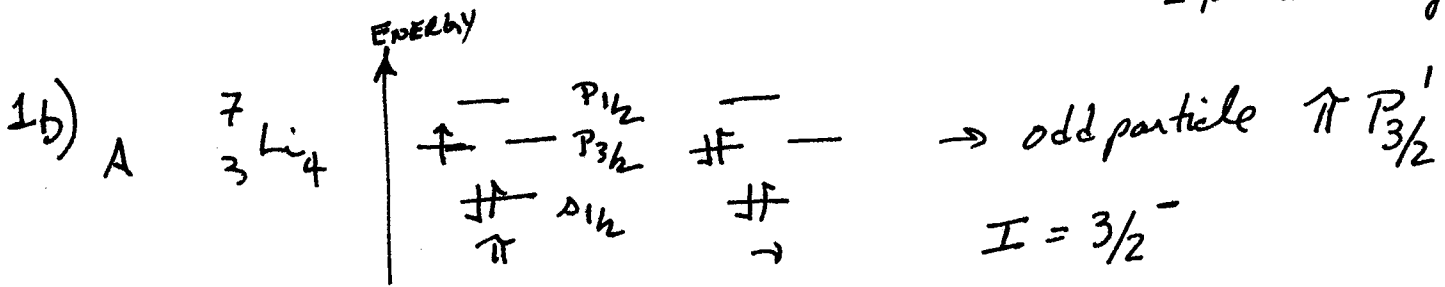


B $Q_{\beta^{+}} = \Delta({}^{40}\text{K}) - [\Delta({}^{40}\text{Ar}) + 2m_e c^2]$
 $= -33.535 - [-35.039 + 2 \times 0.511] = +0.482 \text{ MeV}$
 [β^{+} allowed by $\oplus Q_{\beta^{+}}$]



B ${}^7\text{Li}^*$ would be produced by promoting odd proton up to $p_{1/2}$ state, leave everything else the same
 $I = 1/2^{-}$

1c) A There are no bound nuclei with $A = 5$ which blocks the formation of heavier nuclei. Only a few ^{heavier} nuclei can be made by reactions between He nuclei during the short explosion time.

B Sun runs on the reaction $4p^{+} \rightarrow {}^4\text{He}^{2+} + 2\beta^{+} + 2\gamma$

C Sun produces no iron, fraction = 0

D A-process (slow neutron capture) terminates at ${}^{209}\text{Bi}$ because heavier nuclei decay back by α decay (${}^{210}\text{Po} \rightarrow \alpha + {}^{206}\text{Pb}$)

1d) Recall $\frac{dE}{dx} \propto K \frac{Aq^2}{E}$ | $\frac{dE}{dx} (\text{proton}) = \frac{23 \text{ MeV}}{\text{cm}} = K \frac{1(1)^2}{10}$

$K = 230 \frac{\text{MeV}^2}{\text{cm}}$

$\frac{dE}{dx} ({}^{40}_{20}\text{Ca}^{20+}) = 230 \frac{\text{MeV}^2}{\text{cm}} \left(\frac{40 \times 20^2}{240 \text{ MeV}} \right) = 15,333. \text{ MeV/cm}$

Note that it is important that the same material is used for both ions, otherwise we could not evaluate K

1e) NaI has a much higher ^{average} atomic number than plastic (CH₂) and so will have a much higher attenuation coefficient for rays and have a higher probability of observing rays from a source. DHS detectors cannot detect α or β particles due to their short range.

2) $\frac{N(235)}{N(238)} = \frac{N_0(235)}{N_0(238)} \frac{e^{-\lambda_{235}t}}{e^{-\lambda_{238}t}} = \frac{2.7 \times 10^{-3}}{1}$

assume 1

$(\lambda_{238} - \lambda_{235})t = 2.7 \times 10^{-3}$

$t = \frac{\ln(2.7 \times 10^{-3})}{(\lambda_{238} - \lambda_{235})} = \frac{-5.915}{-8.292 \times 10^{-10} / \text{yr}} = 7.13 \times 10^9 \text{ yr}$

$\lambda_{235} = \frac{\ln 2}{704 \times 10^6 \text{ yr}} = 9.85 \times 10^{-10}$

$\lambda_{238} = \frac{\ln 2}{4.46 \times 10^9 \text{ yr}} = 1.55 \times 10^{-10}$

$$3) A = N\sigma\phi(1 - e^{-\lambda t})$$

$$\lambda_F = \frac{\ln 2}{109.7 \text{ min}} = 6.32 \times 10^{-3} / \text{min}$$

$$t = 5.0 \text{ hr} \times \frac{60 \text{ min}}{\text{hr}} = 300 \text{ min}$$

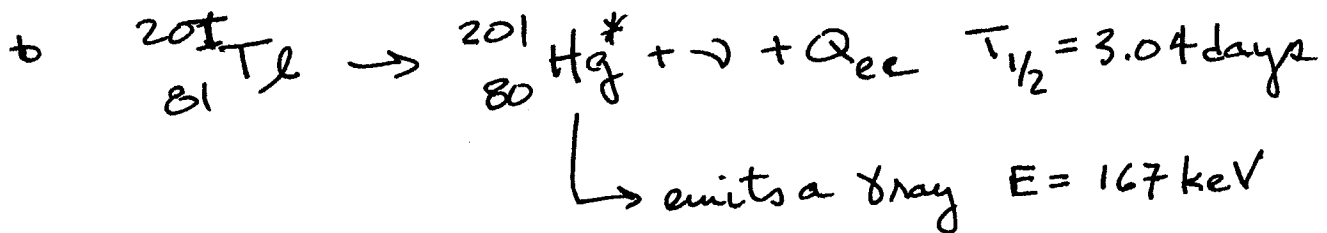
$$(1 - e^{-\lambda t}) = 1 - e^{-1.896} = 0.8498$$

$$A = \left(\frac{0.1 \text{ cm} \times 1.0 \text{ g/cm}^3 \times N_A}{20. \text{ g/mol}} \right) 0.30 \times 10^{-24} \text{ cm}^2 \left(0.5 \times 10^{-6} \text{ A} * \frac{\text{Coul/s}}{\text{A}} * \frac{1 \text{ Pat}}{1.602 \times 10^{-19} \text{ Coul}} \right) \times 0.8498$$

$$A = \left(3.01 \times 10^{21} \frac{\text{H}_2\text{O} \times 10}{\text{cm}^2 \text{ 1 H}_2\text{O}} \right) 3 \times 10^{-25} \text{ cm}^2 (3.121 \times 10^{12} / \text{s}) 0.8498$$

$$A = 2.396 \times 10^9 / \text{s} \rightsquigarrow \text{Bq} \rightsquigarrow 0.0648 \text{ Ci}$$

A) a From the web I found that ^{201}Tl is used in these type of diagnoses



$$c \quad \text{Reduction} = 2^{-10} = \frac{1}{1024}$$

$$N/N_0 = e^{-\lambda t} = e^{-\frac{\ln 2}{T_{1/2}} \times 10 \times T_{1/2}} = e^{-10 \ln 2}$$

$$d \quad t = 10 \times T_{1/2} = 10 \times 3.04 \text{ days} = 30.4 \text{ days}$$