

# The Journal of Organic Chemistry

## Standard Abbreviations and Acronyms

$\alpha$	observed optical rotation in degrees	cod	1,5-cyclooctadiene
[ $\alpha$ ]	specific rotation [expressed without units; the units, (deg·mL)/(g·dm), are understood]	compd	compound
$\text{\AA}$	angstrom(s)	concd	concentrated
Ac	acetyl	concen	concentration
acac	acetylacetone	COSY	correlation spectroscopy
ADP	adenosine 5'-diphosphate	cot	1,3,5,7-cyclooctatetraene
AIBN	2,2'-azobisisobutyronitrile	Cp	cyclopentadienyl
AM1	Austin model 1	<i>m</i> -CPBA	<i>meta</i> -chloroperoxybenzoic acid
AMP	adenosine 5'-monophosphate	CV	cyclic voltammetry
Anal.	combustion elemental analysis	Cy	cyclohexyl
anhyd	anhydrous	$\delta$	chemical shift in parts per million
AO	atomic orbital	d	downfield from tetramethylsilane
aq	aqueous	<i>d</i>	day(s); doublet (spectral); deci
Ar	aryl	DABCO	density
atm	atmosphere(s)	dansyl	1,4-diazabicyclo[2.2.2]octane
ATP	adenosine 5'-triphosphate	DBN	5-(dimethylamino)-1-naphthalenesulfonyl
ATPase	adenosinetriphosphatase	DBU	1,5-diazabicyclo[4.3.0]non-5-ene
av	average	DCC	1,8-diazabicyclo[5.4.0]undec-7-ene
9-BBN	9-borabicyclo[3.3.1]nonyl	DCE	<i>N,N</i> '-dicyclohexylcarbodiimide
9-BBN-H	9-borabicyclo[3.3.1]nonane	DDQ	1,2-dichloroethane
Bn, Bzl	benzyl	DEAD	2,3-dichloro-5,6-dicyano-1,4-benzoquinone
bpy	2,2'-bipyridyl	DEPT	diethyl azodicarboxylate
BOC, Boc	<i>tert</i> -butoxycarbonyl	DFT	distortionless enhancement by polarization transfer
bp	boiling point, base pair	DIBALH	density functional theory
br	broad (spectral)	DMA	diisobutylaluminum hydride
Bu, <i>n</i> -Bu	normal (primary) butyl	DMAP	dimethylacetamide
<i>s</i> -Bu	<i>sec</i> -butyl	DMDO	4-( <i>N,N</i> -dimethylamino)pyridine
<i>t</i> -Bu	<i>tert</i> -butyl	DME	dimethyldioxirane
Bz	benzoyl (not benzyl)	DMF	1,2-dimethoxyethane
B3LYP	3-parameter hybrid Becke exchange/ Lee–Yang–Parr correlation functional	DMPU	dimethylformamide
°C	degrees Celsius	DMSO	1,3-dimethyl-3,4,5,6-tetrahydro-2(1 <i>H</i> )-pyrimidinone
calcd	calculated	DMT	dimethyl sulfoxide
cAMP	adenosine cyclic 3',5'-phosphate	DNA	4,4'-dimethoxytrityl (4,4'-dimethoxyltriphenylmethyl)
CAN	ceric ammonium nitrate	DPS	deoxyribonucleic acid
CASSCF	complete active space self-consistent field	dr	<i>tert</i> -butyldiphenylsilyl
CASPT2	complete active space with second-order perturbation theory	DTT	diastereomeric ratio
cat	catalytic	E1	dithiothreitol
CBZ, Cbz	benzyloxycarbonyl (preferred over the abbreviation Z)	E2	unimolecular elimination
CC	coupled cluster	ED <sub>50</sub>	bimolecular elimination
CD	circular dichroism	EDTA	dose effective in 50% of test subjects
cDNA	complementary deoxyribonucleic acid	EI	ethylenediaminetetraacetic acid
CI	chemical ionization; configuration interaction	EPR	electron impact
CIF	crystallographic information file	eq	electron paramagnetic resonance
CIDNP	chemically induced dynamic nuclear polarization	equiv	equation
cm	centimeter(s)	er	equivalent
cm <sup>-1</sup>	wavenumber(s)	ESI	enantiomeric ratio
		Et	electrospray ionization
		FAB	ethyl
			fast atom bombardment

FD	field desorption	Me	methyl
FID	flame ionization detector; free induction decay	MEM	(2-methoxyethoxy)methyl
Fmoc	9-fluorenylmethoxycarbonyl	Mes	2,4,6-trimethylphenyl (mesityl) [not methylsulfonyl (mesyl)]
FT	Fourier transform	MHz	megahertz
g	gram(s); prefix to NMR abbreviation denoting gradient-selected (e.g. gCOSY, gHMQC)	min	minute(s); minimum
GC	gas chromatography	mM	millimolar (millimoles per liter)
GTP	guanosine 5'-triphosphate	MO	molecular orbital
h	hour(s)	mol	mole(s); molecular (as in mol wt)
HF	Hartree–Fock	MOM	methoxymethyl
HMBC	heteronuclear multiple bond correlation	mp	melting point
HMPA	hexamethylphosphoric triamide (hexamethylphosphoramide)	MP	Møller–Plesset perturbation theory
HMQC	heteronuclear multiple quantum correlation	MRCI	multi-reference configuration interaction
HOMO	highest occupied molecular orbital	mRNA	messenger ribonucleic acid
HPLC	high-performance liquid chromatography	Ms	methylsulfonyl (mesyl)
HRMS	high-resolution mass spectrometry	MS	mass spectrometry
HSQC	heteronuclear single quantum correlation	MTBE	methyl <i>tert</i> -butyl ether
Hz	hertz	MW, mol wt	molecular weight
ICR	ion cyclotron resonance	<i>m/z</i>	mass-to-charge ratio (not <i>m/e</i> )
INDO	intermediate neglect of differential overlap	N	normal (equivalents per liter)
IP	ionization potential	NAD <sup>+</sup>	nicotinamide adenine dinucleotide
IR	infrared	NADH	reduced NAD
<i>J</i>	coupling constant (in NMR spectrometry)	NBO	natural bond orbital
k	kilo	NBS	<i>N</i> -bromosuccinimide
K	kelvin(s) (absolute temperature)	NCS	<i>N</i> -chlorosuccinimide
L	liter(s)	NICS	nucleus-independent chemical shift
LAH	lithium aluminum hydride	nm	nanometer(s)
LCAO	linear combination of atomic orbitals	NMO	<i>N</i> -methylmorpholine- <i>N</i> -oxide
LD <sub>50</sub>	dose that is lethal in 50% of test subjects	NMP	<i>N</i> -methylpyrrolidone
LDA	lithium diisopropylamide; local density approximation	NMR	nuclear magnetic resonance
LFER	linear free energy relationship	NOE	nuclear Overhauser effect
LHMDS	lithium hexamethyldisilazane, lithium bis(trimethylsilyl)amide	NRT	nuclear Overhauser effect spectroscopy
lit.	literature value (abbreviation used with period)	Nu	natural resonance theory
LTMP	lithium 2,2,6,6-tetramethylpiperidine	obsd	nucleophile
LUMO	lowest unoccupied molecular orbital	OD	observed
$\mu$	micro	ORD	optical density
m	multiplet (spectral); meter(s); milli	PCC	optical rotary dispersion
M	molar (moles per liter); mega	PDC	pyridinium chlorochromate
M <sup>+</sup>	parent molecular ion	PES	pyridinium dichromate
MALDI	matrix-assisted laser desorption ionization	Ph	photoelectron spectroscopy
max	maximum	piv	phenyl
MCD	magnetic circular dichroism	pm	pivaloyl
MCR	multicomponent reaction	PM3	picometer(s)
MCSCF	multi-configuration self-consistent field	PMB	parametric method 3
MD	molecular dynamics	PPA	<i>p</i> -methoxybenzyl
		ppm	poly(phosphoric acid)
		PPTS	part(s) per million
		Pr	pyridinium <i>para</i> -toluenesulfonate
		<i>i</i> -Pr	propyl
		PT	isopropyl
		PTC	perturbation theory
		py	phase-transfer catalysis
		q	pyridine
		QSAR	quartet (spectral)
			quantitative structure–activity relationship

RCM	ring-closure metathesis	TDDFT	time-dependent density functional theory
redox	reduction–oxidation	TEAB	tetraethylammonium bromide
rel	relative	temp	temperature
$R_f$	retention factor (in chromatography)	Tf	trifluoromethanesulfonyl (triflyl)
RHF	restricted Hartree–Fock	TFA	trifluoroacetic acid
ROESY	rotating frame Overhauser effect spectroscopy	TFAA	trifluoroacetic anhydride
ROMP	ring-opening metathesis polymerization	THF	tetrahydrofuran
rRNA	ribosomal ribonucleic acid	THP	tetrahydropyran-2-yl
rt	room temperature	TIPS	triisopropylsilyl
s	singlet (spectral); second(s)	TLC	thin-layer chromatography
SAR	structure–activity relationship	TMAI	tetramethylammonium iodide
SCF	self-consistent field	TMEDA	$N,N,N',N'$ -tetramethyl-1,2-ethylenediamine
SEM	scanning electron microscopy; 2-trimethylsilylethoxymethyl	TMS	trimethylsilyl; tetramethylsilane
SET	single electron transfer	TOF	time-of-flight
$S_{\text{N}1}$	unimolecular nucleophilic substitution	Tr	triphenylmethyl (trityl)
$S_{\text{N}2}$	bimolecular nucleophilic substitution	tRNA	transfer ribonucleic acid
$S_{\text{N}'}$	nucleophilic substitution with allylic rearrangement	$t_{\text{R}}$	retention time (in chromatography)
SOMO	single-occupied molecular orbital	Ts	<i>para</i> -toluenesulfonyl (tosyl)
t	triplet (spectral)	TS	transition state
$t$	time; temperature in units of degrees Celsius ( $^{\circ}\text{C}$ )	UHF	unrestricted Hartree–Fock
$T$	absolute temperature in units of kelvins (K)	UV	ultraviolet
TBAB	tetrabutylammonium bromide	VCD	vibrational circular dichroism
TBAC	tetrabutylammonium chloride	vis	visible
TBAF	tetrabutylammonium fluoride	vol	volume
TBS	<i>tert</i> -butyldimethylsilyl	v/v	volume per unit volume (volume-to-volume ratio)
TBHP	<i>tert</i> -butyl hydroperoxide	wt	weight
TCA	trichloroacetic acid	w/w	weight per unit weight (weight-to-weight ratio)
TCNE	tetracyanoethylene	ZINDO	Zerner parameterization of intermediate neglect of differential overlap