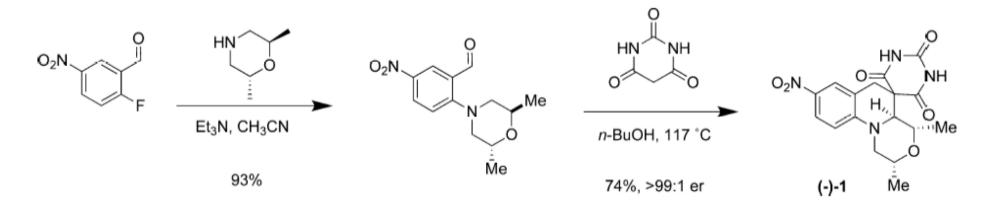
Synthesis of (-)-PNU-286607 by Asymmetric Cyclization of Alkylidene Barbiturates

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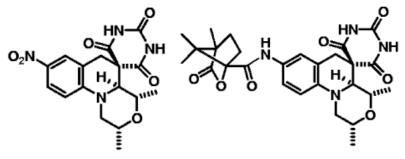


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Charactericstics of PNU-286607

- PNU-286607 was identified during a screening effort at Pharmacia and Upjohn for compounds possessing whole cell antibacterial activity out of ~250,000 compounds.
- A reverse chemical genomics approach (compound-driven target identification) led to the identification of bacterial type II topoisomerase enzymes (DNA gyrase and topoisomerase IV) which are essential for bacterial DNA synthesis. PNU-286607 targets these enzymes via a mechanism of inhibition distinct from the mechanisms of fluoroquinolones and novobiocin. Further, PNU-286607 displays little crossresistance with marketed antibacterial therapies. Since there is a dire need for new molecular entities and antibacterial agents with novel mechanisms of action to counter bacterial resistance, PNU-286607 represents a promising opportunity.
- Upon attempts to resynthesize the originally assigned structure of PNU-286607, it became clear that the structure was incorrect. Extensive NMR studies and an X-ray crystal structure determination revealed that PNU-286607 had the structure shown in Figure 1.



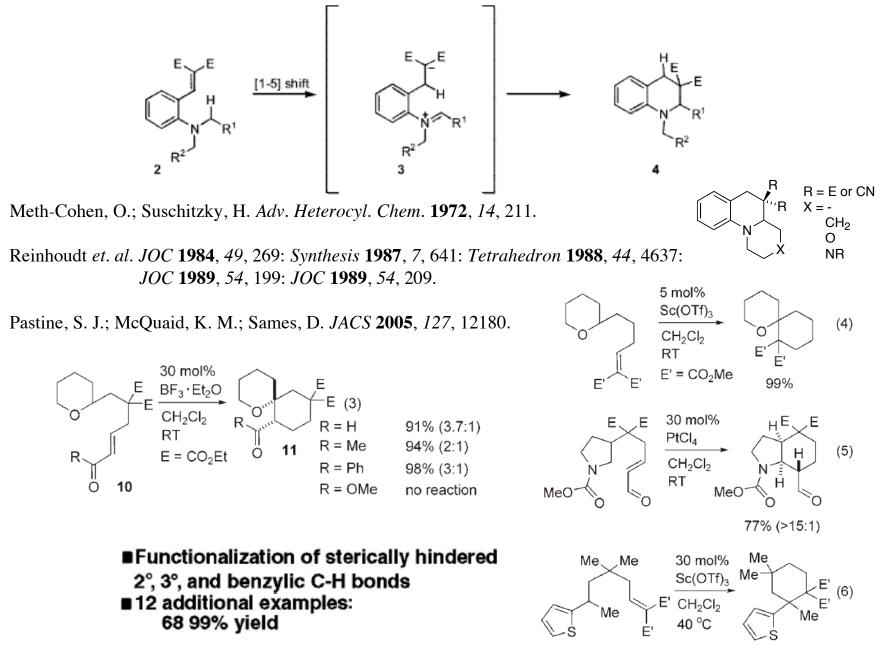
PNU-286607 presented several significant synthetic challenges due to its structural complexity. In addition to the unusual spirocyclic barbituric acid moiety, the compound possesses three stereogenic centers around the fused morpholine portion of the tetrahydroquinoline core. Diastereocontrol of these stereogenic centers was of particular concern. Further, since the antibacterial activity resides solely in the (-)-enantiomer of PNU-286607, an asymmetric synthesis was given high priority.

tert-Amino Effect/Cyclizations

1)

2)

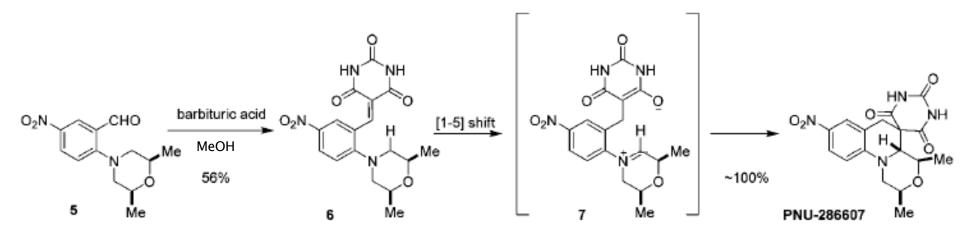
3)

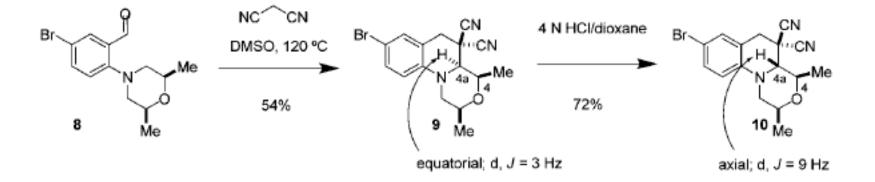


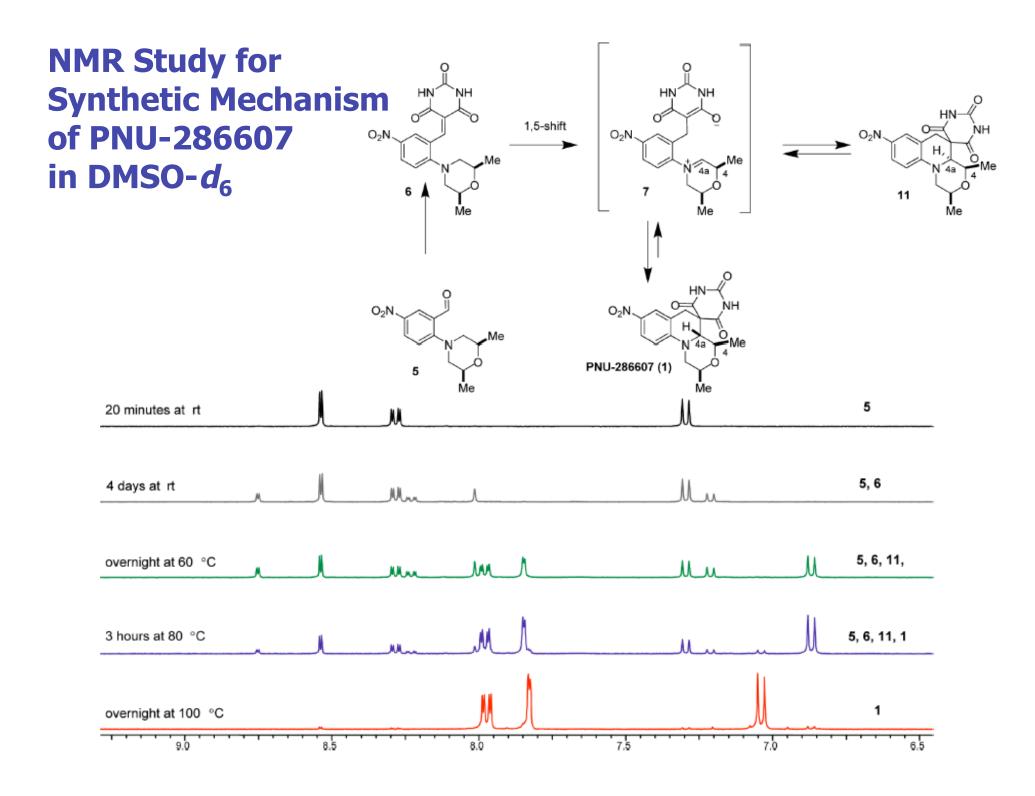
68%

Preliminary Synthesis of PNU-286607

There were very few diastereoselective tert-amino reactions wherein the resident stereogenic centers were located on the pendant azacycle, and none with a stereogenic center located adjacent to the migrating hydrogen.

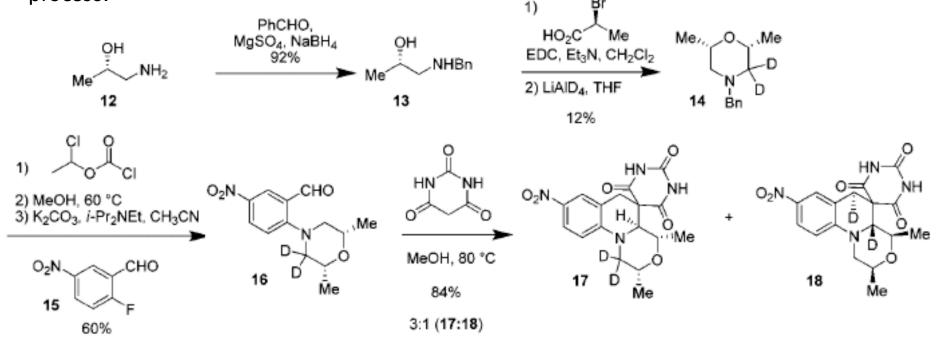






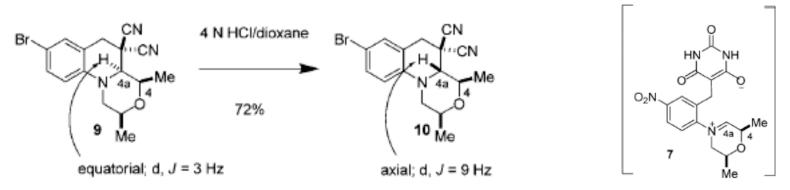
Study for [1,5]-shift

- ★ Reinhoudt and co-workers have investigated the [1,5]-shift with pyrrolidine-2,2,5,5- d_4 , determining the isotope effect of 3.0 ± 0.3 at 91.2 °C in DMSO- d_6 . This result significantly supports [1,5]-shift is the rate-determining step.
 —Tetrahedron 1988, 44, 4637.
- Authors were interested in studying a nonsymmetrically labeled morpholine for deeper mechanistic insight and ultimately to examine the possibility of developing an asymmetric process.

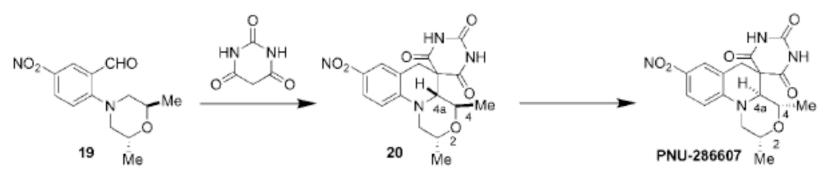


- Deuterium migration was stereoselective in compound 18. The ratio of 17 to 18 was ~3:1, in accordance with the reported isotope effect. Products derived from double deuterium migration were not detected, which indicates that 1,5-shift might not be reversible.
- First asymmetric synthesis of deuterated 17 reconfirmed the absolute stereochemistry and 17 has shown the equal antibacterial activity as (-)-PNU-286607; 18 was devoid of activity.

Study for Stereoselective Cyclization

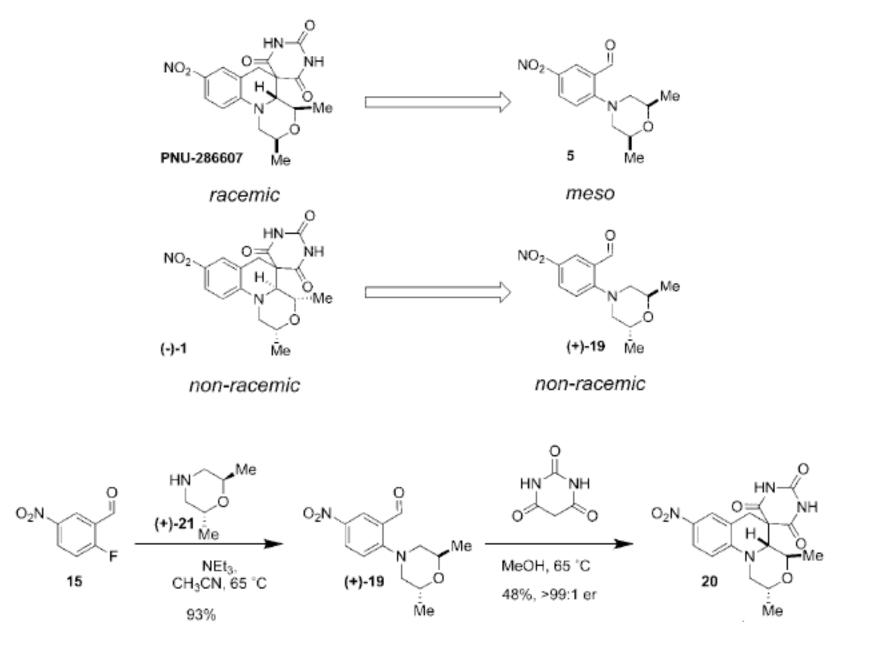


C(4a) was stereochemically labile dependent on the configuration of adjacent methyl group on C(4). However, examination of the C(4a) isomerization mechanism led authors to consider whether the C(4) methyl-bearing stereocenter was labile as well, via the putative iminium ion intermediate 7.

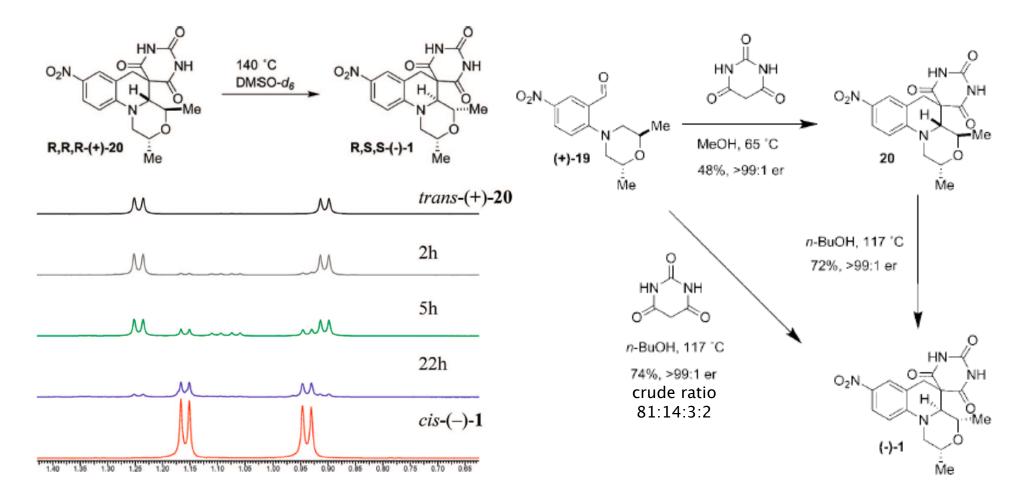


- A small amount of the cis-product (PNU-286607) was observed during the isolation of transproduct (20), despite rigorous purification of aldehyde 19.
- Both the C(4) and C(4a) centers were stereochemically labile and that the configuration of the desired product may be controlled by a single persistent stereocenter at C(2).

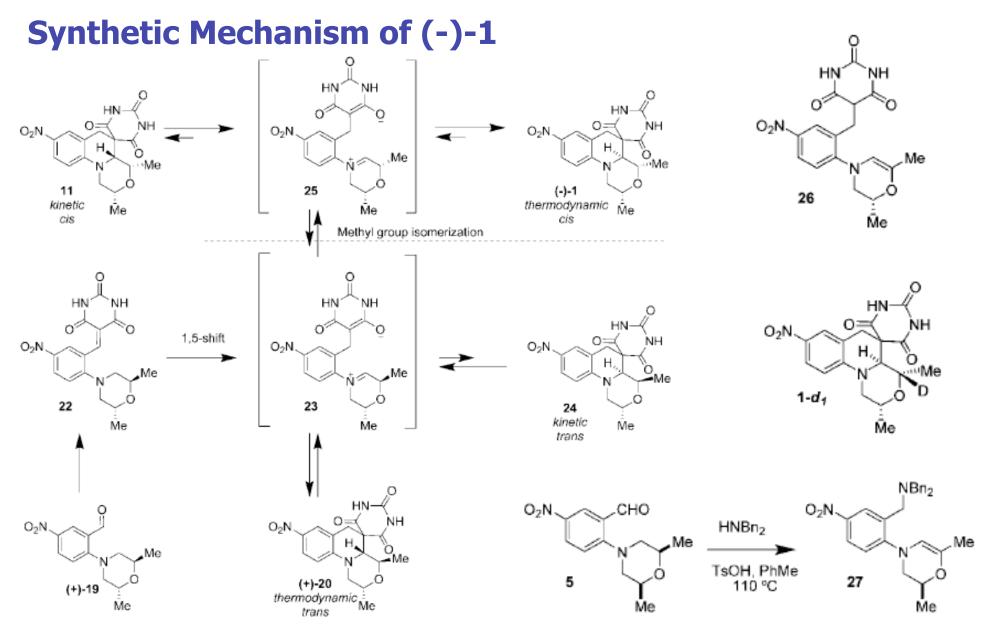
Asymmetric Synthesis of (-)-PNU-286607



Asymmetric Synthesis of (-)-PNU-286607



- In NMR study, equilibrium was established as an ~8:1 mixture of cis/trans isomers after 22 hours.
- Practically, protic solvents were found to be optimal, providing adequate temperature control and favorable reaction rates.



Heating of isomerically pure (-)-1 or (+)-20 to 117 °C in *n*-butanol yields the identical ratio of diastereomers further indicating that the process is under thermodynamic control. However, 1,5-shift appears to be irreversible, as there is no erosion of the enantiomeric ratio of recovered (-)-1.