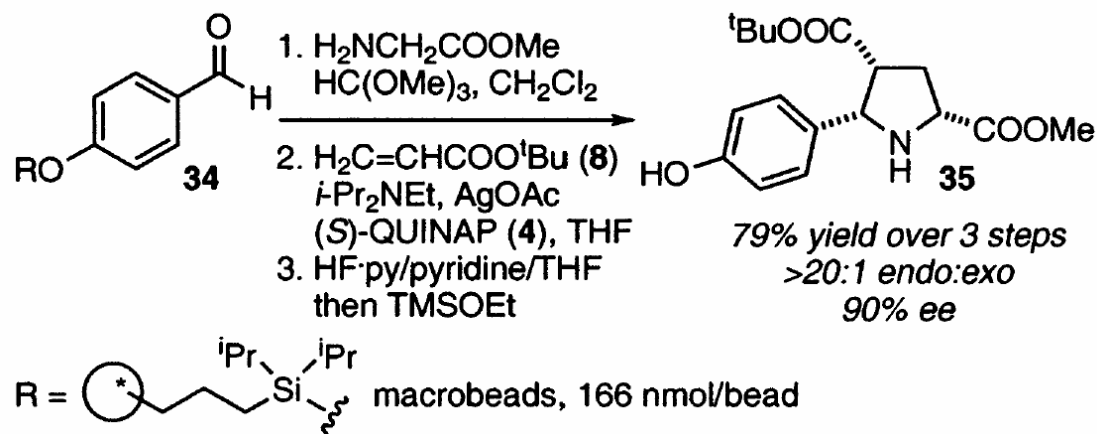
- Extending of the scope of the silver(I)-QUINAP-catalyzed [3+2] azomethine ylide cycloadditions.



entry	R	time	yield ^b	pyrrolidine	ee ^c
1	methyl (26)	24 h	98%	30	80%
2	<i>iso</i> -butyl (27)	48 h	77% ^d	31	80%
3	benzyl (28)	48 h	93%	32	77%
4	3-indolylmethyl (29)	96 h	47% ^e	33	81%



^a Catalyst loading: 10 mol %. ^b Isolated yield. ^c Determined by HPLC.
^d 85% conversion. ^e 50% conversion.



Catalytic Asymmetric 1,3-DC of Azomethine Ylides

- ◆ Brief comparison of three catalytic systems:

	Zhang	Jorgensen	Schreiber
Salt	AgOAc	Zn(OTf) ₂	AgOAc
Ligand	Xylyl-FAP	<i>t</i> -Bu-BOX	QUINAP
Loading	3 mol%	10 mol%	3-10 mol%
ee	70-97%	Up to 91%	60-96%
Substrate Scope	Aromatic, aliphatic ylide; Limited dipolarphile.	Aromatic ylide; Limited dipolarphile.	Aromatic ylide*; Limited dipolarphile.

- ◆ Limitation and possible future works of current systems:
 - None of the system can catalyze the aliphatic ylide highly enantioselectively.
 - Each system is only efficient for certain substrates.
 - Understanding of the mechanism/transition state limited.
 - ...



Summary of Asymmetric [3+2] Cycloaddition of Azomethine Ylides

- ◆ The [3+2] cycloaddition of azomethine ylide with alkene is an extremely versatile and atom economical process.
- ◆ Many highly efficient and stereoselective methods have been developed for the synthesis of various highly-substituted pyrrolidines.
- ◆ While chiral auxiliary approach has been studied extensively, asymmetric catalysis is still a brand-new area and deserves more attention.

“Compared to the development of the asymmetric metal-catalyzed carbo- and hetero-Diels-Alder reactions, the development of the analogous approach to asymmetric 1,3-DC reactions is several years behind.”

-- Jorgensen, *Chem. Rev.* **1998** 98, 863