

Key

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Chemistry 483

Exam 4

December 5, 2008

1. (50 points) define and/or characterize

a. The 5 symmetry elements

$$E, \sigma, \bar{\sigma}, \hat{S}, \hat{C}_N$$

b. Degrees of freedom

We require $3N$ coordinates to specify the location & orientation of a molecule. We say that the molecule has $3N$ degrees of freedom.

c. Rotation reflection operator

\hat{S}_N : rotate through $\frac{2\pi}{N}$ degrees & reflect through the σ_h plane.

d. Normal coordinate

the set of coordinates that allow the potential energy for vibrational motion to be written as

$$\Delta V = \frac{1}{2} \sum_{j=1}^N f_j Q_j^2$$

e. Trace of a matrix

The sum of the diagonal elements

f. Centrifugal distortion

When a diatomic molecule undergoes a transition from $v=0$ to $v=1$ its bond length increases & $\tilde{B}_1 \ll \tilde{B}_0$. This effect is called centrifugal distortion.

g. Selection rule

constraints on the allowed changes of quantum numbers when absorbing or emitting a photon.
 For vibration $\Delta v = \pm 1$, for rotation $\Delta J = \pm 1$

h. Transition dipole

$$\mu_{if} = \int \psi_i \vec{\mu} \psi_f dV$$

i. Rotation matrix

relates the coordinates of a point in one coordinate system to those on a rotated coordinate system. For two dimensions

$$\begin{pmatrix} \cos\theta & \sin\theta \\ -\sin\theta & \cos\theta \end{pmatrix}$$

j. Basis for an irreducible representation

For 1 dm. A function ψ which is an eigenfunction of the symmetry group operators $\hat{R}\psi = \chi_i(R)\psi$

For irreducible representations of dimension 2 or higher the representation of the operators in the basis are the irreducible representation matrices.

2. (25 points) The SO_2 molecule has C_{2v} symmetry. Suppose the molecule is in the yz plane with S at the origin and the two O atoms at $(0, \pm y_0, z_0)$. Use the p_x orbitals on S and the two O atoms as a basis and

- determine the representation matrices of C_{2v} .
- what are the characters of this representation?
- what irreducible representations are contained in this reducible representation?
- if we denote the p_x orbitals on the O atoms as p_L and p_R for left and right respectively what irreducible representations does $p_L + p_R$ belong to?
- what irreducible representation does the p_x orbital on S belong to.

a. First we need to order the basis

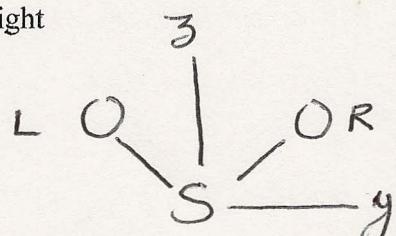
For example

$$\begin{pmatrix} x_L \\ x_R \\ x_S \end{pmatrix}$$

$x_L = p_x$ on Left O

$x_R = p_x$ on right O

$x_S = p_x$ on S



$$\text{Then } E = \begin{pmatrix} 1 & 0 & 0 \\ 0 & 1 & 0 \\ 0 & 0 & 1 \end{pmatrix}; C_2 = \begin{pmatrix} 0 & -1 & 0 \\ -1 & 0 & 0 \\ 0 & 0 & -1 \end{pmatrix}; \sigma_v = \begin{pmatrix} 0 & 1 & 0 \\ 1 & 0 & 0 \\ 0 & 0 & 1 \end{pmatrix}; \sigma_v' = \begin{pmatrix} -1 & 0 & 0 \\ 0 & -1 & 0 \\ 0 & 0 & -1 \end{pmatrix}$$

$$b. \chi = \begin{matrix} 3 & & -1 & & 1 & & -3 \end{matrix}$$

Note: if we write basis as $\begin{pmatrix} x_L \\ x_S \\ x_R \end{pmatrix}$ we get

$$E = \begin{pmatrix} 1 & 0 & 0 \\ 0 & 1 & 0 \\ 0 & 0 & 1 \end{pmatrix}; C_2 = \begin{pmatrix} 0 & 0 & -1 \\ 0 & -1 & 0 \\ -1 & 0 & 0 \end{pmatrix}; \sigma_v = \begin{pmatrix} 0 & 0 & 1 \\ 0 & 1 & 0 \\ 1 & 0 & 0 \end{pmatrix}; \sigma_v' = \begin{pmatrix} -1 & 0 & 0 \\ 0 & -1 & 0 \\ 0 & 0 & -1 \end{pmatrix}$$

$$b. \chi = \begin{matrix} 3 & & -1 & & 1 & & -3 \end{matrix}$$

So the order doesn't matter

$$c. Q_i = \frac{1}{4} (3 - \chi_i(C_2) + \chi_i(\sigma_v) - 3\chi_i(\sigma_v'))$$

$$Q_{A_1} = \frac{1}{4} (3 - 1 + 1 - 3) = 0 \quad Q_{B_1} = \frac{1}{4} (3 + 1 + 1 + 3) = 2$$

$$Q_{A_2} = \frac{1}{4} (3 - 1 - 1 + 3) = 1 \quad Q_{B_2} = 0 \quad 3$$

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d. $\hat{E} (P_L + P_R) = + (P_L + P_R)$

$$\hat{C}_z (P_L + P_R) = - (P_L + P_R)$$

$$\hat{\sigma}_v (P_L + P_R) = + (P_L + P_R)$$

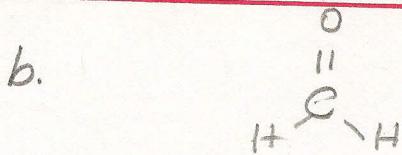
$$\hat{\sigma}'_v (P_L + P_R) = - (P_L + P_R)$$

$$\Gamma = \begin{matrix} 1 & -1 & 1 & -1 & \dots & \boxed{B_1} \end{matrix}$$

e. as above $P_S \Rightarrow \boxed{B_1}$

3. (25 points) Formaldehyde, H₂CO, has C_{2v} symmetry. How many normal modes of vibration does formaldehyde have? What are their symmetries? Which are IR active?

a. $3N - 6 = 12 - 6 = 6$



	E	C ₂	σ_v	σ_v'
χ	12	-2	2	4

$$\alpha_1 = \frac{1}{4} (12 - 2\chi_e(C_2) + 2\chi_e(\sigma_v) + 4\chi_e(\sigma_v'))$$

$$\alpha_{A_1} = \frac{1}{4} (12 - 2 + 2 + 4) = 4$$

$$\alpha_{A_2} = \frac{1}{4} (12 - 2 - 2 - 4) = 1$$

$$\alpha_{B_1} = \frac{1}{4} (12 + 2 + 2 - 4) = 3$$

$$\alpha_{B_2} = \frac{1}{4} (12 + 2 - 2 + 4) = 4$$

From character table Translations are A₁, B₁, B₂
 & Rotations are A₂, B₁, B₂

∴ remaining symmetries are vibrations

$3A_1, 1B_1, 2B_2$

They are all IR active.