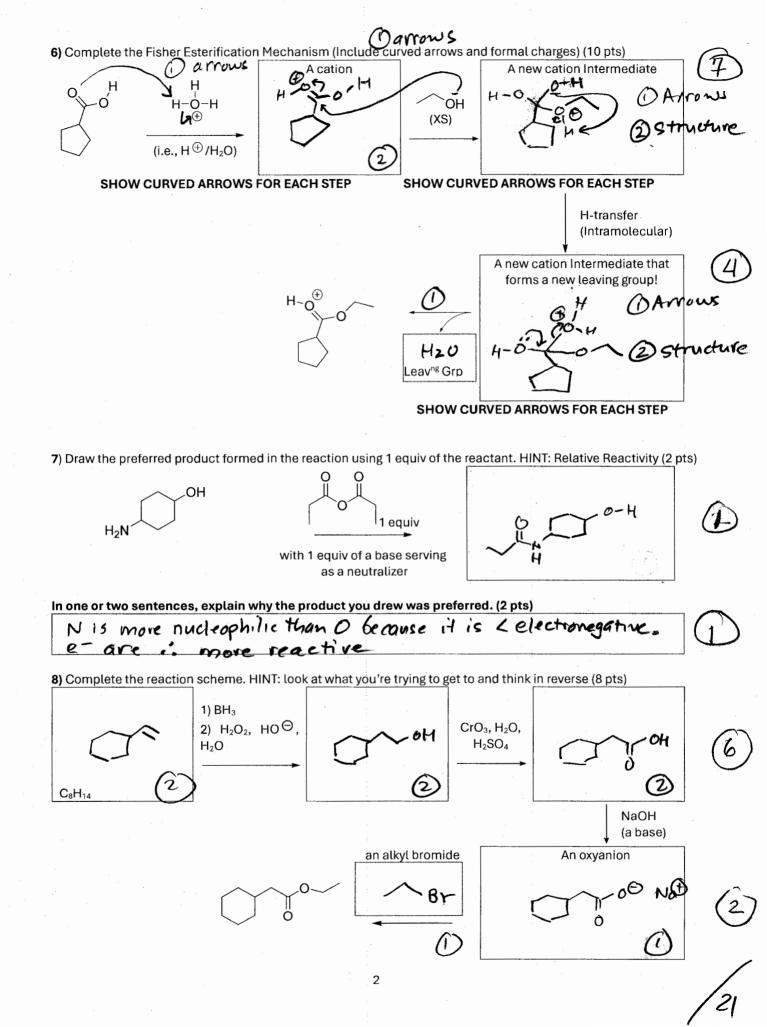
Show the reaction products or starting material in the large rectangles. (10 pts)

1) Na₂Cr₂O₇, H₂O, H₂SO₄ 2 2) Br₂, FeBr₃ (from CEM 351) 2 3) An Anion Mg-Salt MgBr CO_2 2 4) A neutral organic molecule NH₂ + ammonium salt (do not draw) 2 5) MgBr 2 equiv



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2) H₃O [⊕] neutralizer

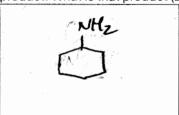


9) Complete the reaction scheme.

Add Lone Electron Pairs, Formal Charges and Line Bonds where necessary in the Lewis Structures

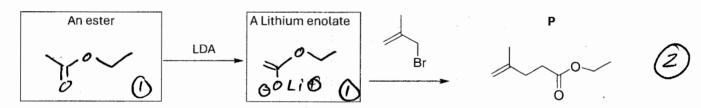
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| Add Lone Electron Pairs, Formal Charges and Line Bonds where necessary | Add Lone Electron Pairs, Formal Charges |
| Add Lone Electron Pairs, Formal Charges | Add Lone Electron Pairs, Formal Charges |

10) After a stable gas is lost, H_3O^+ is added to decompose the LiAl complex, a mild base is added to yield a neutral product. What is that product (2 pts)

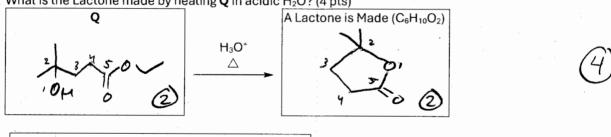


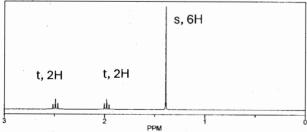
2

11) Complete the following reaction. HINT: Look "downstream" and infer earlier structures from those provided. LDA (Lithium diisopropyl amide) function as a base (2 pts)



12) Product **P** above is treated with 1) Hg(OAc)₂, H₂O, then 2) NaBH₄. These steps convert **P** to **Q**. What is **Q**? What is the Lactone made by heating **Q** in acidic H₂O? (4 pts)

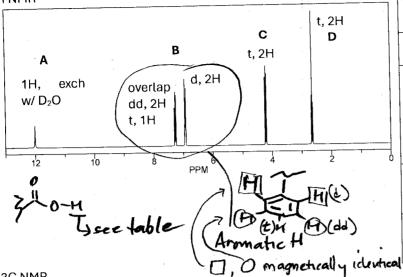




1H NMR of the Lactone

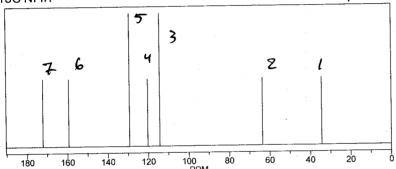
5 Duus

 $C_9H_{10}O_3$ 1H NMR



Draw the fragments/connectivities	5 pt
implied by the spectral data	
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B (detail these H's by showing	
which is(are) the dd, t , and d .	i
the Hold	2
c t sc-chz	(
D 5 CH2 - C (F)	1

13C NMR



Interpret what is implied by the 13C NMR data (1 pt).

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4 pts
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