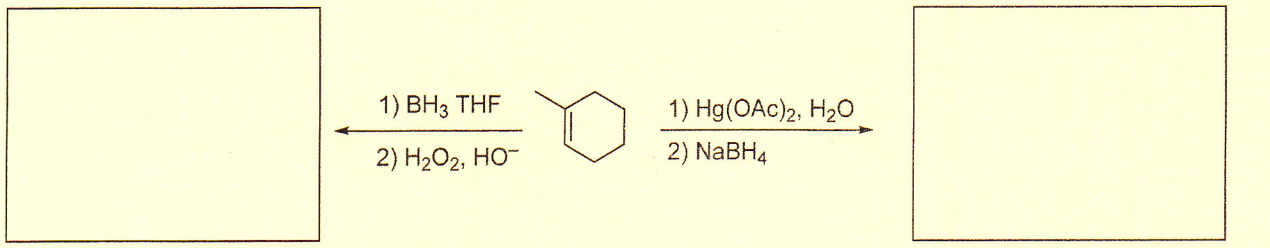
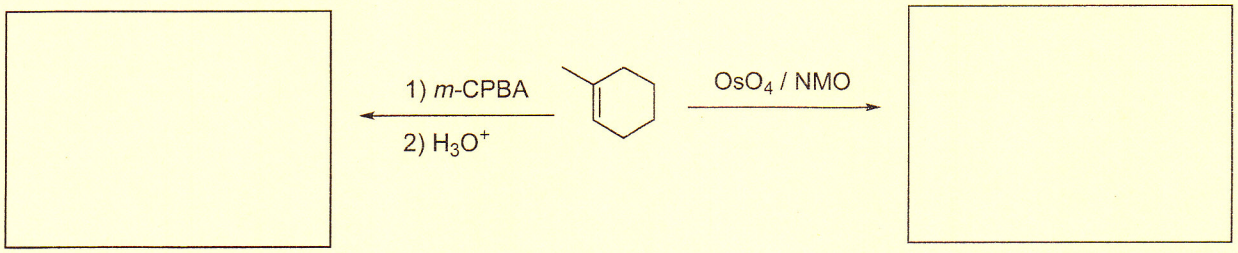
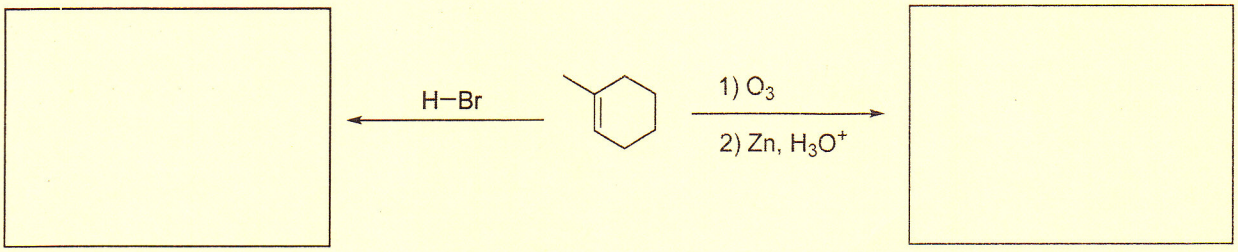
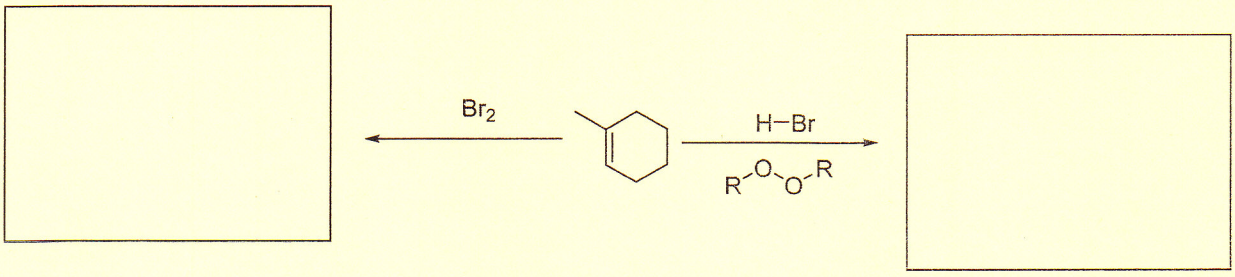
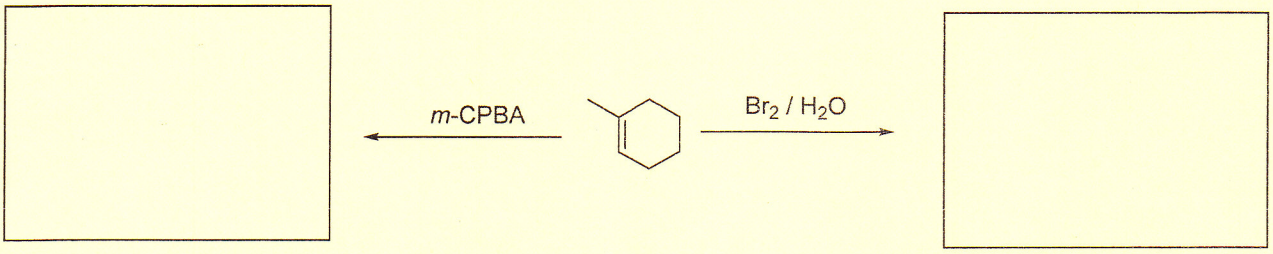
1) Draw the MAJOR (most abundant) product made in the reaction. IF NEW STEREOCENTER(S) IS/ARE MADE, DRAW ONLY ONE ENANTIOMER. (3 pts each, 30 points total)





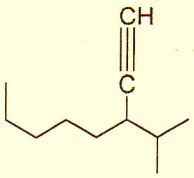






NAME_____

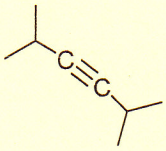
2) Name the following structure (IUPAC). USE iso, sec-, tert, OR neo in naming, if necessary. (2 pts)

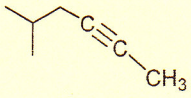


3) Draw the structure. (2 pts)

6-(1,1-dimethylethyl)-4-methyl-2-nonyne

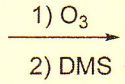
4) Predict the products of the following reaction. (3 pts per box, 6 pts total)

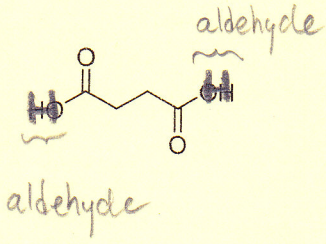




show stereochemistry

5) Draw the alkene reactant. (3 pts)





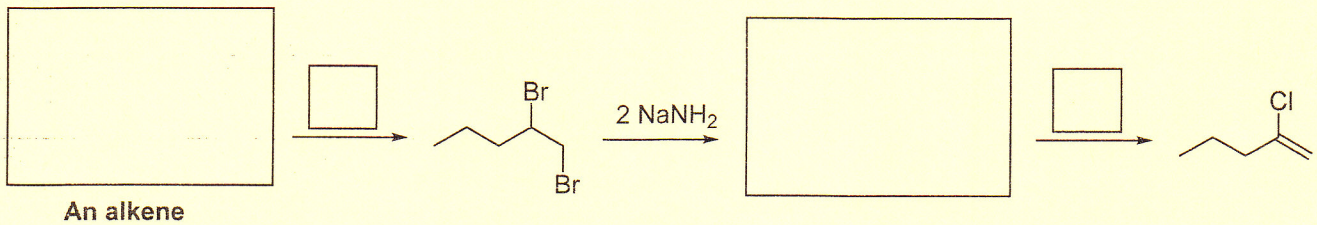
NAME_____

(A) ($NaOCH_3$) **OR** ($NaOEt$)	**(B)** Br_2, excess CH_3OH	**(C)** H_2O	**(D)** H_2, Lindlar catalyst
(E) HBr	**(F)** Na, NH_3 (liquid)	**(G)** 1 mol HCl	**(H)** 2 mol HCl
(I) 1) $Hg(OAc)_2$, H_2O 2) $NaBH_4$	**(J)** $KMnO_4$, NaOH (**cold**)	**(K)** H_2, Pd/C **or** H_2, Pt/C **or** H_2, Ni	**(L)** 2 mol Br_2 (in CH_2Cl_2 solvent)
(M) 1) OsO_4 2) $NaHSO_3$, H_2O	**(N)** 1) O_3 2) DMS (required!!) (DMS -= dimethylsufide) **(compare O)**	**(O)** 1) O_3 2) H_2O (no DMS required) dimethylsufide	**(P)** 1 mol Br_2 (in CH_2Cl_2 solvent)
(Q) 1) BH_3, THF 2) $HO^{\ominus}$, H_2O_2, H_2O	**(R)** $CHCl_3$, KOH (base)	**(S)** Tosyl Cl (TsCl), pyridine	**(T)** 1 mol $NaNH_2$
(U) HIO_4	**(V)** Br_2, excess H_2O	**(W)** 1) $HgSO_4$, H_2O, H_2SO_4,	**(X)** $H_3O^{\oplus}$ (23 °C) **(mild addition cond'ns)**
(Y) *m*-chloroperoxybenzoic acid (*m*CPBA)	**(Z)** H_2O, conc. H_2SO_4, Heat	**(AA)** CH_3Br	**(BB)** NBS, *hv* *N*-bromosuccinimide
(CC) 1) *m*-chloroperoxybenzoic acid (*m*CPBA) 2) $H_3O^{\oplus}$	**(DD)** HBr, ROOR (peroxides)	**(EE)** (*t*-BuOK) **OR** DBU **OR** DBN	**(FF)** 1 mol Cl_2 (in CH_2Cl_2 solvent)
(GG) 1) BH_3, THF 2) $HO^{\ominus}$, H_2O_2, H_2O	**(HH)** Na^+Br^- in DMSO	**(II)** 1) disiamylborane **OR** 9-BBN 2) $HO^{\ominus}$, H_2O_2, H_2O	**(JJ)** Cl_2 (in CH_2Cl_2 solvent)
(KK) 2 mol Cl_2 (in CH_2Cl_2 solvent)			

6) Synthesis: Complete the following synthetic schemes using the table above (**25 points**).
Fill in the **small boxes (1 point each; 7 points total)** with a letter corresponding to a reagent (Table above)
Fill in the **larger rectangles (2 points each; 18 points total)** with organic reactant

NEXT PAGE

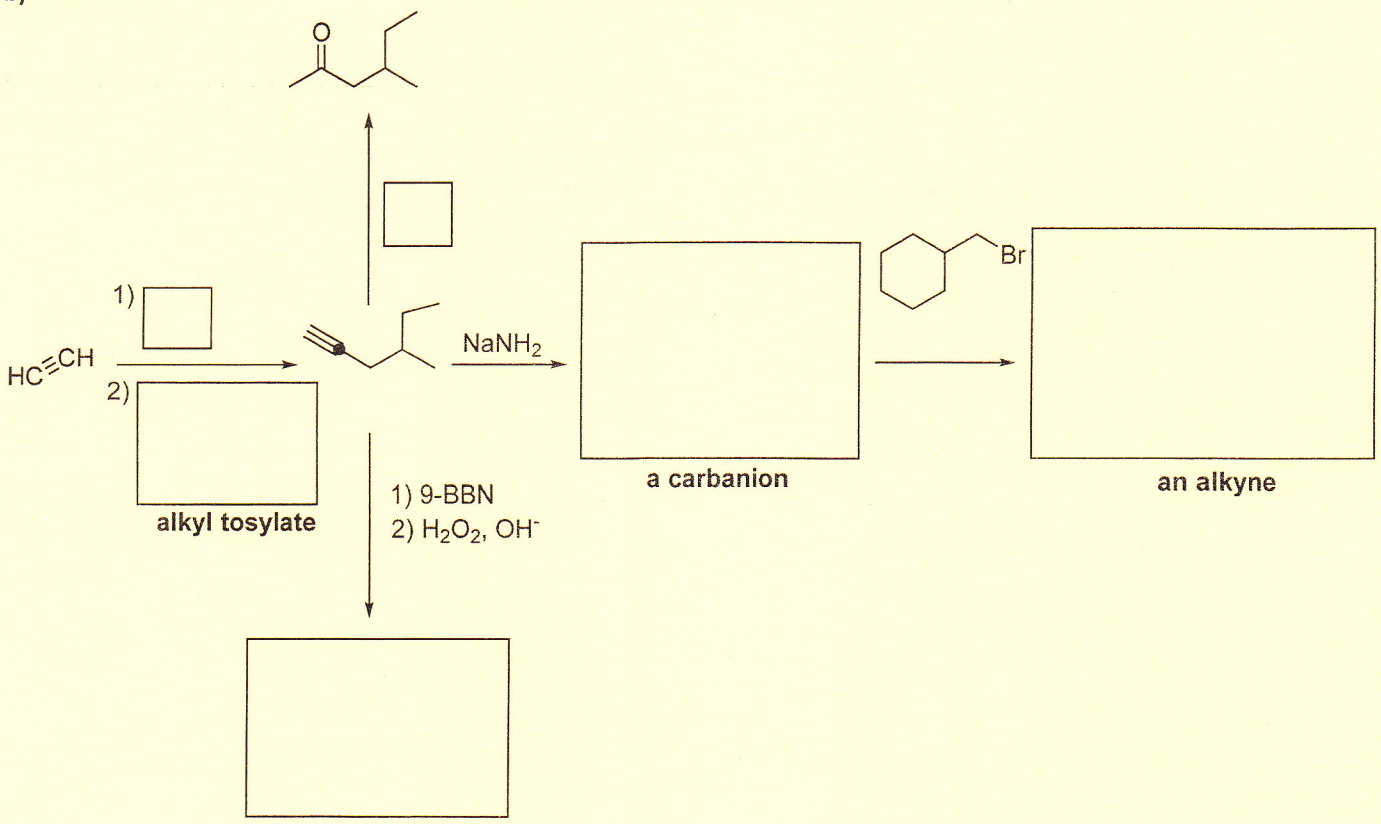
a)



An alkene

NAME_____

b)



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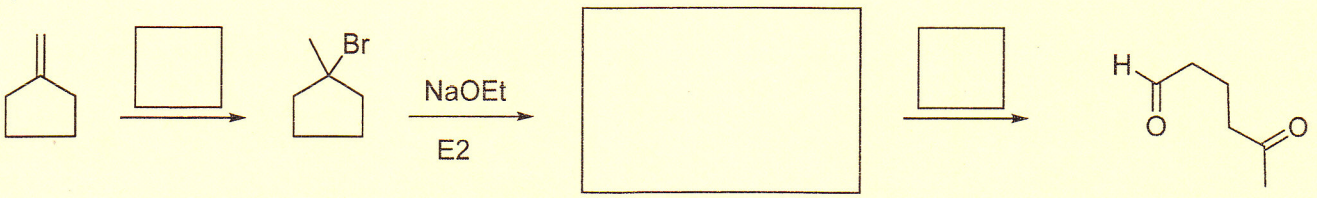
c)



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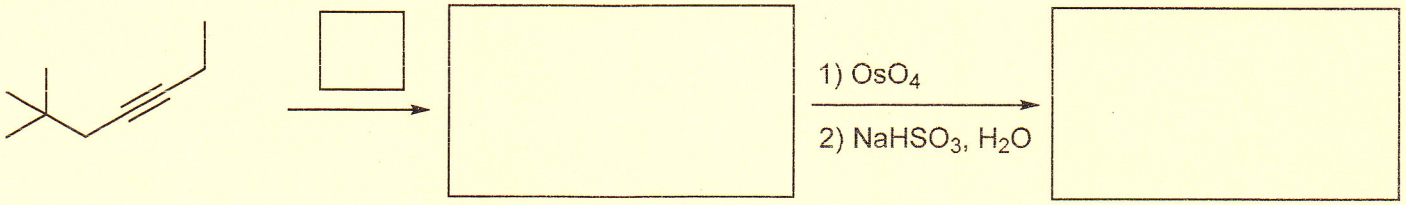
d)

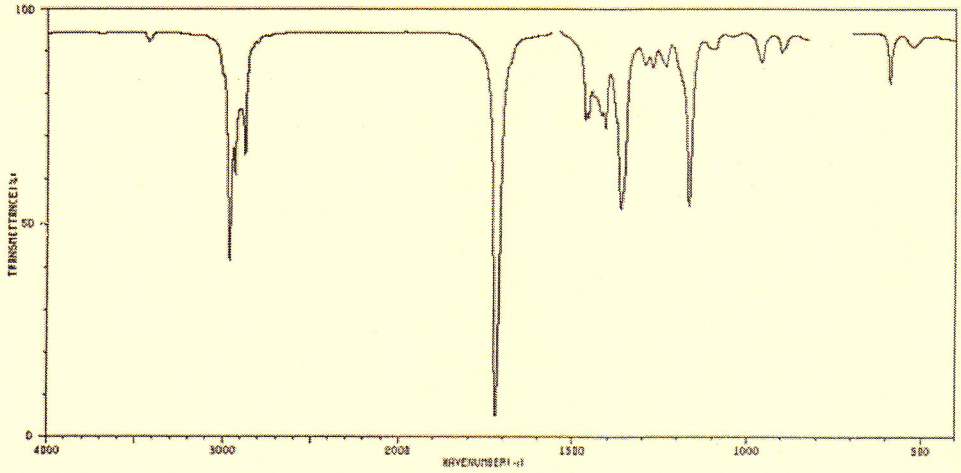


C9 <u>trans</u> alkene

NAME_____

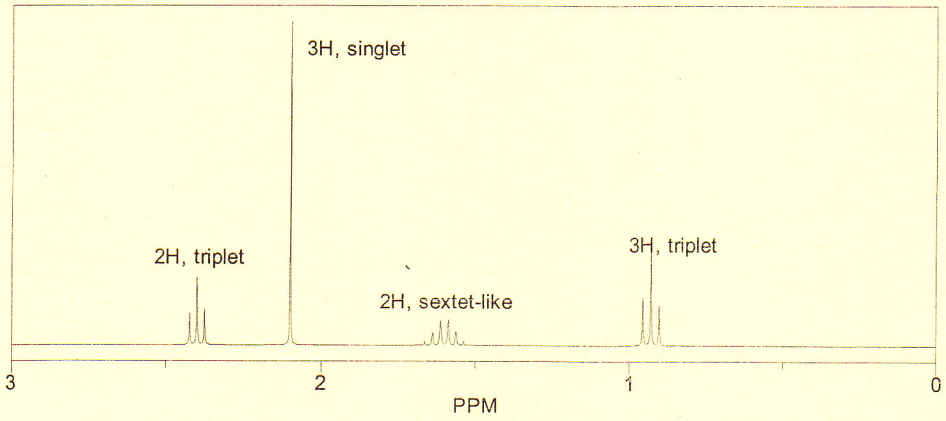
$C_5H_{10}O$ (Dous?)_____ **(1 pt)**

IR



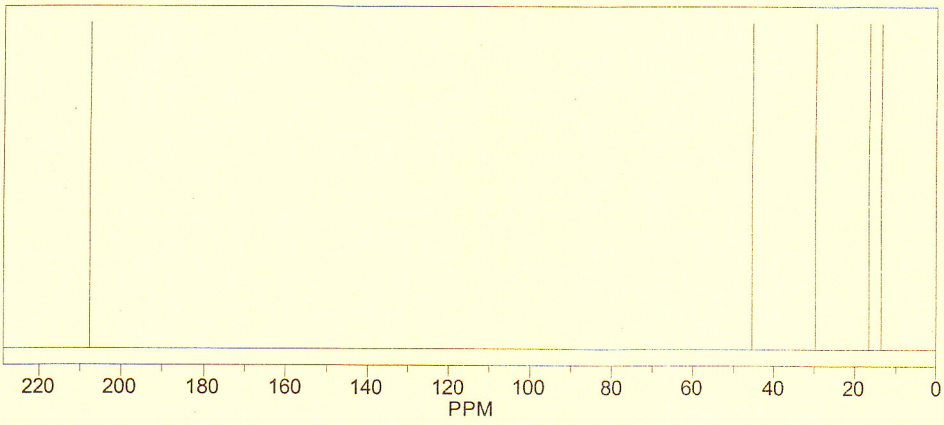
(2 pts)

1H NMR



(2 pts)

13C NMR



(2 pts)

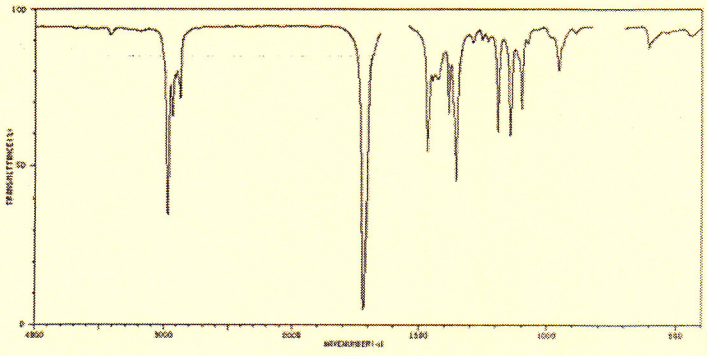
Structure Here

(3 pts)

NAME_____

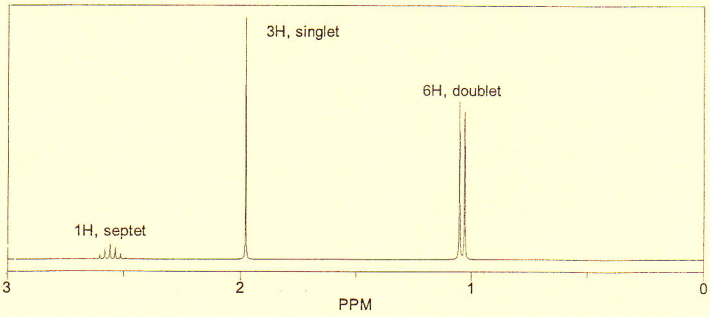
$C_5H_{10}O$ (Dous?)_____ (1 pt)

IR



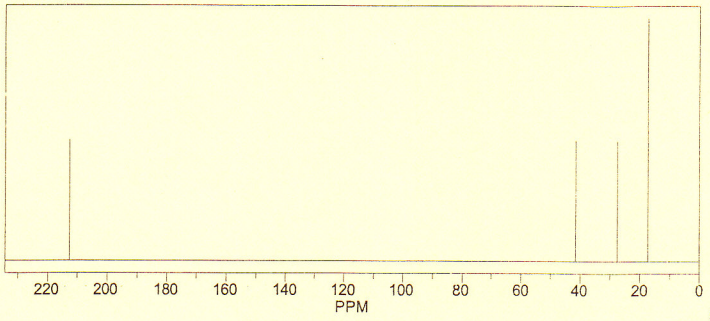
(2 pts)

1H NMR



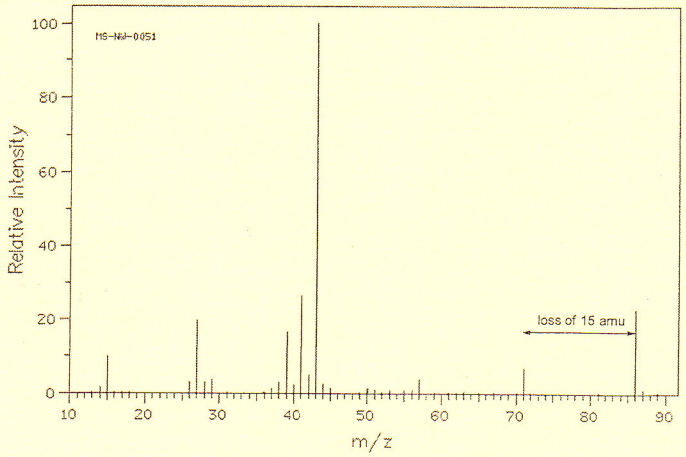
(2 pts)

13C NMR



(2 pts)

Mass Spect. Draw structure of fragment lost from M corresponding to 15 mass units_____ (2 pts)



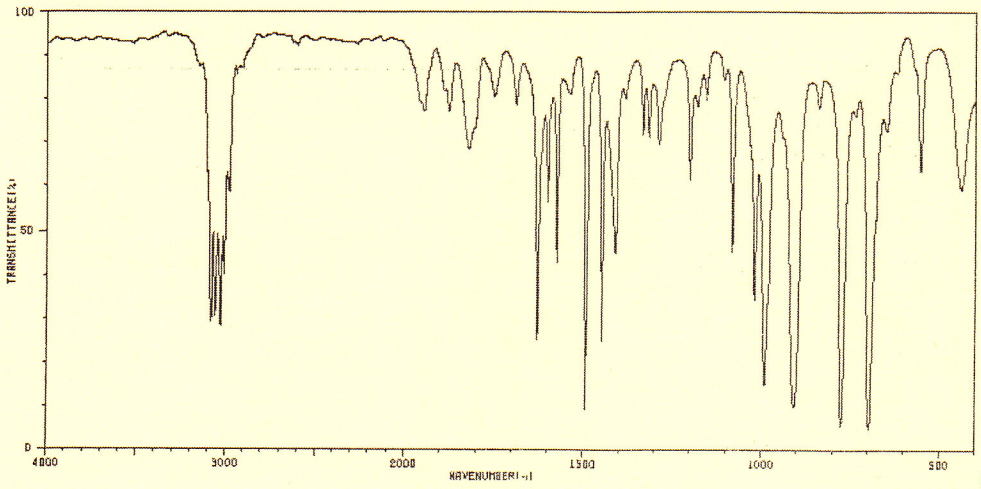
Structure Here

(3 pts)

NAME_____

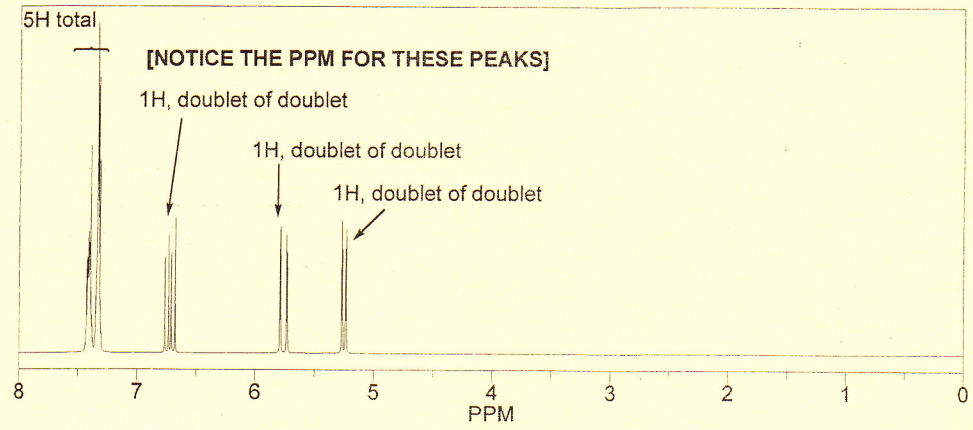
C_8H_8_____ Dous (1 pt)

IR



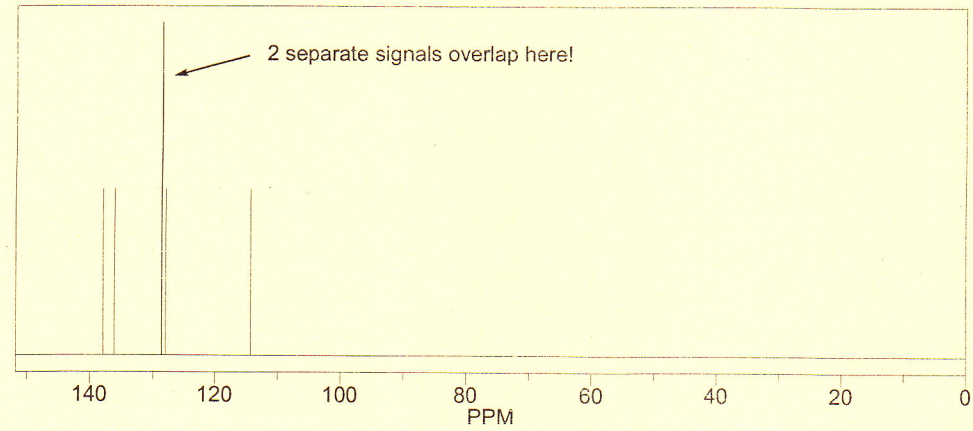
(2 pts)

1H NMR



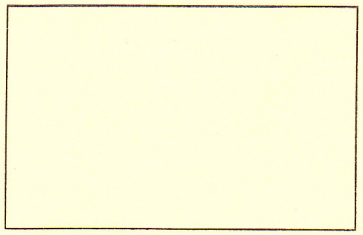
(2 pts)

13C NMR



(2 pts)

Structure Here



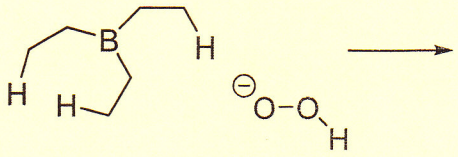
(3 pts)

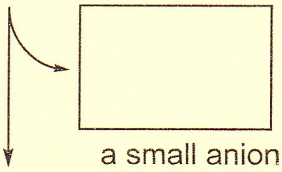
NAME_____

BONUS (MUST BE EXACTLY CORRECT FOR CREDIT!)
5 POINTS

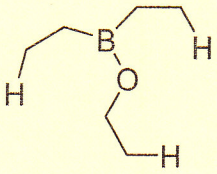
Draw anion intermediate below line

Draw curved mechanism arrow
to show how the deborylation reaction begins





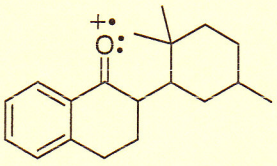
a small anion



DRAW CURVED MECHANISM ARROW ON ANION INTERMEDIATE TO SHOW FORMATION OF PRODUCTS ABOVE

BONUS (SHOW THE MECHANISM AND PRODUCTS OF THE MCLAFFERTY REARRANGEMENT BELOW) 5 POINTS

MECHANISM HERE



CHARGED FRAGMENT

NEUTRAL FRAGMENT

NAME_____