1) (2 pts) How many valence electro does	phosphorus (P) have?
2) (2 pts) What is the valence of Nitrogen	(N)?
3) (6 pts) Place a circle around the more electronegetive element in each of the following pairs.	
a) O or S b) Cs o	or Rb c) Cl or Br
4) (6 pts) Draw a Lewis dot formula for the following compounds.	
a) H ₂ CO ₃	b) H ₂ SO ₃
5) (9 pts) Provide the IUPAC for the followard	ewing compounds.
a) Br	
Br b)	
c)	