

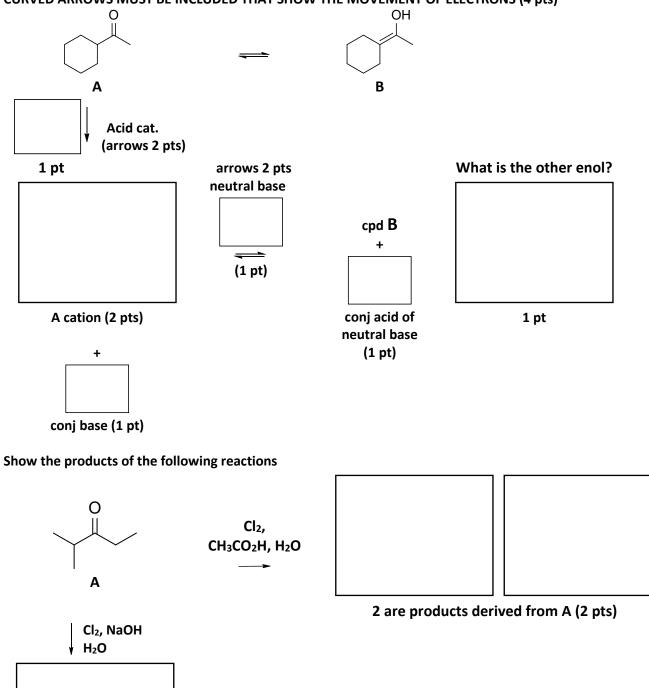
LEAVE THIS COVER SHEET ATTACHED TO THE Quiz!

- 1. ____/15
- 2. ____/16
- 3. ____/12
- 4. ____/11

TOTAL:____/ 50

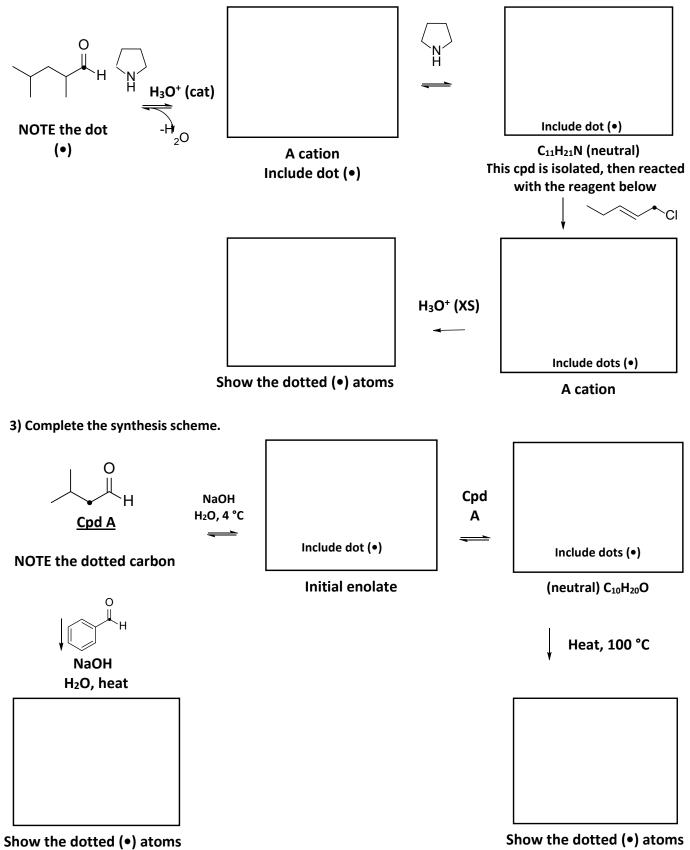
1) Complete the acid catalyzed (H_3O^+) mechanism for the enol/keto tautomerization. Place the appropriate small molecules in the smaller boxes drawn as <u>Lewis (line) structures e.g. (:X-Y^+-Z:)</u> that show bonds, lone electron pairs, and formal charges on the correct atoms. (11 pts)

CURVED ARROWS MUST BE INCLUDED THAT SHOW THE MOVEMENT OF ELECTRONS (4 pts)

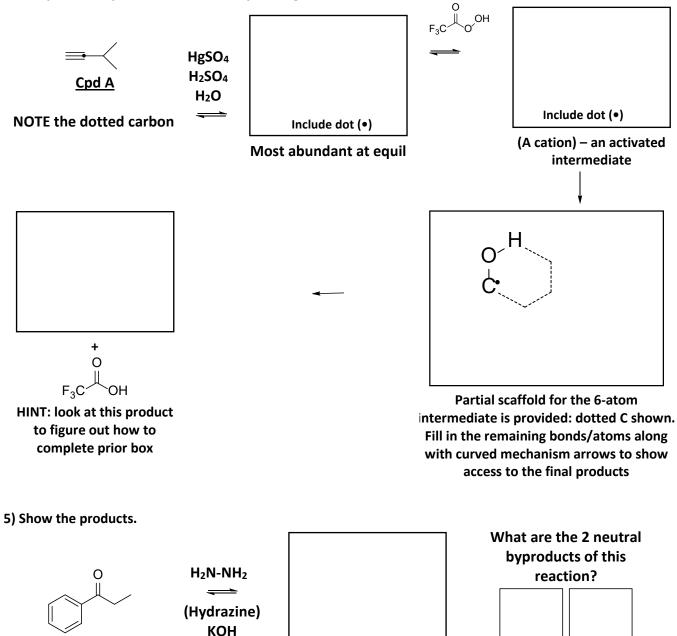


one product, 2 pts

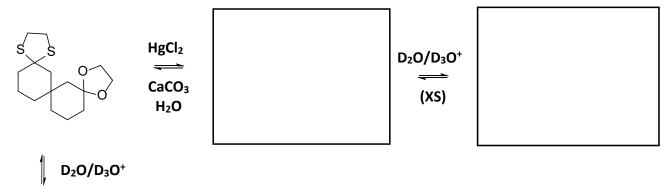
2) Complete the synthesis scheme.



4) Complete the synthesis scheme. (Baeyer-Villiger)

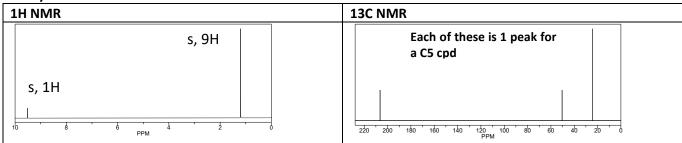


6) Complete the synthesis scheme.



7) Two different aldehydes are mixed in NaOH/H2O and heated at 100 C for 12 h. The NMR spectra for Aldehyde 1 and Aldehyde 2 are shown.

Aldehyde 1

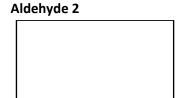


Aldehyde 2

1H NMR		13	CNMR
q, 1H	d, 3H		Each of these is 1 peak; there is no molecular symmetry
10	8 6 4 2 0 PPM	220	' 200 ' 180 ' 160 ' 140 ' 120 ' 100 ' 80 ' 60 ' 40 ' 20 ' 0 PPM

The final aldol condensation product isolated after heating has a molecular formula of C7H12O Derived the structures

Aldehyde 1



Aldol Condensation Product after Δ

