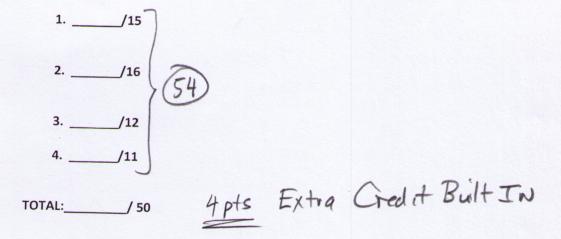
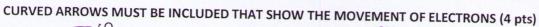
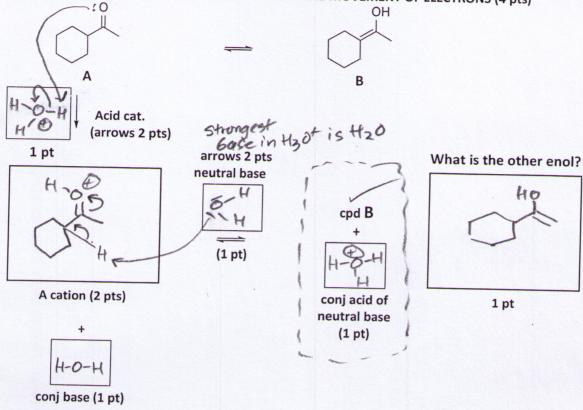


LEAVE THIS COVER SHEET ATTACHED TO THE Quiz!

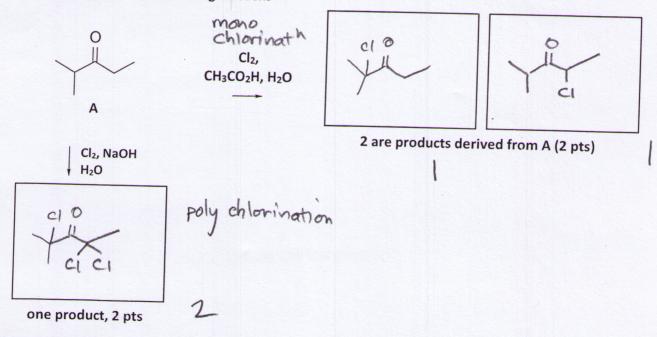


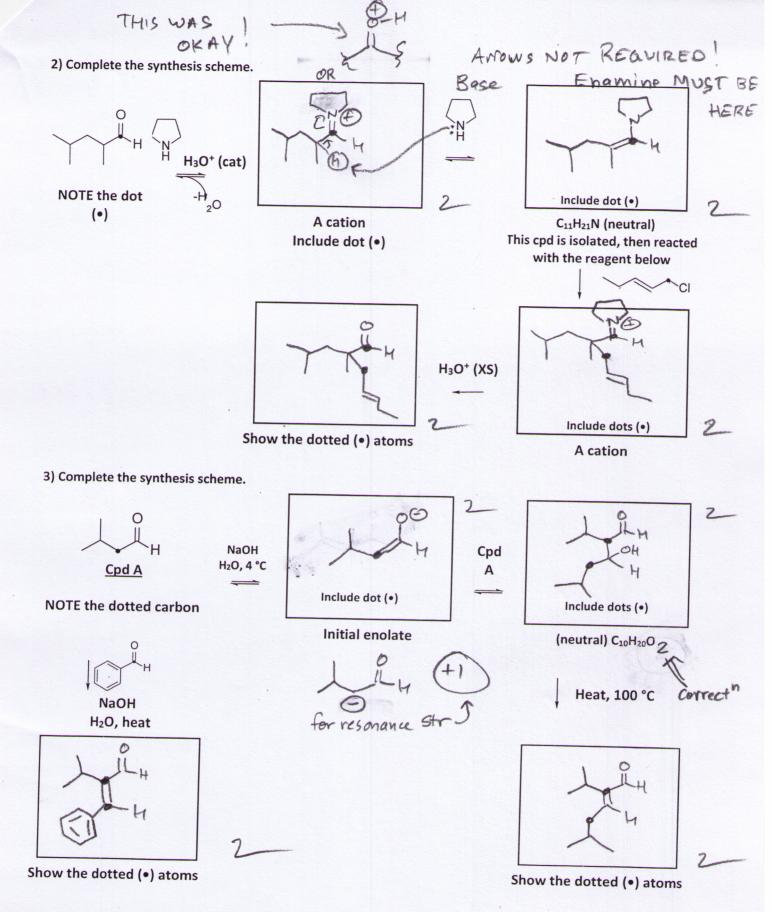
1) Complete the acid catalyzed (H_3O^+) mechanism for the enol/keto tautomerization. Place the appropriate small molecules in the smaller boxes drawn as <u>Lewis (line) structures e.g. (:X-Y^+-Z:)</u> that show bonds, lone electron pairs, and formal charges on the correct atoms. (11 pts)





Show the products of the following reactions





4) Complete the synthesis scheme. (Baeyer-Villiger)

HgSO₄ H₂SO₄ H₂O

2

Include dot (•)

F₃C O OH

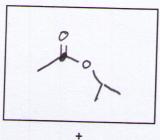
Include dot (•)

NOTE the dotted carbon

Most abundant at equil

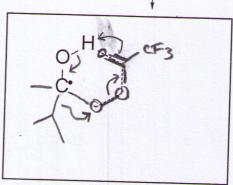
(A cation) – an activated intermediate

2



OHOC OH

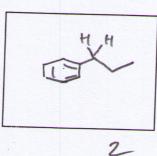
HINT: look at this product to figure out how to complete prior box



Partial scaffold for the 6-atom intermediate is provided: dotted C shown. Fill in the remaining bonds/atoms along with curved mechanism arrows to show access to the final products

5) Show the products.

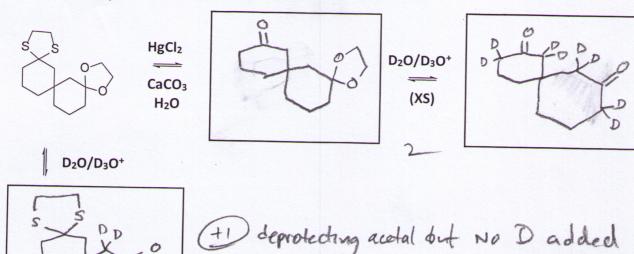
H₂N-NH₂ (Hydrazine) KOH



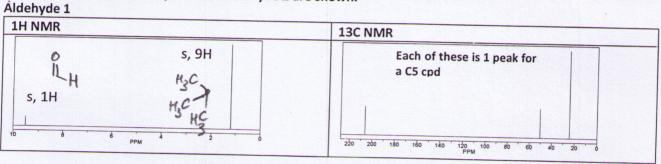
reaction?
N₂ H₂O

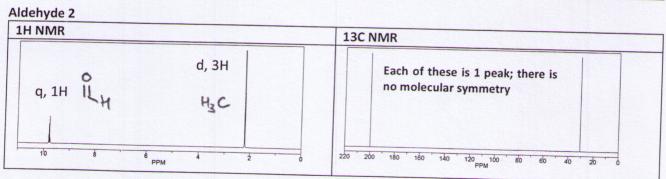
What are the 2 neutral byproducts of this

6) Complete the synthesis scheme.



7) Two different aldehydes are mixed in NaOH/H2O and heated at 100 C for 12 h. The NMR spectra for Aldehyde 1 and Aldehyde 2 are shown.





The final aldol condensation product isolated after heating has a molecular formula of C7H12O Derived the structures

