

1 a) $1.0 \mu\text{Ci} \times 10^{-6} \frac{\text{Ci}}{\mu\text{Ci}} \times 3.7 \times 10^{10} \frac{\text{Bq}}{\text{Ci}} e^{-\lambda t}$ $t = 50 \text{ yr}$
 $3.7 \times 10^4 \text{ Bq} e^{-\frac{\ln 2 \times 50}{351}}$ $T_{1/2} = 351 \text{ yr}$
 $3.7 \times 10^4 \text{ Bq} (0.9059) = 3.35 \times 10^4 \text{ Bq}$

- b) ① alpha decay to 0.388 keV state (84%)
 ② gamma decay to ground state 0.388 keV (81% of α)

c) $E_\alpha = \frac{245}{249} (Q_\alpha - E_\gamma^*) = \frac{245}{249} (6.295 - 0.388) \text{ MeV}$
 $E_\alpha = 5.812 \text{ keV}$

from figure 2.7 $R \sim 28 \mu\text{m}$
in text

d) 388 keV γ is relatively penetrating radiation with ~~very~~ high intensity \rightarrow use a portable inorganic scintillation detector \rightarrow eg NaI(Tl)

2 a) $RC = 50 \Omega (300 \times 10^{-12} \text{ F}) = 1.5 \times 10^{-8} \text{ s} \rightarrow 15 \text{ ns}$

$$\tau_{\text{collection}} \approx \frac{300 \mu\text{m}}{\text{drift velocity}} = \frac{300 \times 10^{-6} \text{ m}}{10^7 \text{ cm/s} \times 10^2 \frac{\text{cm}}{\text{cm}}} = 3 \times 10^{-9} \text{ s} \rightarrow 3 \text{ ns}$$

\rightarrow electronics are "slow" relative to signal time so the pulse will be integrated and the max voltage will be produced

2 b) Eq 4.24 $n_{TRUE} = \frac{n_{OBS}}{1 - n_{OBS} Z}$, $n_{OBS} = \frac{123456 \text{ cts}}{60.2} = 2057.6 \text{ cps}$

$$n_{TRUE} = \frac{2057.6 /s}{1 - (2057.6 \times 5 \times 10^{-6})} = 2079 /s$$

3) $Q = P S_{eff}$ $Q = q \times (\text{Area}) = 10^{-5} t^{-1.3} \frac{W}{m^2} [2\pi R^2 + 2\pi R h]$

$$Q = 10^{-5} t^{-1.3} [2\pi (\frac{1}{2})^2 + 2\pi (\frac{1}{2}) 0.5 \text{ m}^2]$$

$$Q = \frac{2\pi}{4} \times 10^{-5} t^{-1.3} W, t = 1 \text{ hr}$$

$$\frac{1}{S_{eff}} = \frac{1}{S} + \frac{1}{C} = \frac{1}{250 \frac{L}{s}} + \frac{1}{300 \frac{1}{s}}$$

$$\hookrightarrow S_{eff} = \frac{1}{[\frac{1}{250} + \frac{1}{300}]} = 136.4 \frac{L}{s}$$

$$P = \frac{Q}{S_{eff}} = \frac{\pi \times 10^{-5} W}{136.4 \frac{L}{s} \times 10^{-3} \frac{m^3}{L}} = 2.30 \times 10^{-4} Pa$$

$$\left[= 2.30 \times 10^{-4} Pa \times \frac{1 \text{ mbar}}{10^2 Pa} \right] = 2 \times 10^{-6} \text{ mbar}$$

$$\left[= 2.30 \times 10^{-4} Pa \times \frac{1 \text{ torr}}{133.3 Pa} \right] = 1.7 \times 10^{-6} \text{ torr}$$

$$4a) \quad \bar{p} = \frac{3 \text{ events}}{1000 \text{ hrs}} = 3 \times 10^{-3} \frac{\text{evt}}{\text{hr}} \quad \mu = 48 \text{ hr}$$

$$\bar{x} = \bar{p} \mu = 3 \times 10^{-3} / \text{hr} \times 48 \text{ hr} = 0.144$$

$$\text{Prb that you get 1 event} = P(1) = (\bar{x})^1 e^{-\bar{x}} / 1!$$

$$P(1) = (0.144) e^{-0.144} / 1 = 0.125$$

which is ~~7~~ to 1 against seeing an event

$$4b) \quad \lambda_{\text{obs}} = \frac{1990 \text{ evts}}{1000 \text{ hr} \times 60 \frac{\text{min}}{\text{hr}}} = 0.0332 \text{ evts/min}$$

$$\text{from Eq 3.70} \quad P(0) = e^{-\lambda t} = e^{-0.0332 \times 1 \text{ min}}$$

Prb of no events
during interval

$$P(0) = 0.967$$

[same result obtains from integration of interval distribution]

$$5a) \quad \Delta E = 0.1 \text{ MeV in BC-400} \quad 65\% \text{ of anthracene Table 8.1}$$

$$\text{NaI(Tl) is } 230\%$$

$$\text{Table 8.2 NaI(Tl) gives } 38000 \frac{\text{photons}}{\text{MeV}}$$

$$\# \text{ photons} = 0.1 \text{ MeV} \times \frac{65}{230} \times 38000 \frac{\text{photons}}{\text{MeV}} = 1074 \text{ photons}$$

$$b) \quad \lambda_{\text{peak}} = 423 \text{ nm Table 8.1}, \text{ radiant sensitivity } 95 \text{ mA/W}$$

from Table 9.1

→ use dimensional analysis to convert current/power into electrons/energy, ~~use~~ ~~use~~ ~~use~~ energy of photon

$$\begin{aligned}
 \text{Radiant Sensitivity} &= 95 \frac{\text{mA}}{\text{W}} \times 10^{-3} \frac{\text{A}}{\text{mA}} \times \frac{1 \text{ Coul/s}}{\text{A}} \times \frac{1 \text{ W}}{\text{J/s}} \\
 &= 9.5 \times 10^{-2} \frac{\text{Coul}}{\text{J}} \times \frac{1 \text{ electron}}{1.602 \times 10^{-19} \text{ Coul}} \\
 &= 5.93 \times 10^{17} \frac{\text{electrons}}{\text{Joule}}
 \end{aligned}$$

$$E_{\text{photon}} = \frac{hc}{\lambda} = \frac{6.626 \times 10^{-34} \text{ J}\cdot\text{s} \times 3 \times 10^8 \text{ m/s}}{423 \times 10^{-9} \text{ m}} = 4.699 \times 10^{-19} \text{ J}$$

$$\# \text{ electrons/photon} = 5.93 \times 10^{17} \frac{e^-}{\text{J}} \times 4.699 \times 10^{-19} \text{ J} = 0.279$$

5c) $\# \text{ photo electrons needed} = 10 = \# \text{ photons created} * \epsilon_{\text{collection}} * \epsilon_{\text{QUANTUM}}$

$$\epsilon_{\text{collection}} = 10 / 1074 * 0.279 = 0.033$$

need to collect only 3.3% of photons ...