

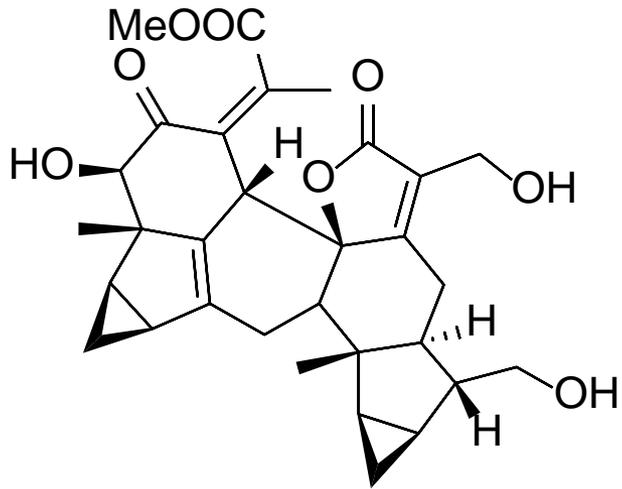
Total Synthesis of *Sarcandrolide J* and *Shizukaol D*

*Changchun Yuan+, Biao Du+, Heping Deng, Yi Man, and
Bo Liu**

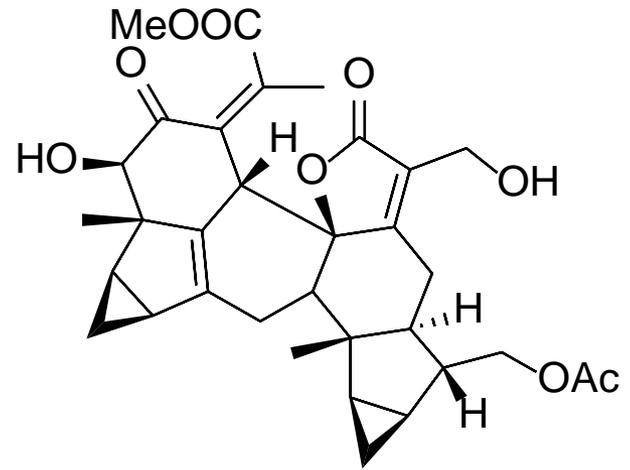
Angew. Chem. Int. Ed. **2017**, *56*, 637–640

Pohan Lin

Structure



sarcandrolide J



shizukaol D

Isolation & Biological Application

- Shizukaol D can activate AMP-activated protein kinase, increase ACC phosphorylation in HepG2 cells, and repress the growth of human liver cancer cells.
- Interestingly, sarcandrolide J shares the same
- molecular architecture as Shizukaol D, although they were isolated from *Sarcandra glabra* and *Chloranthus serratus*, respectively.

Isolation

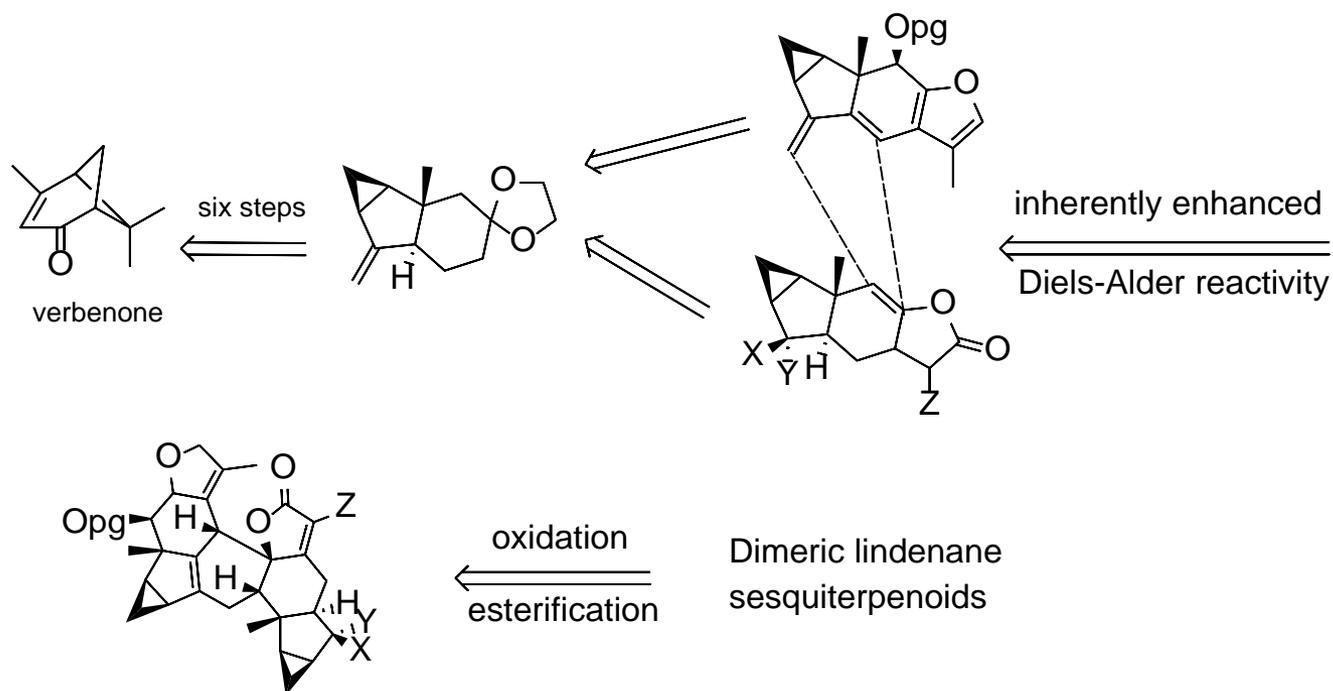


Sarcandra glabra

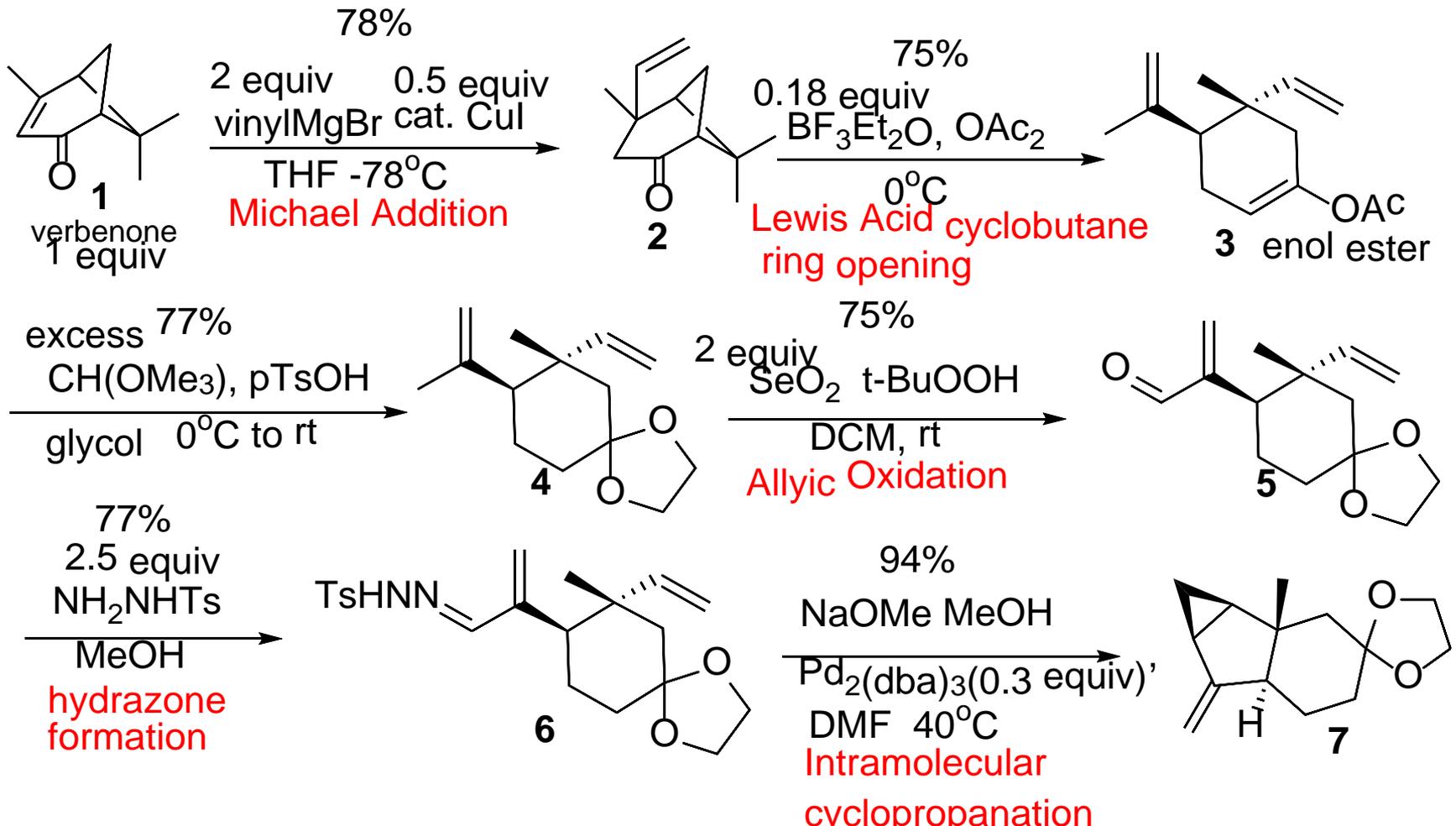


Chloranthus serratus

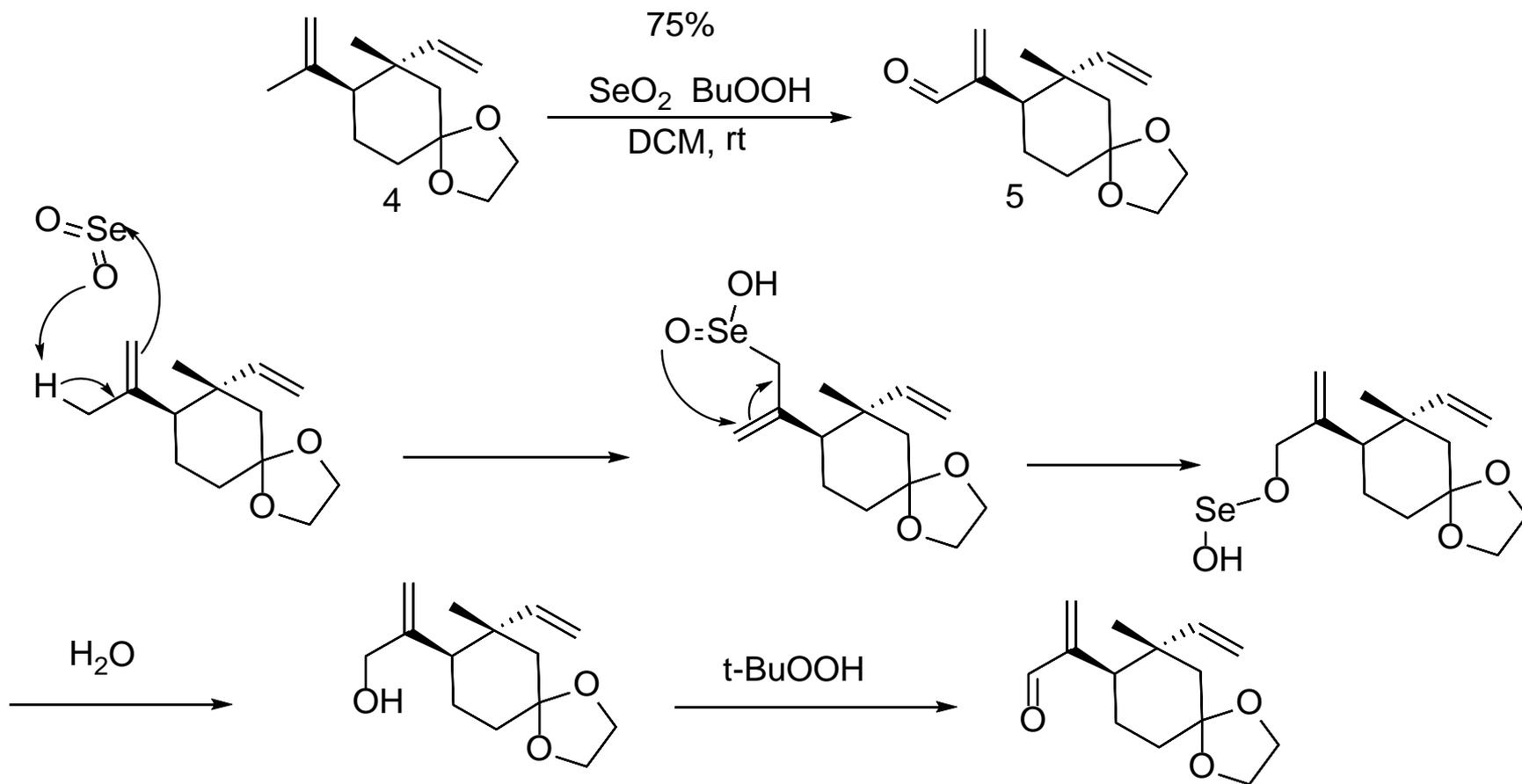
Retrosynthetic Analysis



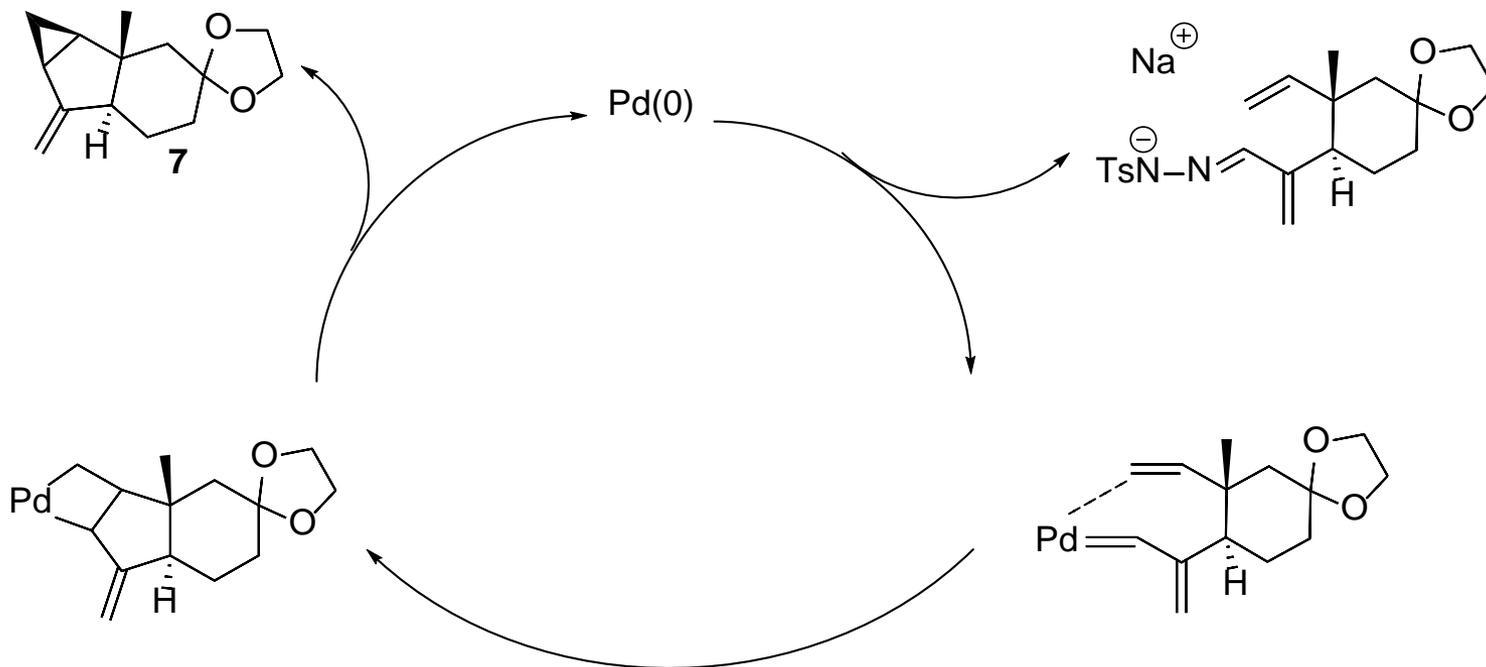
Precursor 7



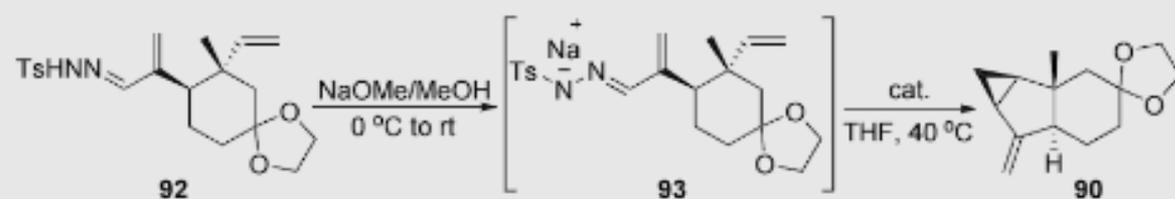
Allylic Oxidation



Cyclopropanation Mechanism



- *Chem. Commun.*, 2015,**51**, 6179-6182

TABLE 4 Optimization of Intramolecular Cyclopropanation of **92**^a

Entry	Base (1.5 eq)	Cat. (15 mol%)	Solvent	Yield (%) ^b
1	NaOMe	–	DMF	0
2	NaOMe	AgOTf ^c	DMF	0
3	NaOMe	Cu(acac) ₂ ^c	DMF	0
4	NaOMe	Ir(cod)Cl ^c	DMF	37
5	NaOMe	Rh ₂ (OAc) ₄	DMF	63
6	NaOMe	Pd(cod)Cl ₂ ^c	DMF	55
7	NaOMe	Pd(MeCN) ₂ Cl ₂ ^c	DMF	48
8	NaOMe	Pd ₂ dba ₃ ·CHCl ₃	DMF	73
9	NaOMe	Pd(dba) ₂	DMF	42
10	NaOMe	Pd(PPh ₃) ₄	DMF	0
11	NaOMe	Pd₂dba₃	DMF	81
12	NaOMe	Pd ₂ dba ₃ ^d	DMF	45
13	NaOMe	Pd₂dba₃^e	DMF	88

14	NaH ^f	Pd ₂ dba ₃	DMF	65
15	NaHMDS ^f	Pd ₂ dba ₃	DMF	49
16	KHMDS ^f	Pd ₂ dba ₃	DMF	63
17	LiHMDS ^f	Pd ₂ dba ₃	DMF	52
18	NaOMe	Pd ₂ dba ₃	DMA	36
19	NaOMe	Pd ₂ dba ₃	NMP	55
20	NaOMe	Pd ₂ dba ₃	Ether	31
21	NaOMe	Pd₂dba₃^e	THF	65
21	NaOMe	Pd ₂ dba ₃	DCM	50
22	NaOMe	Pd ₂ dba ₃	Toluene	39

^aConditions: (1) **92** base/MeOH, 5 min at 0 °C then 15 min at rt, remove solvent; (2) 15 mol% metal catalyst, DMF, 35 °C, TLC monitoring until full conversion of **92**.

^bIsolated yield of **90**.

^c30 mol%.

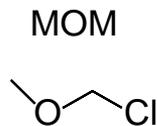
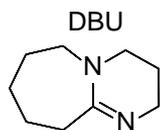
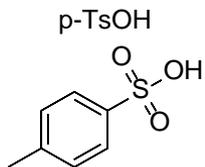
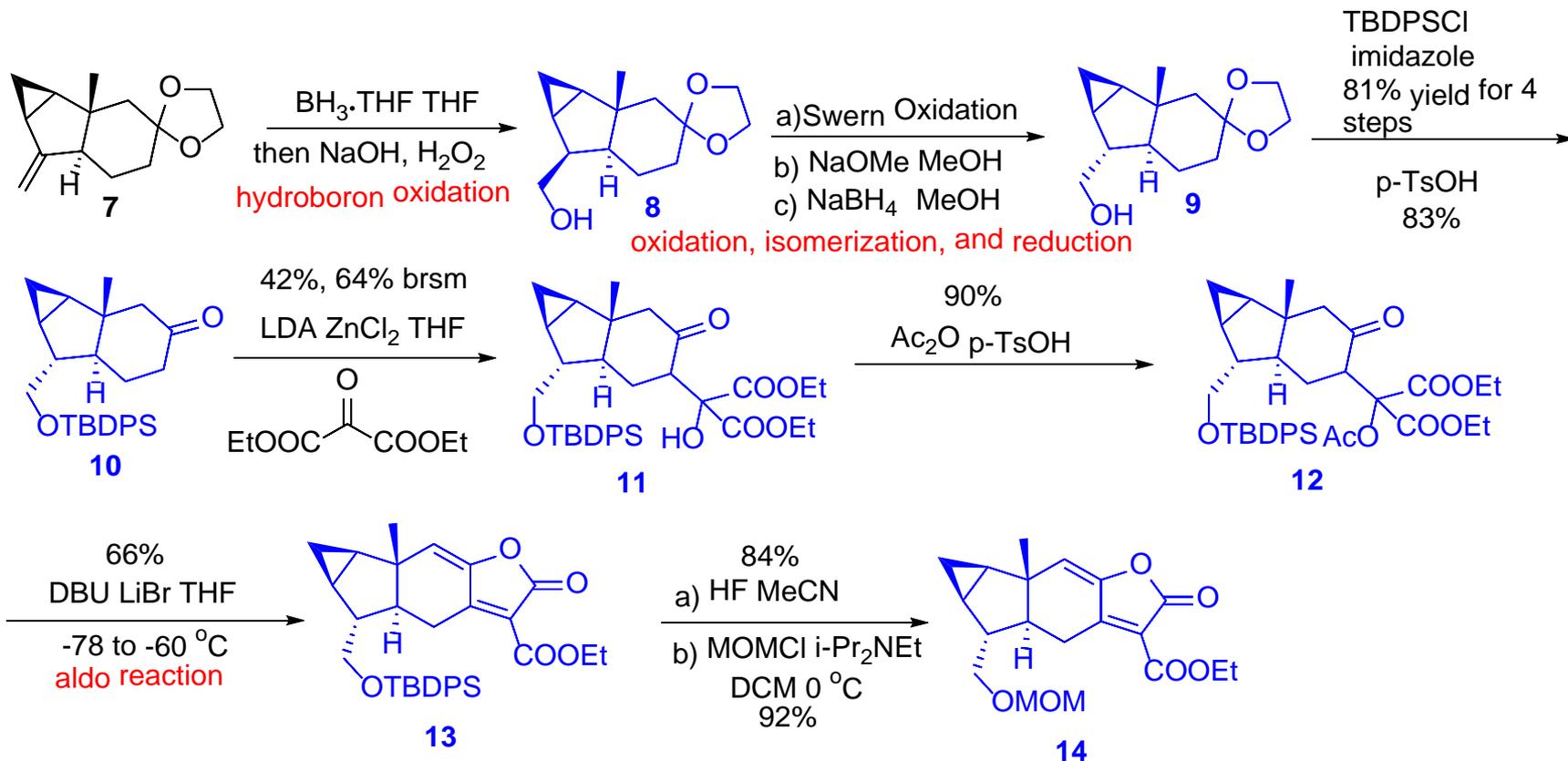
^d7 mol%.

^e50 mol%.

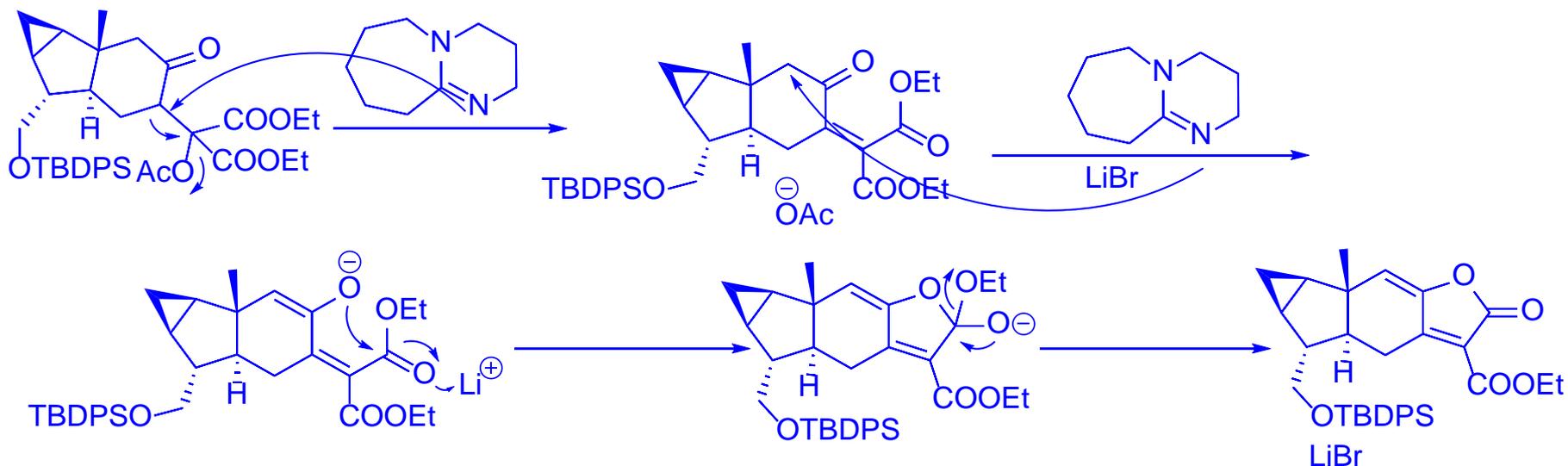
^fTHF was used instead of MeOH.

Chem. Commun., 2015,**51**, 6179-6182

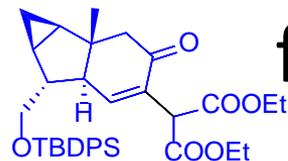
Precursor 14



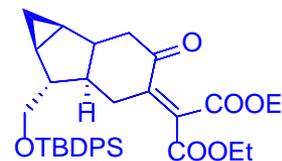
Ring formation Mechanism



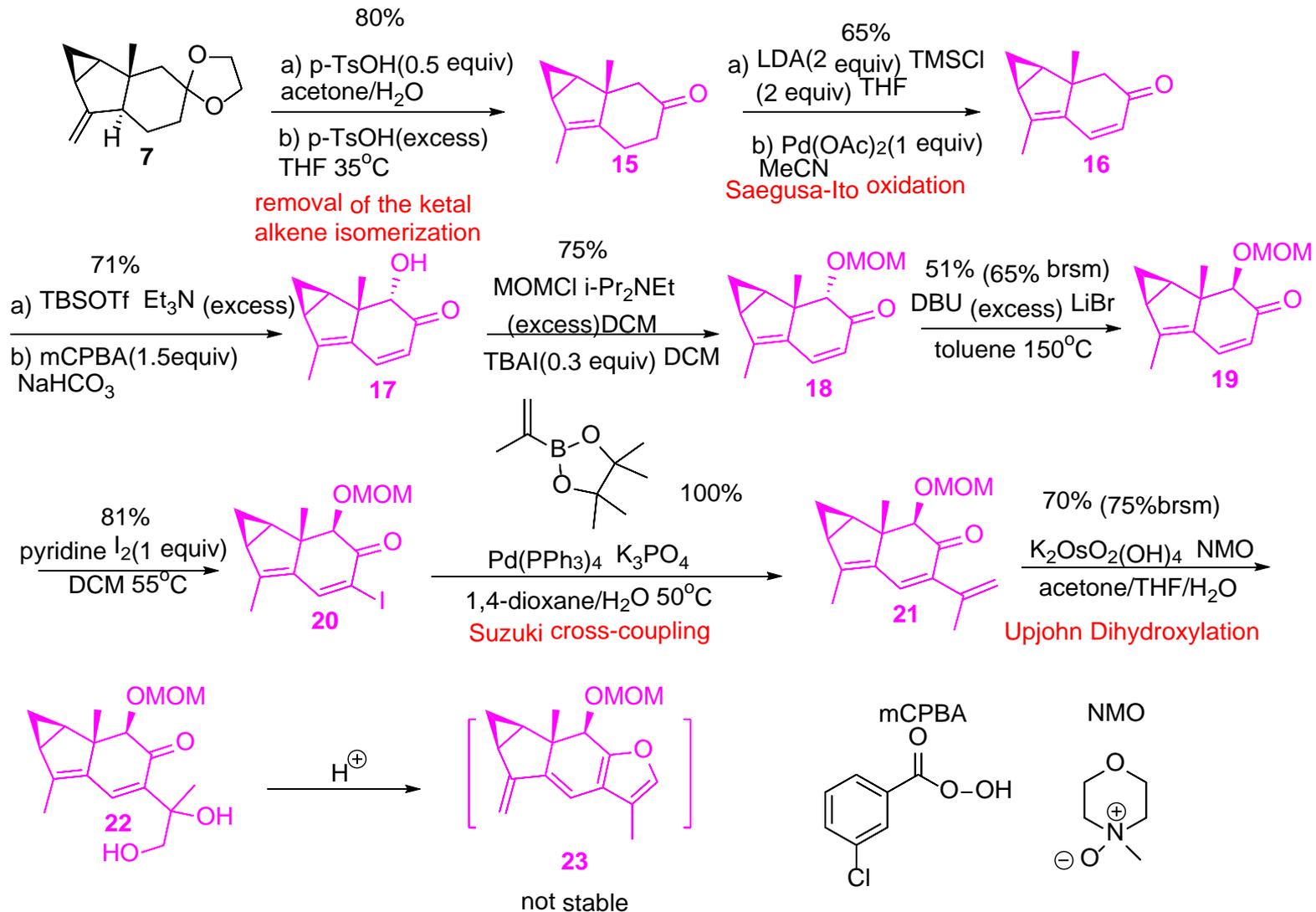
The presence of LiBr is crucial to avoid the competitive formation of



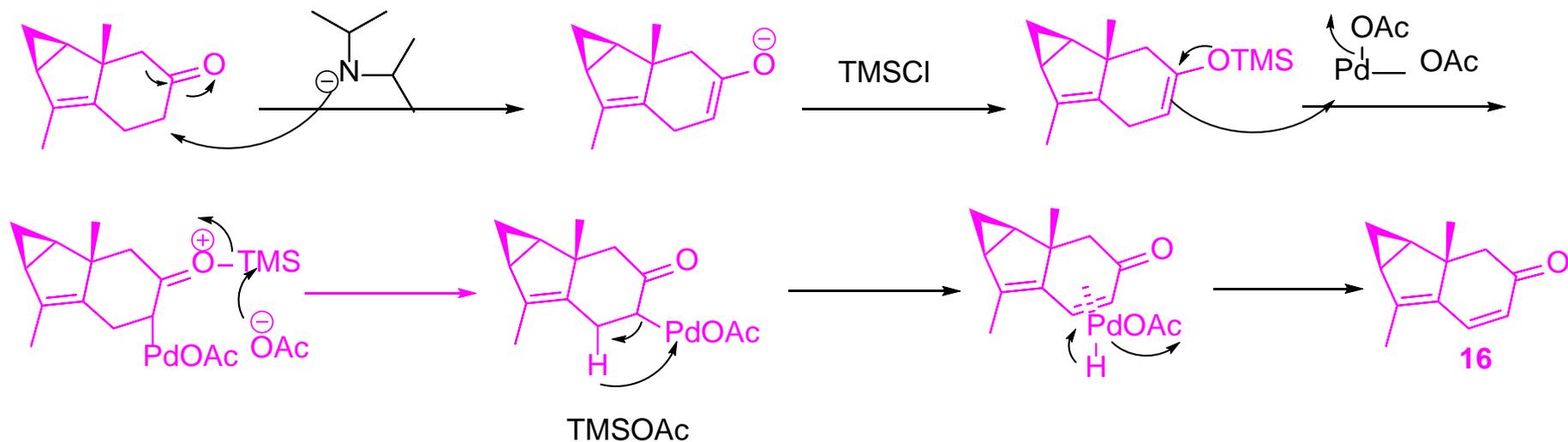
from



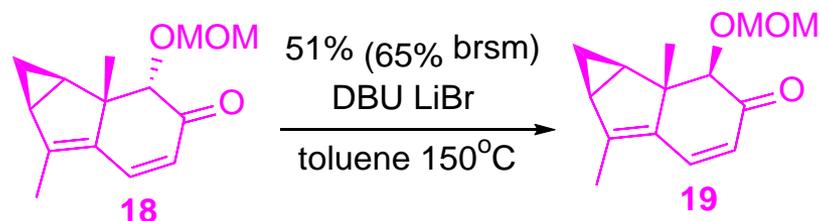
Precursor 22



Saegusa-Ito oxidation *Atsushi Nishida Modification*

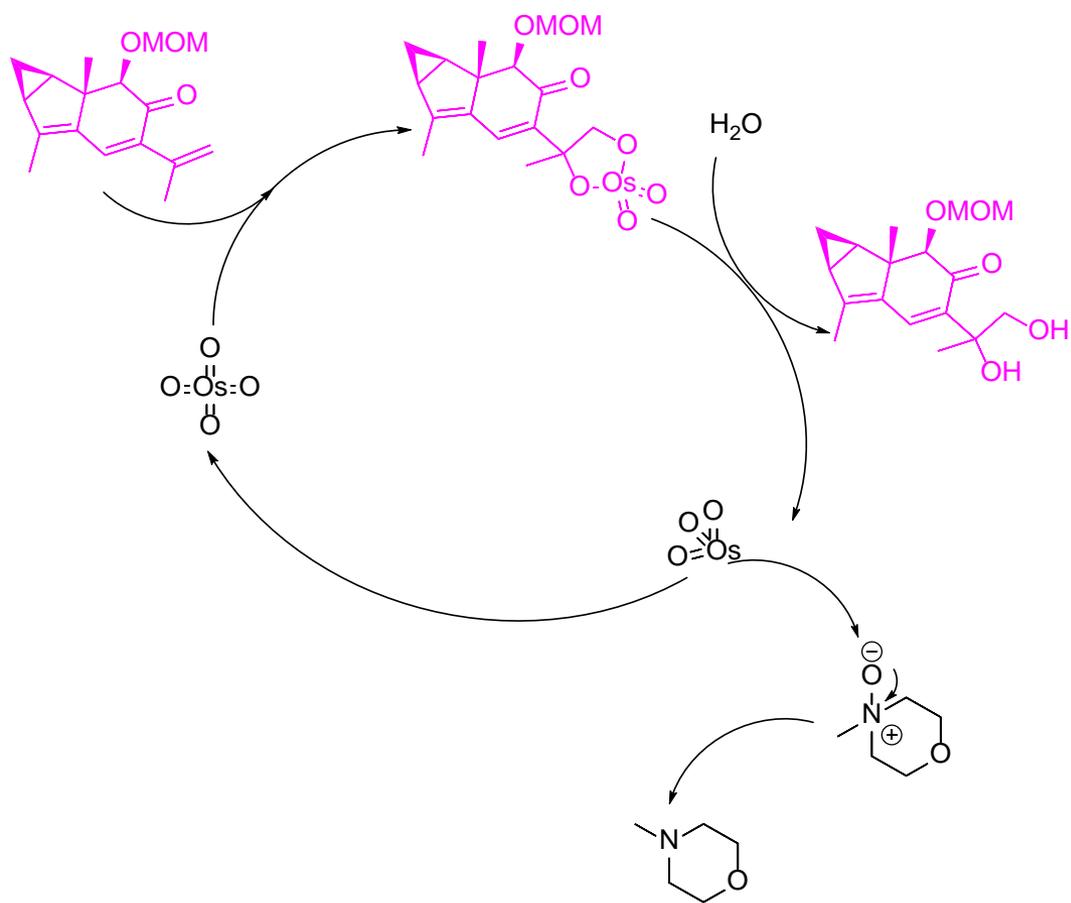
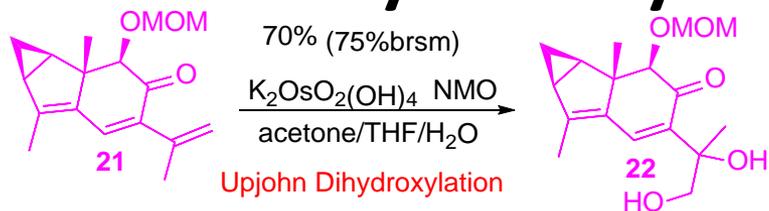


Stereochemistry Inversion

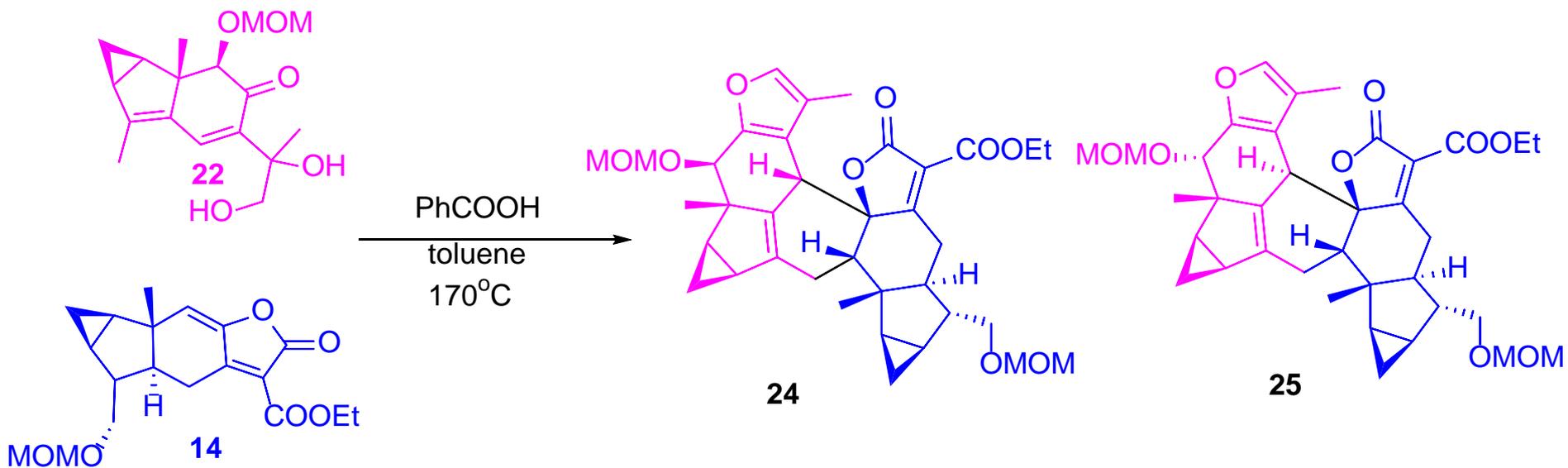


- 18 was subjected to thermal conditions in the presence of both DBU and LiBr to achieve equilibrium.
- The presence of a catalytic amount of a lithium salt (20 mol%) was essential. Otherwise, only trace amounts of 19 could be obtained, even with 10 equivalents of DBU

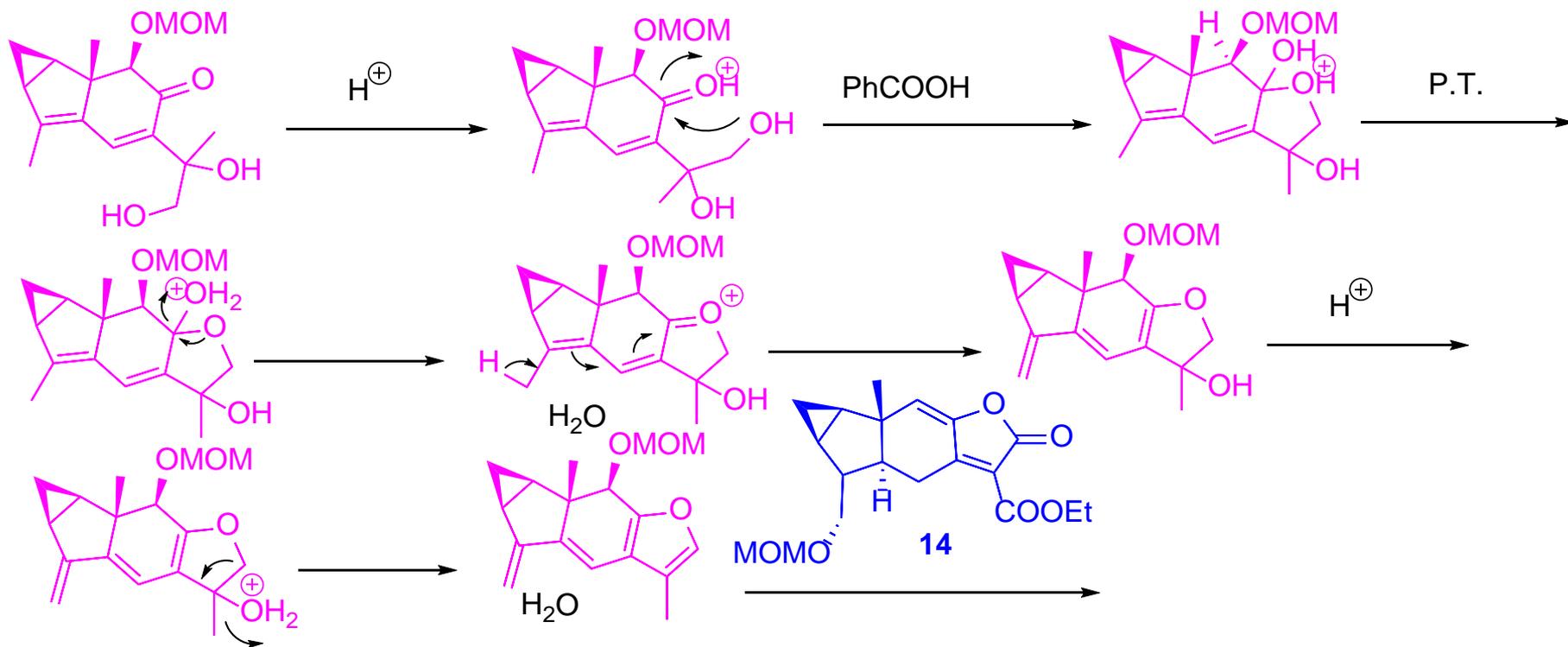
Upjohn Dihydroxylation



Diels-Alder[4+2]

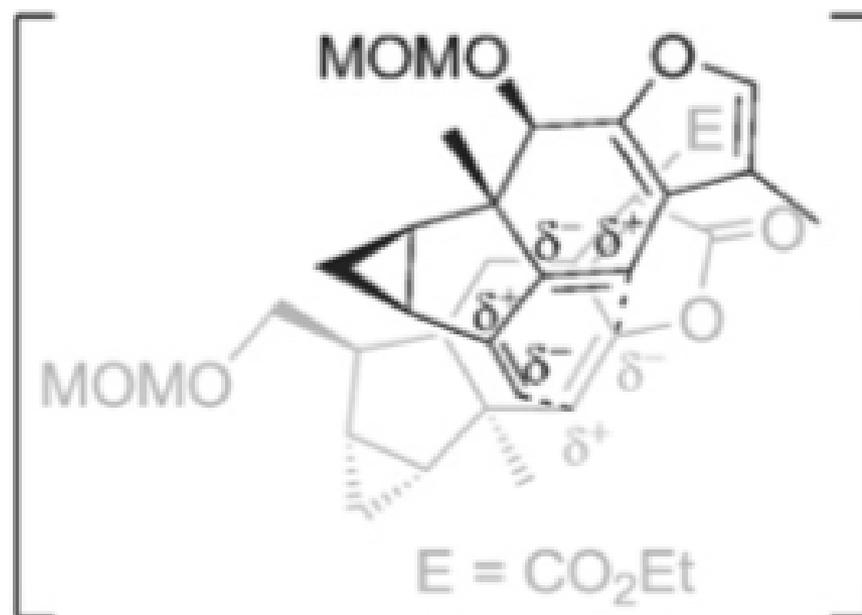


Mechanism of the product 24



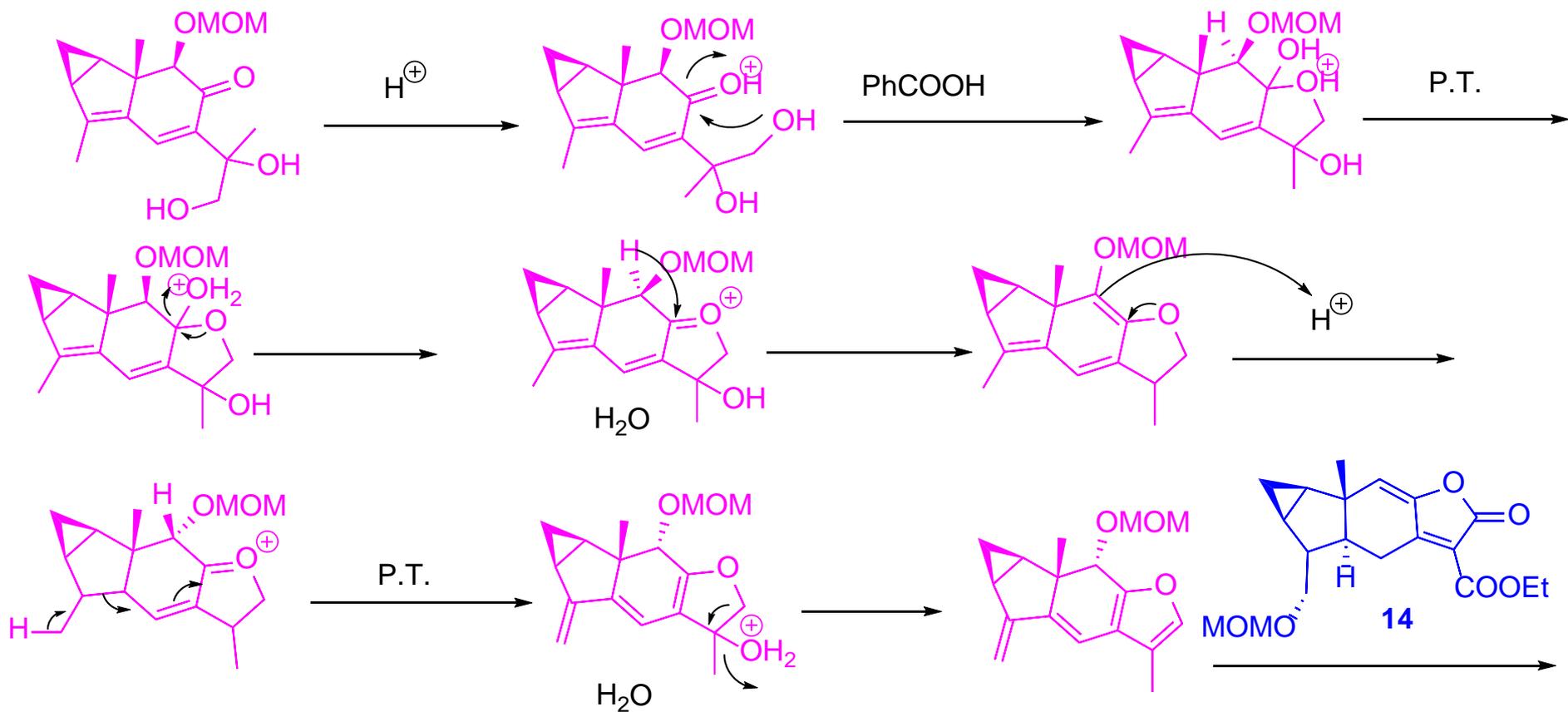
Mechanism of the product 24

endo-type Diels-Alder



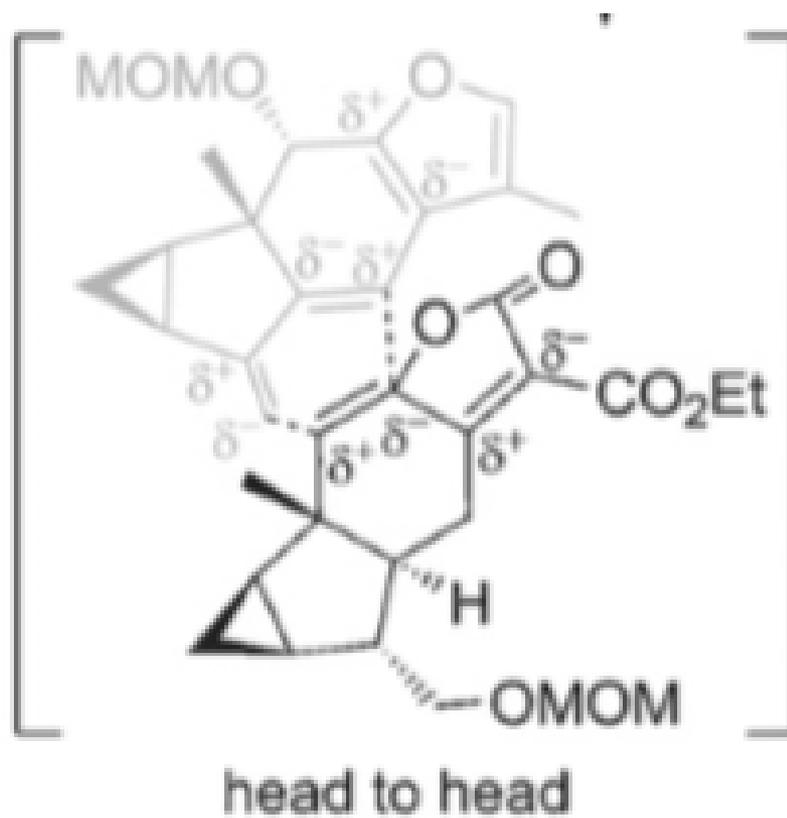
head to head

Mechanism of the product 25

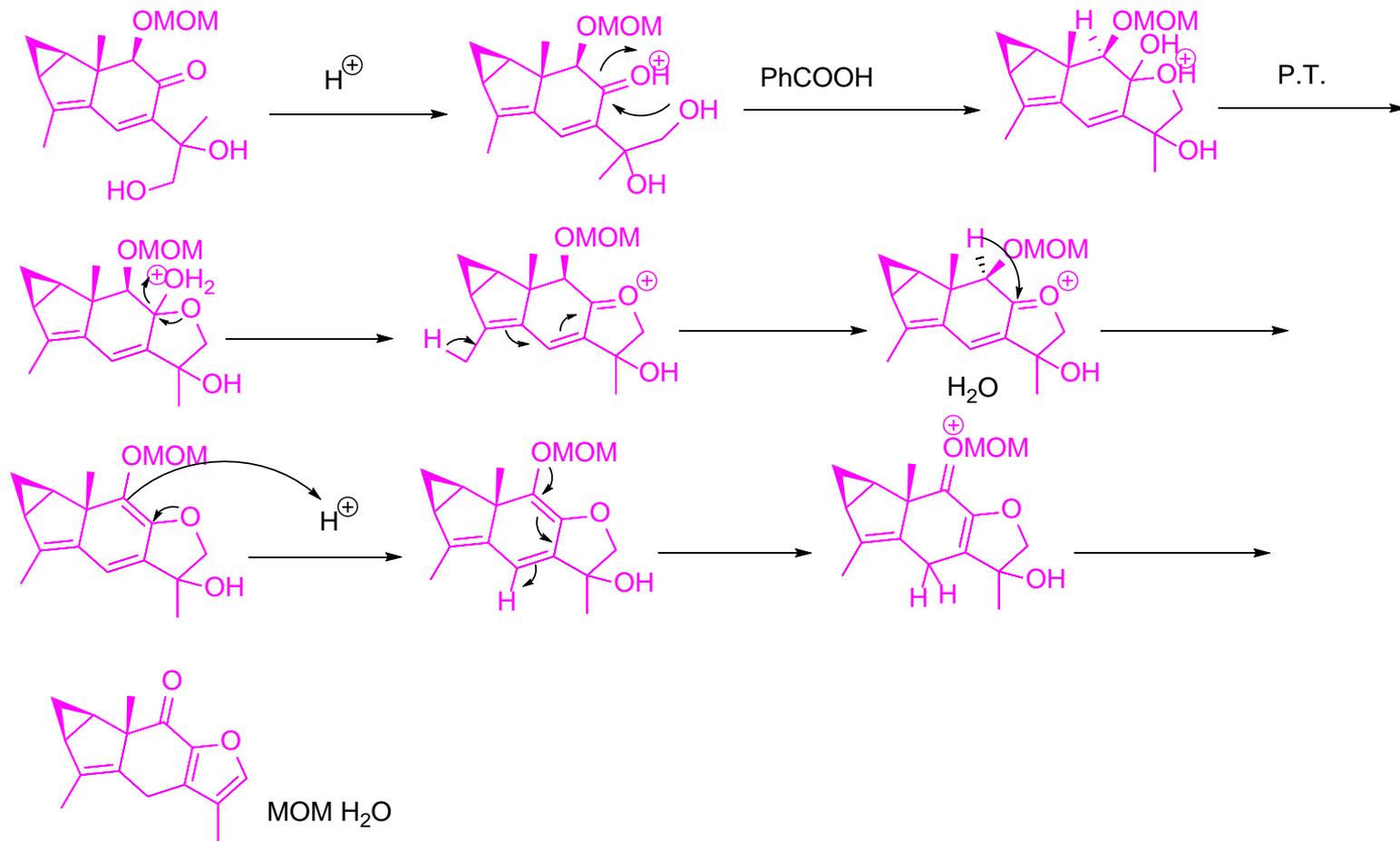


Mechanism of the product 25

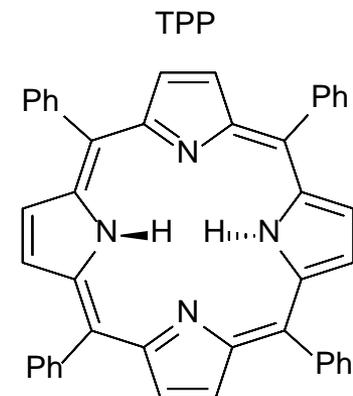
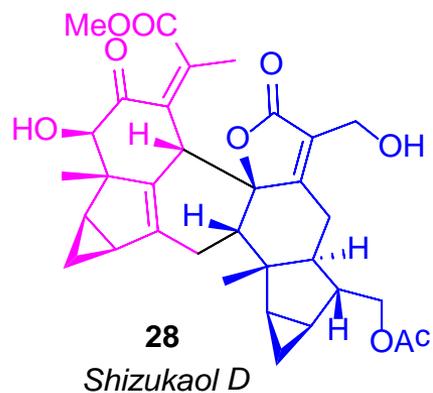
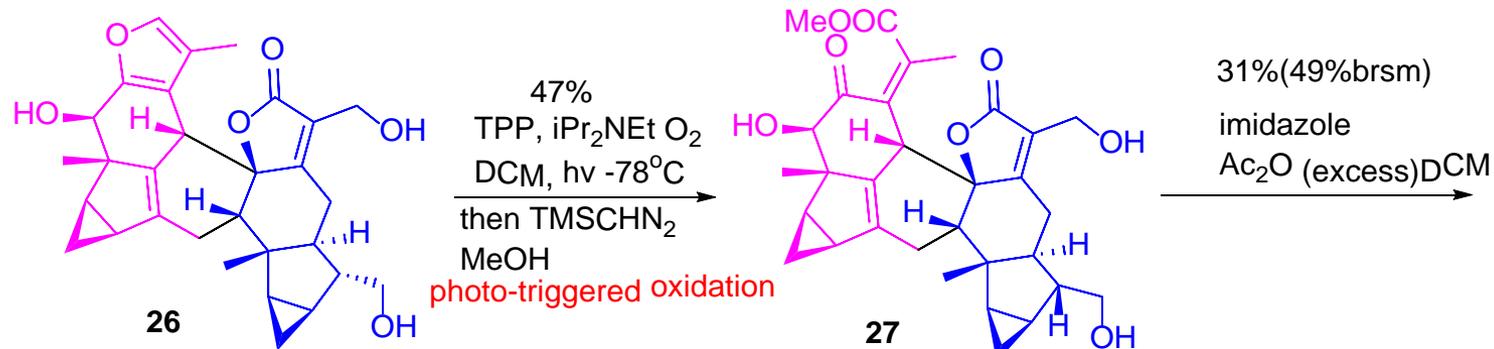
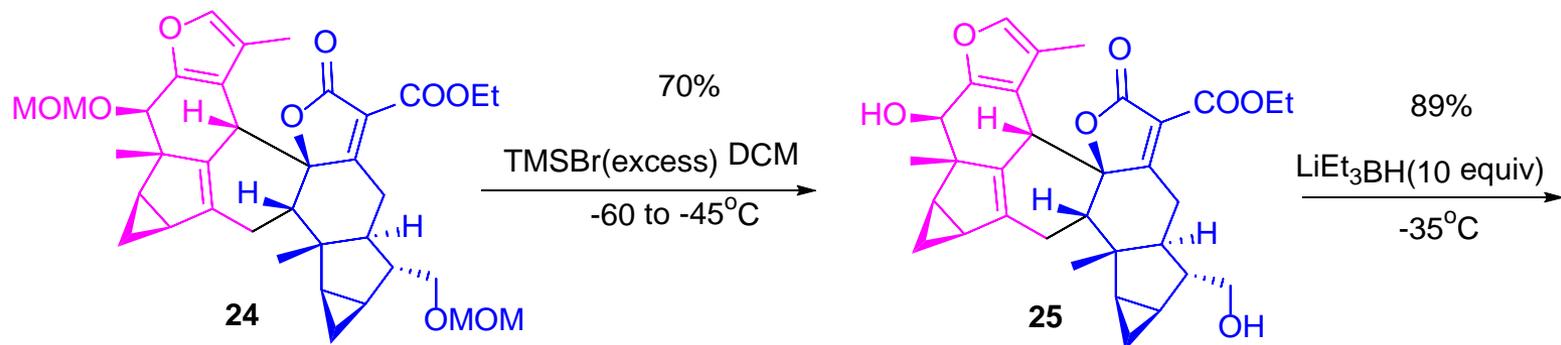
exo-type Diels-Alder



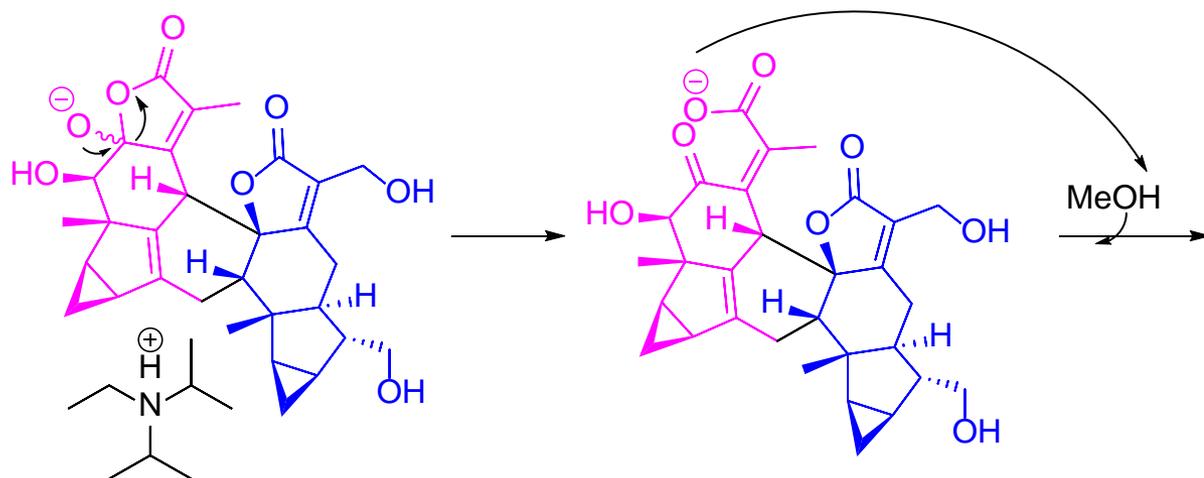
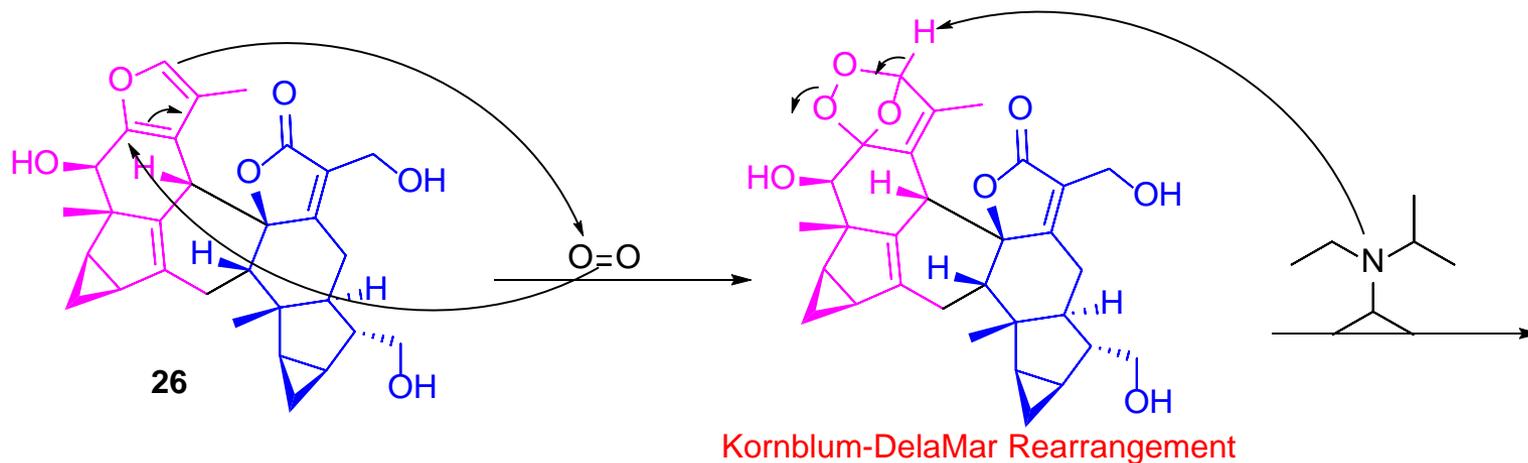
Another Byproduct Mechanism



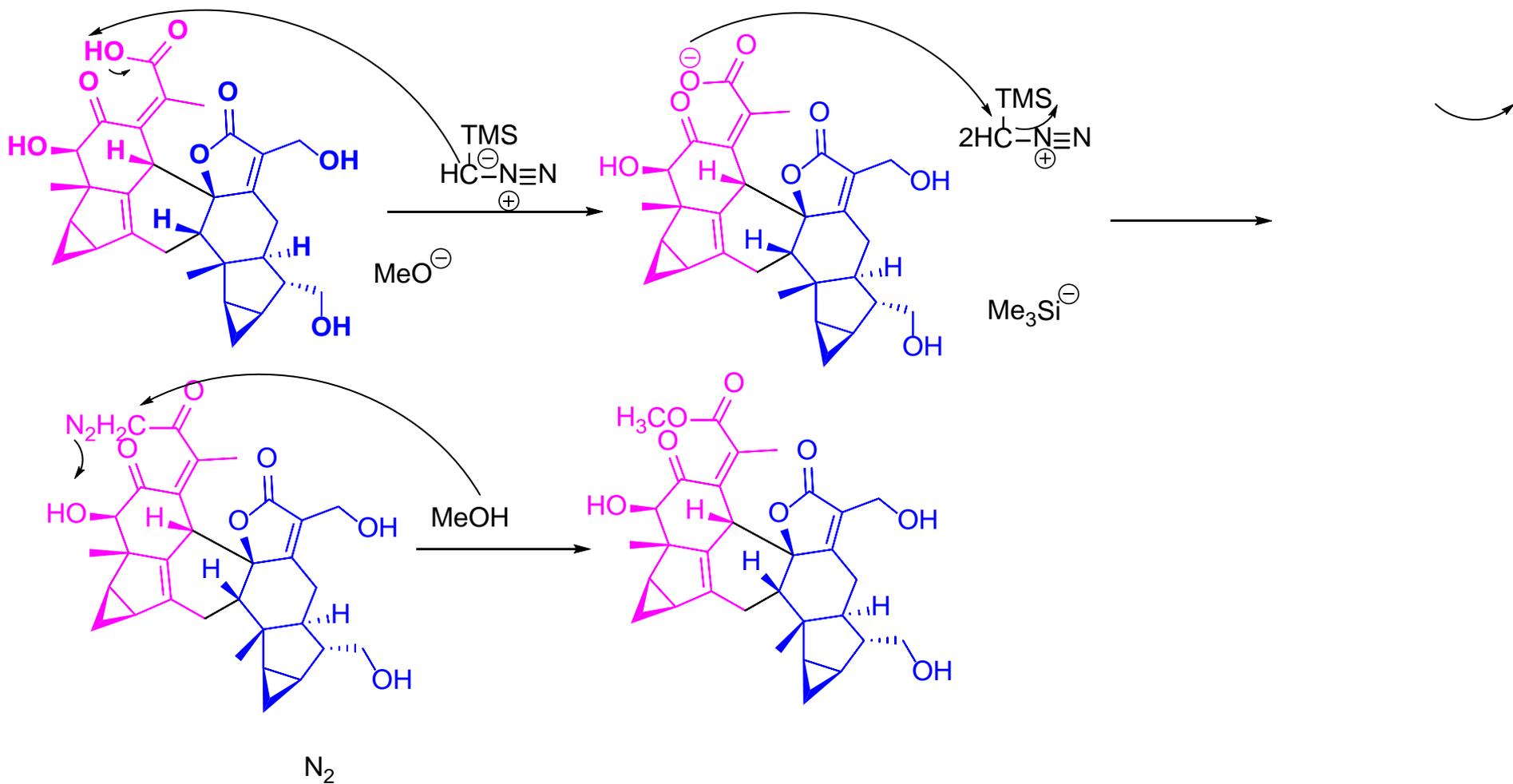
Final steps



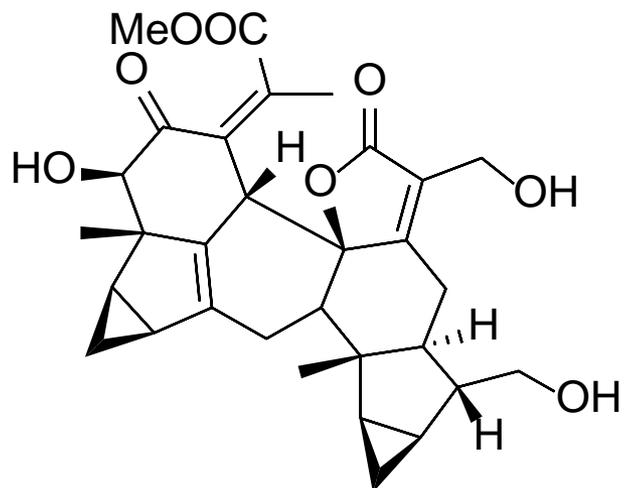
Mechanism of the furan ring opening



Mechanism of the furan ring opening

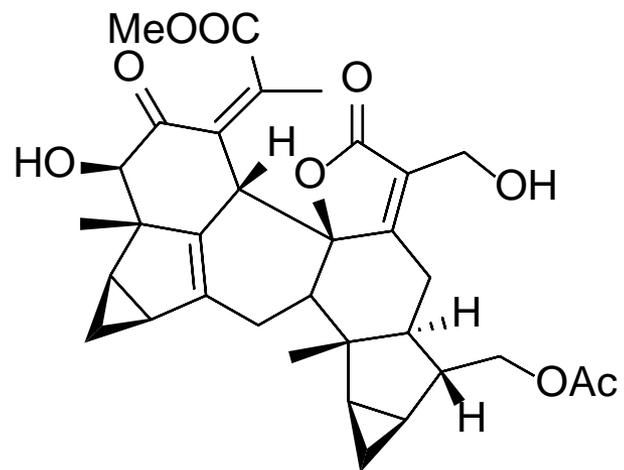


Overall Yield



sarcandrolide J

1.008%



shizukaol D

0.665%

- Thank You!