1. Name the following compounds (3 points each, 6 total)

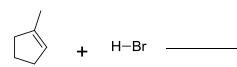
(1R,2R)-1,2-dichlorocyclopentane

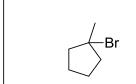
Include cis/trans



cycloocta-1,3-diene

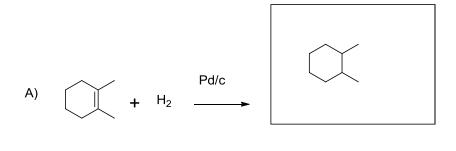
2. Complete to following addition reaction (3 Points).





3. Draw the energy diagram for the above reaction. Include the following; transition states, ΔH , reactants, products, and activation energies assuming an exothermic reaction (6 points).

4. Fill in the products for the following reactions (2 points each, 10 points total).



D)
$$+$$
 Br_2 $+$ Br Br

1,2 addition product

1,4 addition product