

Ch 13 #2, Ch 14 #2, 4, 7, 8

Min Chung Lee

$$13/\#2 \quad E = 2000V / 4 \times 10^{-1} \text{ cm} = 5 \times 10^4 \text{ V cm}^{-1}$$

From figure 11.2 (b)  $\rightarrow v_d = 6.5 \times 10^6 \text{ cm/s}$  (At 300K)

$$\Rightarrow \text{Max. Collection Time} = 4 \times 10^{-1} / 6.5 \times 10^6 = 0.062 \mu\text{sec} \quad \#$$

$$14/\#2 \quad V = \frac{Q}{C} = \frac{M n_0 e}{C} \quad (n + {}^3\text{He} \rightarrow p + {}^3\text{H})$$
$$= 10^3 \left( \frac{0.764 \times 10^6 \text{ eV}}{42.7 \text{ eV}} \right) \frac{1.602 \times 10^{-19} \text{ Col}}{10^2 \times 10^{-12} \text{ Farad}}$$
$$= 0.029 \text{ Volt} \quad \#$$

$$\#4 \quad \Sigma(E) = 1 - e^{-\Sigma(E)X}$$

According to the text in P. 515:

For  $\text{BF}_3$  tube (96% enriched in  ${}^{10}\text{B}$ )

$\Sigma(E) = 91.5\%$  if length is 30 cm

$$\Rightarrow \Sigma(E) = 0.0822 \text{ cm}^{-1}$$

When length is 10 cm

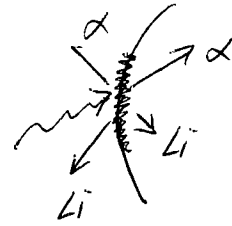
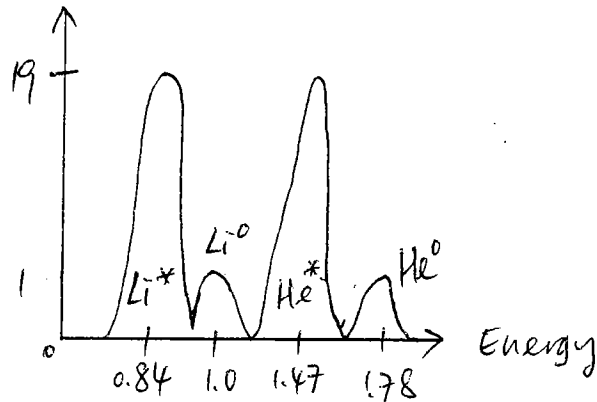
$$\Sigma(E) = 1 - e^{-0.822}$$

$$= 0.56$$

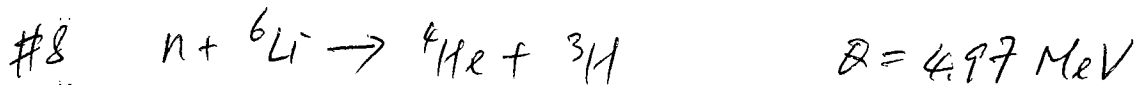
(or use  $\sigma$  from P. 507 in text)

$$\Rightarrow \Sigma(E) = 56\% \quad \#$$

#7/



\* from excited state  
 ° from ground state



For  $\text{Li}^{\circ}$  = Wavelength of max. emission = 470 nm

Scintillation efficiency = 2.8%

→ Energy needed to generate a scintillation photon = 
$$\frac{(6.626 \times 10^{-34} \times 3 \times 10^8)}{(470 \times 10^{-9} \times 0.028)}$$
  
 = 94.288 eV

⇒ No of photons produced =  $4.97 \times 10^6 / 94.288 = 52 \text{ k photons}^{\#}$

For  $\text{Li}^*$  = Wavelength of max. emission = 395 nm

Scintillation efficiency = 0.45%

Energy need to generate a photon = 
$$\left( \frac{6.626 \times 10^{-34} \times 3 \times 10^8}{395 \times 10^{-9} \times 0.0045} \right)$$
  
 = 698.07 eV

⇒ No of photons produced =  $4.97 \times 10^6 / 698.07 = 7 \text{ k photons}^{\#}$